Supplemental materials for:

P-Hydroxybenzenesulfonic Acid-Formaldehyde Resin Solid Acid for Conversion of Fructose and Glucose to 5-Hydroxymethylfurfural Wenzhi Li^a, Tingwei Zhang^{a*}, Haosheng Xin^b, Mingxue Su^c, Longlong Ma^{d*}, Hason Jameel^e, Hou-min Chang^e, Gang Pei^a

^a Department of Thermal Science and Energy Engineering, University of Science and

Technology of China, Hefei 230026, PR China

^b Institute of Materials and Chemical Engineering , Anhui Jiangzhu University, Hefei

230022, PR China

^c Department of Chemistry, University of Science and Technology of China, Hefei

230026, PR China

^d CAS Key Laboratory of Renewable Energy, Guangzhou Institute of Energy

Conversion, Chinese Academy of Sciences, Guangzhou 510640, PR China

^e Department of Forest Biomaterials, North Carolina State University, Raleigh, NC

27695-8005, USA

*Corresponding author. Tel.: +86 0551 63600786

E-mail address: ztwei@mail.ustc.edu.cn (Tingwei Zhang), mall@ms.giec.ac.cn (Longlong Ma).



Scheme S1 Synthesis route of SPFR solid acids

Scheme S1 was obtained according to Yu's PFR gels synthesis route with a slight modification.



Fig.S1 SEM and TEM images of SPFR-0.26 (a, b) and SPFR-0.38 (c, d).



Fig.S2 HMF selectivity from fructose.

Entry	Water	Temperature and time	HMF yield	Furfural
-	concentration (wt.	(°C/min)	(%)	yield (%)
	%)			
1	0	140/30	59.3	16.3
2	5	140/20	79.3	7.0
3	5	140/30	80.3	7.0
4	5	140/40	77.2	7.1
5	10	140/20	77.5	4.2
6	10	140/30	82.6	4.4
7	10	140/40	78.6	4.7
8	20	140/20	58.7	1.7
9	20	140/30	65.5	1.9
10	20	140/40	74.6	2.3
11	30	140/30	46.1	0.9

Table S1 Effect of water concentration on HMF production from glucose [a]

[a] Reaction conditions: 0.2 g SPFR-0.38, 0.40 g fructose, 15.0 ml GVL, different water concentration (based on GVL).