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Electronic Supplementary Information

Amphiphilic Acrylamide Block Copolymer: RAFT Block

Copolymerization and Monolayer Behavior

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Kinetic Plot Showing ptPA Synthesis



Figure S1. (a) First-order kinetic plot of $ln([M]_0/[M])$ versus time, (b) evolution of polydispersity index, and (c) number-average molecular weight for RAFT polymerization of tPA. The broken line in (c) represents the theoretical prediction. Reaction conditions: [tPA]/[CTA-Me]/[AIBN] = 219/5/1 in toluene.

¹H NMR Spectra of pDDA-*b*-ptPA



Figure S2. ¹H NMR spectra of pDDA-*b*-ptPA. Lower panel shows amplified images for the signals at around 2 ppm.

AFM Image of pDDA-b-ptPA LB Film

Preparation of pDDA-*b*-ptPA Langmuir-Blodgett (LB) films was conducted using an automatically controlled Langmuir trough (HBM-AP; Kyowa Interface Science Co. Ltd.). The monolayer was transferred onto a Si wafer at a surface pressure of 15 mN m⁻¹. The substrate was made hydrophobic with *n*-octyltrichlorosilane before use. The pDDA-*b*-ptPA LB film was investigated using atomic force microscopy (AFM) measurements (SPA-400; Seiko Instruments Inc.) under dynamic mode. The measurement was carried out using an Alcoated silicon cantilever (SI-DF20; Hitachi High-Tech Science Corp.).



Figure S3. An AFM image of 2-layer pDDA-b-ptPA LB film.

Monolayer Stability at a Constant Pressure

Figure S4 shows temporal changes in the surface area of pDDA-*b*-ptPA with constant compression at 30 mN m⁻¹ (red). The area remains unchanged with a minor decrease of ca. 5% over 1000 min, which indicates that pDDA-*b*-ptPA takes a stable monolayer at 30 mN m⁻¹. However, the surface area decreases with compressive force of 40 mN m⁻¹ (blue), suggesting collapse of the monolayer at the air–water interface.



Figure S4. Temporal change in the surface area (*A*) of pDDA-*b*-ptPA on the compression at 30 (red) and 40 mN m⁻¹ (blue): A_0 denotes the initial surface area.