

Supplementary Information

Coumarin centered Copper(II) complex with Appended-Imidazole as Cancer Chemotherapeutic Agents against Lung Cancer: Molecular insight via DFT-based Vibrational Analysis

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Table S1. Selected bond lengths (Å) of complex 1.

| Bond lengths | (Å) |
|--------------|-----------|
| N1 Cu1 | 1.990(2) |
| N2 Cu1 | 1.943(2) |
| N3 Cu1 | 1.949(2) |
| O1 Cu1 | 1.918(19) |
| N5 O6 | 1.243(3) |
| N5 O4 | 1.250(3) |
| N5 O5 | 1.261(3) |

Table S2. Selected bond angles of complex 1

| Bond Angle | [deg] |
|------------|------------|
| O6 N5 O4 | 121.0(2) |
| O6 N5 O5 | 119.1(2) |
| O4 N5 O5 | 119.9(2) |
| C19 O1 Cu1 | 127.7(2) |
| O1 Cu1 N2 | 92.83(9) |
| O1 Cu1 N3 | 88.58(9) |
| N2 Cu1 N3 | 171.91(10) |
| O1 Cu1 N1 | 159.69(9) |
| N2 Cu1 N1 | 83.67(10) |
| N3 Cu1 N1 | 97.69(10) |

Table S3. Hydrogen bond metrics for 1

| D – H…A | H…A | D…A | ∠ D – H…A | Symmetry operation |
|----------------------|------|----------|-----------|--------------------|
| N(4) --H(4) ..O(5) | 2.02 | 2.86(18) | 167 | 1/2-x,1/2+y,1/2-z |
| C(5) --H(5) ..O(5) | 2.52 | 3.40(2) | 158 | 3/2-x,-1/2+y,1/2-z |
| C(7) --H(7) ..O(3) | 2.47 | 3.38(2) | 164 | 1-x,-y,-z |
| C(10) --H(10) ..O(3) | 2.45 | 2.78(17) | 101 | |
| C(10) --H(10) ..O(3) | 2.49 | 3.41(2) | 173 | 1-x,-y,-z |
| C(15) --H(15) ..O(4) | 2.40 | 3.29(2) | 161 | -x,1-y,-z |
| C(23) --H(23) ..O(5) | 2.31 | 3.16(2) | 153 | |
| C(23) --H(23) ..O(6) | 2.55 | 3.11(19) | 120 | 1/2-x,1/2+y,1/2-z |

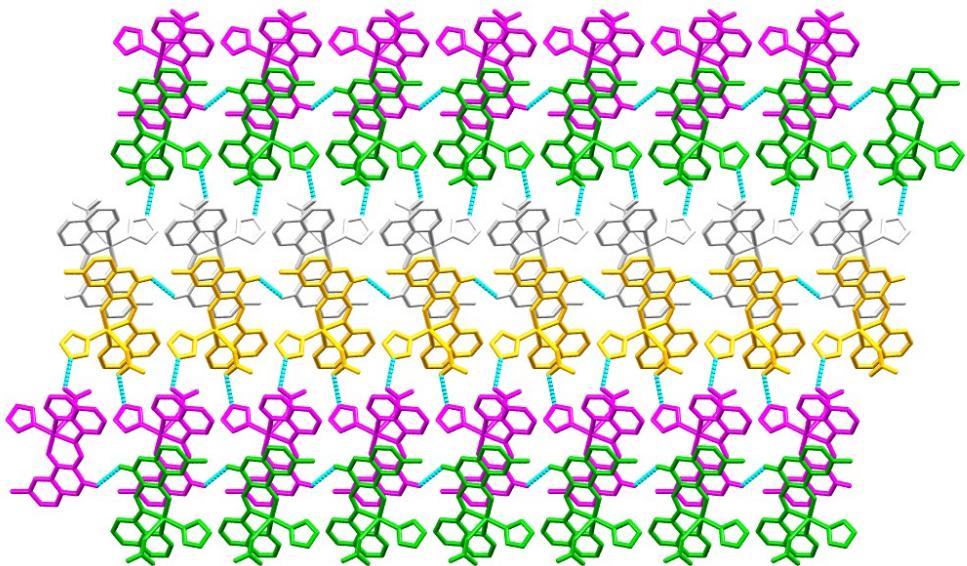


Fig. S1. Representation of 2D view along crystallographic a-axis of complex 1.

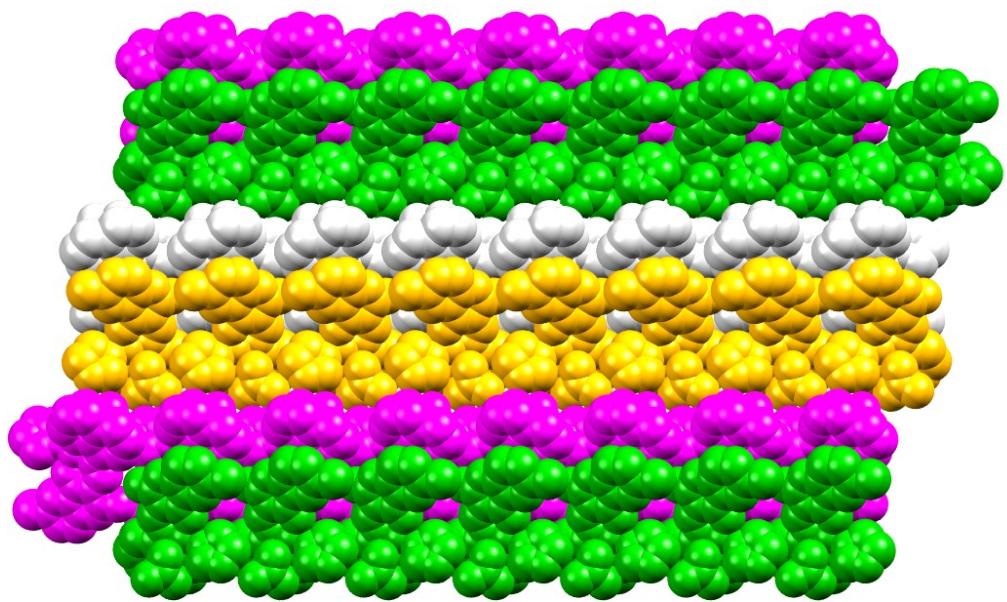


Fig. S2. 2D view in spacefill model of complex 1.

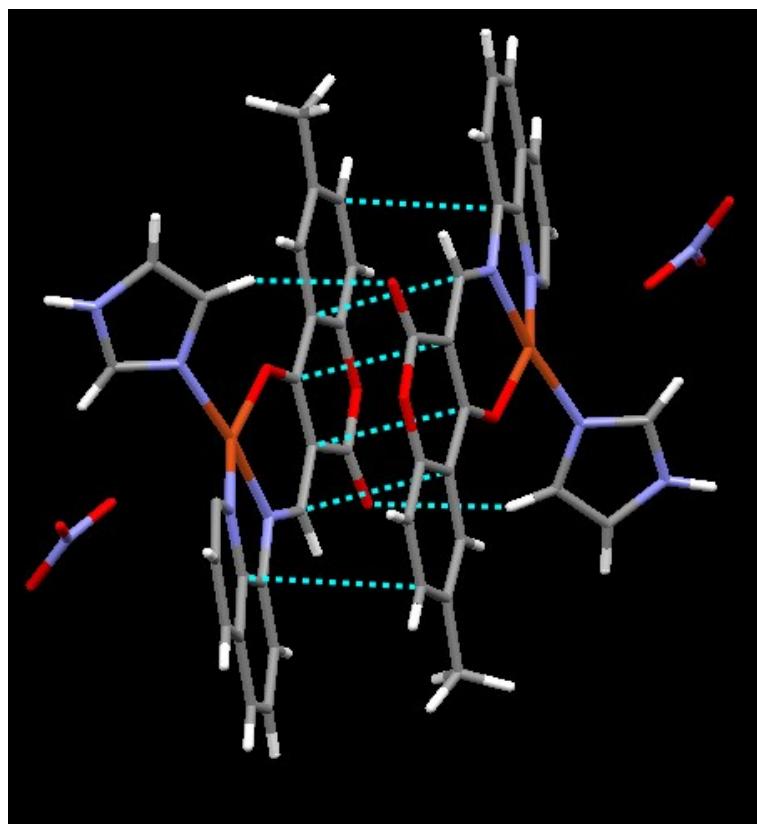


Fig. S3. Non-covalent ($\text{CH}\cdots\pi$, $\pi\cdots\pi$ and Hydrogen bonding) interactions of complex **1**.

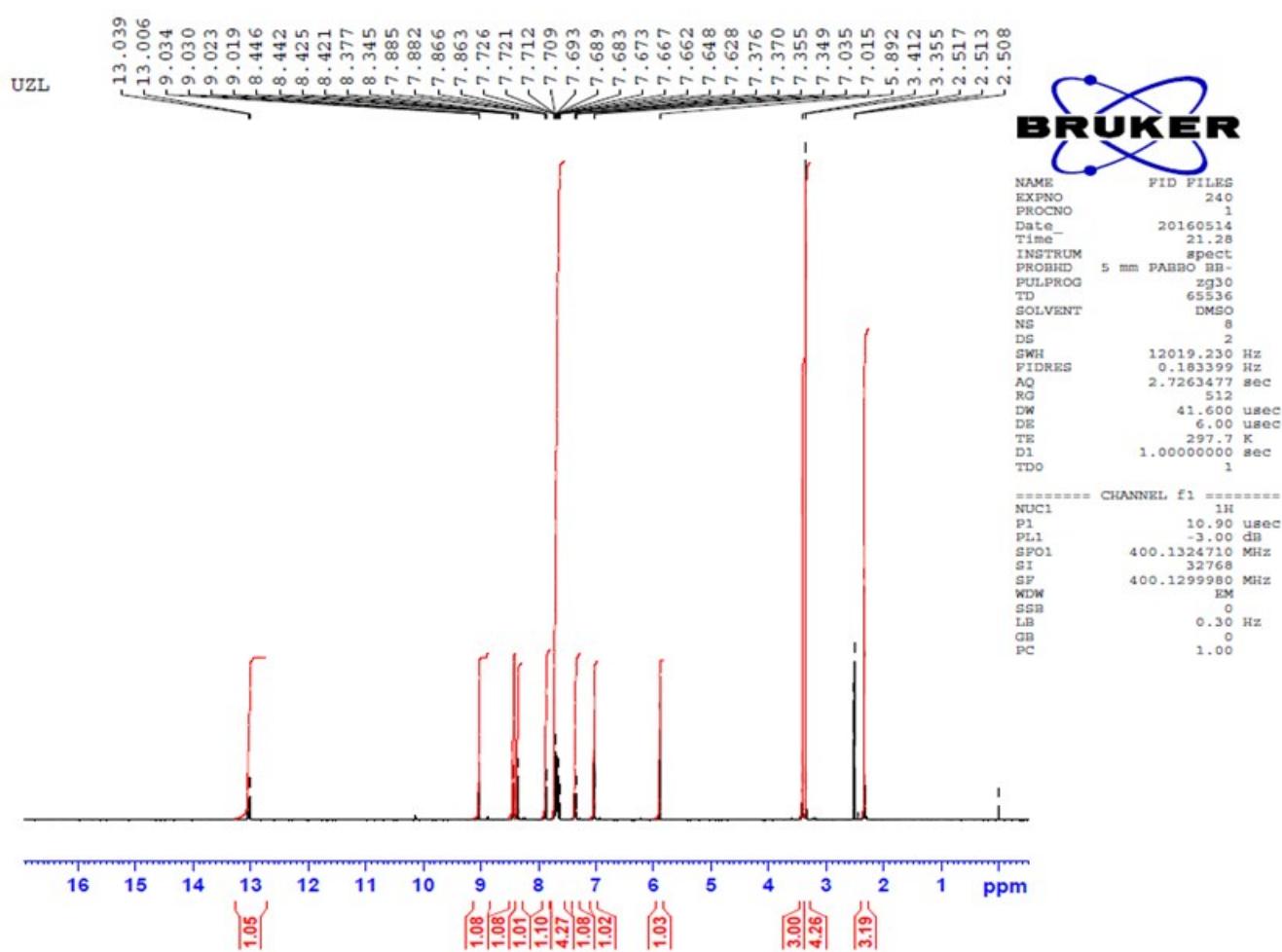


Fig. S4. ^1H NMR spectrum of ligand (HL).

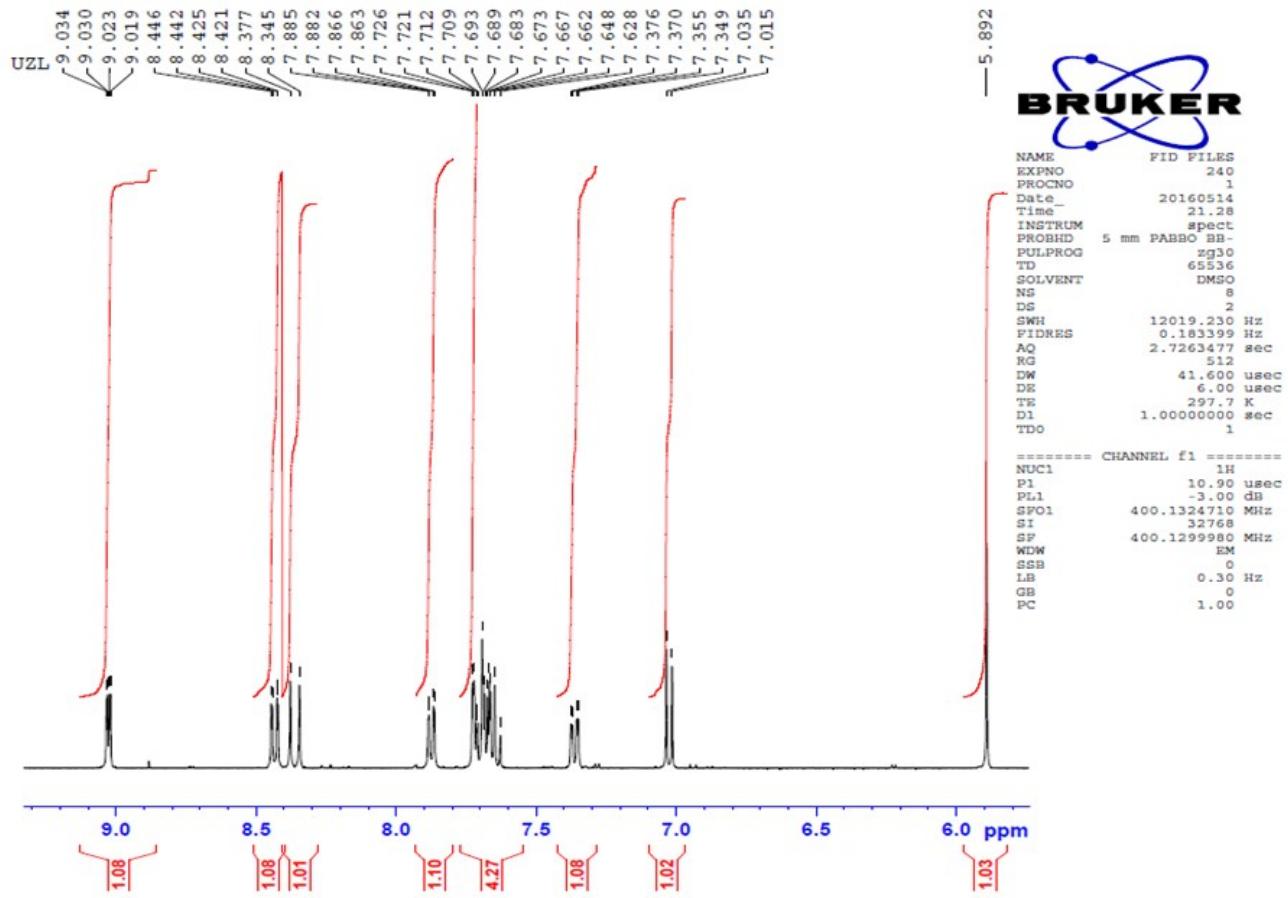


Fig. S5. ¹H NMR spectrum of ligand (HL).

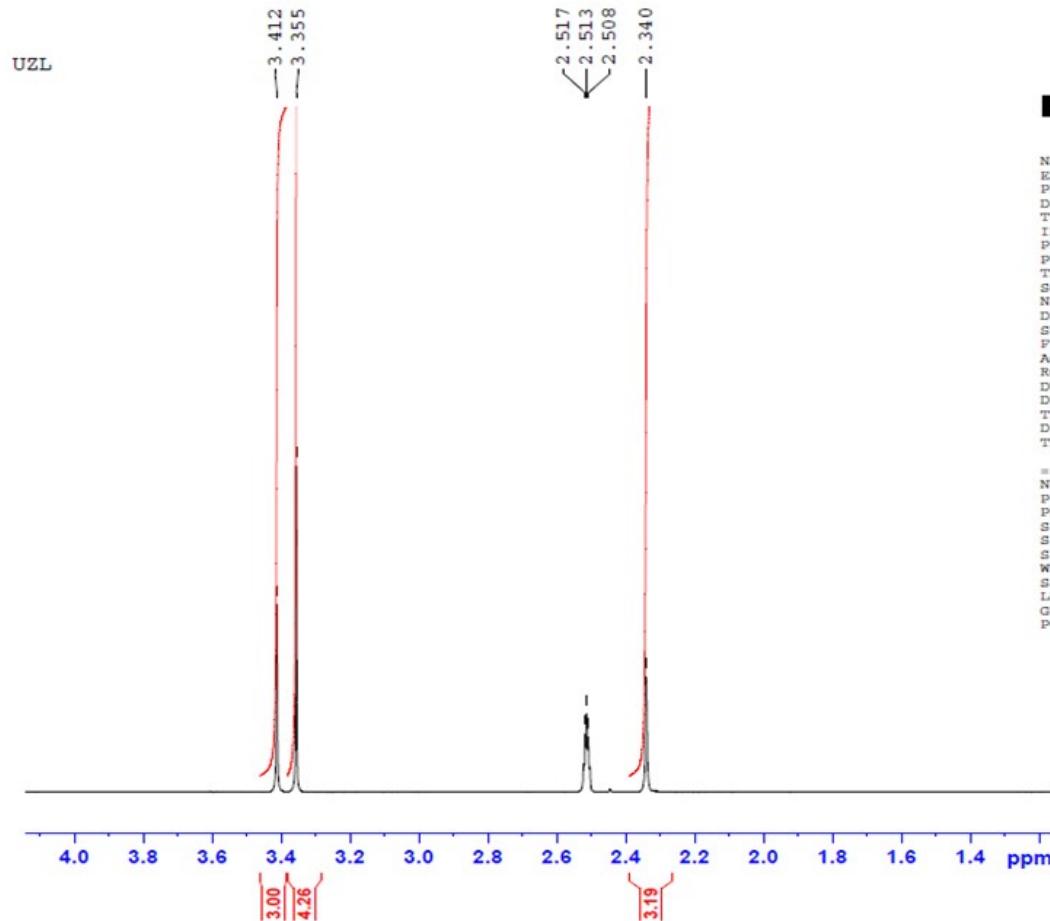


Fig. S6. ^1H NMR spectrum of ligand (HL).

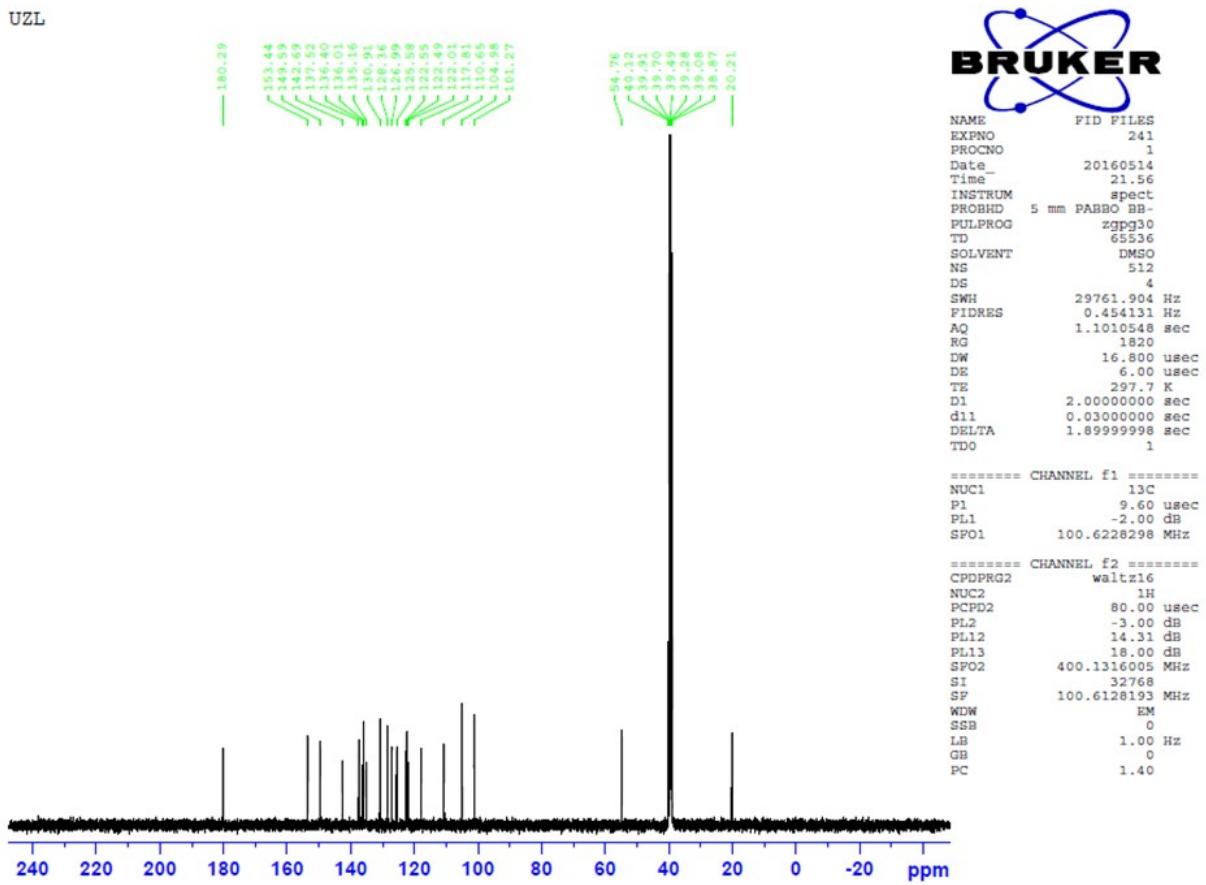


Fig. S7. ^{13}C NMR spectrum of ligand (HL).

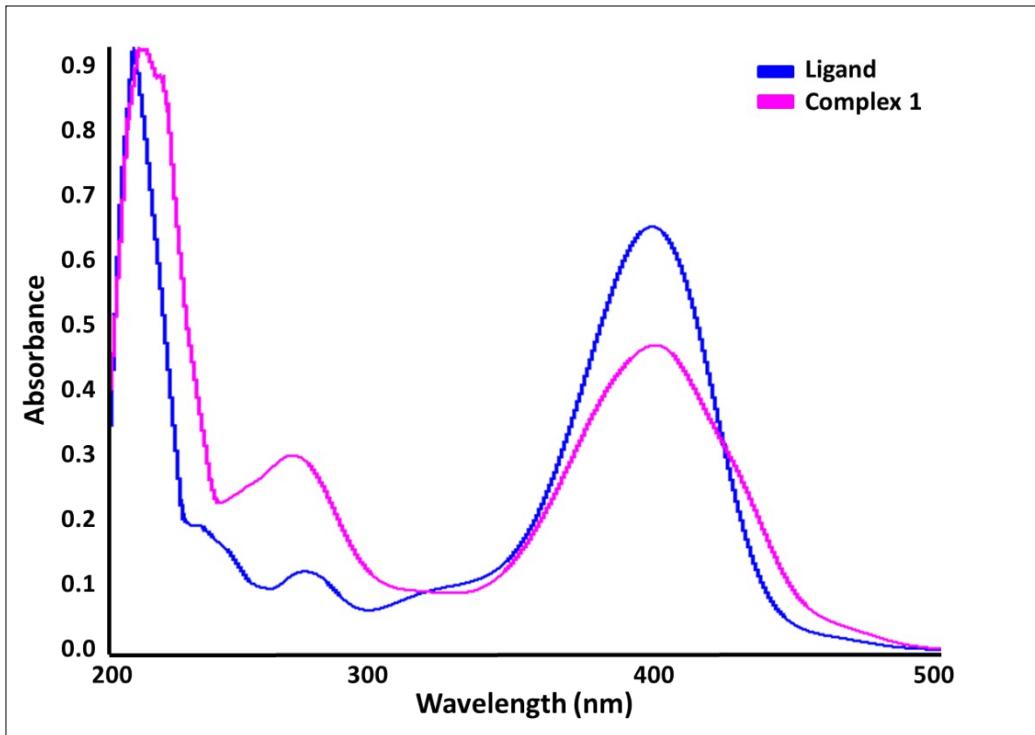


Fig. S8. UV-vis spectra in MeOH (1×10^{-4}).

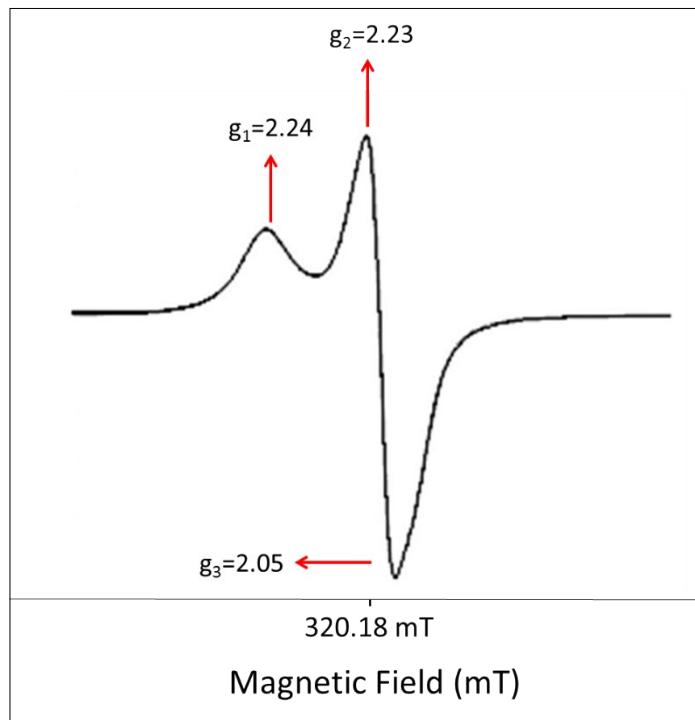


Fig. S9. X-band Polycrystalline EPR spectrum of complex 1 at 77K.

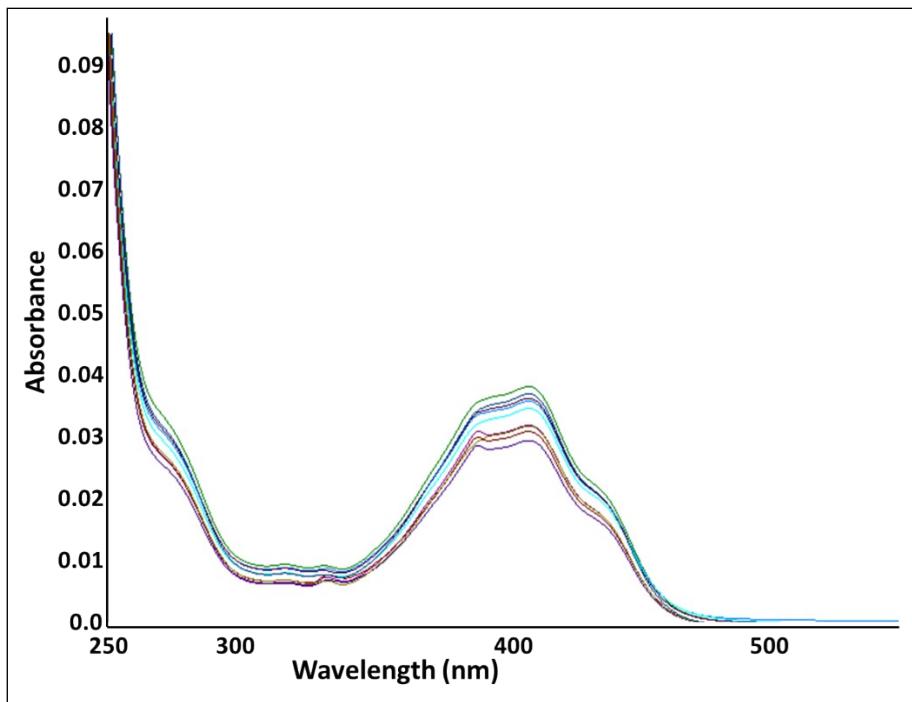
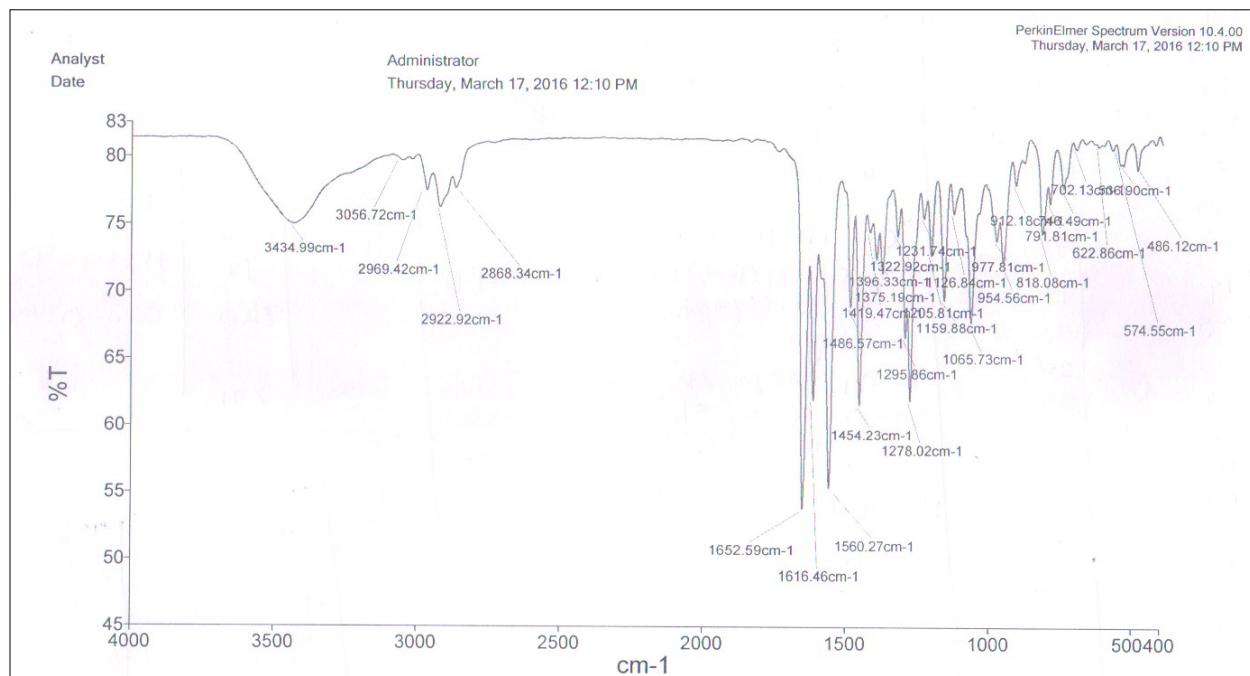
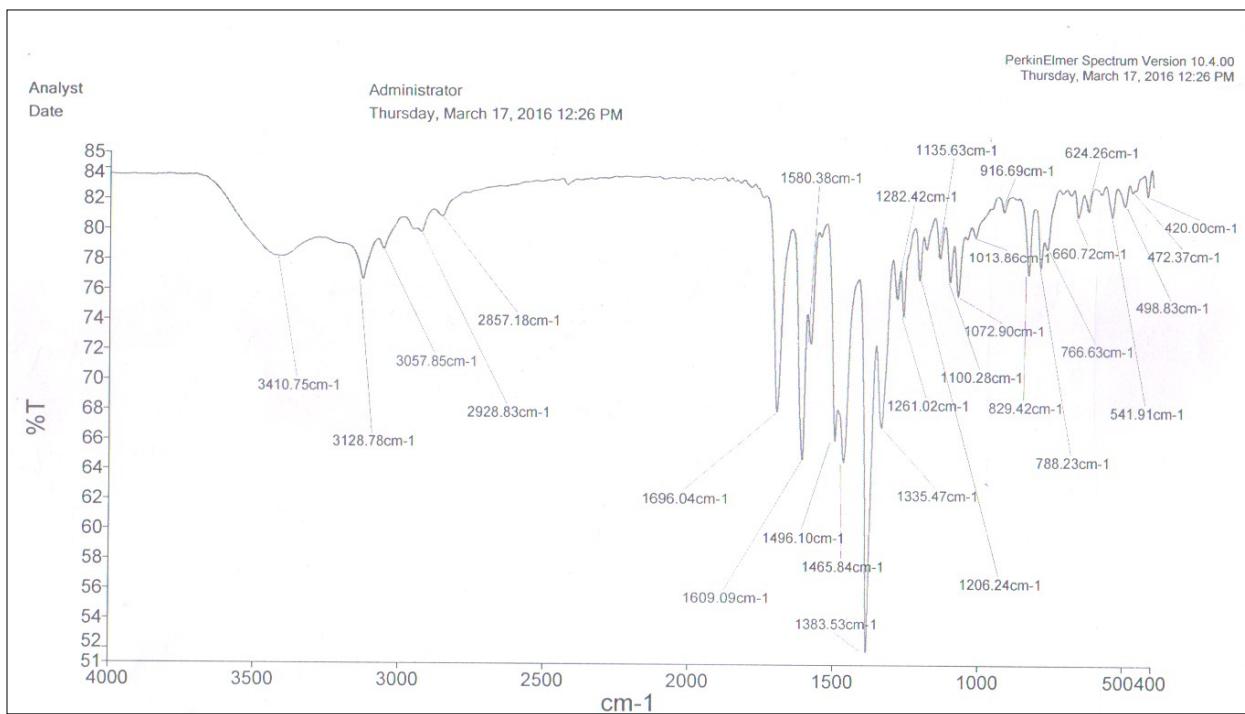


Fig. S10. Time-dependent stability studies of complex **1** in Tris-HCl buffer under physiological conditions (pH = 7.3 & T = 310 K) monitored by UV-vis absorption spectra.

(a)



(b)



(c)

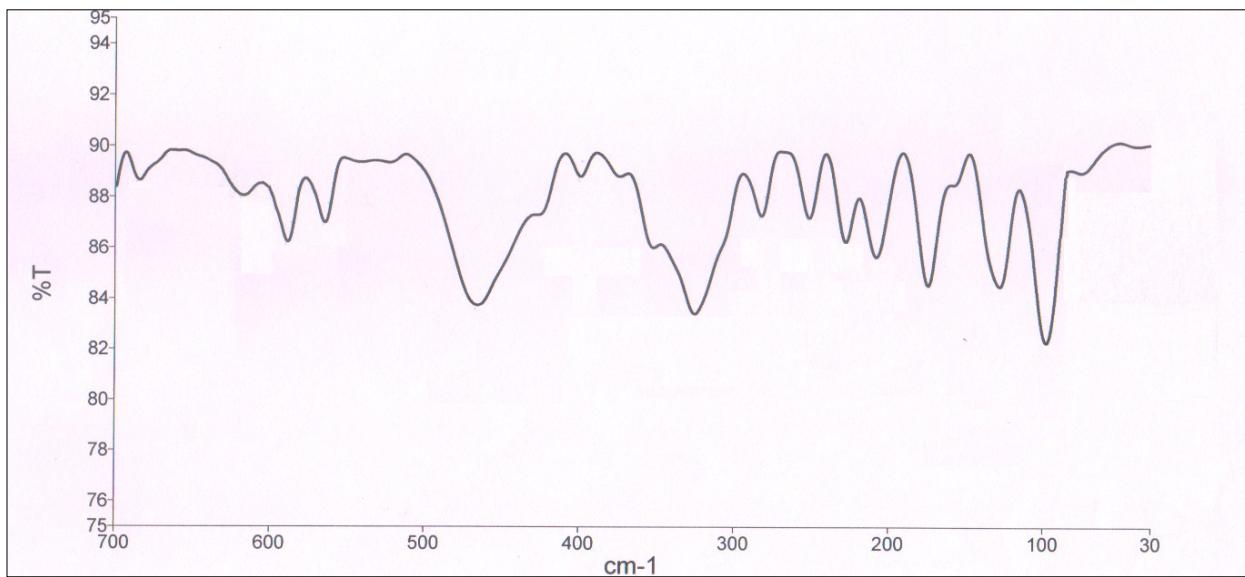


Fig. S11. (a) Experimental FTIR spectra of Ligand (b) complex **1** (c) Experimental Far-IR spectra of complex **1**.

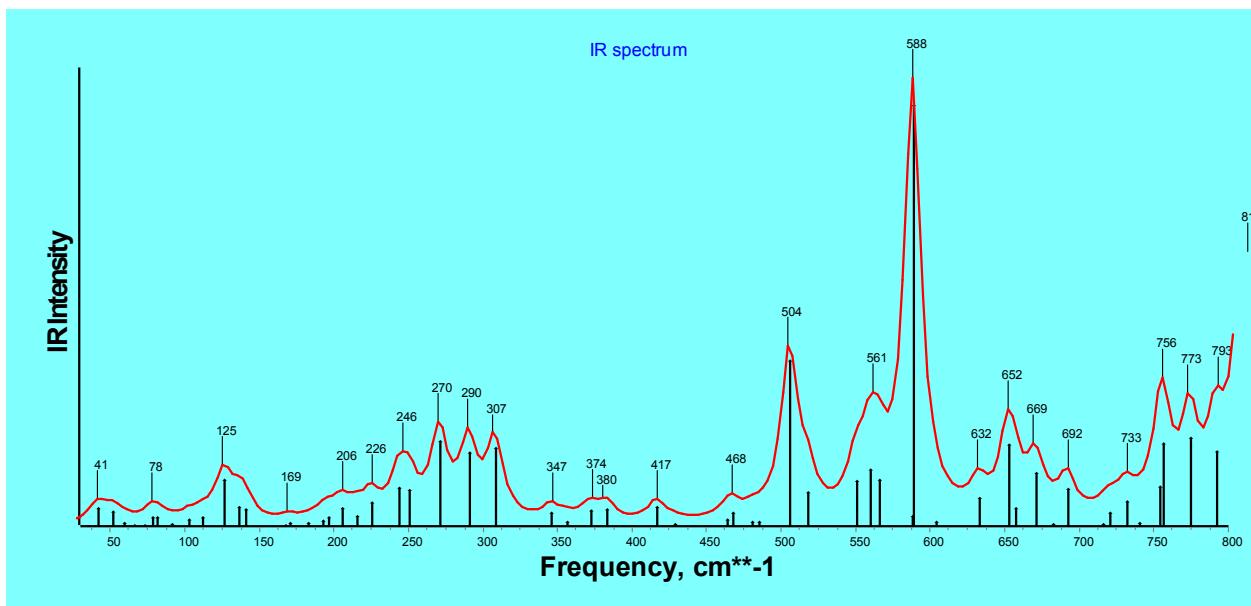


Fig. S12. B3LYP/DFT simulated IR spectrum of complex **1**.

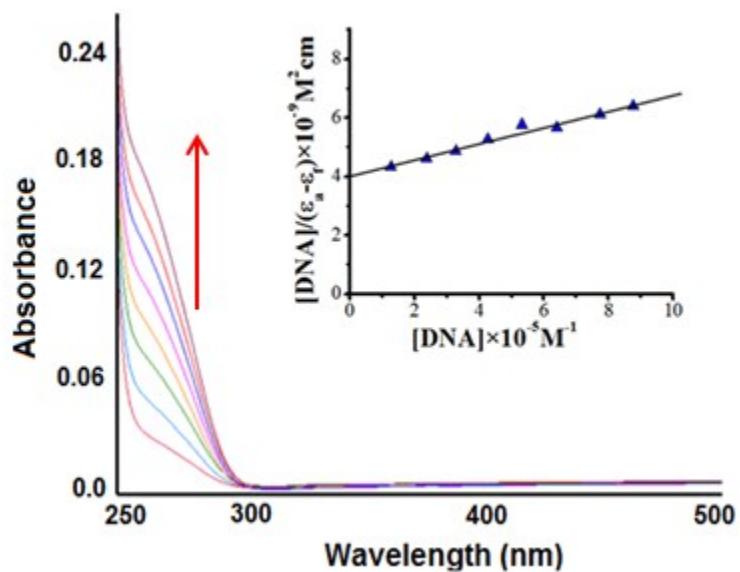


Fig. S13. Absorption spectra for ligand **L** ▲ interaction with CT DNA under the same condition for complex

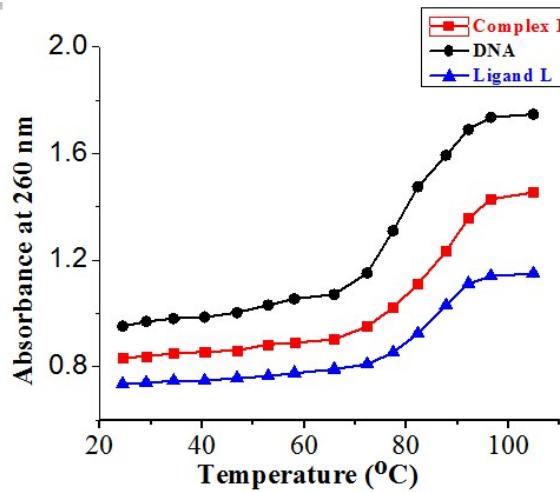


Fig. S14. Thermal melting profile for ligand **L** ▲ and complex **1** ■.

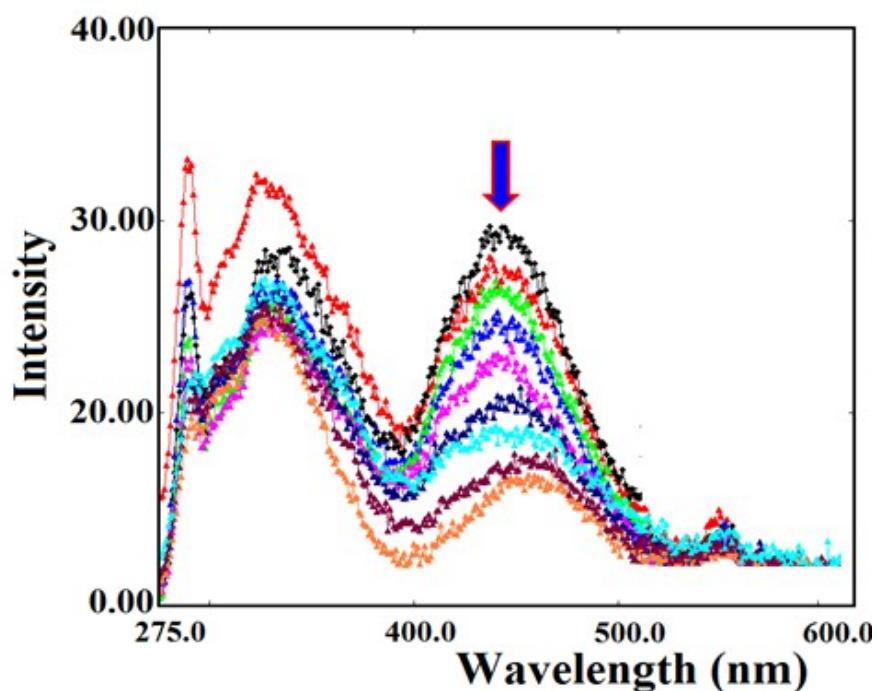


Fig. S15. Emission spectra of ligand (L) in Tris–HCl buffer (pH 7.2) with and without of CT DNA at room temperature. The arrow shows a change in intensity with increasing concentration of DNA.

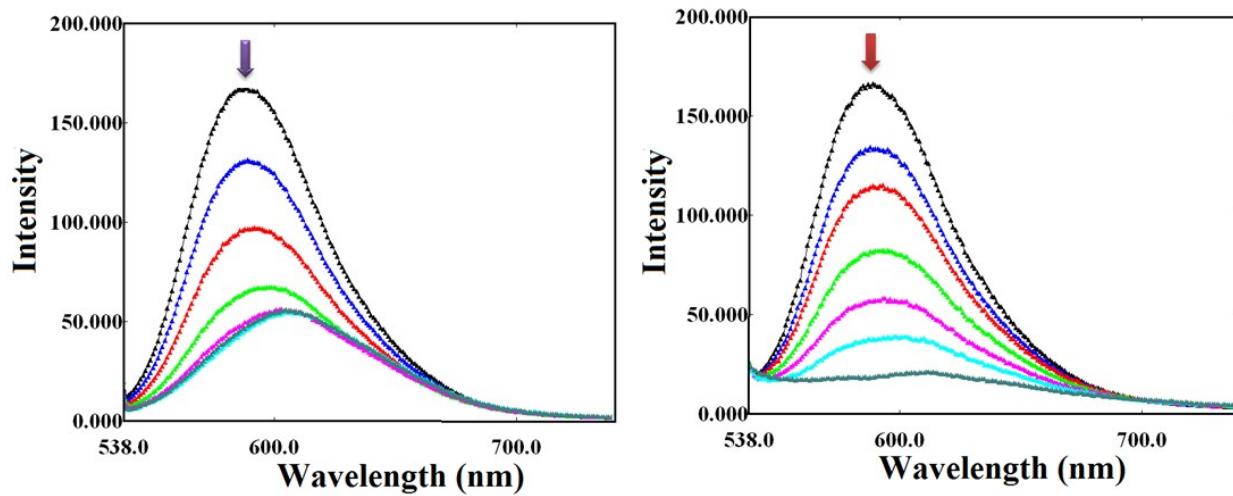


Fig. S16. Emission quenching spectra of CT DNA bound ethidium bromide in the presence of ligand **L** (left) and complex **1** (right) in buffer 5 mM Tris–HCl/50 mM NaCl, pH = 7.2 at 25 °C. Arrow shows change in intensity with increasing concentration of ligand **L**/complex **1**.

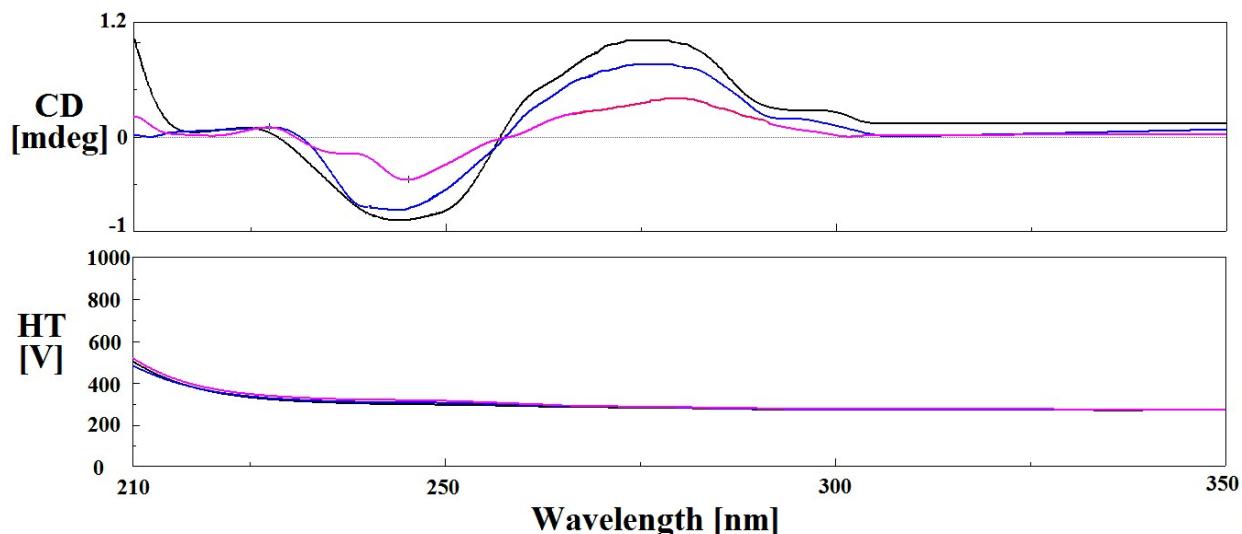


Fig. S17. CD spectra of CT DNA alone in black line (—) and with ligand **L** in blue line (—) Complex **1** in red line (—) in 5 mM Tris–HCl/50 mM NaCl buffer at 25 °C. [Ligand/Complex] = 35 μM, [DNA] = 140 μM.

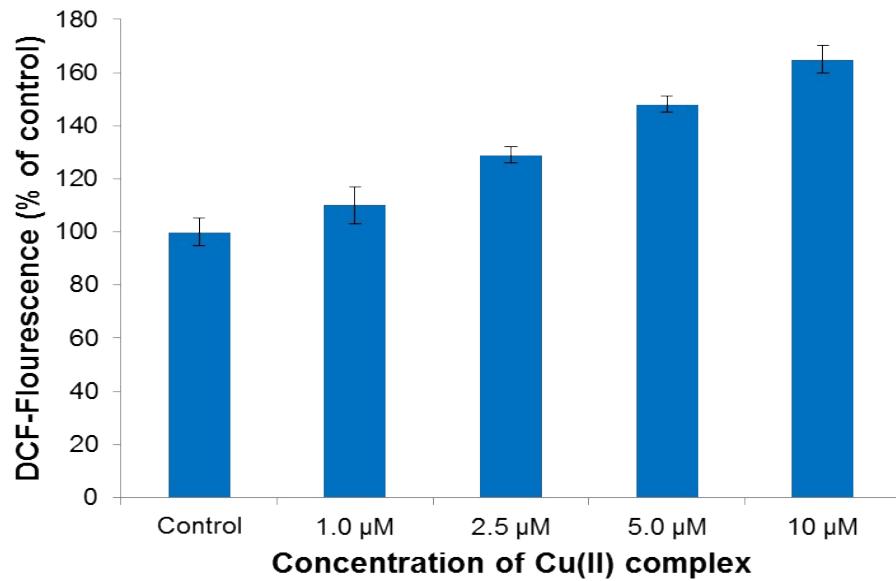


Fig. S18. ROS generation in A549 cells following the exposure of 20 μ M for 24 h.

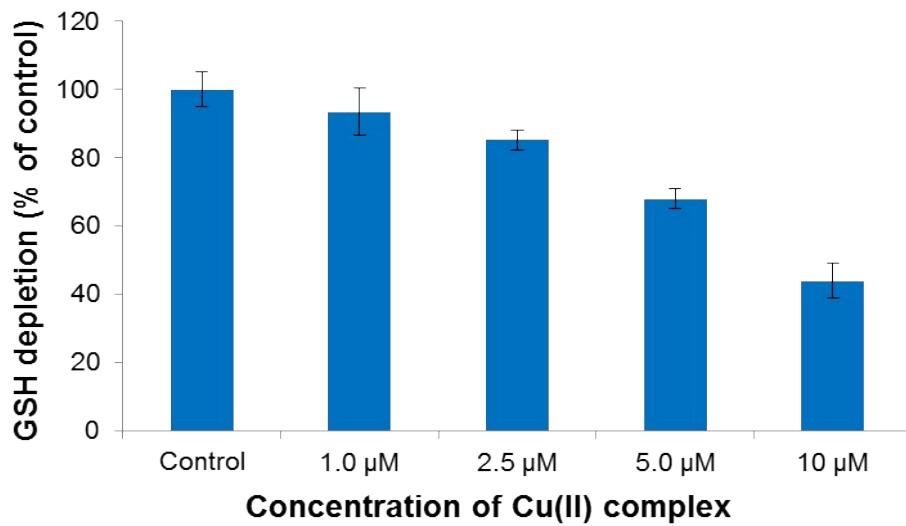


Fig. S19. Glutathione depletion in A549 cells exposed to complex **1** for 24 h.

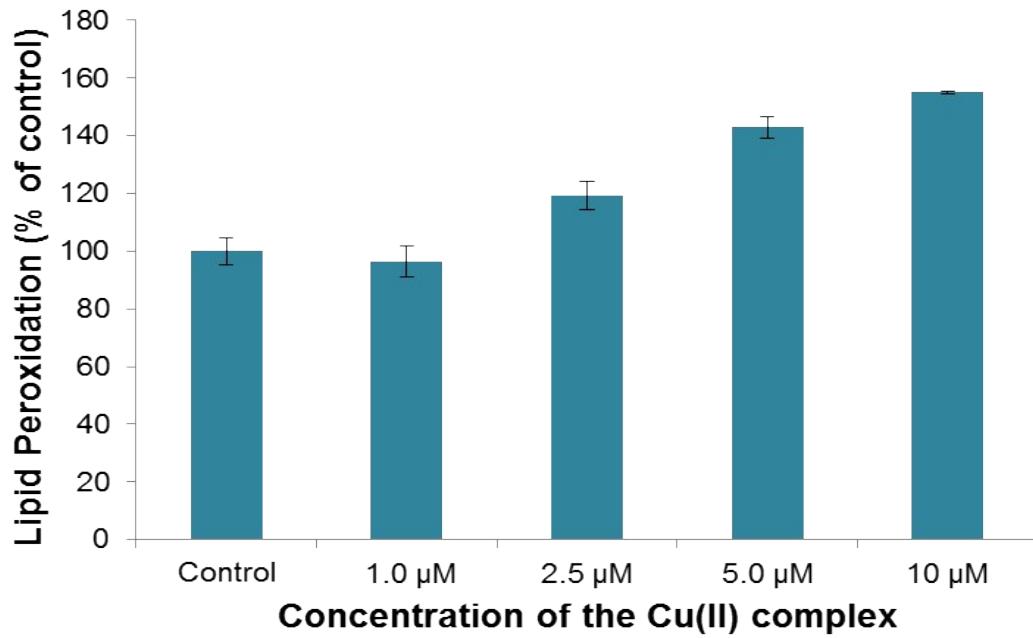


Fig. S20. Lipid peroxidation level in A549 cells exposed to complex **1** for 24 h.