

Supporting Information:

An electrochemical sensor based on Co_3O_4 nanosheets for lead ions determination

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Optimization for electrochemical detection of Pb(II)

Usually, pH value has a considerable effect on the electrochemical behavior of modified electrodes. The influence of pH on the electrochemical responses of the $\text{Co}_3\text{O}_4/\text{ITO}$ electrode to Pb(II) is shown in Fig. S1.A. The pH in the range from 3.0 to 7.0 was tested because the modifier is leaked in under acidic conditions, while metal hydroxide complexes of $\text{Pb}(\text{OH})_2$ and $\text{Pb}(\text{OH})^+$ are formed when pH exceeds 5. The highest response is obtained at pH 5.0. Hence, pH 5.0 is selected as the optimum pH value for the electrochemical measurements of Pb(II) in the subsequent experiments.

Fig. S1.B demonstrates the effect of the Bi(III) concentration on the stripping response of Pb(II). The peak currents increased with the adding concentration of Bi(III) from 0 to 400 $\mu\text{g L}^{-1}$, the bismuth film deposited on the surface of the modified electrode was employed to facilitate the procedure of nucleation for metal ions and enhance the stripping peak currents due to its ability to form alloys with heavy metals^{17,18}. However, the stripping peak currents decreased at the concentrations of Bi(III) exceeded 400 $\mu\text{g L}^{-1}$, indicating that an excess concentration of Bi(III) would be disadvantageous to the performance of the nanocomposite. Consequently, we chose 400 $\mu\text{g L}^{-1}$ as the optimal Bi(III) concentration.

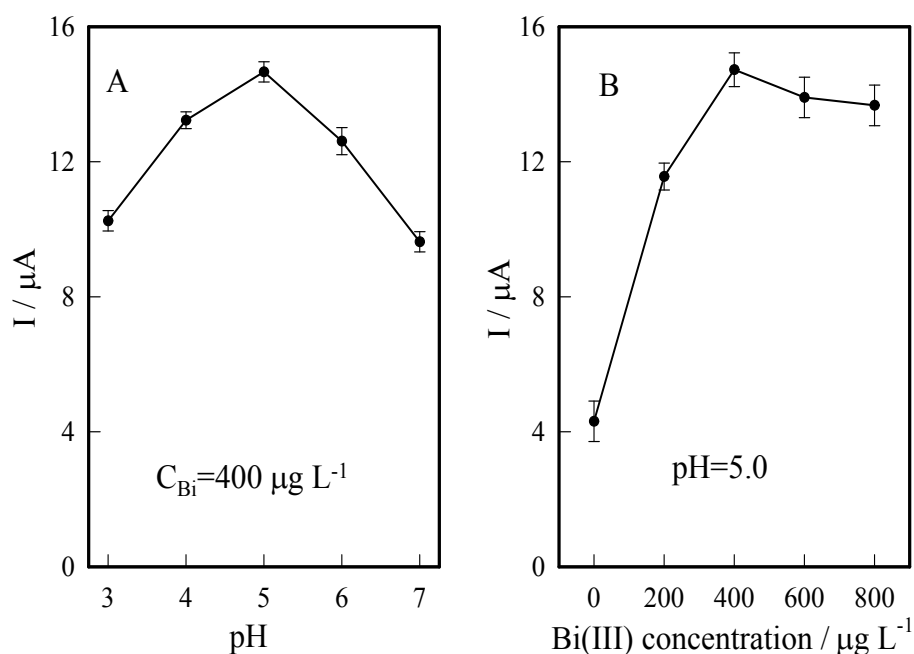


Fig. S1 The effect of pH (A) and Bi(III) concentration (B) on the stripping peak currents of 20 $\mu\text{g L}^{-1}$ Pb(II) at the $\text{Co}_3\text{O}_4/\text{ITO}$ electrode.

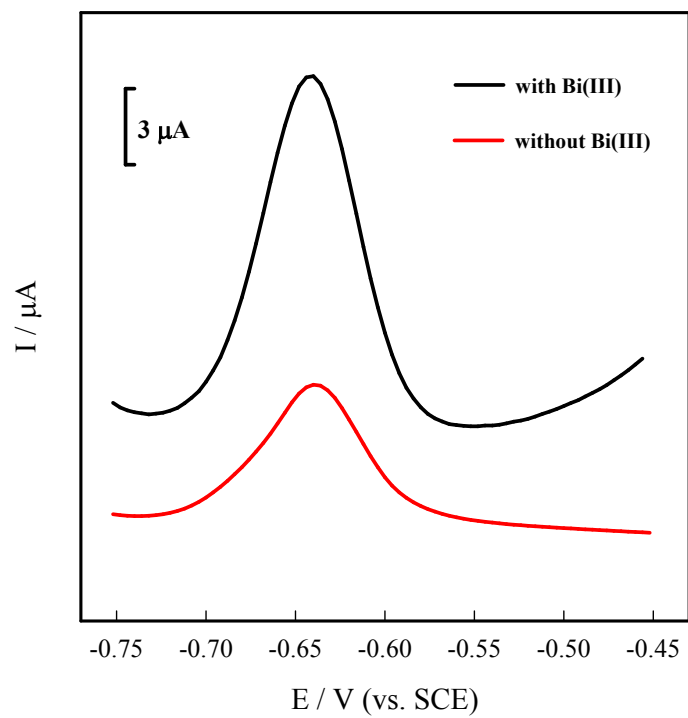


Fig. S2 DPASV of the $\text{Co}_3\text{O}_4/\text{ITO}$ electrode for electroanalysis of Pb(II) with and without Bi(III) in 0.1 M NaAc-HAc buffer (pH 5.0) containing $20 \mu\text{g L}^{-1}$ Pb(II) .

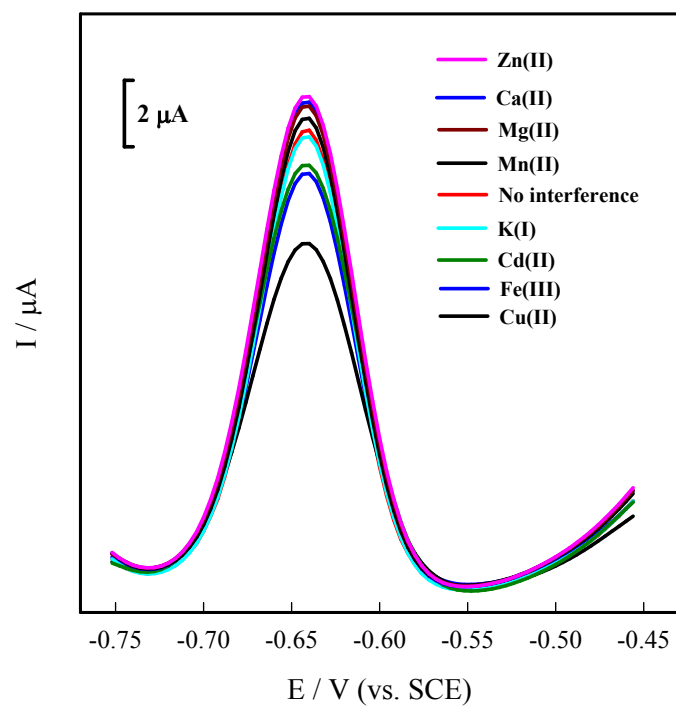


Fig. S3 Typical DPASV responses of the $\text{Co}_3\text{O}_4/\text{ITO}$ electrode towards $\text{Pb}(\text{II})$ in presence of interferences.

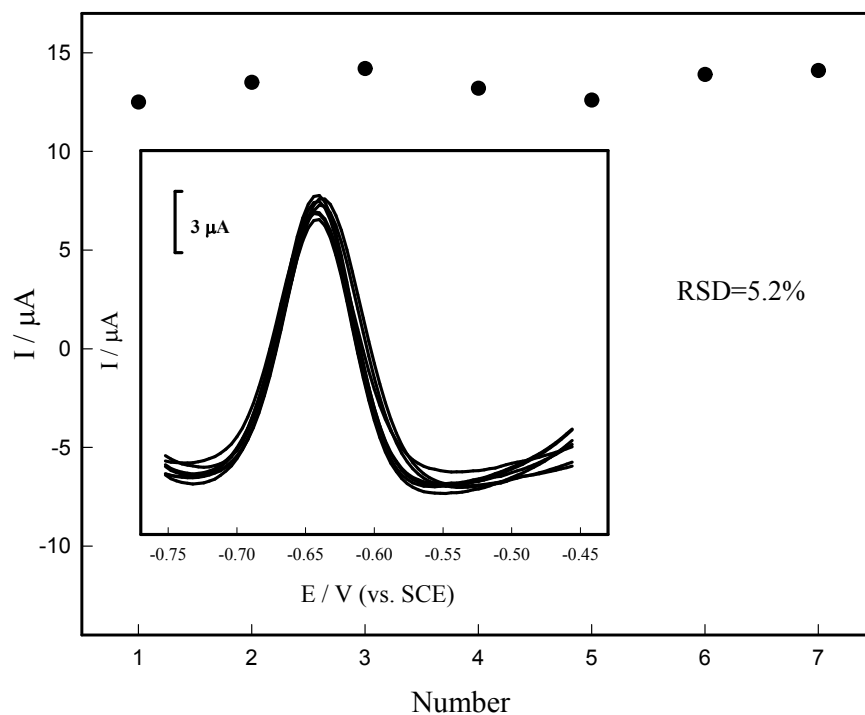


Fig. S4 Electrochemical responses of seven different $\text{Co}_3\text{O}_4/\text{ITO}$ electrode in 0.1 M NaAc-HAc buffer (pH 5.0) containing $20 \mu\text{g L}^{-1}$ Pb(II) and $400 \mu\text{g L}^{-1}$ Bi(III). The insets corresponding to data collected from every DPASV response for seven electrodes. RSD referring to relative standard deviation.

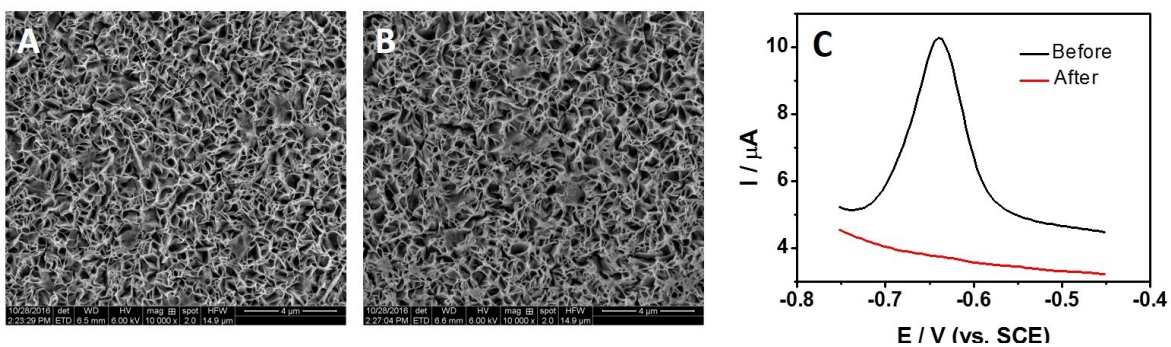


Fig. S5 SEM images of the Co₃O₄ nanosheets before (A) and after (B) the stripping of Pb, and typical DPASV responses of the Co₃O₄/ITO electrode before and after electrochemical stripping of Pb.

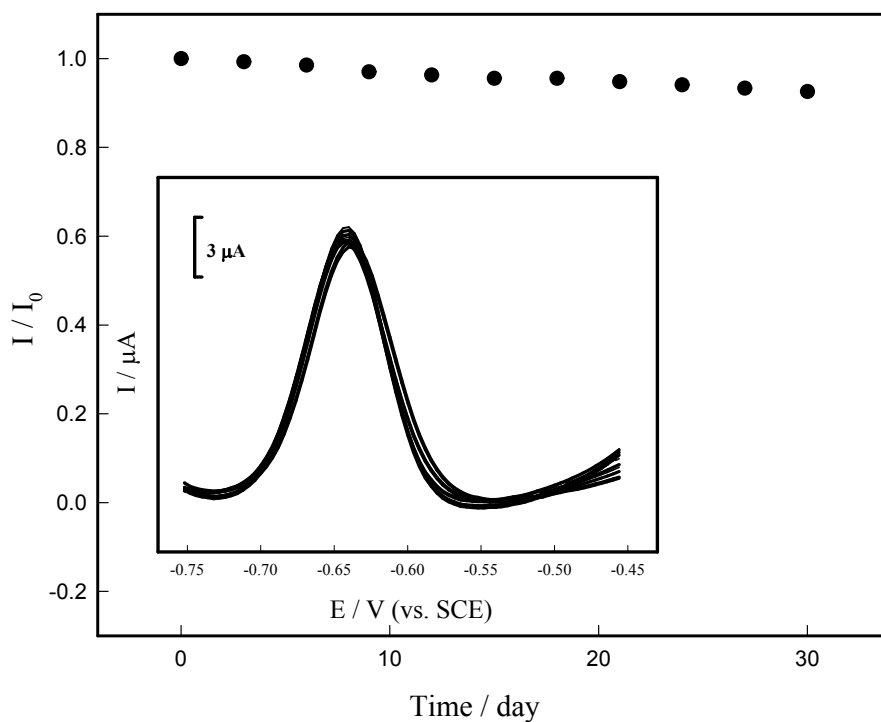


Fig. S6 Stability of the single $\text{Co}_3\text{O}_4/\text{ITO}$ electrode stored at 4°C for several weeks. The insets corresponding to data collected from DPASV response for the same $\text{Co}_3\text{O}_4/\text{ITO}$ electrode.