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Electronic Supplementary information

Synthesis and characterisation of push-pull flavin-based dyes with efficient second harmonic generation (SHG) properties.

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1. ¹H and ¹³C NMR data

Compound 2





S3







S6





2. TGA and DSC data

FLA-A



Figure S1 – Thermal analysis data for compound FLA-A recorded under a nitrogen atmosphere, measured at a heating rate of 10 °C min⁻¹.

FLA-B



Figure S2 – Thermal analysis data for compound **FIa-B** recorded under a nitrogen atmosphere, measured at a heating rate of 10 °C min⁻¹.



Figure S3 – Thermal analysis data for compound Fla-C recorded under a nitrogen atmosphere, measured at a heating rate of 10 °C min⁻¹.

3. Voltammetry data



Figure S4 – Cyclic voltammetry plots of Fla A-C, in $CHCl_3$ (C = $10^{-3}M$). All the experiments were performed using a Pt working electrode, a Pt counter electrode and Ag wire as reference. TBAPF₆ (0.1M) was used as supporting electrolyte and the collected data were referenced to the redox potential of Fc/Fc⁺ couple.

	Eox (V)	Ered (V)	IP (eV)	EA (eV)	Eg,fund (eV)
FLA-A	0.52	-1.25	-5.32	-3.55	1.77
FLA-B	0.69	-0.92	-5.49	-3.88	1.61
FLA-C	0.19	-1.04	-4.99	-3.76	1.23

Table S1 – Summary of electrochemical properties of FLA-A – C. Ionisation potentials (IPs) and electron affinities (EAs)were calculated by applying the empirical equations: $IP = -(E_{OX} + 4.80)$ and $EA = -(E_{RED} + 4.80)$.

4. Theoretical calculations

Solvent dependence computation was performed at DFT level using the hybrid B3LYP level by employing the 6-31+G** basis set using polarizable continuum model (keyword: scrf=*solvent*). Frequency calculations were performed to insure the absence of negative frequencies.

Flavin A	μ	номо	LUMO
1,4-Dioxane	17.0152	-5.6428	-3.1410
Toluene	17.2402	-5.6333	-3.1467
Dichloromethane	20.2001	-5.5245	-3.2142
Dimethylsulfoxide	21.4030	-5.4888	-3.2392
Flavin B	μ	номо	LUMO
1,4-Dioxane	14.3034	-5.5658	-3.2420
Toluene	14.4754	-5.5596	-3.2463
Dichloromethane	16.6393	-5.4913	-3.2937
Dimethylsulfoxide	17.4862	-5.4676	-3.3086
Flavin C	μ	номо	LUMO
1,4-Dioxane	13.6595	-5.8856	-3.3554
Toluene	13.8190	-5.8782	-3.3592
Dichloromethane	15.7836	-5.7818	-3.3954
Dimethylsulfoxide	16.5851	-5.7593	-3.4061

Table S2 – Summary of calculated solvent dependant properties of FLA-A – C.

Gaussian 09 keywords freq=raman, cphf=rdfreq and polar were used to compute the hyperpolarisability. Here the hyperpolarisability β_{tot} is given as the function of the tensor and $\beta_{||}$ is the hyperpolarisability parallel to the ground state CT direction. $\beta_{||}$ is given in the output file whereas β_{tot} is obtained from the different parts of the tensor.

 $\beta_{\rm tot}=(\beta_x{}^2+\beta_y{}^2+\beta_z{}^2)^{1/2}$

 $\beta_{\text{tot}} = [(\beta_{xxx} + \beta_{xyy} + \beta_{xzz})^2 + (\beta_{yyy} + \beta_{yzz} + \beta_{yxx})^2 + (\beta_{zzz} + \beta_{zxx} + \beta_{zyy})^2]^{1/2}$

Fla-A

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Title Card Required

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1064nm

Fla-B

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Title Card Required

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1064nm

Fla-C

#p freq=raman cphf=rdfreq rb3lyp/6-31+g(d,p) scrf=(solvent=1,4-dioxane)
polar

Title Card Required

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Н	1.48029468	2.22//9836	0.61435830
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н	1.50201413	4.69890057	0.32952421
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С	3.35597927	2.93307038	1.37841754
н	3.64114280	1.99134742	1.85810127
Н	4.2/223628	3.42640223	1.03833385
Н	2.88960748	3.56395605	2.14250684

1064nm

5. References

S1 C. M. Cardona, W. Li, A. E. Kaifer, D. Stockdale and G. C. Bazan, Adv. Mater., 2011, 23, 2367-2371.

S2 J.-L. Bredas, Mater. Horiz., 2014, 1, 17-19.