

Adsorption Behavior of Copolymer AM/DMC/APEG/DMAAC- 16 on Carbonate Rock and Its Application for Acidizing

Hongping Quan ^{a, b}, Zhonghao Chen ^{a, b}, Yang Wu ^{*, a, b}, Zhuoke Li ^{a, b}

a. College of chemistry and chemical engineering, Southwest Petroleum University, Chengdu, P. R. of China

b. Oil & Gas Field Applied Chemistry Key Laboratory of Sichuan Province, Southwest Petroleum University, Chengdu, P. R. China.

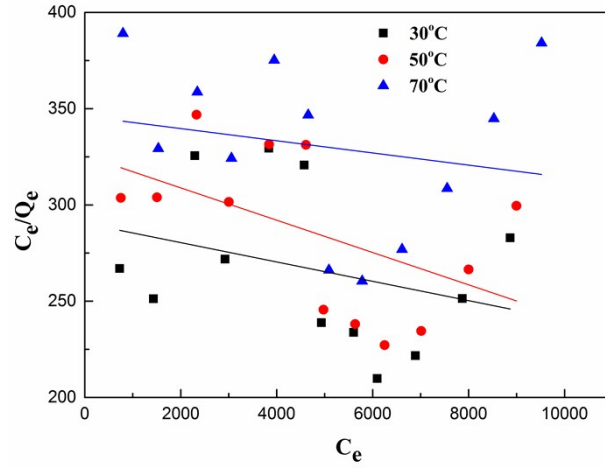


Fig. S1. Langmuir isotherms for PADAD adsorbed on carbonate rock.

Table S1. Langmuir parameters for PADAD adsorbed on carbonate rock.

Temperature (°C)	Langmuir equation	R ²
30	$y = -0.0083x + 301.07$	0.1986
50	$y = -0.0084x + 325.65$	0.2667
70	$y = -0.0032x + 346.01$	0.0384

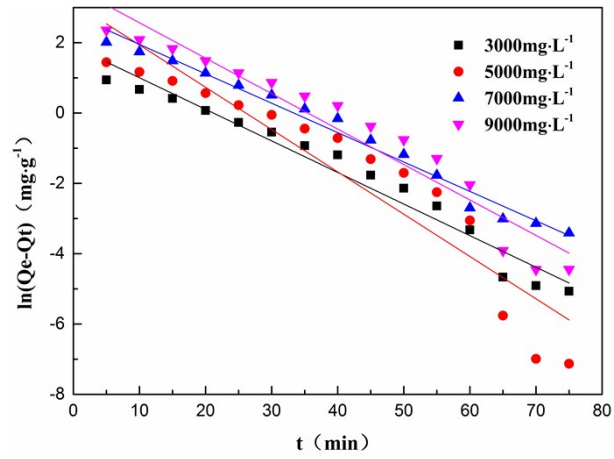


Fig. S2. Pseudo-first-order equation for the adsorption kinetics of PADAD on carbonate rock at different initial concentration

Table S2. Kinetic parameters for the Pseudo-first-order equation

Initial concentration (mg · L ⁻¹)	Q _e (mg · g ⁻¹)	Kinetics equation	R ₁ ²	Q _{e,1} (mg · g ⁻¹)	k ₁ (min ⁻¹)
3000	7.0494	y=-0.0898x+1.9015	0.9626	6.6959	0.2068
5000	11.6421	y=-0.1203x+3.1390	0.8787	23.0808	0.2771
7000	20.6587	y=-0.0836x+2.7871	0.9817	16.2339	0.1925
9000	29.0481	y=-0.1008x+3.5753	0.9337	35.7053	0.2321

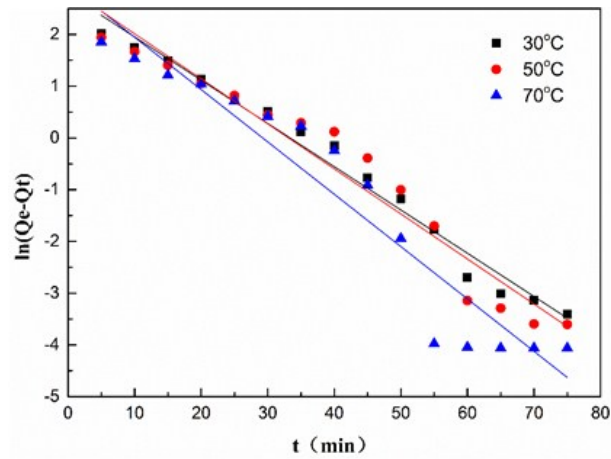


Fig. S3. Pseudo-first-order equation for the adsorption kinetics of PADAD on carbonate rock at different temperature

Table S3. Kinetic parameters for the Pseudo-first-order equation at different temperature

Temperature (°C)	Q_e ($\text{mg} \cdot \text{g}^{-1}$)	Kinetics equation	R_1^2	$Q_{e,1}$ ($\text{mg} \cdot \text{g}^{-1}$)	k_1 (min^{-1})
30	20.6587	$y = -0.0836x + 2.7871$	0.9817	16.2339	0.1925
50	20.2566	$y = -0.0870x + 2.8816$	0.9480	17.8428	0.2004
70	19.1412	$y = -0.1011x + 2.9569$	0.9222	19.2382	0.2328