

## Supporting Information

### Acid-assisted hydrothermal synthesis of red fluorescent carbon dots for sensitive detection of Fe(III)

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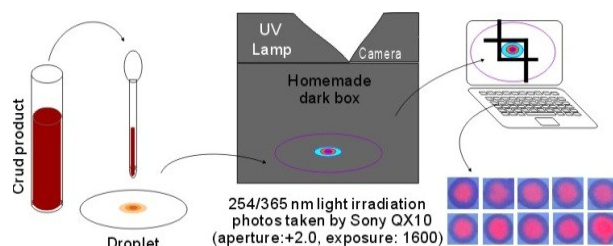
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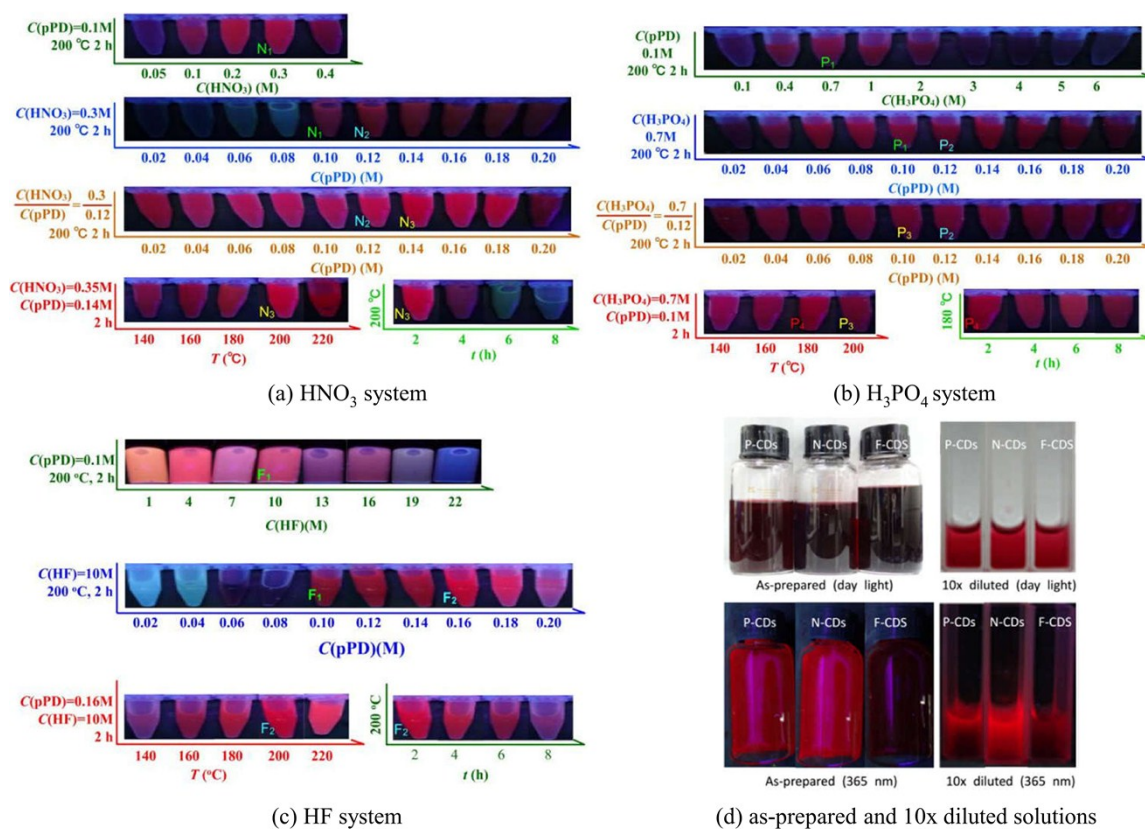
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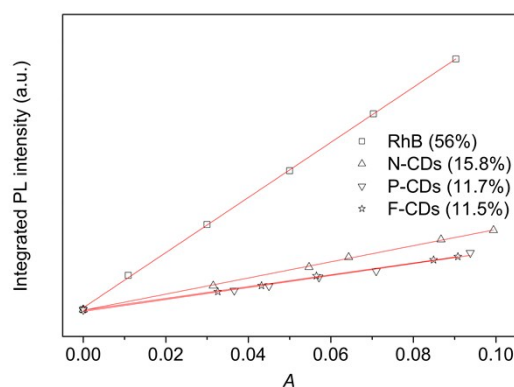
### Figures



**Fig. S1** Paper-test strategy of PL assessment for fluorescence materials. The brightest and the reddest fluorescent C-dots (under 365 nm UV light, in a dark box) were selected via comparing photographs, which were taken by the camera (Sony QX10) with the same parameters (aperture:+2.0, exposure: 1600).



**Fig. S2** Experimental optimization for the preparation of C-dots from p-PD and acid systems of HNO<sub>3</sub> (a), H<sub>3</sub>PO<sub>4</sub> (b), and HF (c). As-prepared C-dots under day light and 365 nm UV light (d). For the optimization experiments, all the as-prepared C-dots aqueous solutions were 10X diluted and filled in 1.5 mL centrifuge tubes, the photographs were taken under 365 nm UV light.



**Fig. S3** Quantum yields data and the PL intensity plots as a function of absorbance for the C-dots and RhB excited at 365 nm.

### Purification of C-dots

After hydrothermal reaction, autoclaves were cooled down to room temperature naturally and the dark-red crude products were obtained. There are 4 methods available to purify these C-dots and the available purification methods for different acid system were shown in **Tab. S1**.

**Tab. S1** Purification methods of C-dots

Methods	N-CDs	P-CDs	F-CDs
dialysis	√	√	√
re-extraction	√	√	√
ethanol-wash	-	√	-
concentration	√	-	-

#### (i) Dialysis

The as-prepared suspension was filtered through a 0.22  $\mu\text{m}$  filter membrane to remove the precipitation and further dialyzed against ultra-pure water through a dialysis bag (molecular weight cut-off: 3500 Da) for 24 h.

#### (ii) Re-extraction

The as-prepared suspension was mixed with ethyl acetate with the volume ratio of 1:1 and shaken for 5 min, C-dots can be extracted into ethyl acetate phase. It is noted that a little amount of water can be resolved by ethyl acetate, so the supernatant contains a little water. Extract the supernatant and add hexane into the supernatant with  $V(\text{supernatant}):V(\text{hexane})=5:1$ . After being shaken for 5 min and let it stand for 10 min, tiny droplets, which means the condensed C-dots aqueous solution, sank and layered. The purified C-dots were then re-extracted.

#### (iii) Ethanol-wash

For C-dots (P-CDs) prepared by non-volatile acid, such as  $\text{H}_3\text{PO}_4$ , add ethanol into the crude products with  $V(\text{product}):V(\text{ethanol})=4:1$ , shake the suspension for 5 min, use centrifuge at 14000 rpm for 30 min and collect the supernatant. Repeat the process for 3-4 cycles.  $\text{H}_3\text{PO}_4$  can be removed by excessive p-PD through forming ammonium salt and by adsorption through poly(p-PD). Supernatant, which contains ethanol and water, can be evaporated to form near-dry paste by a rotary evaporator. The near-dry C-dots paste can be further dried at 80  $^\circ\text{C}$  with a vacuum dryer to make the C-dot powder.

#### (iv) Concentration

For the C-dots (N-CDs and F-CDs) prepared by volatile acid ( $\text{HNO}_3$  and HF), acid-contained crude products can be removed by solvent evaporation. Typically, the crude products were washed (by hexane to remove the unreacted p-PD), centrifuged (at 14000 rpm for 30 min to remove polymer precipitation), and filtered (through a 0.22  $\mu\text{m}$  filter membrane to remove the precipitation). The purified C-dots solutions can be dried to prepare C-dots powders (when necessary) through evaporation by rotary evaporator to near dry state at 80  $^\circ\text{C}$  and a low vacuum condition and further drying in vacuum drying oven.