

Supporting Information

**Phosphomolybdic Acid Supported Atomically Dispersed Transition
Metal Atoms (M = Fe, Co, Ni, Cu, Ru, Rh, Pd, Ag, Os, Ir, Pt, and Au):
Stable Single Atom Catalysts Studied by Density Functional Theory**

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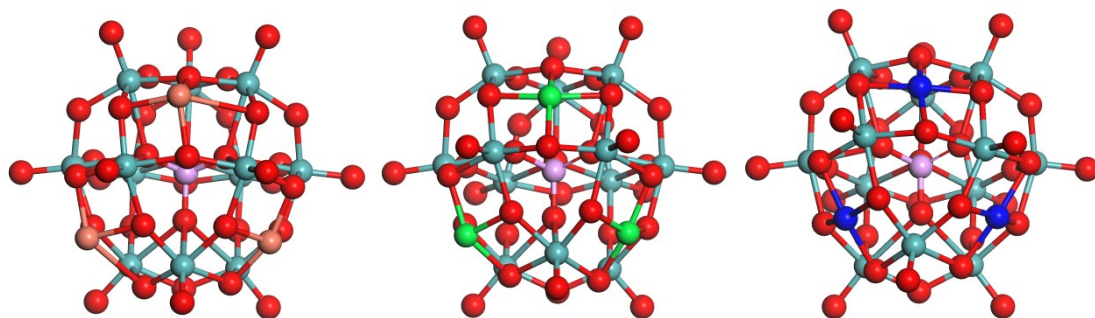


Figure S1 Optimized configurations of M-PMA (M= Cu, Ru and Pt) with the numbers of M equal to 1, 2 and 3 at the 4-H site of PMA.

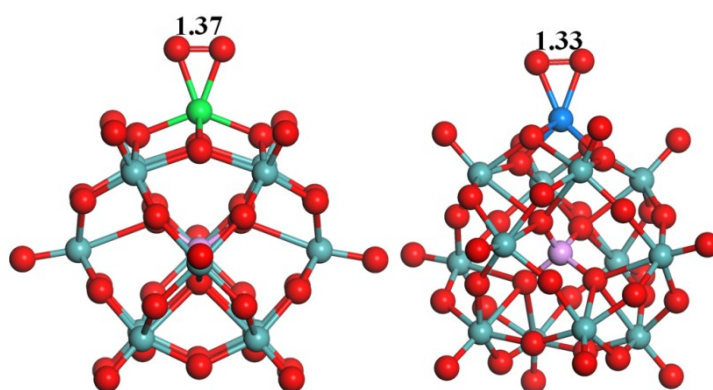


Figure S2 Optimized configurations of an O₂ molecule adsorbed on M-PMA (M= Ru and Fe).

Table S1 Calculated adsorption energies (E_{ads}/eV) for M (M= Cu, Ru and Pt) on PMA with M initially positioned at 4-H, 3H-O_c, 3H-O_{bri}, and B-O_c-O_{bri}, respectively.

M_1	Initial position	$E_{\text{ads}}(\text{eV})$ after optimization
Cu	4-H	-7.50
	3H-O _c	-7.50
	3H-O _{bri}	-5.02
	B-O _c -O _{bri}	-4.49
Ru	4-H	-11.04
	3H-O _c	-11.10
	3H-O _{bri}	-9.70
	B-O _c -O _{bri}	-9.66
Pt	4-H	-7.92
	3H-O _c	-5.95
	3H-O _{bri}	-5.95
	B-O _c -O _{bri}	-5.21

Table S2 Calculated adsorption energies (eV) of M (Pt, Ru and Cu) on PMA at the 4-H site with the numbers of M equal to 1, 2 and 3.

M-PMA	Numbers of M	E_{ads}
Pt	1	-7.92
	2	-6.14
	3	-5.23
Ru	1	-11.69
	2	-9.49
	3	-8.56
Cu	1	-7.23
	2	-5.11
	3	-4.36

Table S3 The comparison between E_{ad} values (eV) and the experimental metal cohesive energies (E_{coh})

M	E_{ads}	E_{coh}
Group VIII		
Fe	-10.90	-4.28
Ru	-11.04	-6.74
Os	-11.24	-8.17
Group IX		
Co	-10.41	-4.39
Rh	-9.11	-5.74
Ir	-10.08	-6.94
Group X		
Ni	-9.35	-4.44
Pd	-6.44	-3.90
Pt	-7.92	-5.85
Group XI		
Cu	-7.50	-3.48
Ag	-4.63	-2.94
Au	-5.04	-3.81