## Highly active and porous single-crystal $In_2O_3$ nanosheet for excellent-response $NO_x$ gas sensor at room temperature

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Figure S1. TG-DSC analyses of In(OH)<sub>3</sub> precursors



Figure S2. The TEM images of the synthesized  $In_2O_3$  with 0.6 g  $In(NO_3)_3$ .



Figure S3. The TEM images of the synthesized  $In_2O_3$  with 0.8 g  $In(NO_3)_3$ .



Figure S4. The TEM images of the synthesized  $In_2O_3$  with 1.2 g  $In(NO_3)_3$ .

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	Sample	S <sub>BET</sub>	pore volume	pore size
		(m² g-1)	(cm <sup>3</sup> g <sup>-1</sup> )	(nm)
	In <sub>2</sub> O <sub>3</sub> -0.6	50.99	0.53	34.73
	In <sub>2</sub> O <sub>3</sub> -0.7	52.89	0.47	36.10
	In <sub>2</sub> O <sub>3</sub> -0.8	65.70	0.27	18.07
	In <sub>2</sub> O <sub>3</sub> -0.9	44.85	0.28	25.51
	In <sub>2</sub> O <sub>3</sub> -1.0	41.09	0.29	28.89
	In <sub>2</sub> O <sub>3</sub> -1.2	36.33	0.16	21.21

Table S1 The results of the surface area, pore volume and pore size.

Response (R_N-R_0)/R_0Sample97.048.5 ppm29.1 ppm9.704.85 ppm0.970.485 ppmppmppmppmppmppmppmln_2O_3-0.714.4513.109.1511.494.212.522.46ln_2O_3-0.820.3713.488.423.52.240.77ln_2O_3-0.989.4880.2675.1245.4922.7211.984.77ln_2O_3-1.014.8813.5911.6212.178.867.951.67ln_2O_3-1.217.009.997.365.395.062.151.73								
Sample97.048.5 ppm29.1 ppm9.704.85 ppm0.970.485 ppmppmppmppmppmppmppmln2O3-0.714.4513.109.1511.494.212.522.46ln2O3-0.820.3713.488.423.52.240.77ln2O3-0.989.4880.2675.1245.4922.7211.984.77ln2O3-1.014.8813.5911.6212.178.867.951.67ln2O3-1.217.009.997.365.395.062.151.73	Response (R <sub>N</sub> -R <sub>0</sub> )/R <sub>0</sub>							
ppmppmln2O3-0.714.4513.109.1511.494.212.522.46ln2O3-0.820.3713.488.423.52.240.77ln2O3-0.989.4880.2675.1245.4922.7211.984.77ln2O3-1.014.8813.5911.6212.178.867.951.67ln2O3-1.217.009.997.365.395.062.151.73	Sample	97.0	48.5 ppm	29.1 ppm	9.70	4.85 ppm	0.97	0.485 ppm
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		ppm			ppm		ppm	
In2O3-0.8 20.37 13.48 8.42 3.5 2.24 0.77    In2O3-0.9 89.48 80.26 75.12 45.49 22.72 11.98 4.77   In2O3-1.0 14.88 13.59 11.62 12.17 8.86 7.95 1.67   In2O3-1.2 17.00 9.99 7.36 5.39 5.06 2.15 1.73	In <sub>2</sub> O <sub>3</sub> -0.7	14.45	13.10	9.15	11.49	4.21	2.52	2.46
In2O3-0.9 89.48 80.26 75.12 45.49 22.72 11.98 4.77   In2O3-1.0 14.88 13.59 11.62 12.17 8.86 7.95 1.67   In2O3-1.2 17.00 9.99 7.36 5.39 5.06 2.15 1.73	In <sub>2</sub> O <sub>3</sub> -0.8	20.37	13.48	8.42	3.5	2.24	0.77	
In2O3-1.014.8813.5911.6212.178.867.951.67In2O3-1.217.009.997.365.395.062.151.73	In <sub>2</sub> O <sub>3</sub> -0.9	89.48	80.26	75.12	45.49	22.72	11.98	4.77
ln <sub>2</sub> O <sub>3</sub> -1.2 17.00 9.99 7.36 5.39 5.06 2.15 1.73	In <sub>2</sub> O <sub>3</sub> -1.0	14.88	13.59	11.62	12.17	8.86	7.95	1.67
	In <sub>2</sub> O <sub>3</sub> -1.2	17.00	9.99	7.36	5.39	5.06	2.15	1.73

Table S2 The gas response of synthesized  $In_2O_3$  with different  $In(NO_3)_3$  addition for 97.0 ppm~0.485 ppm  $NO_x$  at room temperature.

	Response time (s)						
Sample	97.0	48.5 ppm	29.1 ppm	9.70	4.85 ppm	0.97	0.485 ppm
	ppm			ppm		ppm	
In <sub>2</sub> O <sub>3</sub> -0.7	27	74	182	292	27	182	230
In <sub>2</sub> O <sub>3</sub> -0.8	20.6	20.6	16.6	20.0	27.3	27.3	
In <sub>2</sub> O <sub>3</sub> -0.9	16.6	24.0	27.3	36.0	38.6	39.3	59.3
In <sub>2</sub> O <sub>3</sub> -1.0	11.33	74.0	79.3	72.6	133	136	400
In <sub>2</sub> O <sub>3</sub> -1.2	17.3	422.0	470.0	492.6	320.0	130.0	203.3

Table S3 The response time of synthesized  $In_2O_3$  with different  $In(NO_3)_3$  addition for 97.0 ppm~0.485 ppm  $NO_x$  at room temperature.

	Material	Operating temperature	NO <sub>x</sub> Concentration (ppm)	Response	Lowest detectable limit (ppm)
Our work	ps-In <sub>2</sub> O <sub>3</sub> NS	Room temperature	0.485	4.77 <sup>a</sup>	0.485
[S1]	Zn-doped In <sub>2</sub> O <sub>3</sub>	300 °C	5	2.74 <sup>b</sup>	5
[S2]	Pd-loaded In <sub>2</sub> O <sub>3</sub>	110 °C	5	9 <sup>b</sup>	5
[S3]	Porous In <sub>2</sub> O <sub>3</sub>	250 °C	50	164 <sup>b</sup>	1
[S4]	In <sub>2</sub> O <sub>3</sub>	150 °C	100	33.45 <sup>b</sup>	5
[85]	In <sub>2</sub> O <sub>3</sub> -rGO	Room temperature	30	8.25 <sup>b</sup>	

Table S4 The gas sensing performance of  $In_2O_3$  sensors to  $NO_x$  gas.

<sup>a</sup>Response =  $(R_g - R_a)/R_a$ ; <sup>b</sup>Response=  $R_g/R_a$ , where  $R_g$  and  $R_a$  are the resistance values of the sensor measured in the target gas and air, respectively.

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