

Supplementary Information

A novel hexahydroquinazolin-2-amine based fluorescence sensor for Cu²⁺ from isolongifolanone and its biological applications

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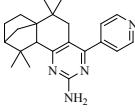
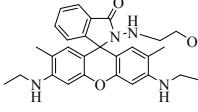
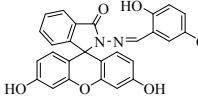
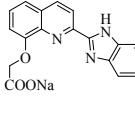
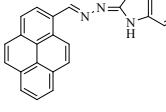
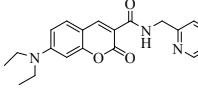
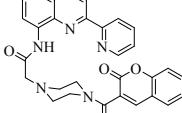
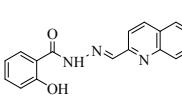
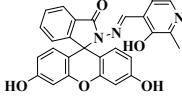
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Table S1. Crystal data and structure refinement for compound **2c**.

Empirical formula	C ₂₂ H ₂₈ N ₄
Formula weight	348.48
Crystal system	monoclinic
a [Å]	7.6253(3)
b [Å]	10.0149(4)
c [Å]	12.9039(5)
α [deg]	95.695(2)
β [deg]	93.482(2)
γ [deg]	97.522(2)
Volume [Å ³]	969.41(7)
Z	2
D _c [g/cm ³]	1.194
F (000)	376
T [K]	140(2)
Theta range for data collection [deg]	2.47 to 26.00
Wavelength [Å]	0.71073
Absorption coefficient [mm ⁻¹]	0.072
Completeness to theta	26.00 (99.9 %)
Reflections collected/unique	24230 / 3808 [R(int) = 0.0940]
Data/restraints/parameters	3808 / 0 / 235
R ₁ , wR ₂ [I>2σ(I)]	0.0692, 0.1510
R ₁ , wR ₂ (all data)	0.1363, 0.1712
Goodness of fit on F ²	1.017
Largest diff. peak and hole [e.Å ⁻³]	0.771 and -0.223

Table S2. Comparison with other reported Cu²⁺ fluorescence probes

Probes	Detection limit (M)	Type(E/Q)	Bio- imaging	Selectivity	Reference
	4.0×10 ⁻⁸	Quenched	Blank mice (in vivo)	Cu ²⁺	this work
	1.8×10 ⁻⁶	Enhanced	MG63 cells (in vitro)	Cu ²⁺	[39]
	2.5×10 ⁻⁵	Quenched	No data	Cu ²⁺	[40]
	1.7×10 ⁻⁷	Quenched	SMMC-7721 cells (in vitro)	Cu ²⁺ , S ²⁻	[41]
	2.73×10 ⁻⁶	Enhanced	RAW 264.7 cells (in vitro)	Cu ²⁺	[42]
	5×10 ⁻⁷	Quenched	LLC-MK2 cells (in vitro)	Cu ²⁺	[43]
	4.6×10 ⁻⁷	Quenched	No data	Cu ²⁺	[44]
	8.68×10 ⁻⁶	Quenched	No data	Cu ²⁺	[45]
	1.4×10 ⁻⁷	Quenched	HepG2 cells (in vitro)	Cu ²⁺ , Zn ²⁺	[46]

Supplemental Figures

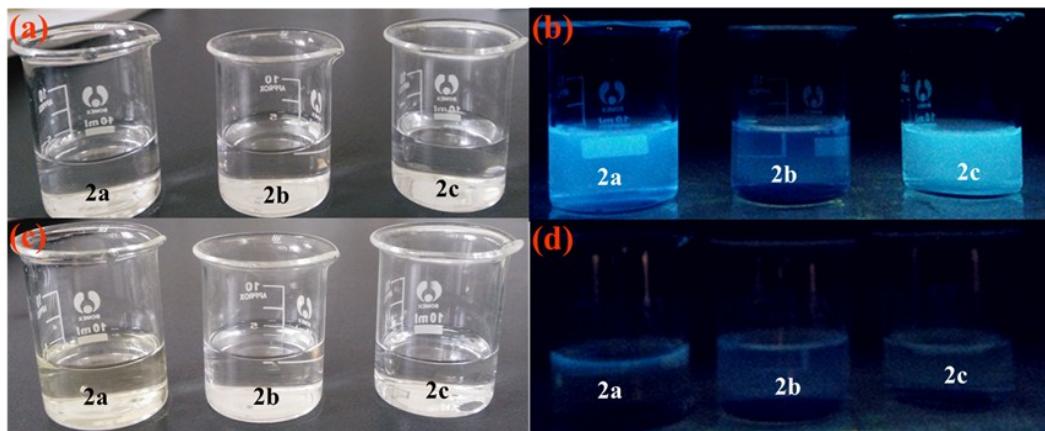


Fig. S1. Photographs of compounds **2a-2c** (1×10^{-4} M) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2) under sunlight (a) and 365 nm UV light (b); Photographs of compounds **2a-2c** (1×10^{-4} M) in presence of Cu^{2+} ion (1.5×10^{-3} M) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2) under sunlight (c) and 365 nm UV light (d).

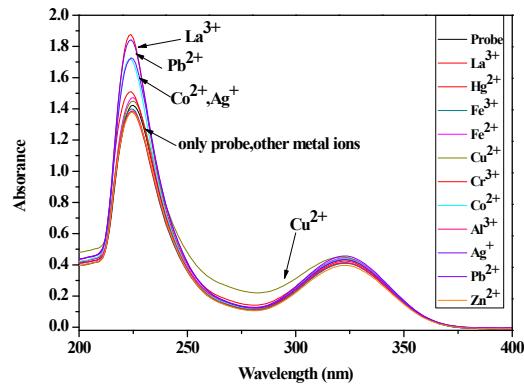


Fig. S2. UV-vis spectral changes of compound **2c** (1×10^{-4} M) upon addition of various metal ions (1×10^{-3} M) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2, 20mM HEPES buffer, pH=7.2) solution.

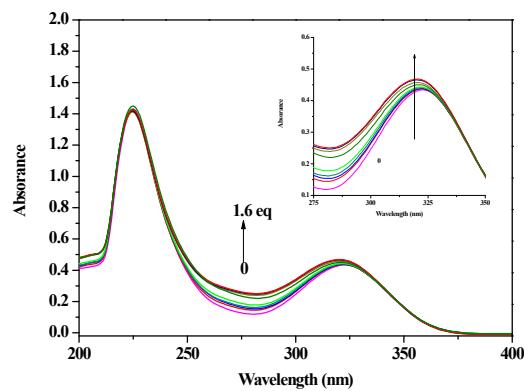


Fig. S3. UV-vis spectral changes of compound **2c** (1×10^{-4} M) with an increasing amount of Cu^{2+} ion in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2, 20mM HEPES buffer, pH=7.2) solution.

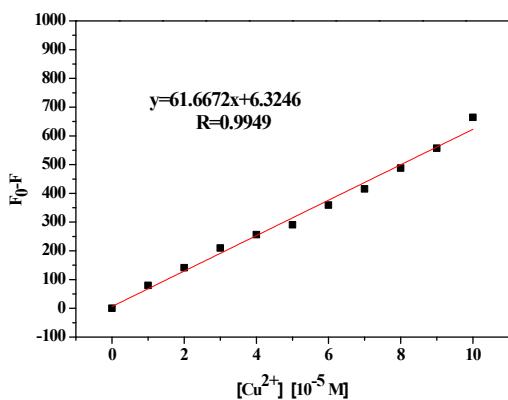


Fig. S4. Fluorescence intensity reduction of compound **2c** before and after quenching versus Cu^{2+} concentration. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

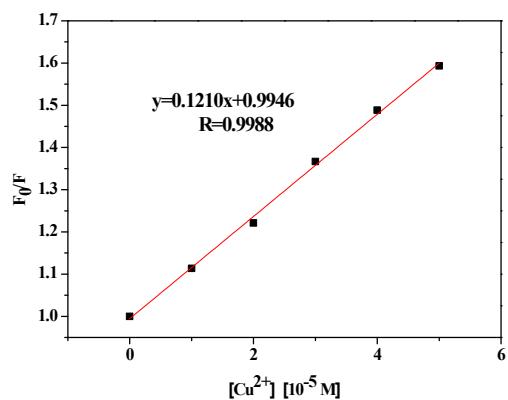


Fig. S5. The Stern-Volmer plots of compound **2c** with Cu^{2+} ion. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

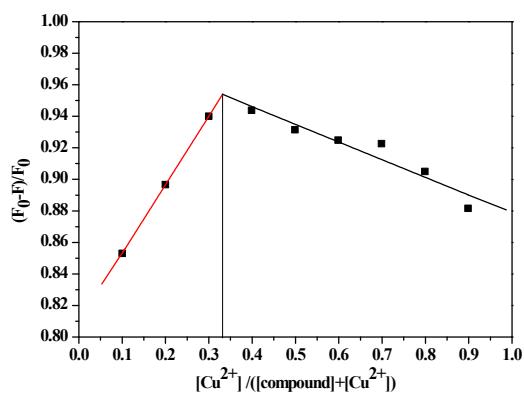


Fig. S6. Job plot of compound **2c** and Cu^{2+} in 8/2, v/v, 20 mM HEPES $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$, $\text{pH}=7.2$ (F_0 indicate fluorescence intensity of $10\mu\text{M}$ compound **2c**). The total concentrations of compound **2c** and Cu^{2+} were $1\times 10^{-5} \text{ M}$. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

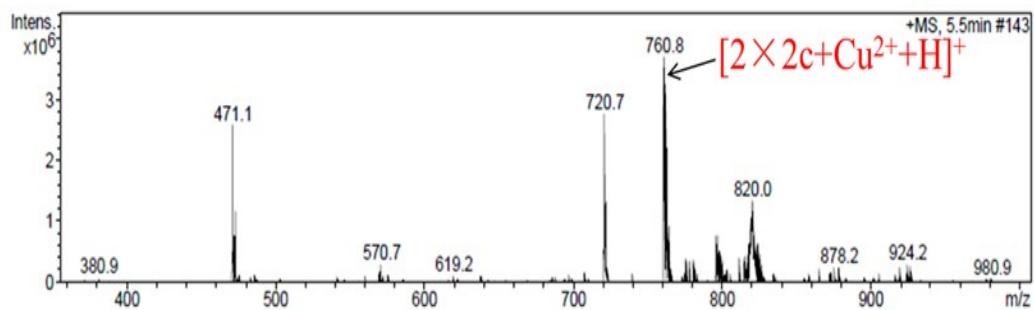


Fig. S7. Mass spectrometry of compound **2c** with Cu^{2+} measured by LC-MS.

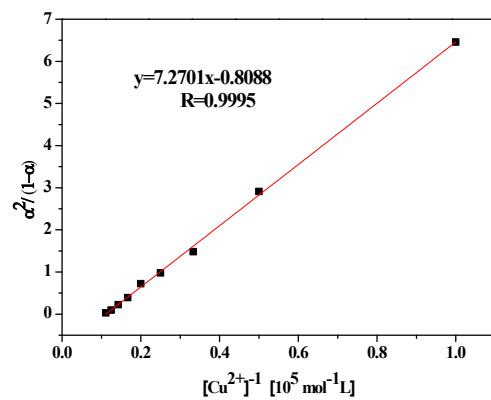


Fig. S8. The Benesi-Hildebrand plots of compound **2c** with Cu^{2+} ion. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

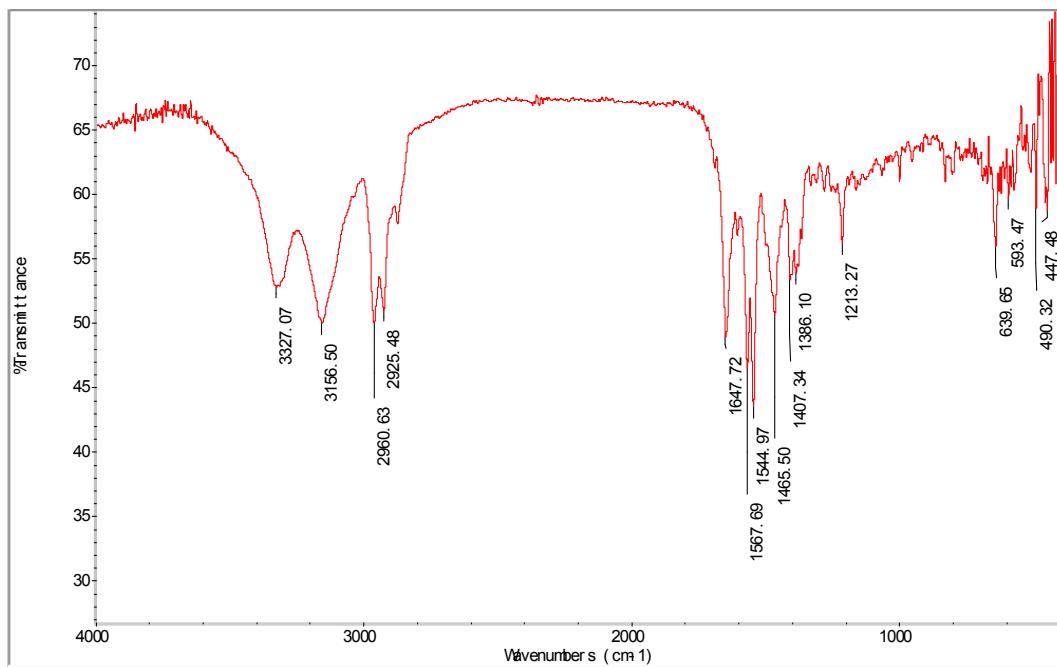


Fig. S9. The infrared spectra (KBr) of compound **2c**.

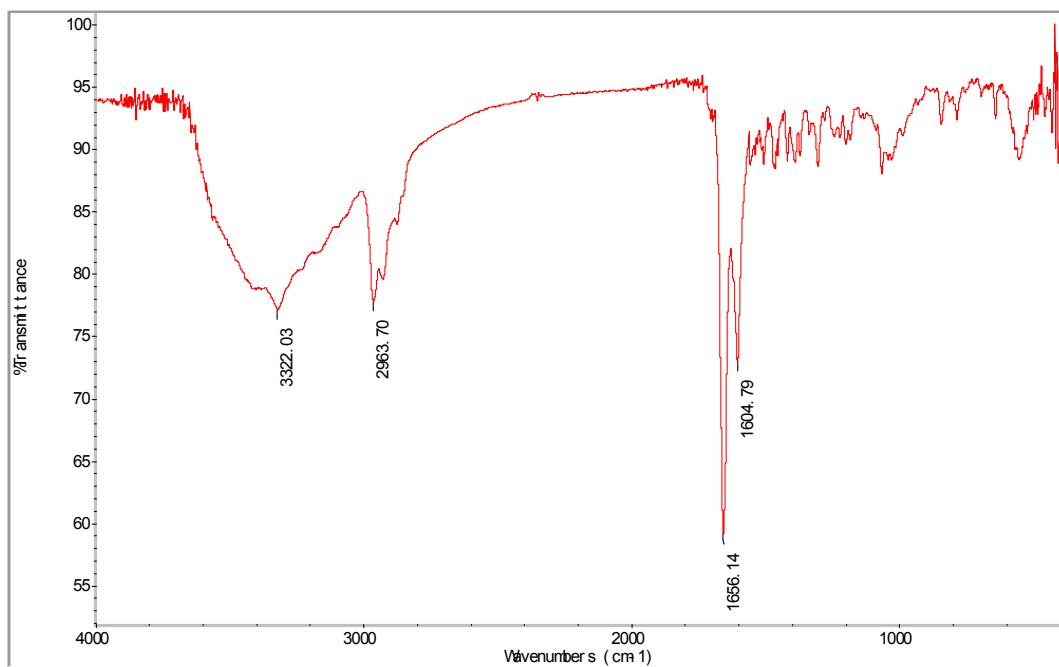


Fig. S10. The infrared spectra (KBr) of compound **2c** with Cu^{2+} .

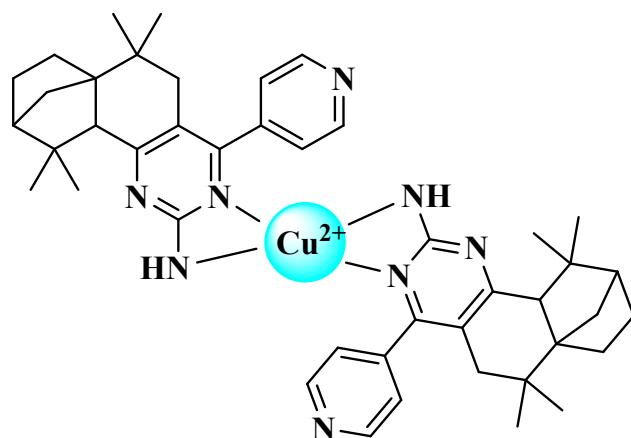


Fig. S11. The optimized binding mode of compound **2c** with Cu^{2+} .

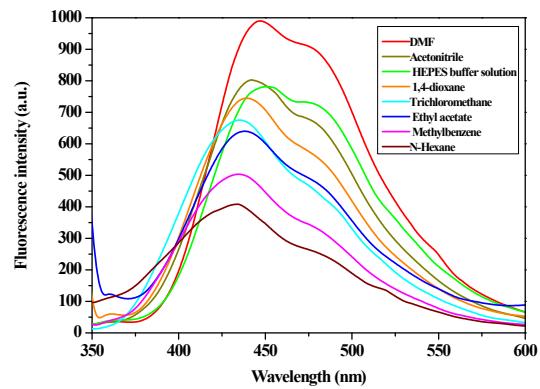


Fig. S12. Fluorescence spectra of compound **2c** (1×10^{-5} M) in different solvents. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

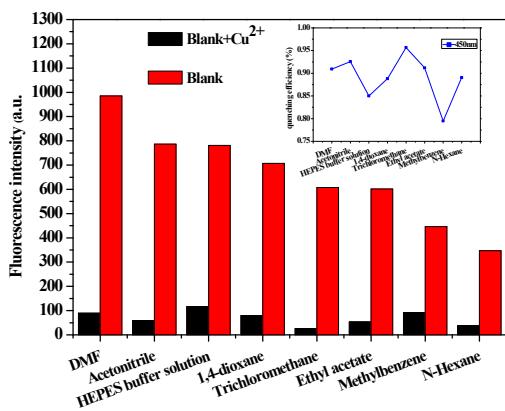


Fig. S13. Fluorescence intensity changes of compound **2c** (1×10^{-5} M) in absence and presence of Cu^{2+} (1×10^{-4} M) at 450nm in different solvents. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

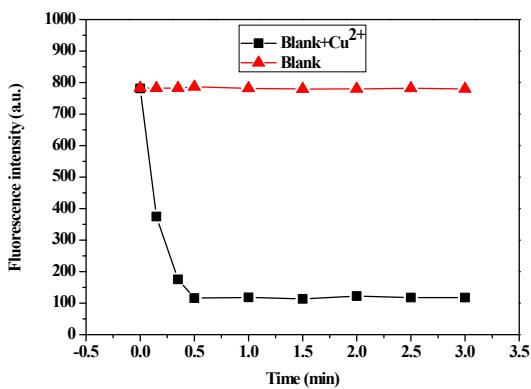


Fig. S14. The response time profiles of compound **2c** (1×10^{-5} M) towards Cu^{2+} ion (1×10^{-4} M) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ ($v/v=8/2$, 20mM HEPES buffer, pH=7.2) solution. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

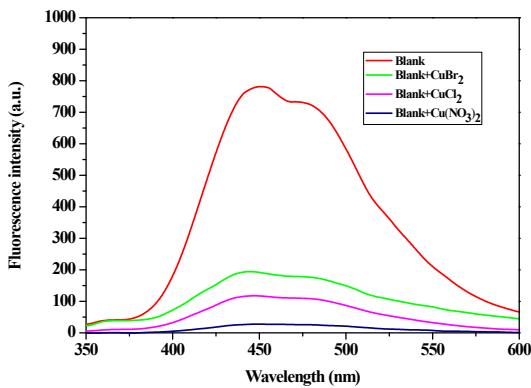


Fig. S15. Fluorescence spectra of compound **2c** (1×10^{-5} M) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ ($v/v=8/2$, 20mM HEPES buffer, pH=7.2) solution upon addition of various copper salts (1×10^{-4} M). Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

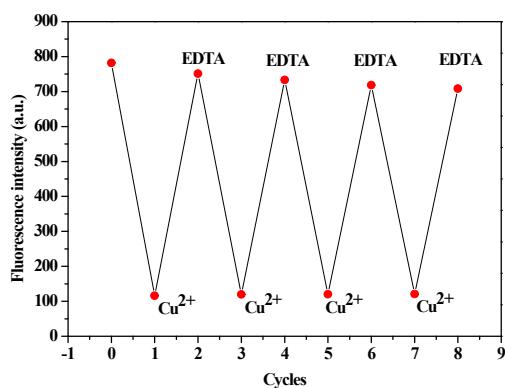


Fig. S16. Fluorescence intensity of compound **2c** (1×10^{-5} M) with Cu^{2+} ion (1×10^{-4} M) in presence of EDTA disodium (0-10 μM) in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2, 20mM HEPES buffer, pH=7.2) solution. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

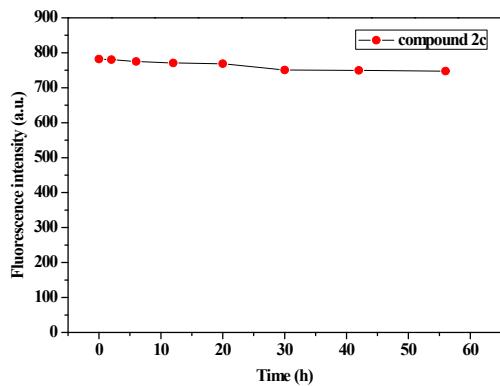


Fig. S17. Fluorescence intensity of compound **2c** in $\text{CH}_3\text{CH}_2\text{OH}-\text{H}_2\text{O}$ (v/v=8/2, 20mM HEPES buffer, pH=7.2) solution under continuous illumination fluorescent lamp. Excitation wavelength at 325 nm; Em. Slit at 12 nm; Ex. Slit at 5 nm.

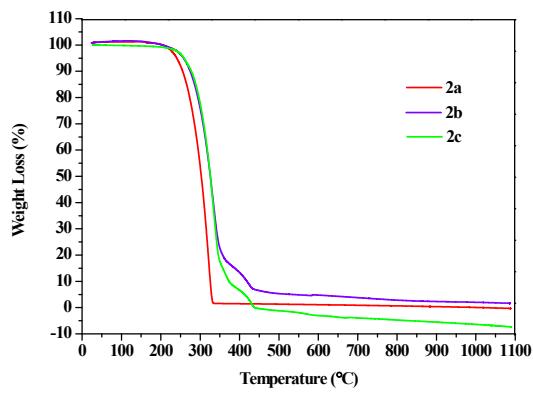


Fig. S18. TGA curves of compounds **2a-2c**.

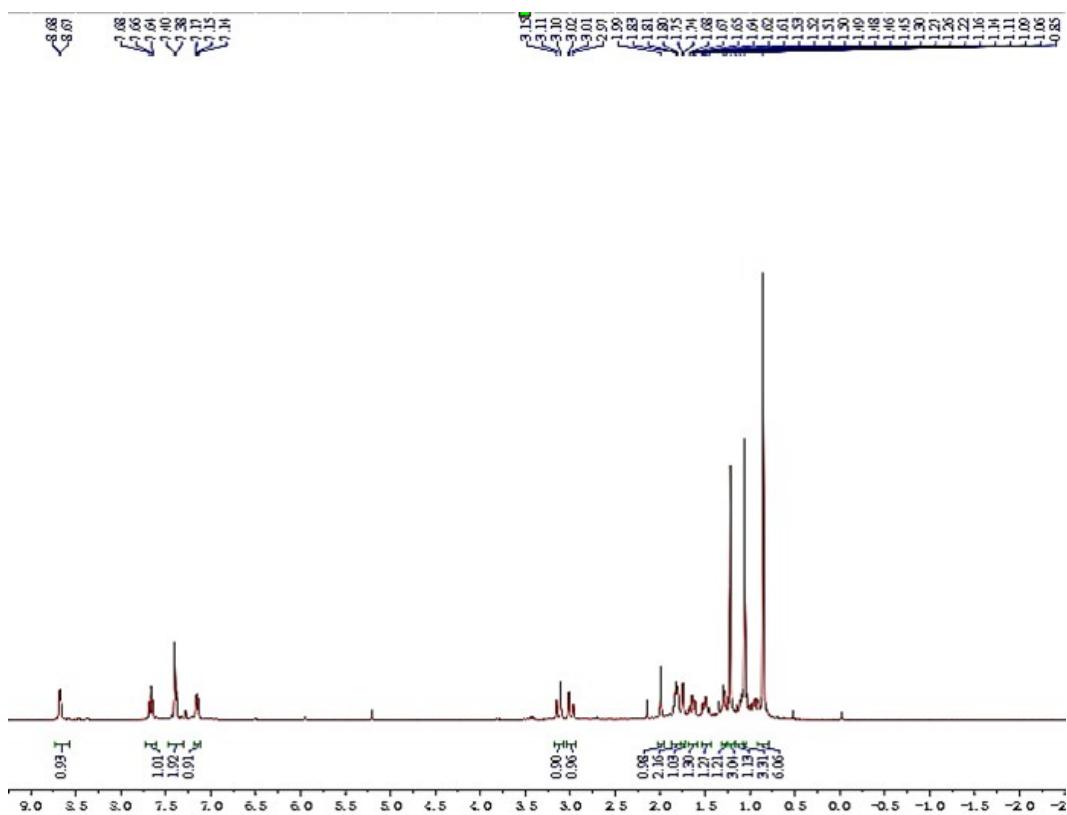


Fig. S19. ^1H NMR spectra of compound **1a** in CDCl_3 .

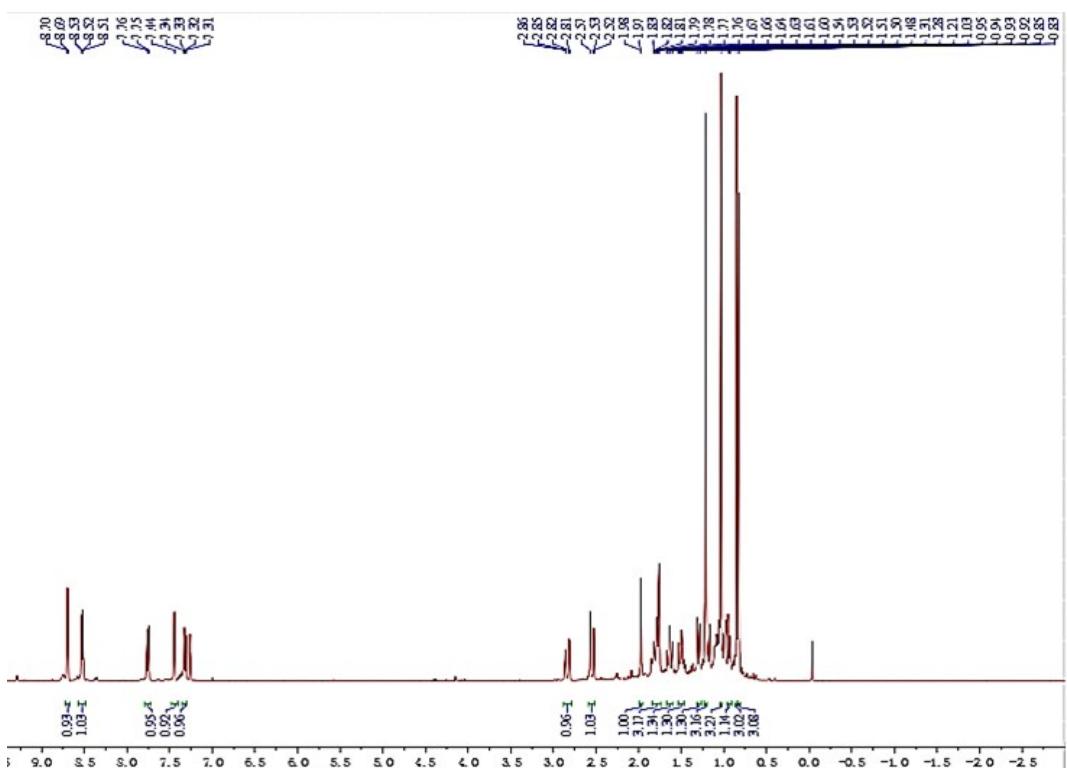


Fig. S20. ^1H NMR spectra of compound **1b** in CDCl_3 .

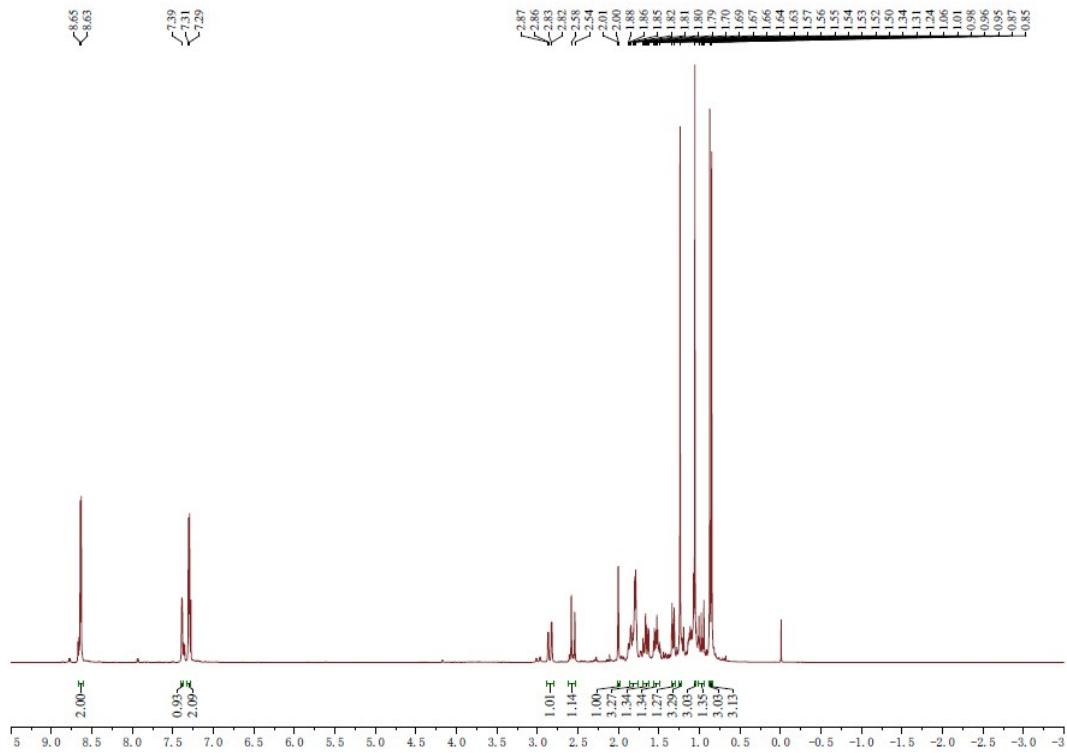


Fig. S21. ^1H NMR spectra of compound **1c** in CDCl_3 .

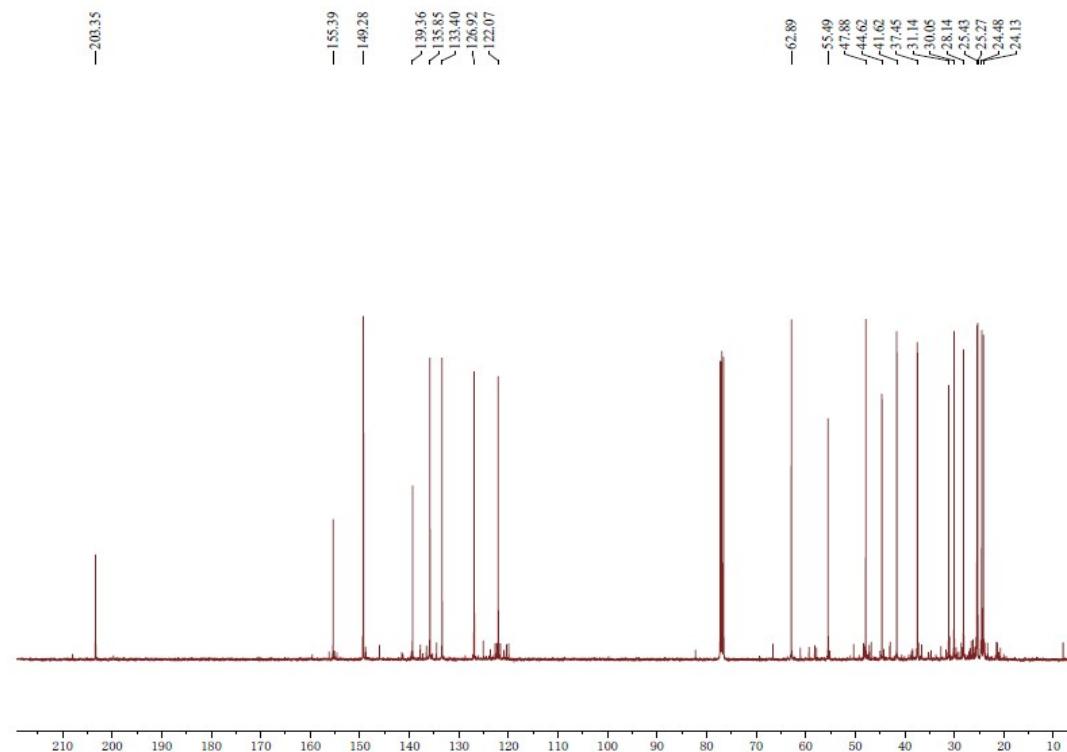


Fig. S22. ^{13}C NMR spectra of compound **1a** in CDCl_3 .

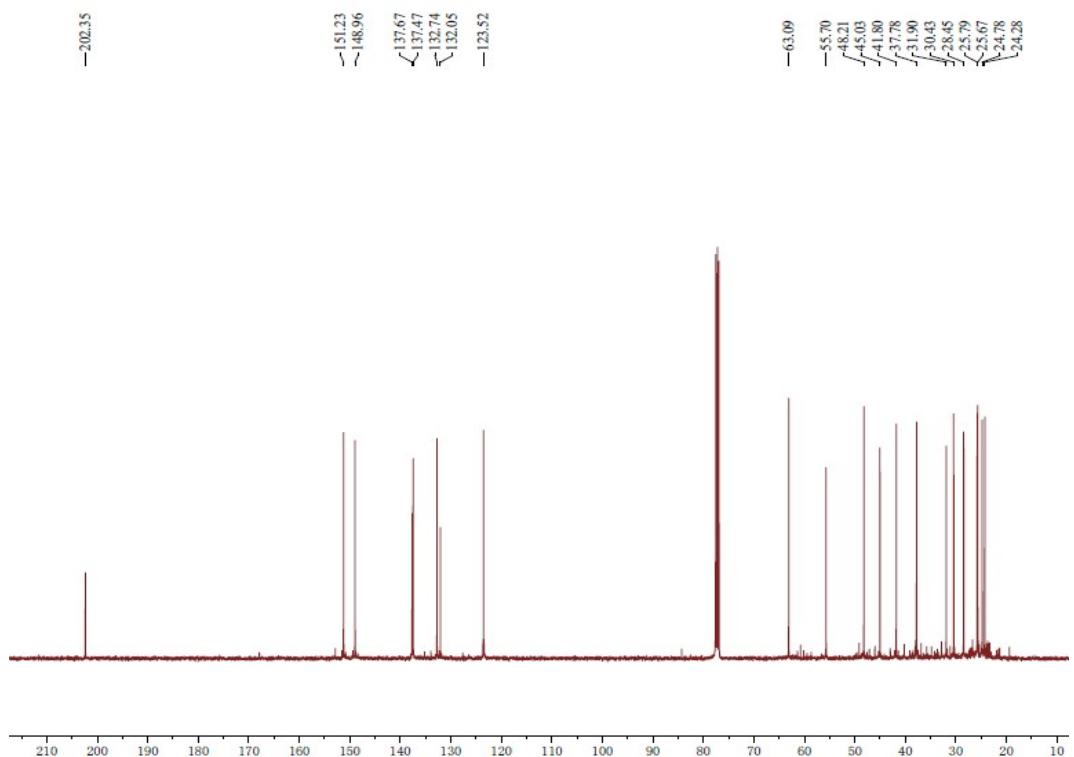


Fig. S23. ^{13}C NMR spectra of compound **1b** in CDCl_3 .

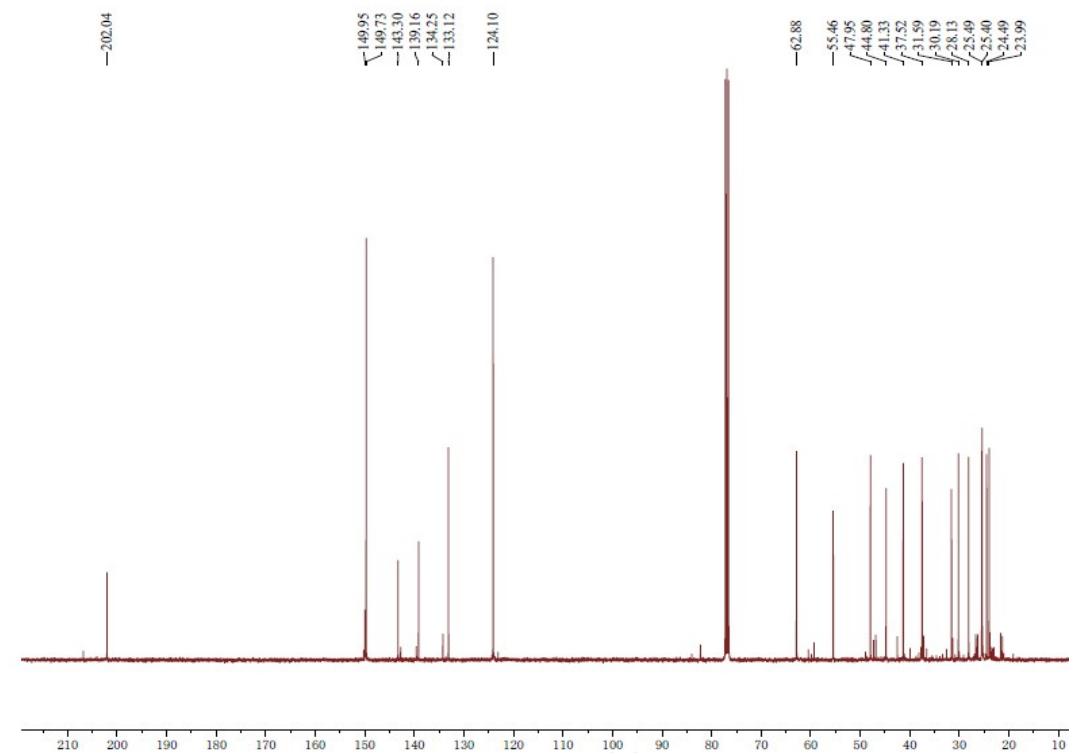


Fig. S24. ^{13}C NMR spectra of compound **1c** in CDCl_3 .

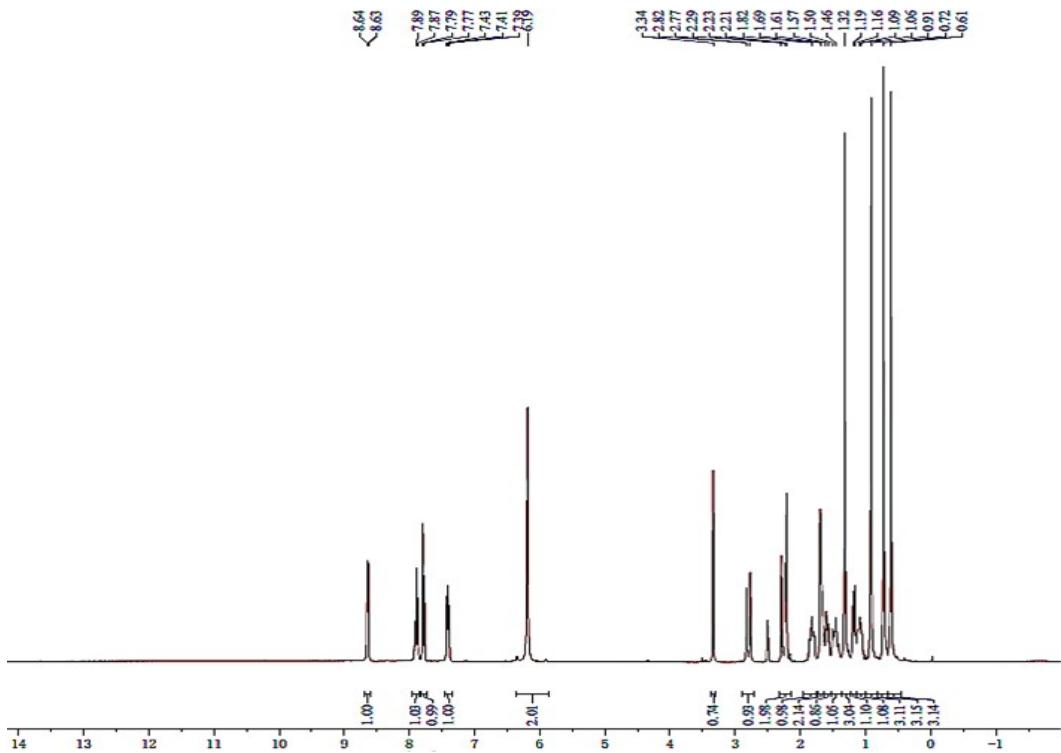


Fig. S25. ^1H NMR spectra of compound **2a** in DMSO-d₆.

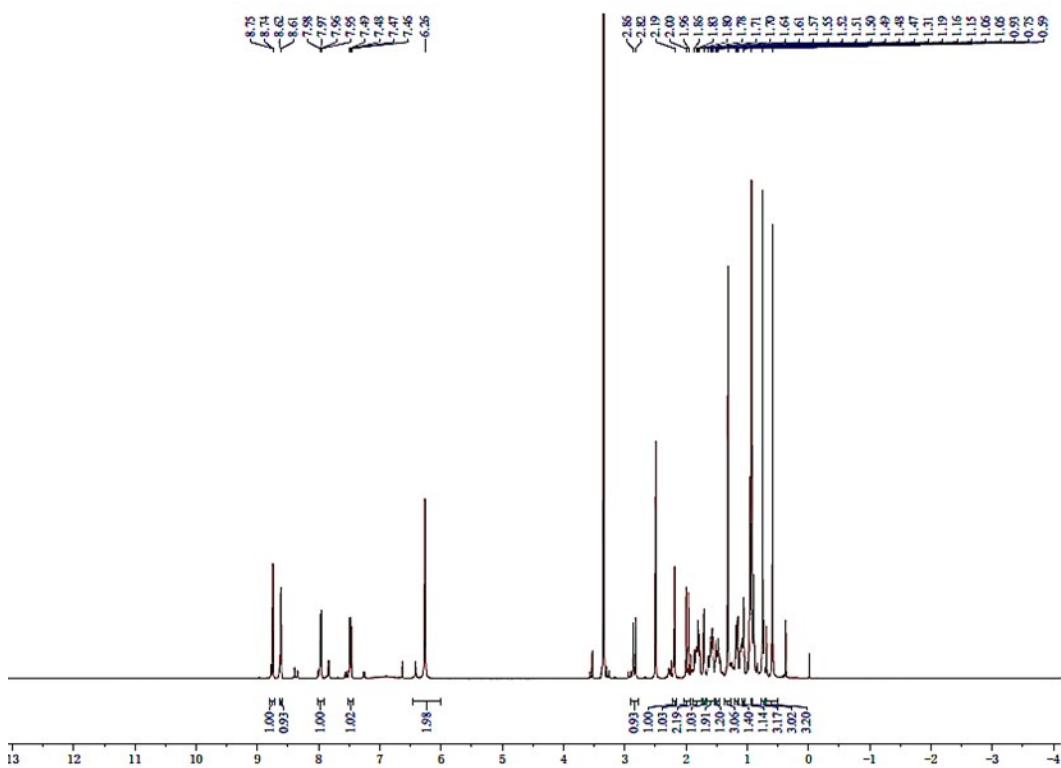


Fig. S26. ^1H NMR spectra of compound **2b** DMSO-d₆.

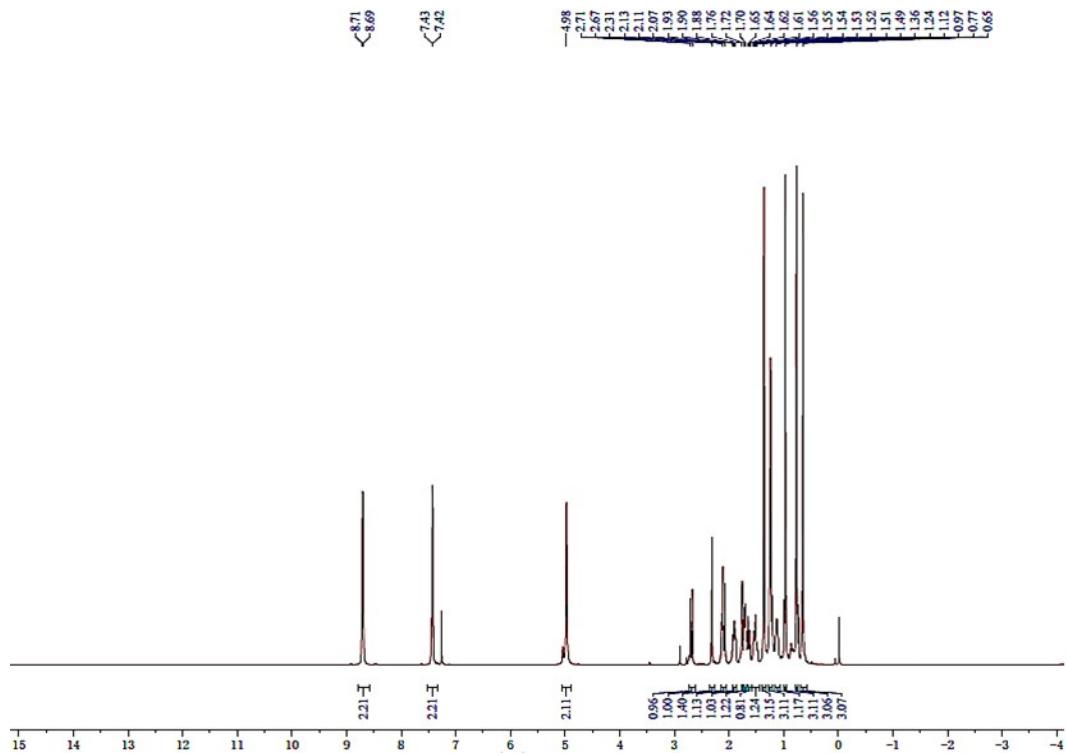


Fig. S27. ^1H NMR spectra of compound **2c** in CDCl_3 .

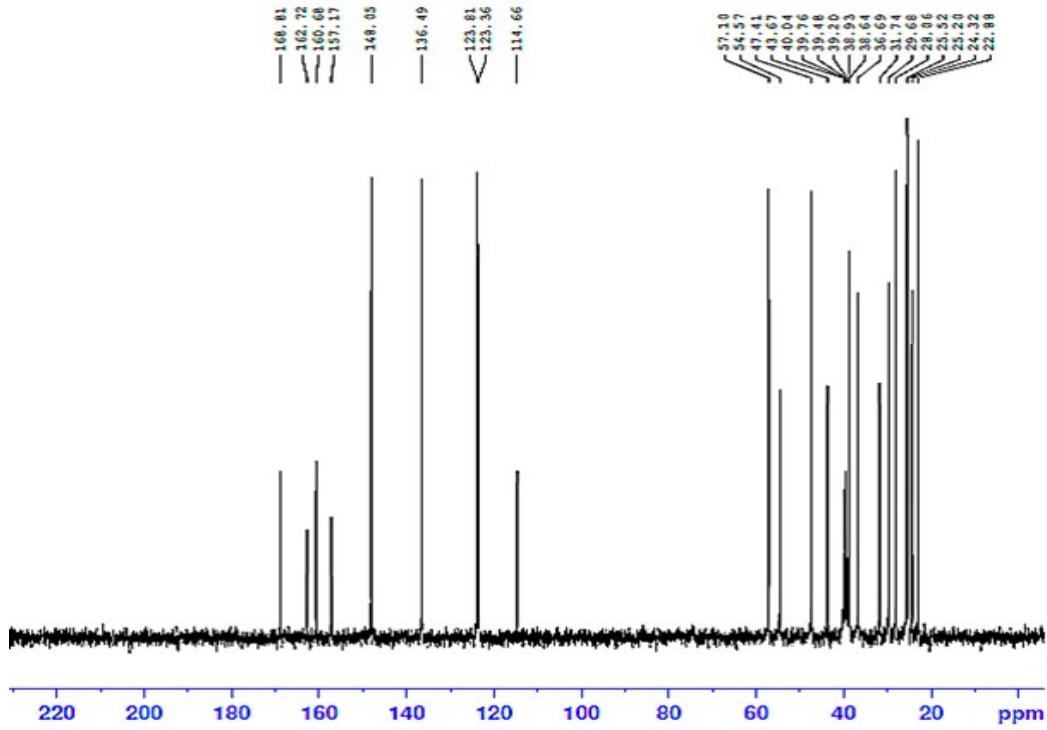


Fig. S28. ^{13}C NMR spectra of compound **2a** in DMSO-d_6 .

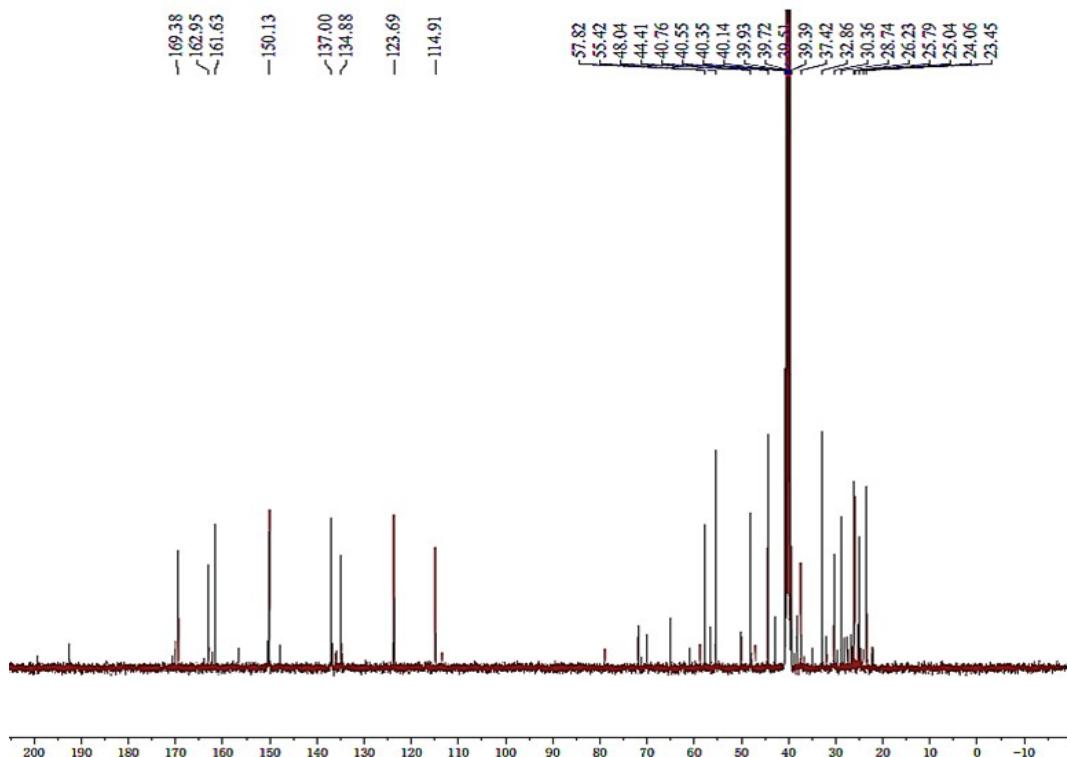


Fig. S29. ¹³C NMR spectra of compound **2b** in DMSO-d₆.

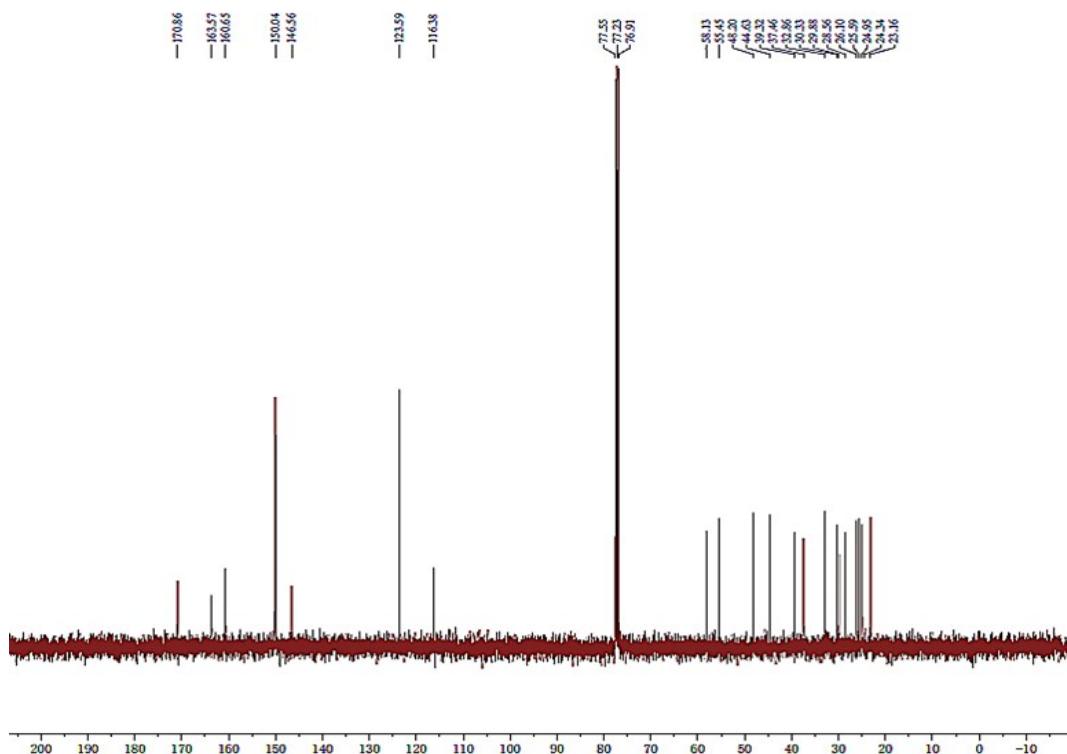


Fig. S30. ¹³C NMR spectra of compound **2c** in CDCl₃.

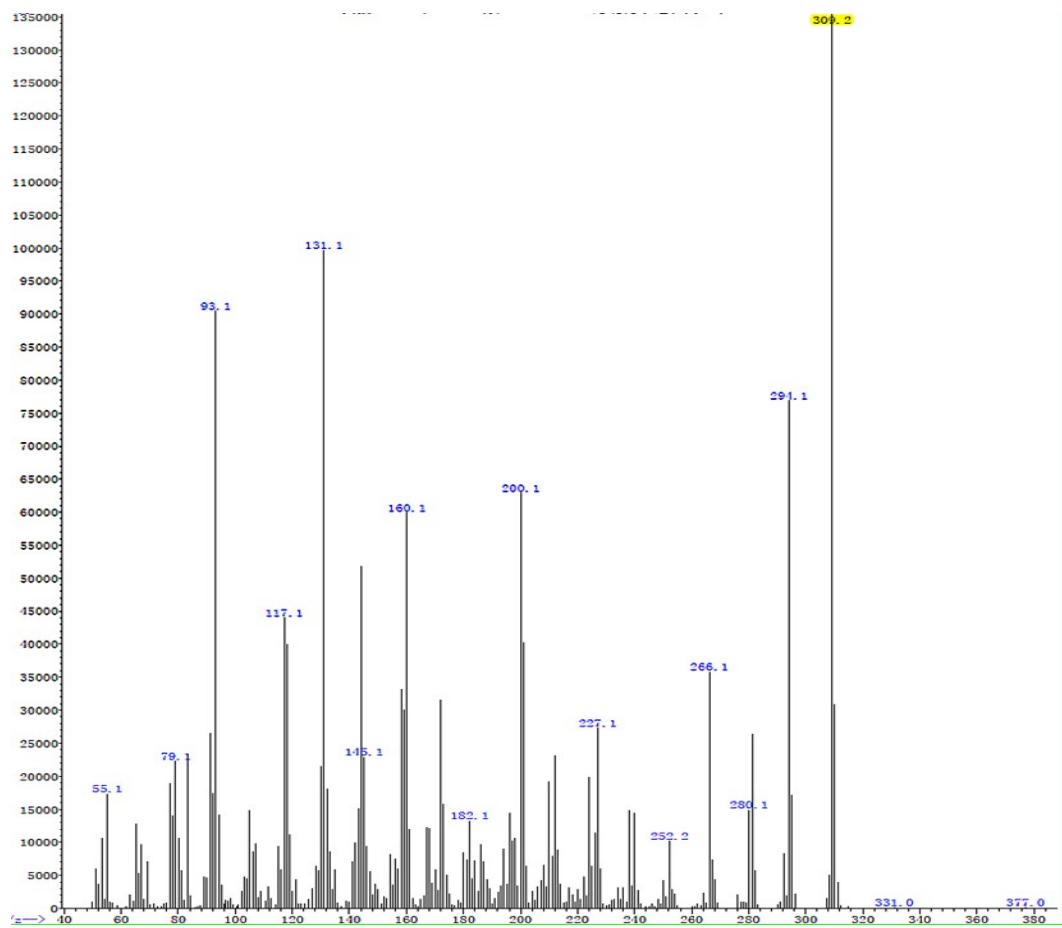


Fig. S31. Mass spectrometry of compound 1a.

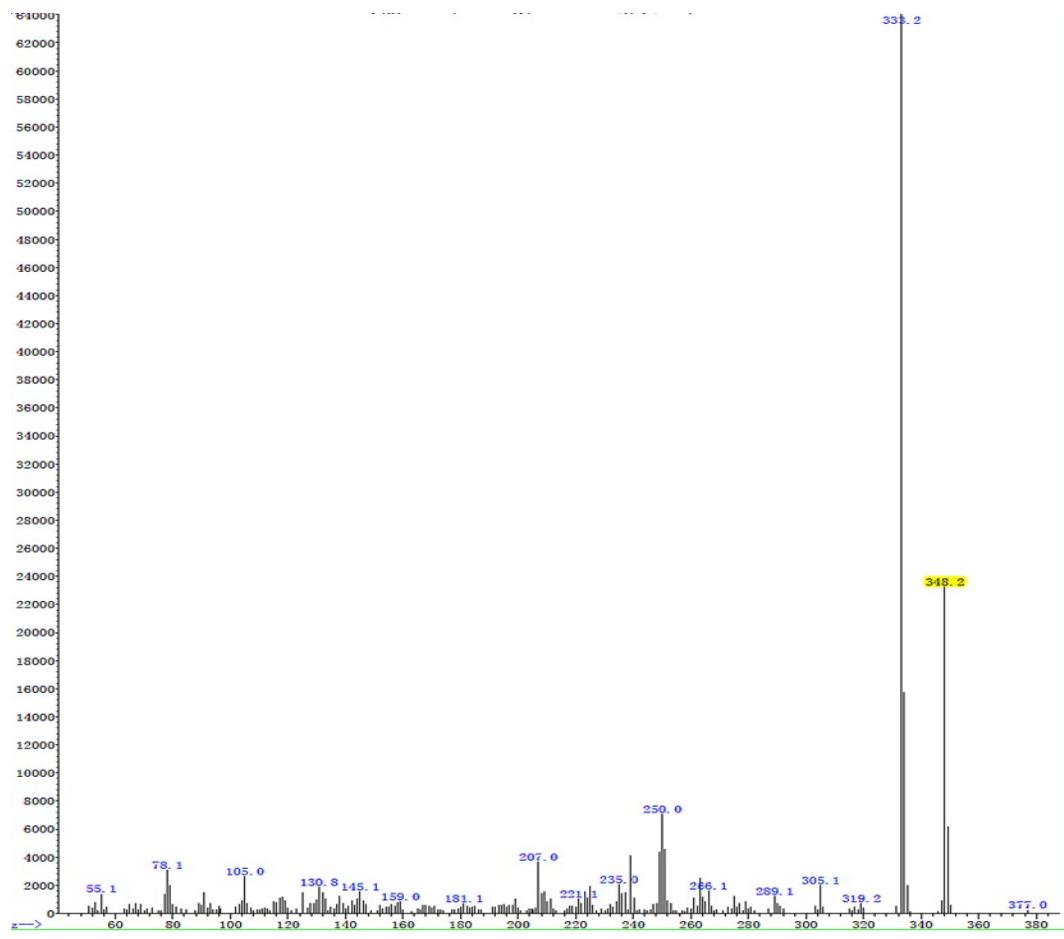


Fig. S32. Mass spectrometry of compound 2a.

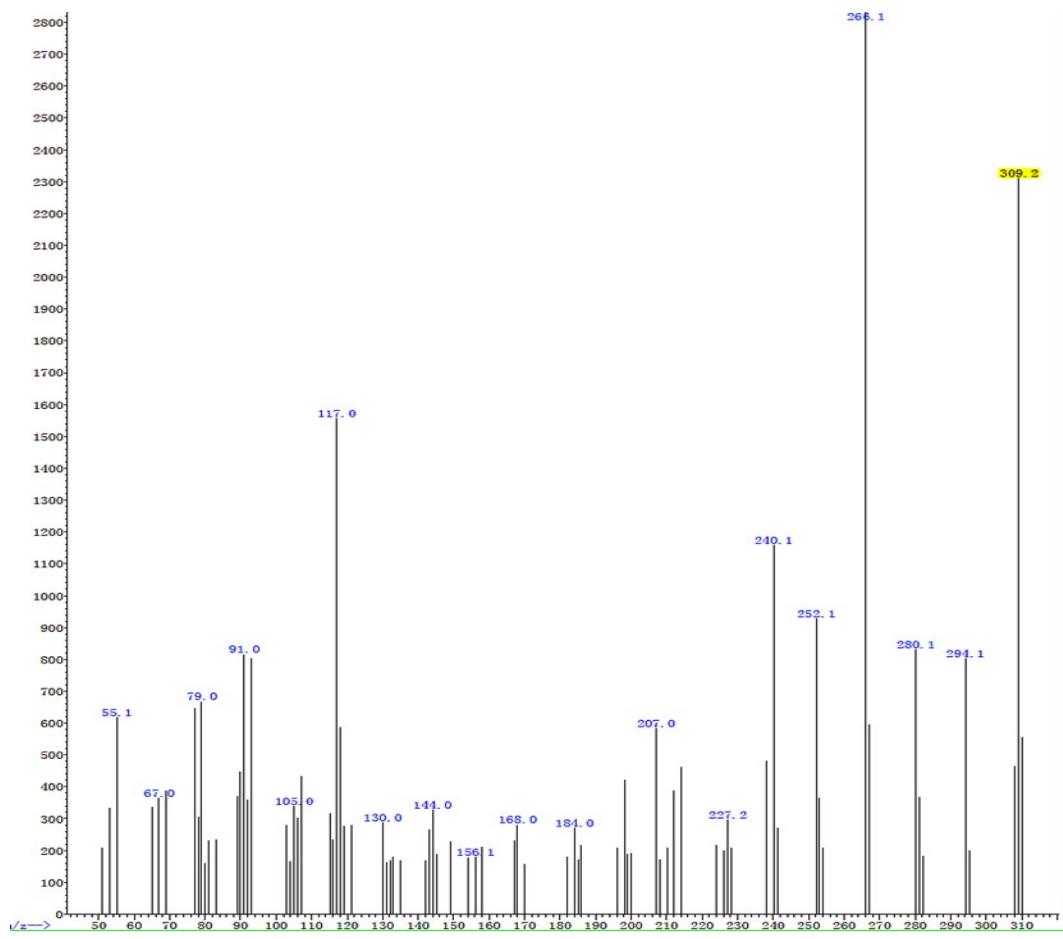


Fig. S33. Mass spectrometry of compound 1b.

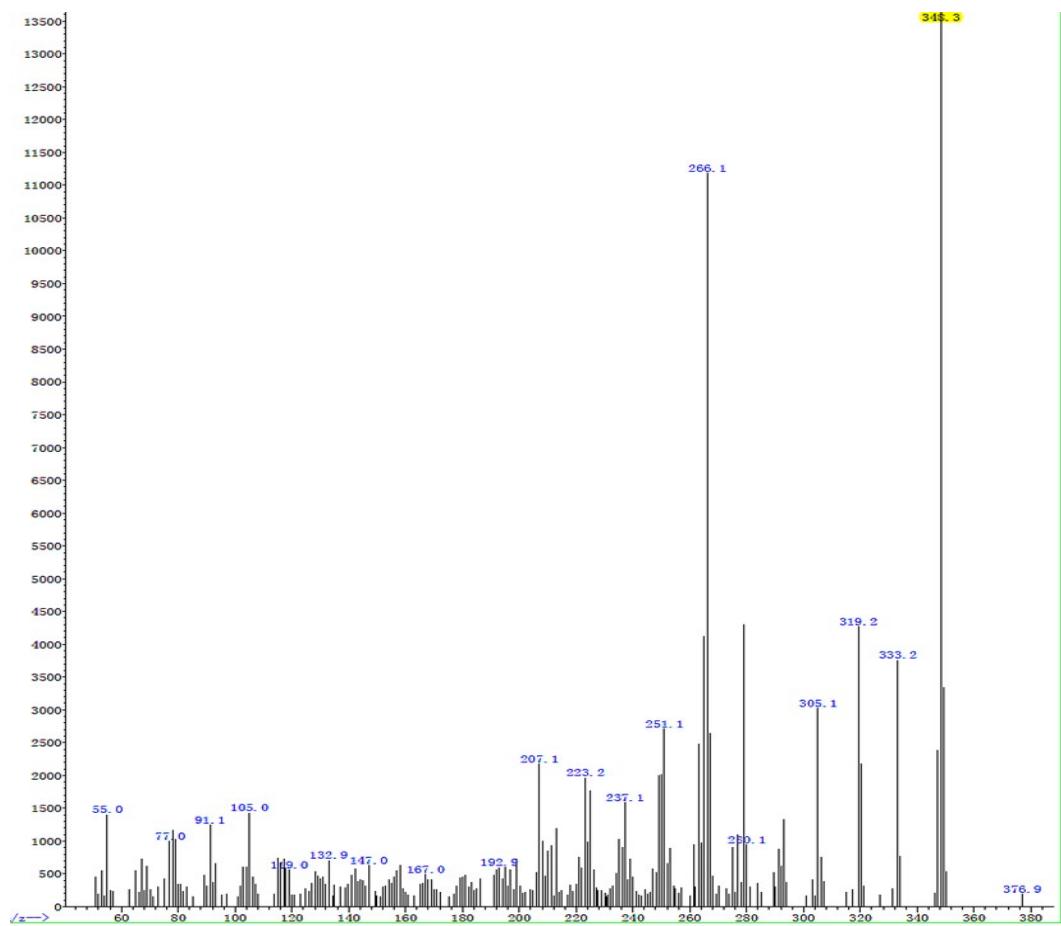


Fig. S34. Mass spectrometry of compound **2b**.

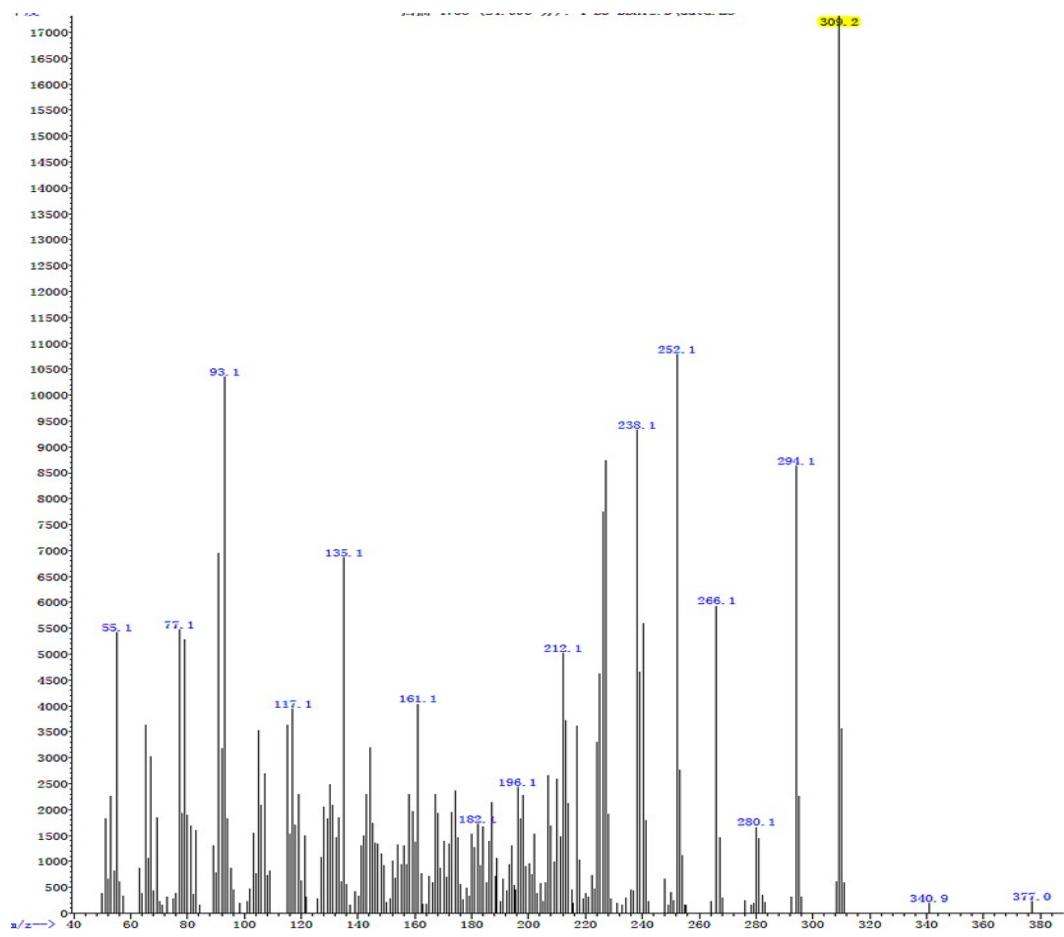


Fig. S35. Mass spectrometry of compound **1c**.

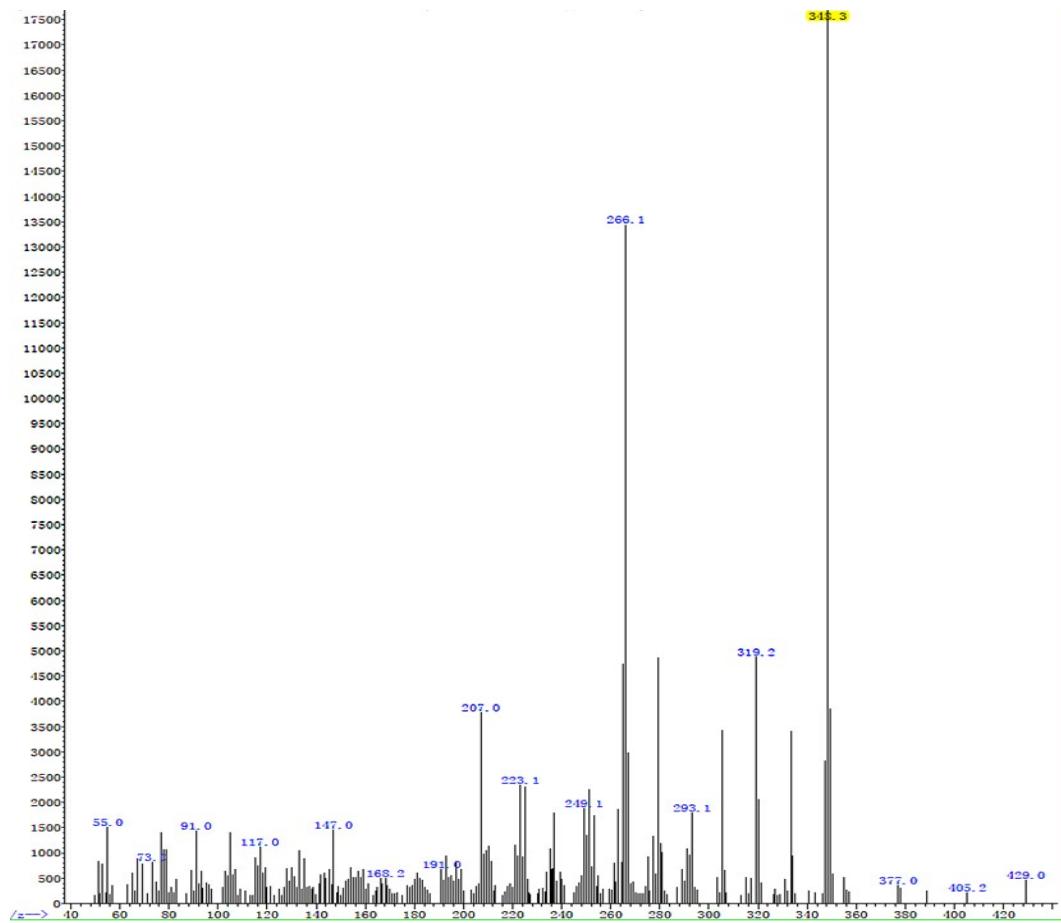


Fig. S36. Mass spectrometry of compound **2c**.

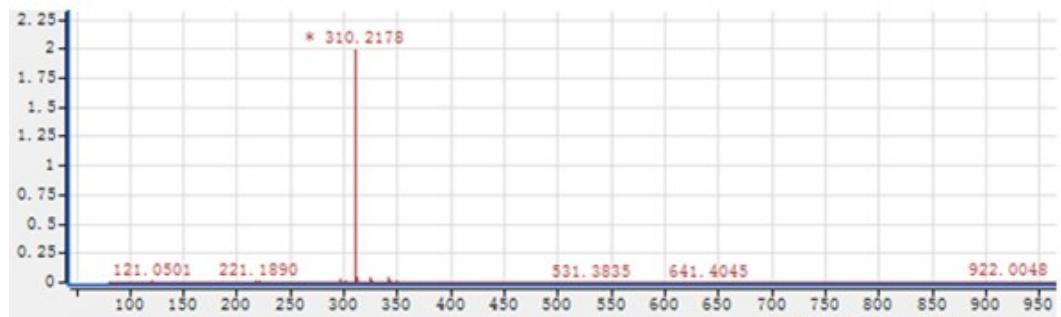


Fig. S37. HRMS of compound **1a**.

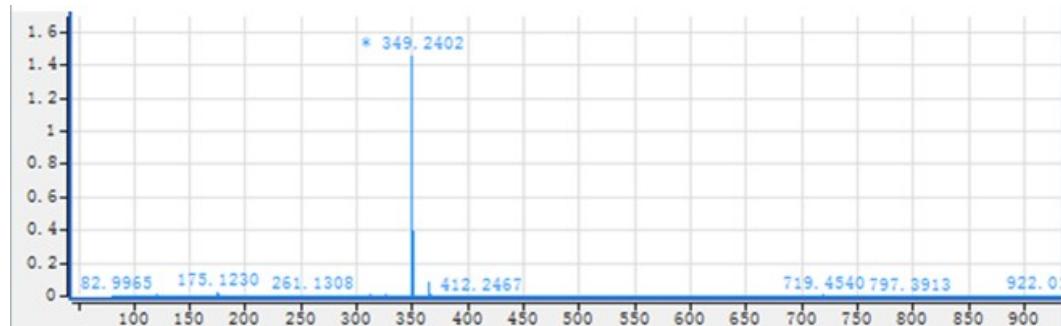


Fig. S38. HRMS of compound **2a**.

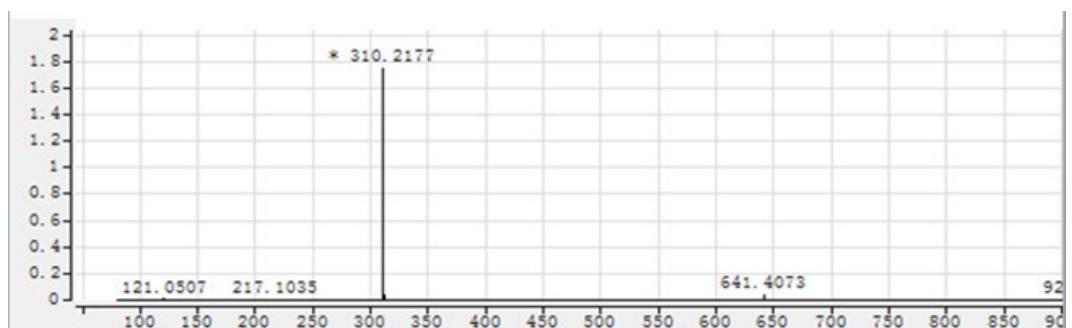


Fig. S39. HRMS of compound 1b.



Fig. S40. HRMS of compound 2b.

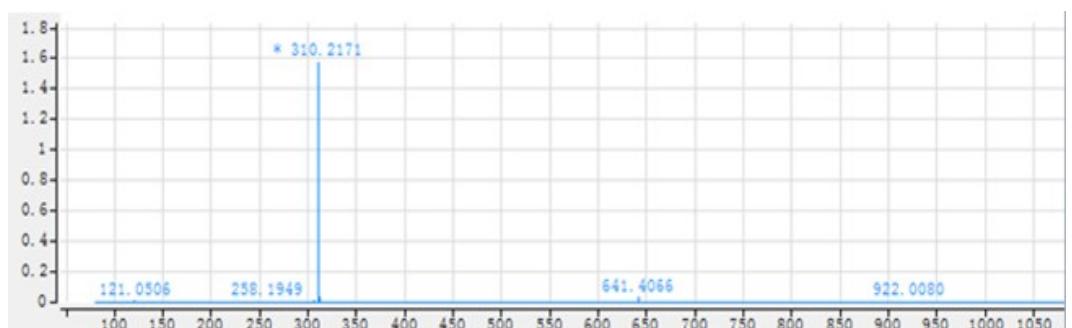


Fig. S41. HRMS of compound 1c.



Fig. S42. HRMS of compound 2c.

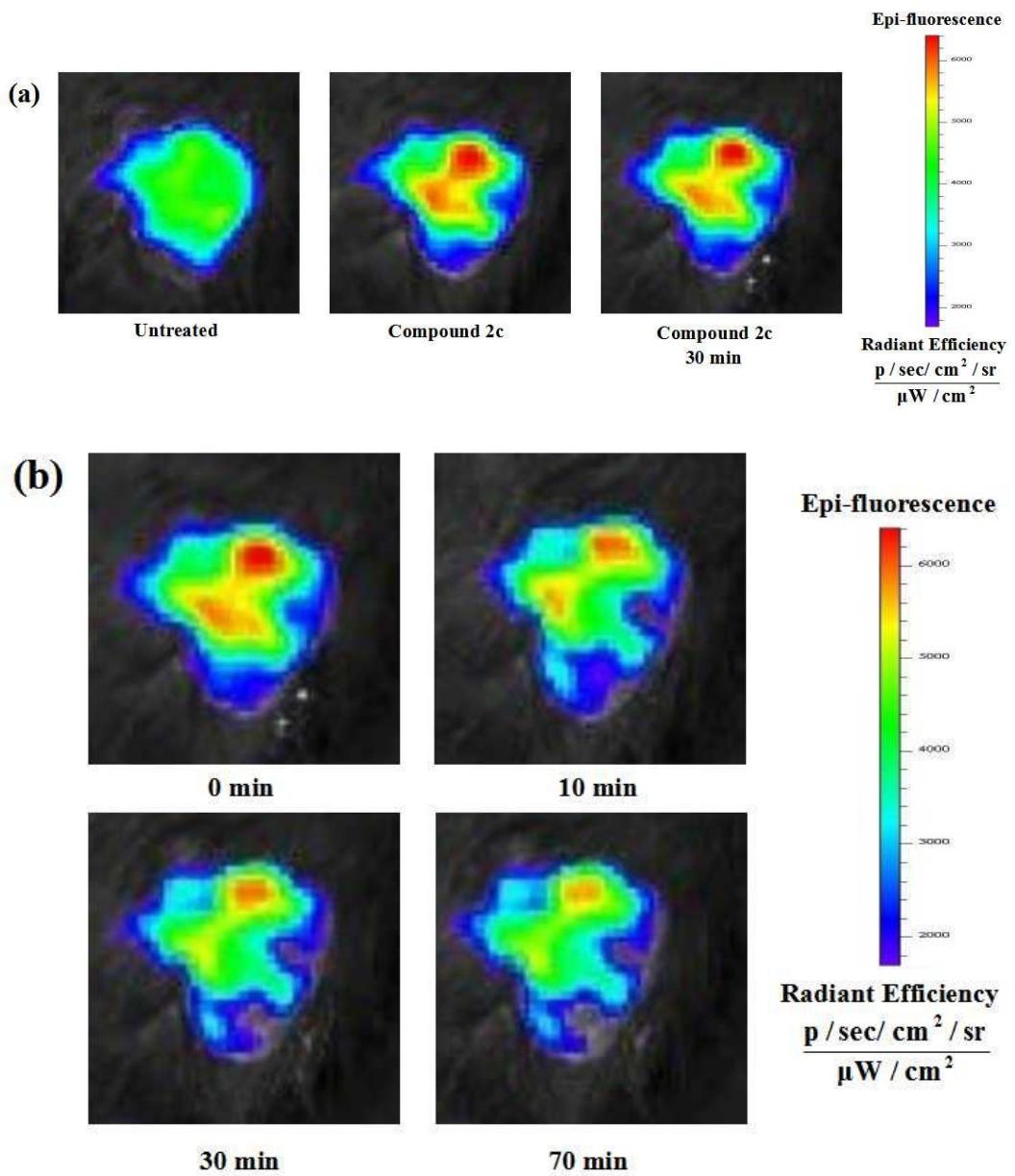


Fig. S43. (a) The zoomed in versions of injected part (the hip of mouse) in **Fig. 10a**, (b) the zoomed in versions of injected part (the hip of mouse) in **Fig. 10b**.