Supplementary Information for

Rigid NON-Donor Pincer Ligand Complexes of Lutetium and Lanthanum: Synthesis and Hydroamination Catalysis

Kelly S. A. Motolko, David J. H. Emslie, * and James F. Britten

*Department of Chemistry, McMaster University, 1280 Main Street West, Hamilton, Ontario, L8S 4M1, Canada. Fax: (905)-522-2509; Tel: (905)-525-9140 x 23307. E-mail:emslied@mcmaster.ca. Website: http://www.chemistry.mcmaster.ca/emslie/emslie.html

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Figure S1 - ¹H NMR Spectrum of $[\rm K_2(\rm XN_2)(\rm dme)_{2.5}]$ (600 MHz, $\rm C_6\rm D_6)$





Figure S3 - $^{13}\mathrm{C}$ NMR Spectrum of $[\mathrm{K_2(XN_2)(dme)_{2.5}}]$ (126 MHz, $\mathrm{C_6D_6})$





 $(600 \text{ MHz}, C_6 D_6)$







Figure S8 - Expanded Region of the 1 H NMR Spectrum of [(XN₂)LuCH₂SiMe₃(THF)] (1) (600 MHz, C₆D₆)



Figure S9 - Expanded Region of the $^1\mathrm{H}$ NMR Spectrum of $[(XN_2)LuCH_2SiMe_3(THF)]$ (1) (600 MHz, C_6D_6)





 $[(XN_2)LuCH_2SiMe_3(THF)]$ (**1)** (126 MHz, C_6D_6)



Figure S13 - ¹H NMR Spectrum of $[{(XN_2)LaCl(THF)}_x]$ (x = 1 or 2) (2) (600 MHz, d₈-THF)



Figure S14 - $^{13}\mathrm{C}$ NMR Spectrum of $[\{(\mathrm{XN}_2)\mathrm{LaCl}(\mathrm{THF})\}_x]$ (x = 1 or 2) (2) (126 MHz, d_8-THF)



Figure S15 - Expanded Region of the ¹³C NMR Spectrum of $[{(XN_2)LaCl(THF)}_x]$ (x = 1 or 2) (2) (126 MHz, d₈-THF)



(600 MHz, C₆D₅Br)



(**3)** (600 MHz, d₈-THF)









Figure S20 - Expanded Region of the ¹H NMR Spectrum of $[Li(THF)_x]^+[(XN_2)La(CH_2SiMe_3)_2]^- \cdot (Tol)$ (3) (600 MHz, d₈-THF)



Figure S21 - $^{13}\mathrm{C}$ NMR Spectrum of [Li(THF)_x]+[(XN_2)La(CH_2SiMe_3)_2]^- (Tol) (**3**) (600 MHz, d_8-THF)



Figure S22 - Expanded Region of the $^{13}\mathrm{C}$ NMR Spectrum of $[\mathrm{Li}(\mathrm{THF})_{x}]^{+}[(\mathrm{XN}_{2})\mathrm{La}(\mathrm{CH}_{2}\mathrm{SiMe}_{3})_{2}]^{-}\bullet(\mathrm{Tol})$ (3) (600 MHz, d₈-THF)













Figure S26 – Expanded Reagion of the Variable Temperature ¹H NMR Spectrum of $[Li(THF)_3]^+[(XN_2)La(CH_2SiMe_3)_2]^- \cdot (Tol)$ (3) (500 MHz, d₈-THF)



Figure S27 - In Situ ¹H NMR Spectrum of $[(XN_2)LaCl(THF)_{1.5}]$ (2) with excess MeLi after 20 min at 24 °C (600 MHz, d₈-THF)



Figure S28 - In Situ ¹H NMR Spectrum of the Intramolecular Hydroamination of 1 mol % (1) with 1-Amino-2,2-diphenylpent-4-ene after ~10 min at 24 °C (600 MHz, C_6D_6)



after ~20 min at 24 °C (600 MHz, C_6D_6)







Figure S32 - *In Situ* ¹H NMR Spectrum of the *In Situ* Intermolecular Hydroamination of (1) with ^tBuAniline and 1-octene. Evaporated to dryness *in vacuo* to remove excess 1-octene before NMR (600 MHz, d₈-Tol)



Hydroamination of (1) with 'Bu-BenzylAmine and 1-octene. Evaporated to dryness *in vacuo* to remove excess 1-octene before NMR (600 MHz, d_8 -Tol)



Hydroamination of (1) with Octyl-Amine and 1-octene. Evaporated to dryness *in vacuo* to remove excess 1-octene before NMR (600 MHz, d_8 -Tol)



Figure S35 - In Situ ¹H NMR Spectrum of the Intermolecular Hydroamination of (1) with ^tBu-Aniline and Diphenylacetylene (600 MHz, d₈-Tol)





Hydroamination of (1) with Octyl-Amine and Diphenylacetylene (600 MHz, d₈-Tol)



Figure S38 - *In Situ* ¹H NMR Spectrum of the Intramolecular Hydroamination of 1 mol % (**3**) with 1-Amino-2,2-diphenylpent-4-ene after ~45 h at 24 °C (600 MHz, d₈-THF)

GC-MS Data for Intermolecular Hydroamination Reactions

Note: MSTFA (N-methyl-N-(trimethylsilyl)trifluoroacetamide) is a common derivatizing reagent and was used for the following GC-MS studies. It will silylate OH, COOH, NH_2 groups by replacing one or two hydrogen atoms with a trimethylsilyl, (TMS / Si(CH₃)₃) group, to make it less polar and therefore more amendable for GC separation. This is particular important for the starting reagents as they would not otherwise be detected using the chosen GC column without derivatization. The product of the reaction between octylamine and 1-octene was partially derivatized, while all other products did not get derivatized in the process (possibly the result of steric hinderance).

Anthrancenemethanol was used as a quality control to check if the derivatization did occur since both its underivatized and derivatized form can be detected with GC-MS.

Mass Spec. Results for the reaction of 10 mol % of 1 with ^tBuAniline and 1-octene after 24 h at 110 °C (Table 2, Entry 1 in Manuscript)

Sample A was diluted 10 fold in toluene and then derivatized 1:4 with MSTFA. Sample B was used to determine the selectivity for Markovnikov product formation and was pumped down prior to obtaining the ¹H NMR spectra and submitting for GC-MS. Sample B was diluted 10 fold in toluene, and then derivatized 1:1 with MSTFA. Samples were analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C.



Figure S39: Total ion chromatogram of Sample A

Figure S40: Mass spectra for compound 4 (target) in the TIC m/z 246 (-CH₃), m/z 176 (-C₆H₁₃)



Figure S41: Mass spectra for compound 1 in the TIC and library mass spectra







Figure S42: Mass spectra for compound 2 in the TIC





Figure S43: Mass spectra for compound 3 (MW 149-TMS) in the TIC



Figure S44: Mass spectra for compound 5 in the TIC



Figure S45: Mass spectra for compound 6 (H₂XN₂ proligand) in the TIC





Figure S47: Mass spectra for compound 3 (Anti) in the TIC for sample B







Figure S49: Mass spectra for compound 1 in the TIC for sample B



Figure S50: Mass spectra for compound 2 in the TIC for sample B



Figure S51: Mass spectra for compound 5 (H₂XN₂ proligand) in the TIC for sample B

Mass Spec. Results for the reaction of 10 mol % of 1 with 'BuBenzylAmine and 1-octene after 24 h at 110 °C (Table 2, Entry 2 in Manuscript)

Sample A was diluted 10 fold in toluene, and then derivatized 1:1 with MSTFA. Sample B was used to determine the selectivity for Markovnikov product formation and was pumped down prior to obtaining the ¹H NMR spectra and submitting for GC-MS. Sample B was diluted 10 fold in toluene and then derivatized 1:4 with MSTFA. Samples were analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C.



Figure S52: Total ion chromatogram of Sample A

Figure S53: Mass spectra for compound 2 (Target) in the TIC



m/z 260 (-CH₃), m/z 190 (-C₆H₁₃, -CH₃)



Figure S54: Mass Spectra for compound 1 in the TIC



Figure S56: Mass spectra for compound 1 (Target) in the TIC for Sample B m/z 260 (-CH₃), m/z 190 (-C₆H₁₃, -CH₃)





Figure S57: Mass spectra for compound 2 (H₂XN₂ proligand) in the TIC for Sample B

Mass Spec. Results for the reaction of 10 mol % of 1 with OctylAmine and 1-octene after 24 h at 110 °C (Table 2, Entry 3 in Manuscript)

Samples A and B were diluted 10 fold in toluene, and then derivatized 1:1 with MSTFA. Sample B was used to determine the selectivity for Markovnikov product formation and was pumped down prior to obtaining the ¹H NMR spectra and submitting for GC-MS. Samples were analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C.

Figure S58: Total ion chromatogram of Sample A



#	RT [min]	Area	
1	8.1	655964	MW 129-TMS
2	14.8	23309753	MW 241 (Target)
3	16.0	10063755	Target-TMS
4	38.7	3976270	H ₂ XN ₂ proligand

Figure S59: Mass spectra for compound 2 (Target) in the TIC

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m/z 226 (-CH3), m/z 156 (-C<sub>6</sub>H<sub>13</sub>)
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Figure S60: Mass spectra for compound 1 in the TIC



Figure S61: Mass spectra for compound 3 in the TIC



Figure S62: Mass spectra for compound 4 (H₂XN₂ pro-ligand) in the TIC

Figure S63: Total ion chromatogram of Sample B



#	RT [min]	Area	
1	14.8	13826204	Target
2	16.0	9599033	Target-TMS
3	38.7	6199069	H_2XN_2 proligand

Figure S64: Mass spectra for compound 1 (Target) in the TIC for sample B



m/z 226 (-CH3), m/z 156 (-C₆H₁₃)



Figure S65: Mass spectra for compound 2 in the TIC for sample B



Figure S66: Mass spectra for compound 3 (H₂XN₂ proligand) in the TIC for sample B

Mass Spec. Results for the reaction of 10 mol % of 1 with ^tBuAniline and Diphenylacetylene after 24 h at 110 °C (Table 2, Entry 4 in Manuscript)

The sample was diluted 10 fold in toluene, and then derivatized 1:1 with MSTFA. The sample was analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C.

Did not detect starting reagent ^tBuAniline (MW 149). The target is not included in the library. Compound 3 is assigned to be the target since the observed mass spectrum is most consistent with the target structure.



Figure S68: Mass spectra for compound 3 (likely the product) in the TIC m/z 312 (-CH₃), m/z 236 (-C₇H₇, characteristic for alkyl benzene)





Figure S69: Mass spectra for compound 1 in the TIC and library mass spectra



Figure S70: Mass spectra for compound 2 in the TIC and library mass spectra

Figure S71: Mass spectra for compound 4 in the TIC m/z 312 (-CH₃), m/z 270 (-C₄H₉)



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Figure S73: Mass spectra for compound 6 in the TIC m/z 312 (-CH₃), m/z 270 (-C₄H₉)



Mass Spec. Results for the reaction of 10 mol % of 1 with ^tBuBenzylAmine and Diphenylacetylene after 24 h at 110 °C (Table 2, Entry 5 in Manuscript)

The sample was diluted 10 fold in toluene, and then derivatized 1:1 with MSTFA. The sample was analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C.

Did not detect starting reagent ^tBuBenzylAmine (MW 163). The target is not included in the library. Compound 4 is assigned to be the target since the observed mass spectrum is most consistent with the target structure.





	RT [min]	Area	
1	18.2	329695471	MW 178
			Diphenylacetylene
2	18.6	6945087	Stilbene
3	30.0	3269646	
4	30.2	30017498	Target

Figure S75: Mass spectra for compound 4 (Target) in the TIC m/z 250 (- $C_7H_{7,}$ characteristic for alkyl benzene), m/z 147 (- $C_{14}H_{12}N$)





Figure S76: Mass spectra for compound 1 in the TIC and library mass spectra



Figure S77: Mass spectra for compound 2 in the TIC and library mass spectra

Figure S78: Mass spectra for compound 3 in the TIC m/z 326 (-CH₃), m/z 147 (-C₁₄H₁₂N)



Mass Spec. Results for the reaction of 10 mol % of 1 with OctylAmine and Diphenylacetylene after 24 h at 110 °C (Table 2, Entry 6 in Manuscript)

The sample was diluted 500 fold in toluene, and then derivatized 1:1 with MSTFA. The sample was analyzed with Agilent GCMS on DB-5MS column after an one hour incubation at 60 °C. Did not detect starting reagent OctylAmine (MW 129).

Figure S79: Total ion chromatogram





Figure S80: Mass spectra for compound 3 (Target) in the TIC





Figure S82: Mass spectra for compound 2 in the TIC and library mass spectra



