

Supporting information

Coating of Pd/C catalysts with Lewis-acidic ionic liquids and liquid coordination complexes – SCILL induced activity enhancement in arene hydrogenation

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Batch autoclave setup

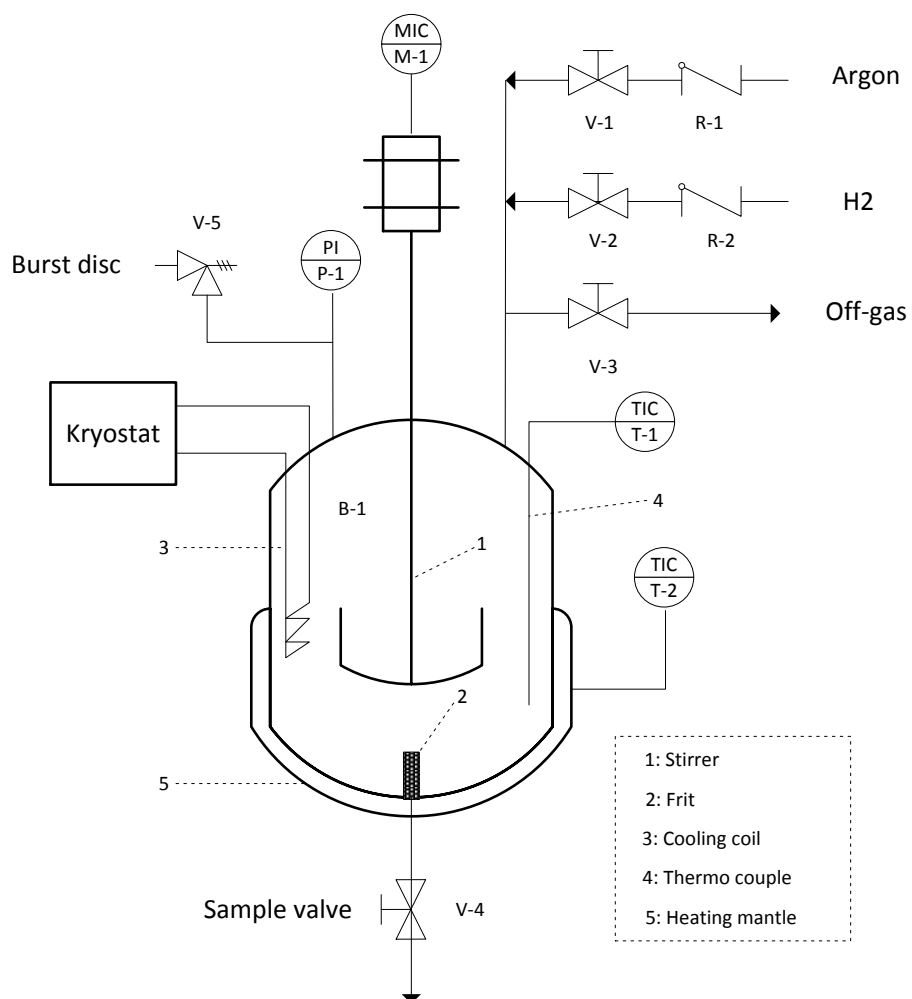


Figure S1. Schematic flow sheet (left) and photograph (right) of the batch reactor utilized.

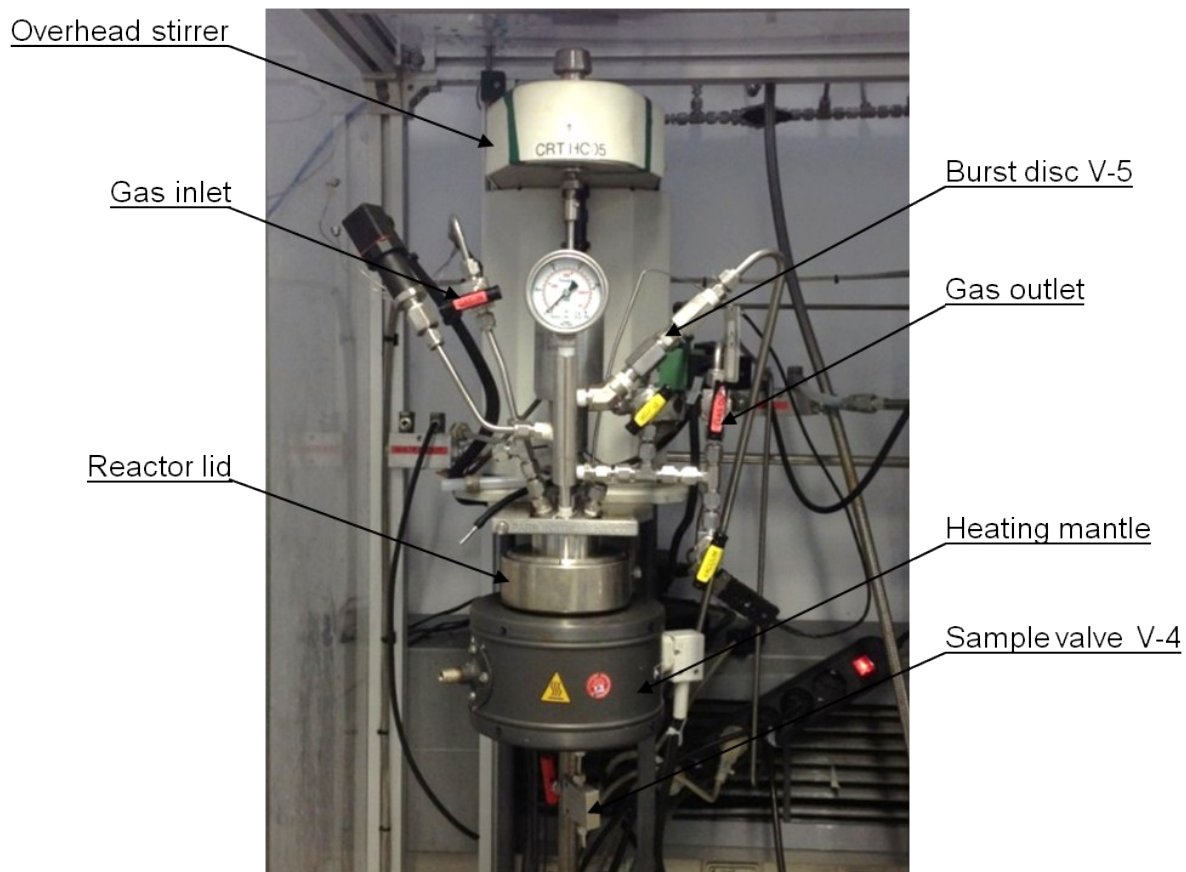


Figure S2. Photograph of the assembled reactor.

Comparison between addition of AlCl₃ and [BMIM]Cl/AlCl₃ in bulk and SCILL

Figure S3 shows the conversion time plots for uncoated Pd/C without and with addition of [BMIM]Cl/AlCl₃ ($x_{\text{AlCl}_3} = 0.67$). The amount of ionic liquid added was adjusted to the same amount of AlCl₃ present in a Pd-SCILL-1 system, namely approx. 13 % or 41.3 mg.

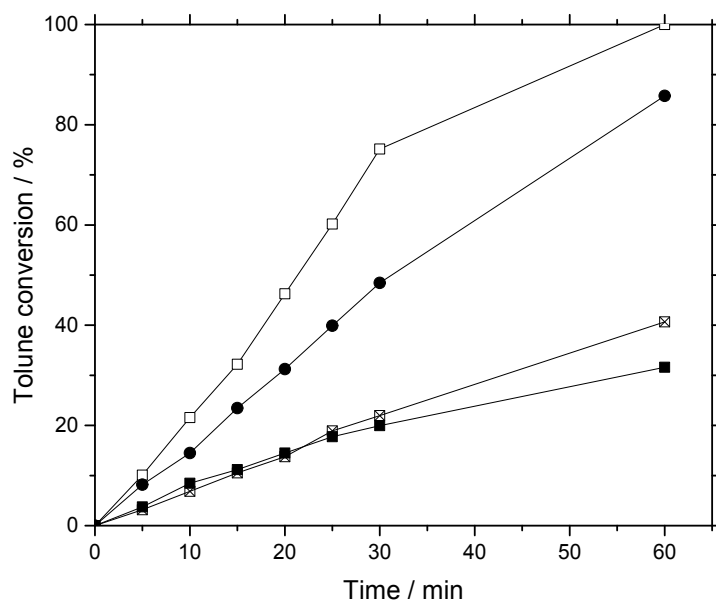


Figure S3. Conversion over time for toluene hydrogenation using Pd/C, with 330 mg (open squares), 41.3 mg (open squares intersected) and without (filled squares) addition of AlCl₃, as well as Lewis-acidic chloroaluminate Pd-SCILL-1 catalysts (filled circles). Reaction conditions: T = 60 °C, $p_{\text{hydrogen}} = 15$ bar, $c_{\text{toluene}} = 0.25$ mol L⁻¹, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1$ g, $w_{\text{Pd}} = 10$ wt%, stirring speed = 1000 min⁻¹. Ionic liquid = [BMIM]Cl/AlCl₃ ($x_{\text{AlCl}_3} = 0.67$).

Textural analysis of the Pd-SCILL-1 catalyst

The Pd-SCILL-1 catalyst, coated with chloroaluminate ionic liquid, was analyzed by means of N₂ sorption at 77 K on a Quadrasorb-SI from Quantachrome. The data are compiled in Table S1.

Table S1. Textural data for the Pd-SCILL-1 catalyst containing varying amounts of chloroaluminate ionic liquid [BMIM]Cl/AlCl₃ ($x_{\text{AlCl}_3} = 0.67$).

Catalyst	Ionic liquid wt%	Surface area m ² g ⁻¹	Pore volume mL g ⁻¹	Av. Pore diameter nm
Pd/C	-	921	0.67	2.9
Pd-SCILL-1	6	393	0.47	4.8
Pd-SCILL-1	17	283	0.31	4.4
Pd-SCILL-1	45	146	0.18	4.8

Temperature variation in toluene hydrogenation

The uncoated Pd/C catalyst was tested in the temperature range between 40 and 80 °C.

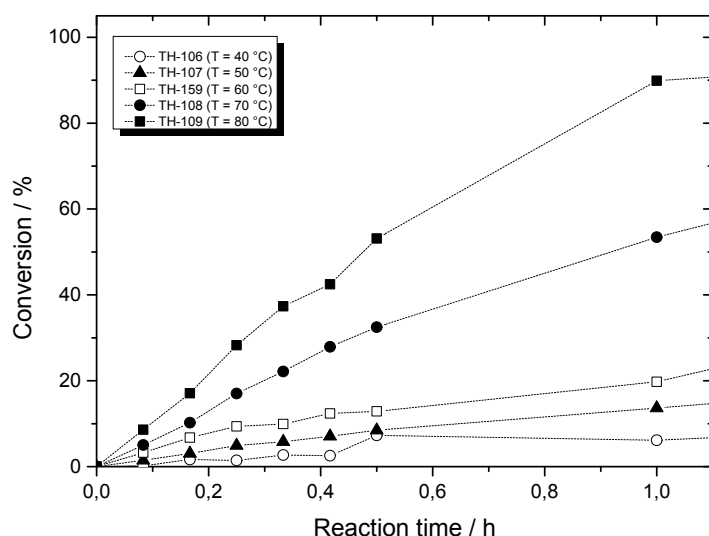


Figure S4. Conversion over time plot for the hydrogenation of toluene using Pd/C catalyst. Reaction conditions: $p_{\text{hydrogen}} = 15 \text{ bar}$, $c_{\text{toluene}} = 0.25 \text{ mol L}^{-1}$, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1 \text{ g}$, $w_{\text{Pd}} = 10 \text{ wt\%}$, stirring speed = 1000 min^{-1} , ionic liquid coating = 13 wt%.

The chloroaluminate coated Pd-SCILL-1 catalyst was tested in the temperature range between 40 and 80 °C.

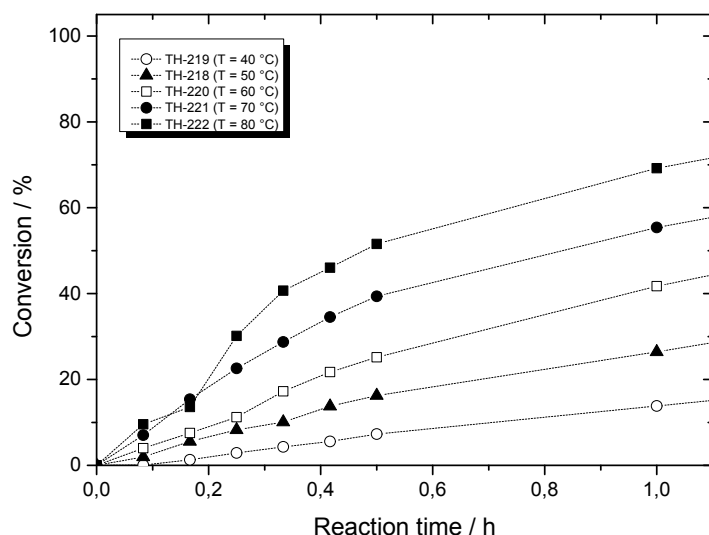


Figure S5. Conversion over time plot for the hydrogenation of toluene using Pd-SCILL-1 catalyst. Reaction conditions: $p_{\text{hydrogen}} = 15 \text{ bar}$, $c_{\text{toluene}} = 0.25 \text{ mol L}^{-1}$, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1 \text{ g}$, $w_{\text{Pd}} = 10 \text{ wt\%}$, stirring speed = 1000 min^{-1} , ionic liquid coating = 13 wt%.

The LCC coated Pd-SCILL-2 catalyst was tested in the temperature range between 40 and 90 °C.

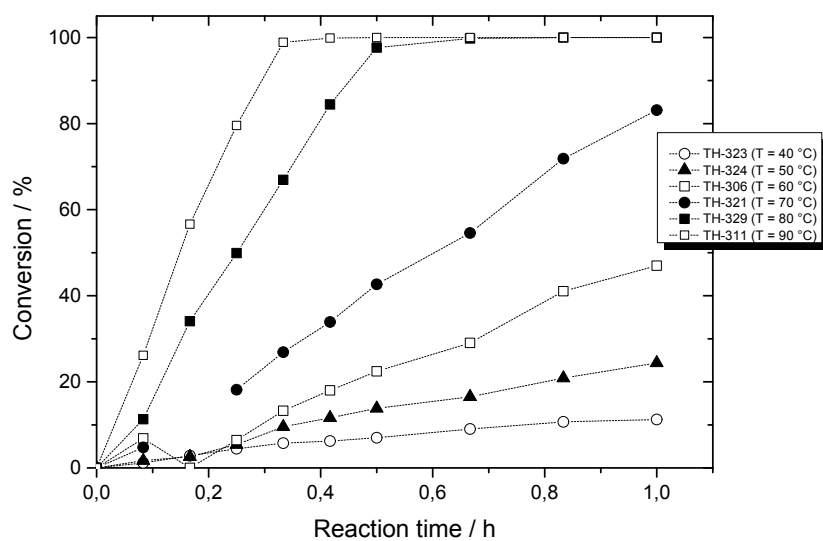


Figure S6. Conversion over time plot for the hydrogenation of toluene using Pd-SCILL-2 catalyst. Reaction conditions: $p_{\text{hydrogen}} = 15 \text{ bar}$, $c_{\text{toluene}} = 0.25 \text{ mol L}^{-1}$, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1 \text{ g}$, $w_{\text{Pd}} = 10 \text{ wt\%}$, stirring speed = 1000 min^{-1} , ionic liquid coating = 13 wt%.

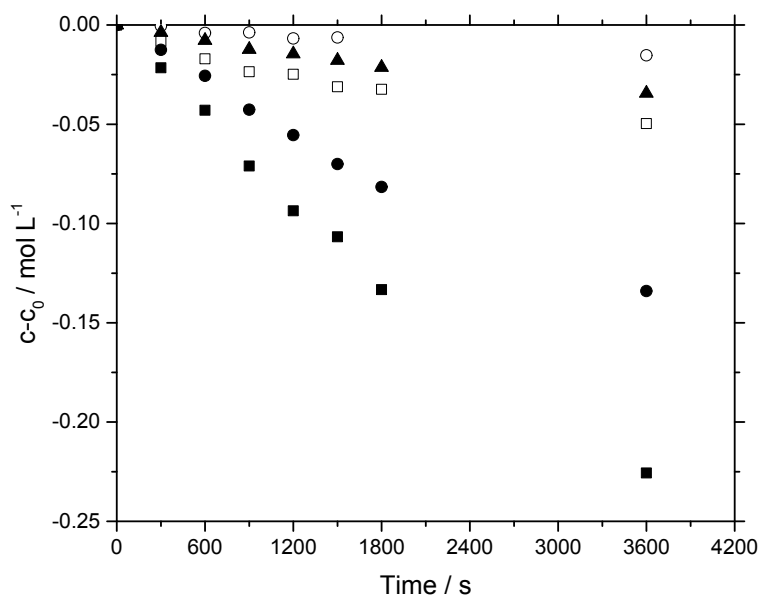


Figure S7. Integral analysis of toluene hydrogenation using Pd/C catalysts. Temperatures as indicated in Figure S3.

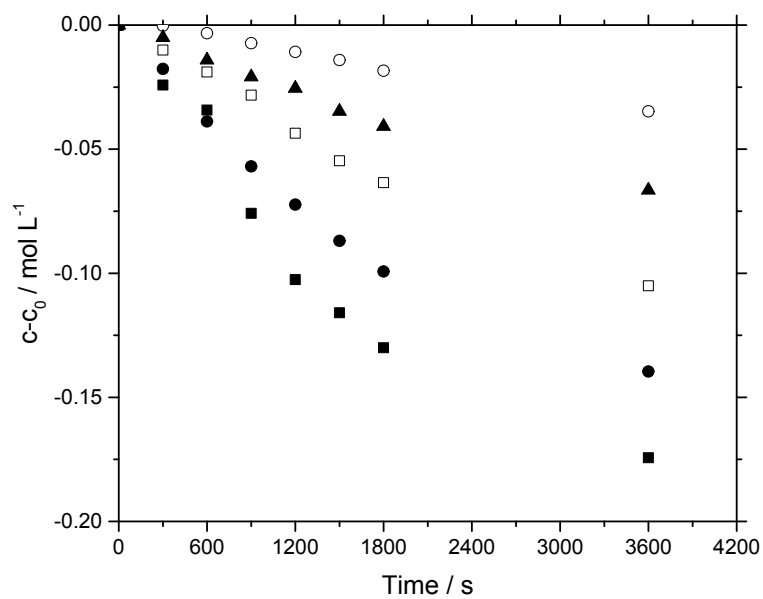


Figure S8. Integral analysis of toluene hydrogenation using Pd-SCILL-1 catalysts. Temperatures as indicated in Figure S4.

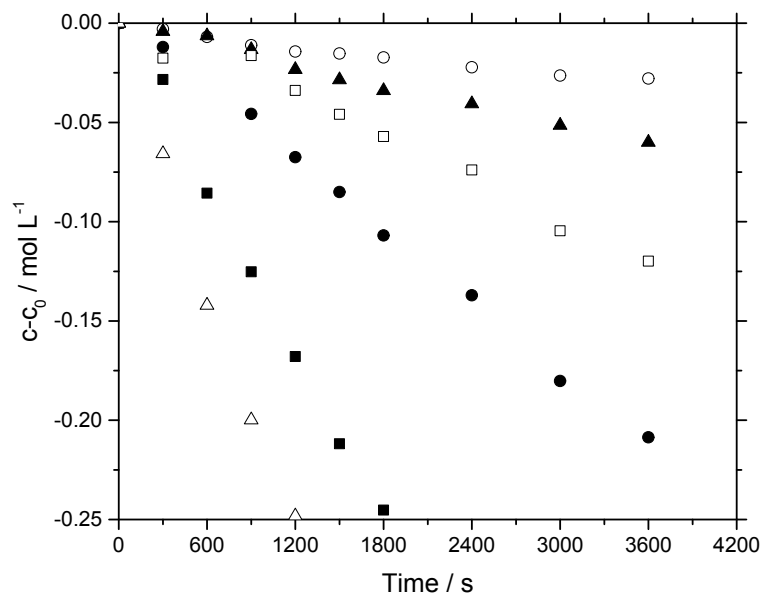


Figure S9. Integral analysis of toluene hydrogenation using Pd-SCILL-2 catalysts. Temperatures as indicated in Figure S5.

Agitation speed variation

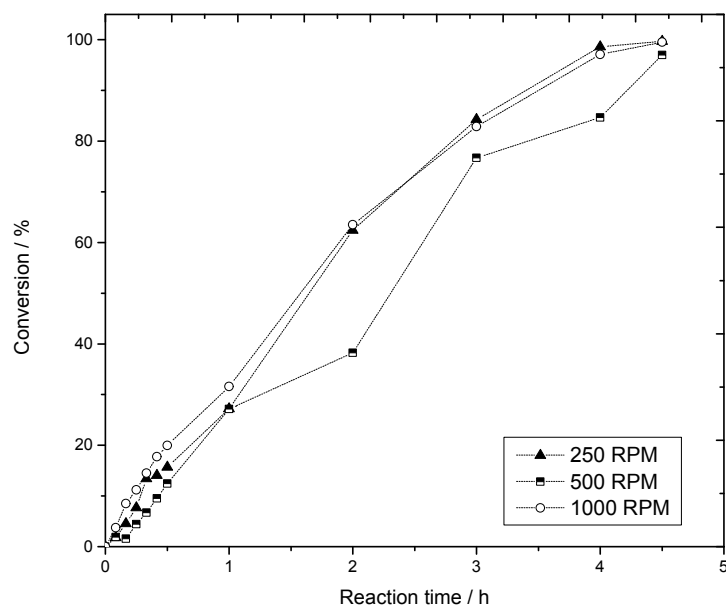


Figure S10. Conversion over time plot for the agitation speed variation of the hydrogenation of toluene using Pd/C catalyst. Reaction conditions: $p_{\text{hydrogen}} = 15$ bar, $c_{\text{toluene}} = 0.25$ mol L⁻¹, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1$ g, $w_{\text{Pd}} = 10$ wt%, stirring speed = 250-1000 min⁻¹.

Metal leaching studies

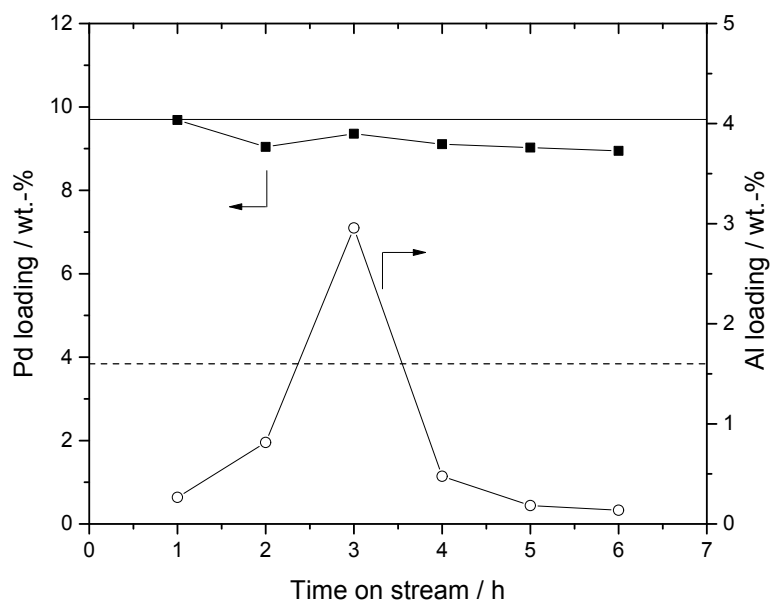


Figure S11. Leaching studies determined via ICP analysis of the Pd-SCILL samples. Reaction conditions: $T = 60$ °C, $p_{\text{hydrogen}} = 1$ bar, $c_{\text{toluene}} = 0.25$ mol L⁻¹, solvent = 100 mL cyclohexane, $m_{\text{cat}} = 1$ g, $w_{\text{Pd}} = 10$ wt%, stirring speed = 1000 min⁻¹, ionic liquid coating = 13 wt%. The solid line indicates the initial Pd loading, the dashed line indicates the initial Al loading.