

## **One-step synthesis of magnetically recyclable Co@BN core-shell nanocatalysts for catalytic reduction of nitroarenes**

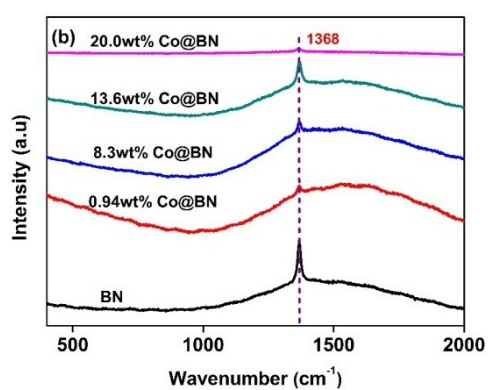
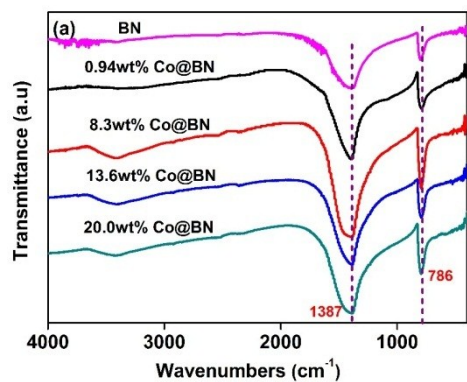
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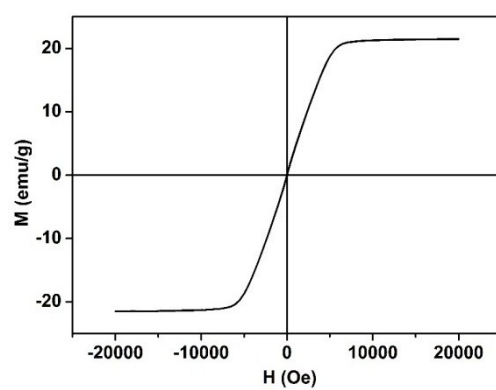
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**Fig. S1** (a) FTIR and (b) Raman spectra of *h*-BN and the Co@BN samples with various Co contents.

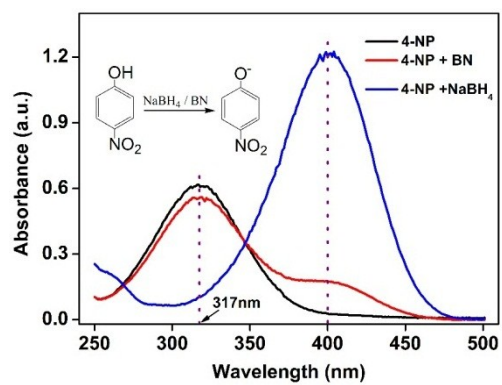
(a)



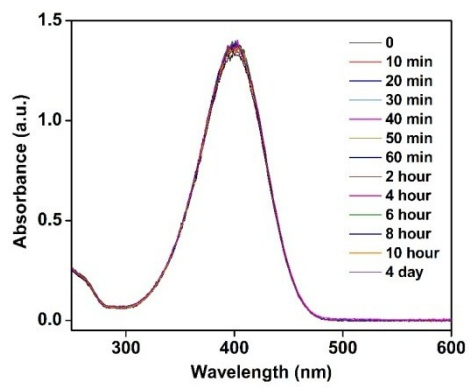
(b)



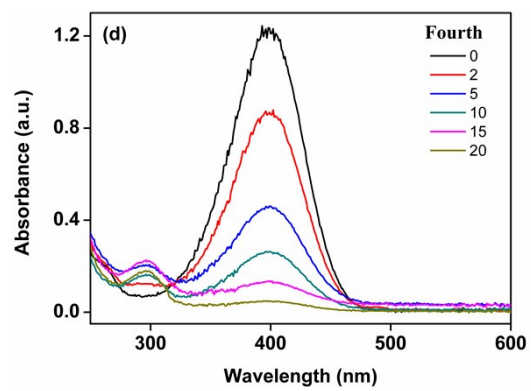
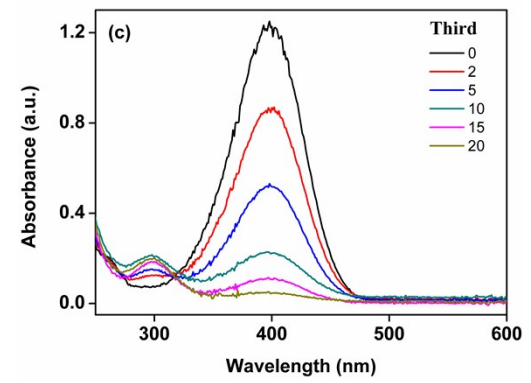
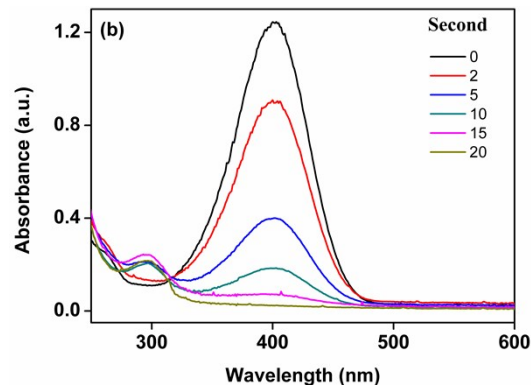
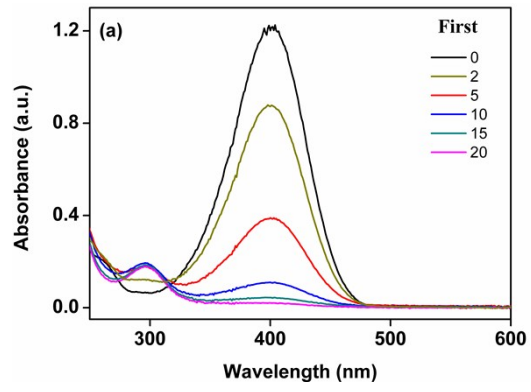
**Fig. S2.** (a) Hysteresis loops of the 13.6 wt% Co@BN nanocatalyst measured at 300 K. (b) photographs of the Co@BN dispersion before (left) and after (right) the magnetic separation.

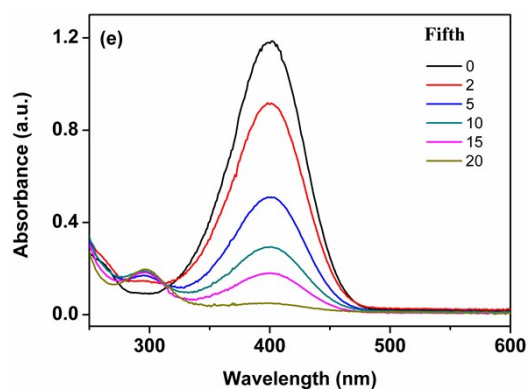


**Fig. S3.** Evolution of UV-vis spectra of the 4-NP solution added with NaBH<sub>4</sub> or *h*-BN.

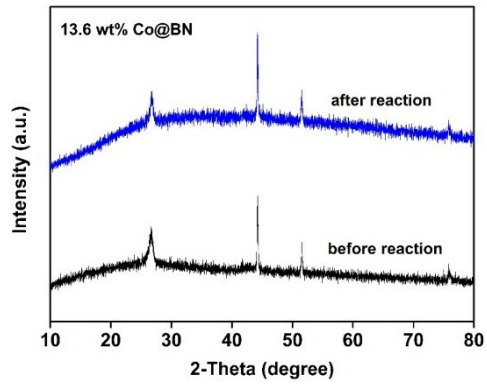


**Fig. S4.** Catalytic reduction of 4-nitrophenol in the presence of NaBH<sub>4</sub>.



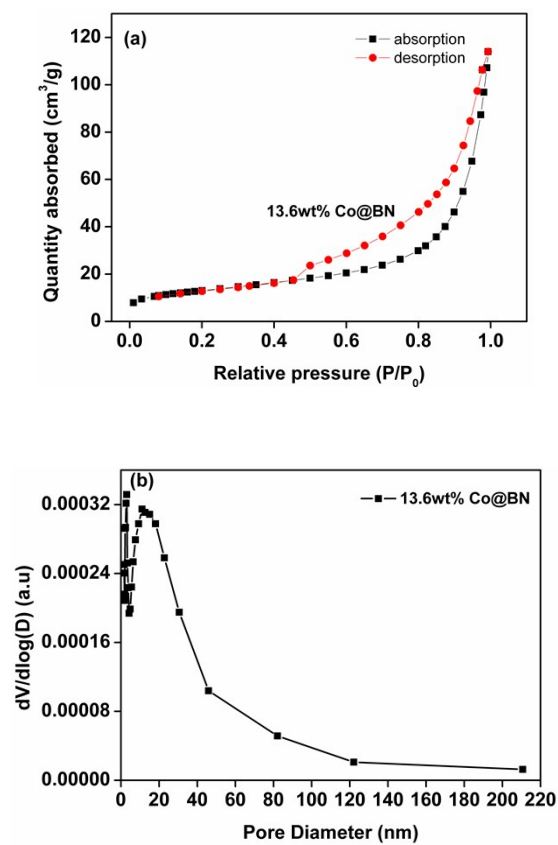


**Fig. S5.** Successive UV-vis spectra for the reduction reaction of 4-NP by NaBH<sub>4</sub> using the catalyst of 13.6 wt% Co@BN nanoparticles in each catalysis recycle.

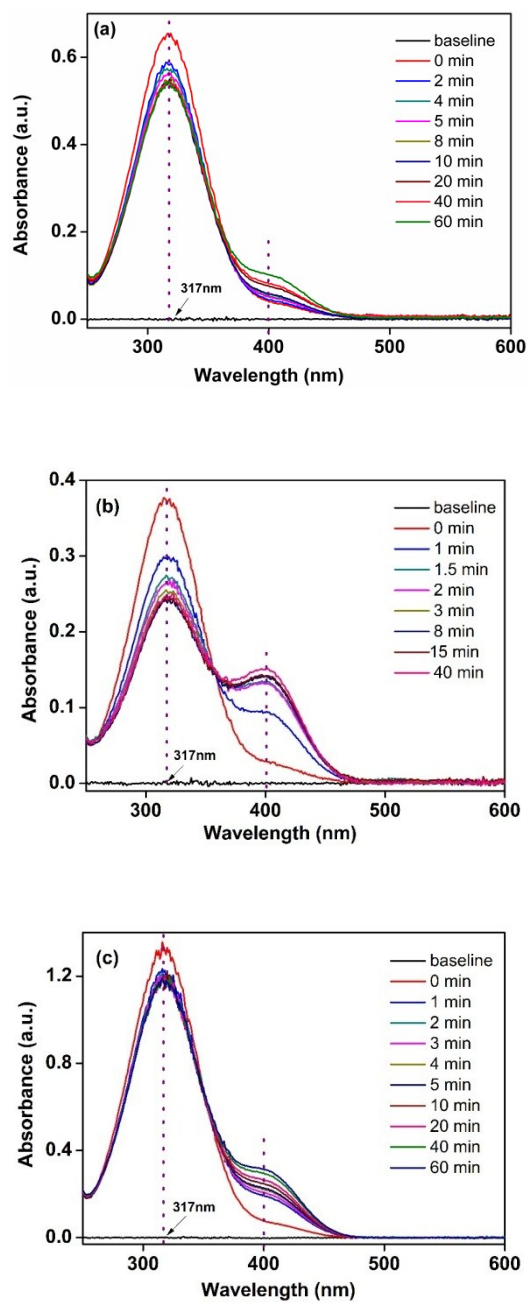


**Fig. S6.** XRD patterns of 13.6 wt% Co@BN nanoparticles before and after the reaction.

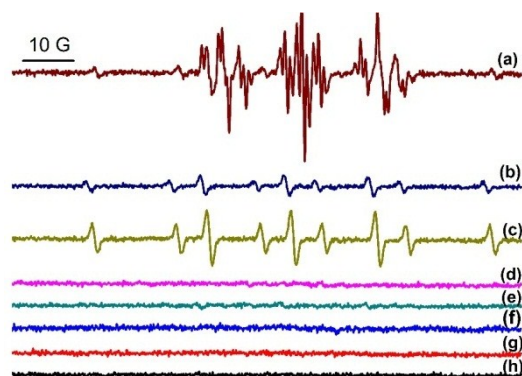




**Fig. S7.** (a)  $N_2$  adsorption–desorption isotherms and (b) the corresponding pore size distributions of 13.6 wt% Co@BN.



**Fig. S8.** Absorbance of 4-NP at a concentration of (a) 100 ppm, (b) 50ppm and (c) 200ppm in the presence of *h*-BN ( 10 mg).



**Fig. S9.** Spin adducts formed in the presence of (a) 4-NP+NaBH<sub>4</sub>+catalyst+DMPO, (b) 4-NP+NaBH<sub>4</sub>+DMPO, (c) NaBH<sub>4</sub>+DMPO, (d) catalyst + DMPO, (e) 4-NP+DMPO, (f) catalyst, (g) NaBH<sub>4</sub> and (h) 4-NP.

**Table S1.** Reduction of 4-nitrophenol to 4-aminophenol by some cobalt-based catalysts.

| Entry | catalyst            | Yield (%)          | Con. (%) | TONs | Select. (%) <sup>a</sup> | Ref <sup>c</sup> |
|-------|---------------------|--------------------|----------|------|--------------------------|------------------|
| 1     | FeCo@N-doped C      | N. P. <sup>b</sup> | >95      | 10   | N. P.                    | 1                |
| 2     | Co@N-C 700          | N. P.              | >99      | 6    | N. P.                    | 2                |
| 3     | p(AMPS)-Co          | N. P.              | >97      | 5    | N. P.                    | 3                |
| 4     | Co@BN               | >96                | >96      | 5    | >99                      | This work        |
| 5     | Co@SiO <sub>2</sub> | N. P.              | >99      | 4    | N. P.                    | 4                |
| 6     | meso-Co-150         | N. P.              | >99      | 4    | N. P.                    | 5                |
| 7     | Co-NC               | N. P.              | >93      | 4    | N. P.                    | 6                |

<sup>a</sup> The selectivity can reach >99% because 4-aminophenol is the sole product for this model reaction.

<sup>b</sup> N. P.: Not provided.

<sup>c</sup> Refs:

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2. Y. Yusran, D. Xu, Q. Fang, D. Zhang and S. Qiu, *Micropor. Mesopor. Mat.*, 2017, **241**, 346-354.
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**Table S2.** The load of Co on the recycled catalyst 13.6 wt% Co@BN.

| Recycle | Initial | First run | Second run | Third run | Fourth run | Fifth run |
|---------|---------|-----------|------------|-----------|------------|-----------|
| Co wt%  | 13.6    | 13.3      | 13.2       | 12.9      | 12.5       | 11.5      |