

**Molecular fractionation of a soil fulvic acid (FA) and competitive sorption of trace metals
(Cu, Zn, Cd, Pb) in hematite-solution systems: effect of the FA-to-mineral ratio.**

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Supporting information: 4 figures

Fig. S1: Percentage of PPH organic carbon sorbed onto hematite ($r = 1.4 \text{ mgC/m}^2$).

Fig. S2: Negative ion ESI mass spectra in the ranges 120-400 m/z and 400-798 m/z of (a) PPH native solution (excerpted from Fleury et al. ²¹), and (b) the supernatant of PPH sorption experiment onto hematite at pH 3.9 and $r = 1.4$.

Fig. S3: VK diagram showing elemental formulae assigned to ESI(-) FTMS peaks detected at pH 3.9 on the ranges 120-400 m/z (a) and 400-798m/z (b) for the PPH native solution. Excerpted from Fleury et al. ²¹.

Fig. S4: Results of experiments of competitive Cu sorption (with Pb, Zn and Cd as possible competitors) onto 0.375g/L hematite in the absence of PPH, and of competitive and single Cu sorption for an initial PPH-to-mineral ratio of 0.1 mgC/m^2 .

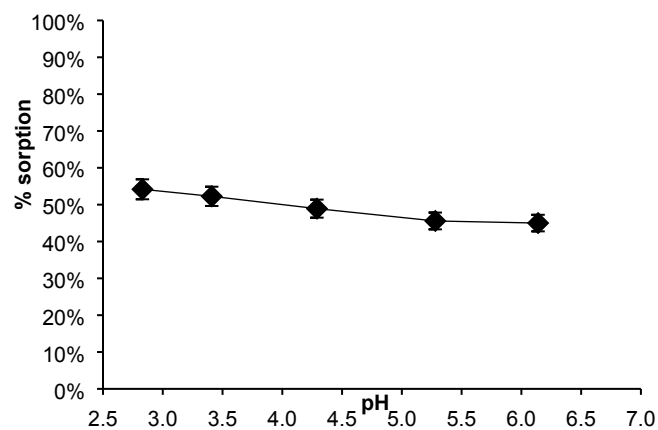


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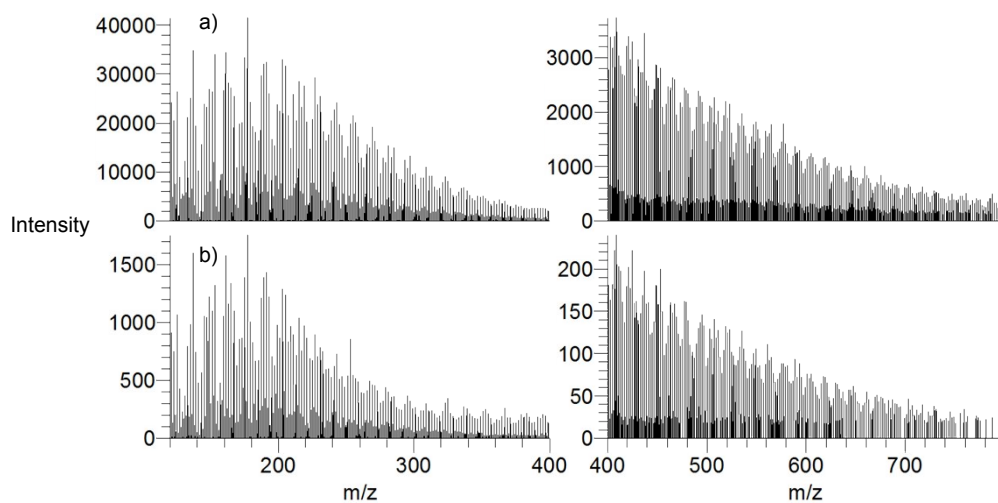


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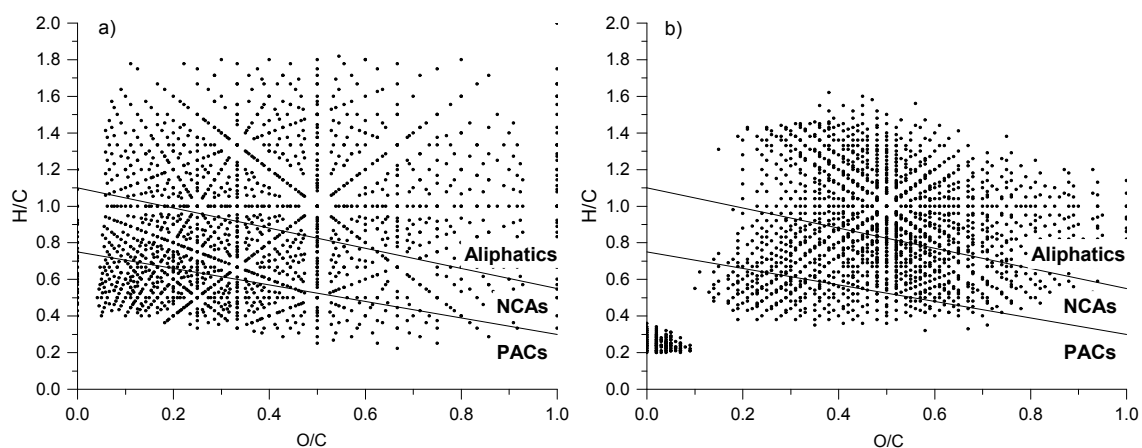


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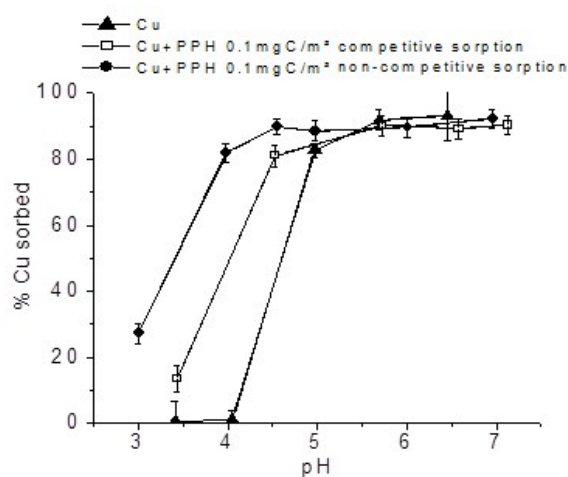


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