

Enhanced Photocatalytic CO₂ Reduction over Co-Doped NH₂-MIL-125(Ti) under Visible Light

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Supporting Information

Table S1. Produced HCOOH in the photocatalytic reduction of CO₂ using TEOA as the sacrificial agent upon visible-light irradiation^a

Entry	Catalyst	HCOOH ($\mu\text{mol g}_{\text{cat}}^{-1} \text{h}^{-1}$)
1	NH ₂ -MIL-125(Ti)	16.3
2 ^b	Co(NO ₃) ₂ ·6H ₂ O + NH ₂ -MIL-125(Ti)	15.8
3 ^c	CoO + NH ₂ -MIL-125(Ti)	17.2
4	1 wt.% Co/NH ₂ -MIL-125(Ti)	38.4

^a Reaction conditions: 50 mg of photocatalyst, a MeCN/TEOA volume ratio of 5 to 1, and a total solution volume of 60 mL upon visible-light irradiation ($800 \text{ nm} \geq \lambda \geq 420 \text{ nm}$) for 10 h.

^b With Co(NO₃)₂·6H₂O (0.5 mg) and NH₂-MIL-125(Ti) (49.5 mg).

^c With CoO (0.5 mg) and NH₂-MIL-125(Ti) (49.5 mg).

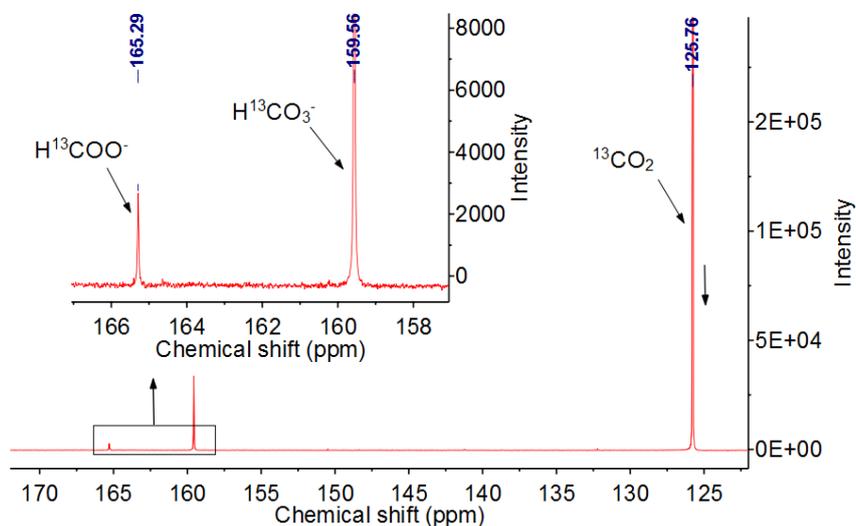


Figure S1. ¹³C NMR spectrum of the product solution from the photocatalytic reduction of ¹³CO₂ over 1.0 wt.% NH₂-MIL-125(Ti) for 20 h upon visible-light irradiation.

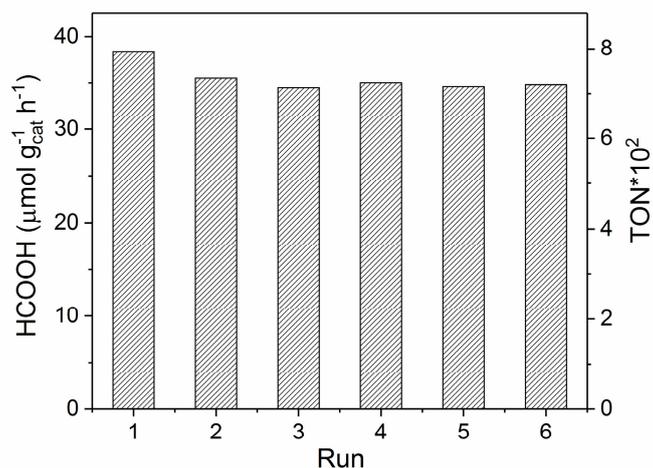


Figure S2. Reusability of 1.0 wt.% Co/NH₂-MIL-125(Ti) in the photocatalytic reduction of CO₂ upon visible-light irradiation. TON (turnover number) was based on HCOOH produced per mole of the Ti species in the catalyst and the reaction of 10 h. Reaction conditions: 50 mg of photocatalyst, a MeCN/TEOA volume ratio of 5 to 1, and a total solution volume of 60 mL upon visible-light irradiation ($800 \text{ nm} \geq \lambda \geq 420 \text{ nm}$) for 10 h.

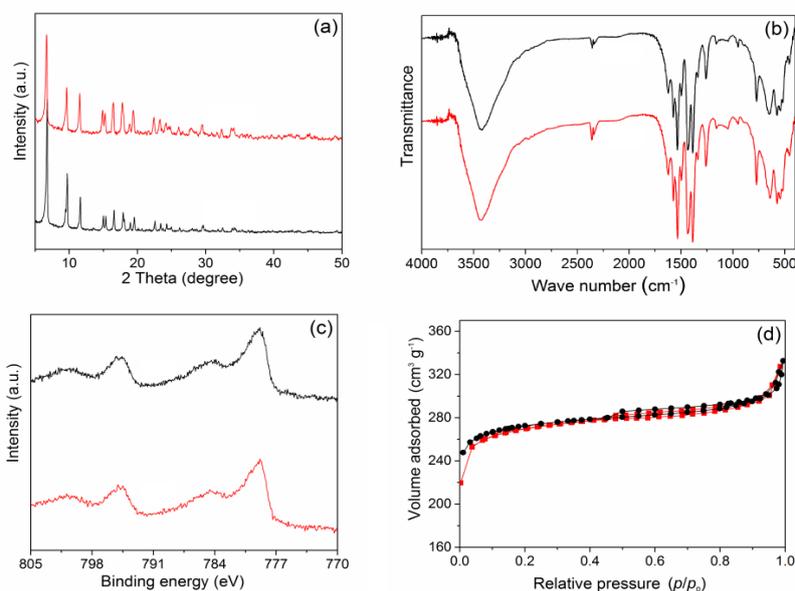


Figure S3. XRD patterns (a), IR spectra (b), N₂ adsorption-desorption isotherms at -196 °C (c), and high-resolution XPS spectra (d) of the fresh (black) and used (red) 1.0 wt.% Co/NH₂-MIL-125(Ti) after the irradiation of visible light.

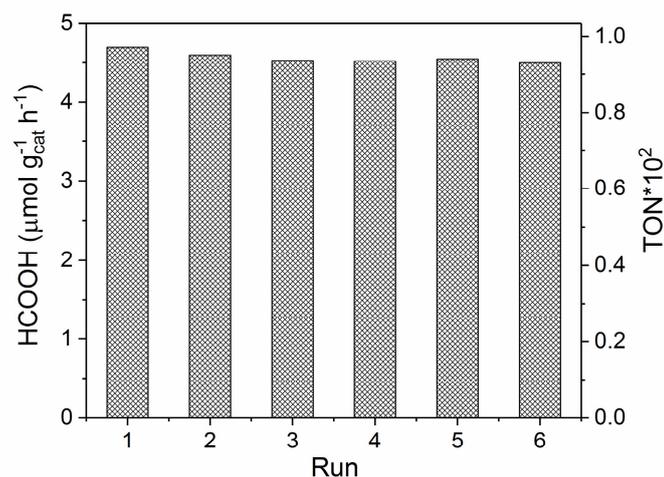


Figure S4. Reusability of 1.0 wt.% Co/NH₂-MIL-125(Ti) in the photocatalytic reduction of CO₂ upon UV-light irradiation. TON (turnover number) was based on HCOOH produced per mole of the Ti species in the catalyst and the reaction of 10 h. Reaction conditions: 50 mg of photocatalyst, a MeCN/TEOA volume ratio of 5 to 1, and a total solution volume of 30 mL upon UV-light irradiation (four 4 W UV lamps with a wavelength centered at 365 nm used as an irradiation source) for 10 h.

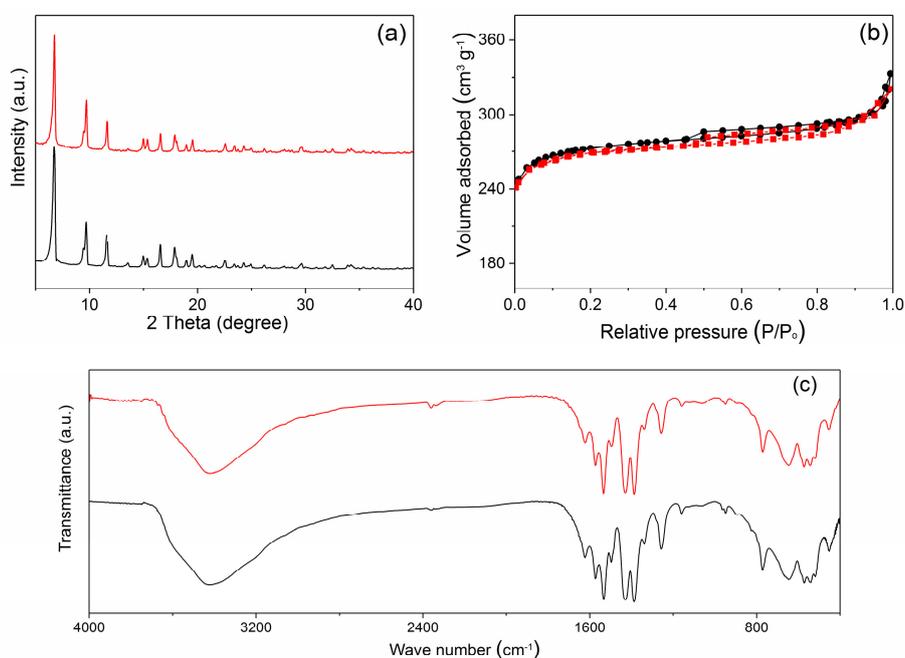


Figure S5. XRD patterns (a), N₂ adsorption-desorption isotherms at -196 °C (b), and FT-IR spectra (c) of the fresh (black) and used (red) 1.0 wt.% Co/NH₂-MIL-125(Ti) after the irradiation of UV light.

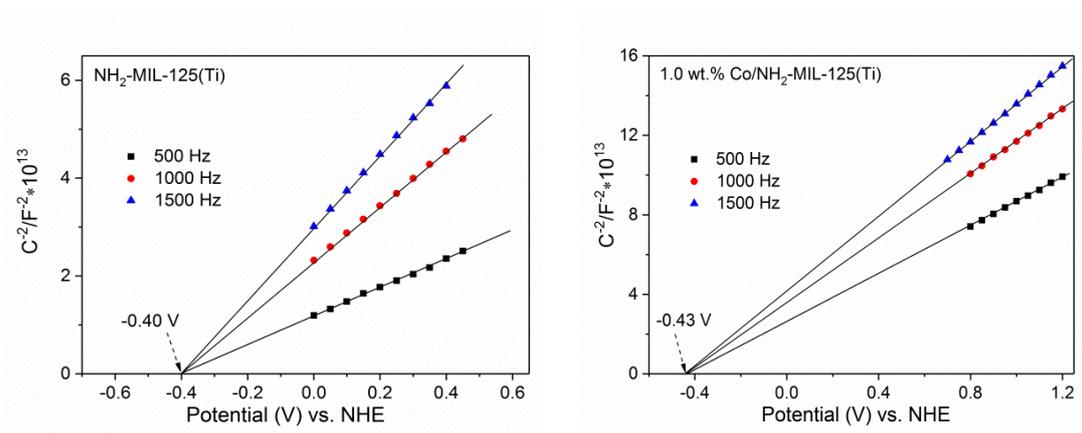


Figure S6. Mott-Schottky plots of NH₂-MIL-125(Ti) (left) and 1.0 wt.% Co/NH₂-MIL-125(Ti) (right) measured at different frequencies.

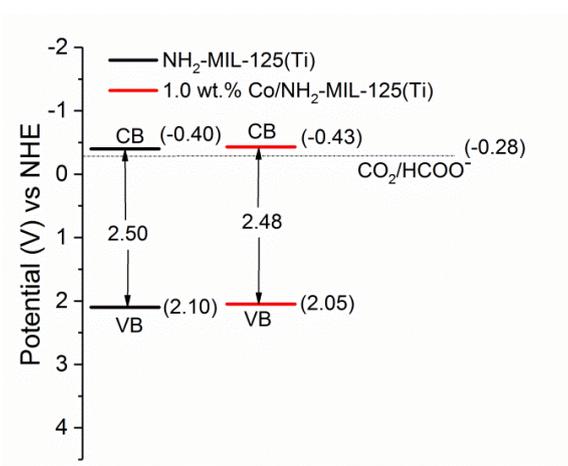


Figure S7. Calculated band positions for NH₂-MIL-125(Ti) and 1.0 wt.% Co/NH₂-MIL-125(Ti).

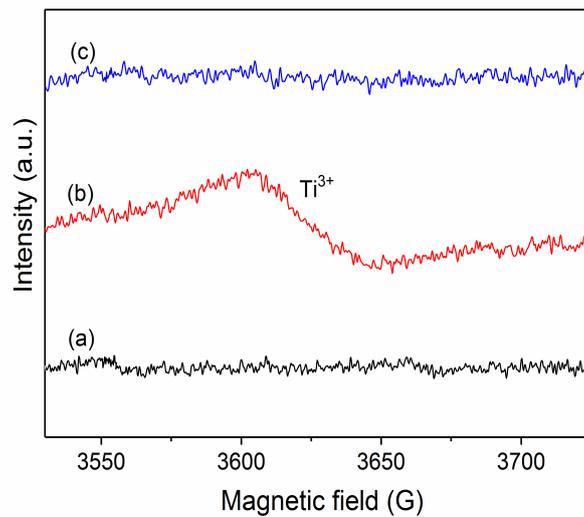


Figure S8. ESR spectra of 1.0 wt.% Co/NH₂-MIL-125(Ti) with TEOA in the dark (a), upon visible-light irradiation for 1 min in N₂ (b), and after the introduction of CO₂ without visible-light irradiation (c).