### Liquefaction of waste pine wood and its application in the synthesis of a flame

# retardant polyurethane foam

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#### 1. The carbon NMR analysis of liquefied polyol

Position	Chemical shift δ (ppm)						
glycouse	C1	C2	C3	C4	C5	C6	$\Box C1$
levulinate	100.2 L1	72.3 L2	72.6 L3	71.3 L4	73.5 L5	63.1 □L1	66.7
Formic eater	174.1 F1	28 □F1	37.9	207.4	29.9	66.7	
glycerol	163.4 G1	173.1 G2					
PEG400	72.7 P1	63.1 P2					
	70.2	60.6					

Table S1.<sup>13</sup>CNMR chemical shift of liquefied polyol

### 2. Thermal behaviour and flammability properties of MWPU foam



**Fig.S1**.The TGA spectra of samples P2 to P5 (The five independent samples measurement are R1, R2, R3, R4, R5 respectively)



Fig.S2. T onset standard values of samples P2 to P5



## 3. FTIR-TGA analysis of evolved gases from MP

Fig.S3A: FTIR-TGA spectra of evolved gases from MP at different temperatures



**Fig.S3B:** 3D view FTIR-TGA spectra of evolved gases from MP at different temperatures

From the FTIR-TGA spectra of MP at different temperatures as shown in Fig.S1A and S1B, no absorptions peak was below 204°C, indicating no gases evolved from MP. At 314°C, new peaks appear at 3566, 3451, 1450, 827 and 750 cm<sup>-1</sup> which can be attributed to absorptions of melamine formed from the decomposition of MP. It can be observed that after 406°C, new peaks appears at 3330, 1625, 1500, 956 and 926 cm<sup>-1</sup> are due to NH<sub>3</sub> absorptions, and the peaks at 3724,3648 and 1080 cm<sup>-1</sup> are attributed to absorptions of water. <sup>1-3</sup>

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# 4. The combustion process of foams



Fig.S4.The combustion process of (A) wood based foam and (B) MWPU foam with 10wt% MP at oxygen concentrations of 18.5vol % and 21.7vol %, respectively





Fig.S5. The SEM-EDS analysis of the char of sample P5.

6. The mechanical property of MWPU foams



**Fig. S6.** The tensile strength of P2 to P5