Supporting Information

Chemoenzymatic One-Pot Reaction of Noncompatible Catalysts: Combining Enzymatic Ester Hydrolysis with Cu(I) /Bipyridine Catalyzed Oxidation in Aqueous Medium

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1.¹H-, ¹³C-NMR- and ESI-MS data of synthesized monomers

4'-Methoxy-[2,2'-bipyridin]-4-ol:

To a solution of 4,4'-dimethoxy-2,2'-bipyridine (500 mg, 2.31mmol, 1.0 eq.) in 30 ml absolute acetic acid was given 48% HBraq (0.31 ml, 2.77 mmol, 1.15 eq.) while stirring. After heating for reflux the solution was neutralized with aqueous ammonia solution up to pH 8-9 under ice bath cooling and stirred for 30 min at rt. The solution was extracted with dichloromethane (4 x 30 ml) and the organic layers were dried with MgSO4. The solvent was removed under reduced pressure and the residue was washed with a small amount of chloroform. 4'-Methoxy-[2,2'-bipyridin]-4-ol was yielded as white crystals (328 mg, 1.62 mmol, 70%).



¹**H-NMR (400 MHz, CDCl₃):** δ = 3.91 ppm (s, CH₃O), 6.53 (dd, J = 7.2 Hz, CHCHCOCH₃), 6.88 (dd, *J* = 5.8 Hz, CHCHCOH), 7.11 (s, br, CCHCOH), 7.39 (s, CCHCOCH₃), 7.71 (d, J = 7 Hz, CHCHCOH), 8.42 (d, J = 5.8 Hz, CHCHCOCH₃): ¹³C-NMR (100 MHz, CDCl₃): δ = 55.5 ppm (CH₃O), 106.1 (2 x CCHCOR), 111.2 (2 x CHCHCOR), 150.2 (2 x CHCHCOR), 167.0 (2 x NCCH, 2 x COR); **HR-ESI-MS**: $M_{calculated} = 202.0742$ [M = C₁₁H₁₀N₂O₂]; $M_{measured} = 203.0816$ [M+H]⁺ Known





2-n-Heptyl-2-Oxazolin:

To octanonitrile (24.57 ml, 154.94 mmol, 1.0 eq.) ethanolamine (11.24 ml, 185.93 mmol, 1.20 eq.) and Cd(OAc)₂·2H₂O (825.91 mg, 3.10 mmol, 0.02 eq.) was added. The solution was stirred 30 h at 130 °C and the product was purified by distillation (88 °C, $1.6 \cdot 10^{-1}$ mbar). After addition of CaH₂ the product was distillated again and yielded as colorless liquid (15.5 g, 91.57 mmol, 60%).



¹H-NMR (400 MHz, CDCl₃): δ = 0.86 ppm (m, CH₃), 1.30 (m, 4xCH₂), 1.61 (quin, J = 7.4 Hz, CH₃CH₂), 2.25 (t, J = 7.7 Hz, CCH₂), 3.80 (t, J = 9.4 Hz, CH₂N), 4.20 (t, J = 9.5 Hz, CH₂O); ¹³C-NMR (100 MHz, CDCl₃): δ =

14.0 ppm (*C*H₃), 22.6 (*C*H₃*C*H₂), 25.9 (*C*H₂CH₂CO), 27.9 (*C*H₂CH₂CH₂CH₃), 28.9 (*C*H₂CN), 29.2 (*C*H₂CH₂CH₂CN), 31.6 (*C*H₂CH₂CH₃), 54.3 (*C*H₂N), 67.1 (*C*H₂O), 168.6 (*C*O); **HR-ESI-MS**: M_{calculated} = 169.1467 [M = C₁₀H₁₉NO]; M_{measured} = 170.1538 [M+H]⁺ Known compound.^[S2]



2-(5-Chlorpentyl)-2-oxazolin:

The synthesis was carried out according to *Litt et al.*^[S15] Starting from 3-caprolactone (10 g, 87.61 mmol, 1.0 eq.) 2-(5-chlorpentyl)-2-oxazolin was obtained after three steps with an overall yield of 66%.





2-(5-Azidopentyl)-2-oxazolin:

The synthesis was carried out according to *Lav et al.*^[S3] Starting from 6-bromohexanoic acid (25 g, 128.17 mmol, 1.0 eq) 2-(5-azidopentyl)-2-oxazolin was obtained after three steps with an overall yield of 43%.



¹**H-NMR (500 MHz, CDCl₃):** $\delta = 1.28 - 1.42$ ppm (m, CCH₂CH₂CH₂CH₂), 1.48 - 1.64 (m, CCH₂CH₂CH₂CH₂CH₂CH₂N₃), 2.20 (td, J = 7.46/1.22 Hz, CCH₂), 3.19 (td, J = 6.85/1.47 Hz, CH₂N₃), 3.66 - 3.82 (m, CH₂O), 4.14 (td, J = 9.42/1.71

Hz, CH₂N); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 25.3$ ppm (CH₂CH₂C(O)N), 26.1 (CH₂CH₂CH₂C(O)N), 27.6 (CH₂CH₂CH₂N₃), 28.4 (CH₂C(O)N), 51.1 (CH₂N), 54.2 (CH₂CH₂CH₂N₃), 67.0 (NCH₂CH₂O), 168.1 (C(O)N); HR-ESI-MS: M_{calculated} = 182.1168 [M = C₈H₁₄N₄O]; M_{measured} = 183.1246 [M+H]⁺ Known compound.^[S3]



1-((Prop-2-yn-1-yloxy)methyl)-4-vinylbenzene:

1H-NMR (500 MHz, CDCl₃): $\delta = 2.49$ ppm (t, J = 2.2 Hz, CHC), 4.19 (d, J = 2.4, CHCCH₂), 4.63 (s, OCH₂Ar), 5.27 (d, J = 10.8 Hz, CH₂CH), 5.77 (d, J = 17.6 Hz, CH₂CH), 6.74 (dd, J = 17.6 Hz/11.2 Hz, CH₂CH), 7.31-7.45 (m, Ar); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 56.9$ (CH₂CCH), 71.1 (OCH₂Ar), 74.6 (CH₂CCH), 79.5 (CH₂CCH), 113.9 (CH₂CH), 126.2 (2xCH₂CCH), 128.3 (2xCH₂CCH), 136.4

(CH₂*C*H), 136.7 (CH₂*C*(Ar)), 137.2 (CH*C*(Ar)). Known compound.^[S4]





2-(5-(4-(((4-Vinylbenzyl)-oxy)-methyl)-1H-1,2,3-triazol-1-yl)pentyl)-2-oxazolin:



¹H-NMR (500 MHz, CDCl₃): $\delta = 1.32 - 1.42$ (m, CCH₂CH₂CH₂), 1.62 - 1.71 (m, CCH₂CH₂CH₂CH₂), 1.86 -1.96 (m, CCH₂CH₂), 2.26 (t, J = 7.48 Hz, CCH₂CH₂), 3.79 (t, J=9.46 Hz, OCH₂CH₂N), 4.18 (td, J = 9.42/1.71 Hz, OCH₂CH₂N), 4.28 - 4.38 (m, CH₂NN), 4.58 (s, CHCCH₂),

4.66 (s, OCH₂Ar), 5.23 (d, J = 10.8 Hz, CH₂CH), 5.74 (dd, J = 17.70, 0.61 Hz, CH₂CH), 6.70 (dd, J = 17.6 Hz/11.2 Hz, CH₂CH), 7.28 - 7.42 (m, Ar), 7.53 (s, CHN); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 25.1$ ppm (CH₂CH₂C(O)N), 25.9 (CH₂CH₂CH₂CH₂C(O)N), 27.5 (CH₂CH₂CH₂CH₂N), 29.9 (CH₂C(O)N), 50.0 (CH₂N), 54.2 (CH₂CH₂CH₂N), 63.6 (OCH₂CN), 67.1 (NCH₂CH₂O), 72.2 (OCH₂Ar), 113.8 (CH₂CH), 122.3 (CHN), 126.2 (2xCHCCH), 128.1 (2xCH₂CCH), 136.4 (CH₂CH), 137.0 (CH₂C(Ar)), 137.3 (CHC(Ar)), 145.0 (OCH₂CN), 168.0 (C(O)N); HR-ESI-MS: M_{calculated} = 354.2056 [M = C₂₀H₂6N₄O₂]; M_{measured} = 355.2136 [M+H]⁺



2. Fig. S1. DLS size graph of **NP1** and **NP2** in methanol and water. 1mM solution at room temperature.



3. Fig S2. ¹H-NMR data of **P1** in CDCl₃.



¹**H-NMR (500 MHz, CDCl₃):** $\delta = 0.85$ ppm (s(br), $CH_{3(\text{HepOx})}$), 1.26 (s(br), $4xCH_{2(\text{HepOx})}$), 1.43 – 1.89 (m, $CH_{2(\text{HepOx})}$, $3xCH_{2(\text{BiPyOx})}$, $2xCH_{2(\text{StyOx})}$, $3xCH_{2(\text{Pip})}$), 1.89 – 2.06 (s(br), $CH_{2(\text{StyOx})}$), 2.06 – 2.20 (m, $CH_{3(MeOx)}$, $2xCH_{2(Pip)}$), 2.20 – 2.50 (m, $CH_{2(\text{HepOx})}$, $CH_{2(\text{BiPyOx})}$, $CH_{2(\text{StyOx})}$), 2.94/3.01 ($CH_{3(\text{In})}$), 3.19 – 3.75 (m, $CH_{2-CH_{2(\text{backbone})}$), 3.93 (s(br), $OCH_{3(\text{BiPyOx})}$), 4.10 (s(br), $OCH_{2(\text{BiPyOx})}$), 4.34 (s(br), $CH_{2(\text{StyOx})}$), 4.59 – 4.71 (m, $2xCH_{2(\text{StyOx})}$), 5.20 – 5.26 (m, $CH_{(\text{StyOx})}$), 5.68 – 5.79 (m, $CH_{(\text{StyOx})}$), 6.65 – 6.74 (m, $CH_{(\text{StyOx})}$), 6.82 (s(br), $2xCH_{(\text{BiPyOx})}$), 7.28 – 7.43 (m, $4xCH_{\text{Ar(StyOx)}}$), 7.49 – 7.65 (s(br), $CH_{\text{Triazol(StyOx)}}$), 7.94 (s(br), $2xCH_{(\text{BiPyOx})}$), 8.44 (s(br), $2xCH_{(\text{BiPyOx})}$).

4. Fig S3. ¹H-NMR data of P2 in CDCl₃.



¹**H-NMR (500 MHz, CDCl₃):** $\delta = 0.85$ ppm (s(br), $CH_{3(\text{HepOx})}$), 1.25 (s(br), $4xCH_{2(\text{HepOx})}$), 1.45 – 1.89 (m, $CH_{2(\text{HepOx})}$, $3xCH_{2(\text{BiPyOx})}$, $2xCH_{2(\text{StyOx})}$, $3xCH_{2(\text{Pip})}$), 1.89 – 2.01 (s(br), $CH_{2(\text{StyOx})}$), 2.01 – 2.20 (m, $CH_{3(\text{MeOx})}$, $2xCH_{2(\text{Pip})}$), 2.20 – 2.54 (m, $CH_{2(\text{HepOx})}$, $CH_{2(\text{BiPyOx})}$, $CH_{2(\text{StyOx})}$), 2.94/3.01 ($CH_{3(\text{In})}$), 3.18 – 3.70 (m, $CH_{2-}CH_{2(\text{backbone})}$), 3.92 (s(br), $OCH_{3(\text{BiPyOx})}$), 4.10 (s(br), $OCH_{2(\text{BiPyOx})}$), 4.34 (s(br), $CH_{2(\text{StyOx})}$), 4.52 – 4.72 (m, $2xCH_{2(\text{StyOx})}$), 5.19 – 5.28 (m, $CH_{(\text{StyOx})}$), 5.69 – 5.79 (m, $CH_{(\text{StyOx})}$), 6.64 – 6.74 (m, $CH_{(\text{StyOx})}$), 6.82 (s(br), $2xCH_{(\text{BiPyOx})}$), 7.28 – 7.42 (m, $4xCH_{\text{Ar}(\text{StyOx})}$), 7.49 – 7.65 (s(br), $CH_{\text{Triazol(StyOx)}}$), 7.93 (d(br), $2xCH_{(\text{BiPyOx})}$), 8.44 (s(br), $2xCH_{(\text{BiPyOx})}$).

5 mol% CuBr 5 mol% NP1 -ligand					
	R ²	5 mol% <i>N</i> -Oxyl		kyl R ²	
		10 mol% NMI			
	R' OH	H_2C	D, rt, ai	r R' O	
entry	substrate	N-oxyl	<i>t</i> [h]	conversion ^[b] [%]	yield ^[c] [%]
1a	ОН	ABNO	2	95 (± 3)	92
1b		TEMPO	3	99 (± 0)	96
2a	ОН	ABNO	2	96 (± 3)	91
2b		TEMPO	3	94 (± 4)	90
3a	ОН	ABNO	2	90 (± 3)	85
3b	Ls	TEMPO	3	62 (± 2)	56
4a	ОН	ABNO	2	79 (± 3)	72
4b	Lo	TEMPO	3	58 (± 0)	54
5a	ОН	ABNO	2	91 (± 3)	89
5b		TEMPO	3	87 (± 4)	84
6a		ABNO	2	79 (± 3)	71
6b		TEMPO	3	73 (± 2)	64
7a		ABNO	2	44 (± 2)	37
7b		TEMPO	3	3 (± 1)	n. d.
8a	M- OH	ABNO	2	67 (± 2)	59
8b		TEMPO	3	$9(\pm 1)$	5

5. Table S1. Representative examples of functionalized alcohols in the micellar Cu(I) / N-oxyl catalysed aerobic oxidation with **P1** as polymeric ligand.^{*a*}

[a] 40 mol of CuBr, 5 mM solution of **P1**; [b] average conversion (3 runs) after work-up and isolation determined by ¹H-NMR-spectroscopy; [c] average isolated yield (3 runs) after purification.

5.1 ¹H- and ¹³C-NMR data of the synthesized aldehydes.



¹**H-NMR (500 MHz, CDCl₃):** $\delta = 7.52 - 7.57$ ppm (m, 2xCHCH), 7.63 - 7.67 (m, CHCHCH), 7.88 – 7.90 (m, 2xCCH), 10.03 (s, OCH); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 128.9 \text{ ppm} (2\text{xCCHCH}), 129.7 (2\text{xCCHCH}), 134.4 (CCHCHCH), 136.3 (CCH),$

192.4 (HCO). Known compound.^[S5]



¹**H-NMR (500 MHz, CDCl₃):** δ = 3.88 ppm (s, 3H, OCH₃), 7.00 (d, J = 8.0 Hz, 2xCHCCH), 7.83 (d, J = 8.0 Hz, 2xOCCH), 9.87 (s, OCH); ¹³C-NMR (125 MHz, **CDCl₃**): δ = 55.5 ppm (OCH3), 114.2 (2xOCCH), 129.8 (2xCHCCHCH), 131.9 (CHCCHCH), 164.5 (OCCH), 190.8 (HCO). Known compound.^[S5]



¹**H-NMR (500 MHz, CDCl₃):** δ = 7.23 ppm (dd, J = 4.8, 3.9 Hz, CCH), 7.76 - 7.81 (m, SCHCH), 9.94 (s, OCH); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 128.3$ ppm (SCHCH), 135.1 (CCHCH), 136.3 (SCHCH), 143.9 (CS), 183.0 (HCO). Known

compound.[85]



¹H-NMR (500 MHz, CDCl₃): $\delta = 6.63 - 6.64$ ppm (m, CCH), 7.28 - 7.30 (m, CCHCH), 7.72 – 7.75 (m, OCHCH), 9.68 (s, CCHO); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 127.9$ (OCHCH), 134.8 (CCHCH), 136.1 (OCHCH), 142.7 (COCH), 182.6

(HCO). Known compound.^[S6]



¹**H-NMR** (400 MHz, CDCl₃): $\delta = 6.73$ ppm (dd, J = 15.9, 7.5 Hz, OCHCH), 7.38 - 7.53 (m, 2xCCHCH), 7.58 (m, CHCHCHCCH), 9.72 (d, J = 7.65 Hz, OCH); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 128.4$ ppm (CCHCHCH), 128.4

(2xCCHCHCH), 129.0 (CCHCHC(H)O), 131.2 (CCHCHC(H)O), 152.8 (CCHCHC(H)O), 193.8 (HCO). Known compound.^[S6]



¹H-NMR (400 MHz, CDCl₃): $\delta = 2.30$ (m, CH₃), 3.69 (s, CCCH), 6.29 (dd, J =8.2, 1.6 Hz, OCHCH), 10.02 (d, J = 8 Hz, OCH); ¹³C-NMR (100 MHz, CDCl₃): $\delta = 24.6 \text{ ppm}$ (CH₃), 80.1 (CHCC), 88.2 (CHCC), 136.8 (CHC(H)O), 140.9

(CHCC), 192.3 (HCO). Known compound.^[S7]



¹**H-NMR (500 MHz, CDCl₃):** $\delta = 0.84$ ppm (t, J = 6.4 Hz, CH₂CH₃), 1.20 – 1.36 (m, CH₃CH₂CH₂CH₂), 1.40 - 1.52 (m, CH₃CH₂), 2.28 (q, J = 7 Hz, CHCH₂CH₂), 6.50 (dd, J = 15.6 Hz, 7.9 Hz, CH₂CH₂CH), 6.82 (dt, J = 15.2, 6.6 Hz, OCHCH),

9.44 (d, J = 8.1 Hz, OCH); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 14.0$ ppm (CH₃), 22.5 (CH₂CH₃), 28.3 (CH₂CH₂CH₂CH₃), 31.3 (CH₂CH₂CH₃), 32.6 (CHCH₂), 132.9 (CHCHC(H)O), 158.9 (CHCHC(H)O), 193.9 (HCO). Known compound.^[S8]



¹**H-NMR (500 MHz, CDCl₃):** δ = 1.84 – 1.95 ppm (m, 2xCH₂CH₂C), 2.04 – 2.16 (m, 2xCH₂C); ¹³**C-NMR (125 MHz, CDCl₃):** δ = 23.1 ppm (2xCH₂CH₂C), 38.2 (2xCH₂CH₂C), 220.5 (*C*). Known compound.^[S9]



















6.¹H- and ¹³C-NMR data of the synthesized acetyl esters.



(0.99 g, 5.52 mmol, 95%); ¹H-NMR (**500** MHz, CDCl₃): δ = 2.09 ppm (s, CCH₃), 3.82 (s, OCH₃), 5.05 (s, OCH₂), 6.90 (d, *J* = 8 Hz, 2xCH₂CCH), 7.31 (d, *J* = 8 Hz, 2xOCCH); ¹³C-NMR (**125** MHz, CDCl₃): δ = 21.1 ppm (CCH₃), 55.3 (OCH₃), 66.1 (OCH₂), 113.9 (2xCH₂CCH), 128.0 (2xOCCH),

130.1 (CH₂CCH), 159.6 (OCCH), 171.0 (CCH₃). Known compound.^[S10]



(0.86 g, 5.51 mmol, 95%); ¹H-NMR (**500** MHz, CDCl₃): $\delta = 2.10$ ppm (s, CH₃), 5.27 (s, CH₂), 7.0 (dd, J = 4.9, J = 3.7 Hz, CCH), 7.10 (d, J = 3.4 Hz, CCHCH), 7.33 (d, J = 4.9, 0.9 Hz, SCH); ¹³C-NMR (**125** MHz, CDCl₃): $\delta = 21.0$ ppm (CCH₃), 60.4 (OCH₂), 126.8 (CCHCHCH), 128.2 (SCH), 137.9 (SCCH₂), 170.7

(CCH₃). Known compound.^[S11]



(0.65 g, 4.47 mmol, 77%); ¹**H-NMR (500 MHz, CDCl₃):** δ = 2.08 ppm (s, *CH*₃), 5.06 (s, *CH*₂), 6.36 (dd, *J* = 2.9 Hz, 2 Hz, *CCH*), 6.41 (d, *J* = 2.9 Hz, *CCHCH*), 7.42 (d, *J* = 1.5 Hz, *OCH*); ¹³**C-NMR (125 MHz, CDCl₃):** δ = 20.8 (*CC*H₃), 58.0 (*OCH*₂), 110.5 (*CCHCHCH*), 143.2 (*OCH*), 149.4 (*OCCH*₂), 170.6 (*CC*H₃).

Known compound.^[S10]



(1.01 g, 5.74 mmol, 99%); ¹H-NMR (500 MHz, CDCl₃): δ = 2.12 ppm (s, CCH₃), 4.74 (dd, J = 6.4, 0.9 Hz, OCH₂), 6.30 (dt, J = 15.9, 6.4, 6.4 Hz, OCH₂CH), 6.66 (d, J = 15.9 Hz, CH₂CHCH), 7.22 – 7.45 (m, 5xCHAromat); ¹³C-NMR (125 MHz, CDCl₃): δ = 20.9 ppm (CCH₃), 65.0 (OCH₂), 123.0

(OCH₂*C*H), 126.5 (CCHCH*C*H), 128.0 (2xCCH*C*H), 128.5 (2xC*C*HCH), 134.1 (C*C*H), 136.1 (*C*CH), 170.8 (*C*CH₃). Known compound.^[S11]



(0.74 g, 5.34 mmol, 92%); ¹H-NMR (500 MHz, CDCl₃): δ = 1.88 ppm (s, CHCC*H*₃), 2.06 (s, OCC*H*₃), 2.87 (s, CCC*H*), 4.64 (d, *J* = 6.8 Hz, OC*H*₂), 6.00 (t, *J* = 4 Hz, OCH₂C*H*); ¹³C-NMR (125 MHz, CDCl₃): δ = 17.5 (CHCCH₃), 20.8 (CCH₃), 60.3 (CH₂), 76.0 (CCCH), 85.3 (CCCH), 131.8 (CHCCCH),

170.7 (CCH₃). Known compound.^[S12]



(0.55 g, 3.25 mmol, 56%); ¹H-NMR (500 MHz, CDCl₃): $\delta = 0.89$ ppm (t, J = 7.1 Hz, CH₂CH₃), 1.23 – 1.36 (m, CH₃CH₂CH₂CH₂), 1.39 (dt, J = 6 Hz, CH₃CH₂), 2.03 (d, J = 4 Hz, CHCH₂CH₂), 2.06 (s, CH₃), 4.51 (d, J = 6.8 Hz,

OCH₂), 5.56 (dt, J = 12.1, 3.7 Hz, CH₂CH₂CH), 5.77 (dt, J = 12.1, 3.7 Hz, OCH₂CH); ¹³C-NMR (125 MHz, CDCl₃): $\delta = 14.0$ (CH₂CH₃), 21.0 (CCH₃), 22.5 (CH₂CH₃), 28.5

(*C*H₂CH₂CH₂CH₃), 31.3 (*C*H₂CH₂CH₃), 32.2 (*C*H₂CH₂CH₂CH₂CH₂CH₃), 65.3 (O*C*H₂), 123.6 (OCH₂CH*C*H), 136.8 (OCH₂CH), 170.9 (*C*CH₃). Known compound.^[S13]

 $(0.54 \text{ g}, 2.90 \text{ mmol}, 50\%); {}^{1}\text{H-NMR} (500 \text{ MHz, CDCl}_3): \delta = 0.89 \text{ ppm} (t, J = 6.8 \text{ Hz}, CH_3), 1.18 - 1.42(\text{m}, CH_3CH_2CH_2CH_2CH_2CH_2), 1.62 (qin, J = 6.8 \text{ Hz}, OCH_2CH_2), 2.05 (s, CCH_3), 4.06 (t, J = 6.8 \text{ Hz}, OCH_2); {}^{13}\text{C-NMR} (125 \text{ MHz}, CDCl_3): \delta = 14.0 (CH_2CH_3), 20.9 (CCH_3), 22.6 (CH_2CH_3), 25.9 (OCH_2CH_2CH_2), 28.6$

CDCl₃): $\delta = 14.0$ (CH₂CH₃), 20.9 (CCH₃), 22.6 (CH₂CH₃), 25.9 (OCH₂CH₂CH₂CH₂), 28.6 (OCH₂CH₂), 29.2 (OCH₂CH₂CH₂CH₂CH₂), 29.4 (CH₂CH₂CH₂CH₃), 31.8 (CH₂CH₂CH₂CH₃), 64.6 (OCH₂), 171.2 (CCH₃). Known compound.^[S14]















References

- [S1] H. Sand, R. Weberskirch, *RSC Adv.* **2015**, *5*, 38235-38242.
- [S2] S. Takahashi, H. Togo, *Synthesis* **2009**, *14*, 2329-2332.
- [S3] T.-X. Lav, P. Lemechko, E. Renard, C. Amiel, V. Langlois, G. Volet, *Reactive and Functional Polymers* **2013**, *73*, 1001-1008.
- [S4] E. G. Doyagueez, G. Corrales, A. Fernandez-Mayoralas, J, Rodriguez-Hernandez, A. Gallardo, *Macromolecules* 2012, 45, 7676-7683.
- [S5] R. Prebil, G. Stavber, S. Stavber, Eur. J. Org. Chem. 2014, 2, 395-402.
- [S6] C. Cheng, M. Brookhart, Angew. Chem., Int. Ed. 2012, 51, 9422-9424.
- [S7] B. Muller, J.-P. Ferezou, A. Pancrazi, J.-Y. Lallemand, *Bulletin de la Societe Chimique de France* **1997**, *134*, 13-26.
- [S8] R. A. Fernandes, V. Bethi, RSC Adv. 2014, 4, 40561-40568.
- [S9] G. Zhang, X. Wen, Y. Wang, W. Mo, C. Ding, J. Org. Chem. 2011, 76, 4665-4668.
- [S10] I. Chiarotto, Synthetic Communications 2016, 46, 1840-1847.
- [S11] J. M. Álvarez-Calero, Z. D. Jorge, G. M. Massanet, Org. Lett. 2016, 18, 6344-6347.
- [S12] H. Bader, H. Hopf, H. Jäger, Chemische Berichte 1989, 122, 1193-1198.
- [S13] T. Moragas, J. Cornella, R. Martin, J. Am. Chem. Soc. 2015, 136, 17702-17705.
- [S14] O. Vechorkin, X. Hu, Xile, Angew. Chem., Int. Ed. 2009, 48, 2937-2940.
- [S15] J. Kim, M. H. Litt, J. Polym. Sci. 1989, 27, 2711-2722.