Supporting Information

Jun Xie¹⁺, Yingying Wu¹⁺, Tianyuan Zhang¹, Mengyue Zhang¹, Weiwei Zhu¹,

Elizabeth A Gullen², Zhaojie Wang³, Yung-Chi Cheng², Yixuan Zhang^{1*}

¹School of Life Science and Biopharmaceutics, Shenyang Pharmaceutical University,

Shenyang, 110016, People's Republic of China

²Department of Pharmacology, Yale University School of Medicine, New Haven, CT

06520, United States of America

³Yunnan Provincial Academy of Science and Technology, Kunming, 650051, People's

Republic of China

⁺These authors contributed equally to this work

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Figure S1. ¹H NMR (600 MHz, methanol-d4) spectrum of 1



Figure S2. ¹³C NMR (150MHz, methanol-d4) spectrum of 1



Figure S3. ¹H-¹H COSY spectrum of 1 in methanol-d4







Figure S5. HMBC spectrum of 1 in methanol-d4



Figure S6. NOSEY spectrum of 1 in methanol-d4

Figure S7. HRESIMS spectrum of 1



Figure S8. ¹H NMR (600 MHz, methanol-d4) spectrum of 2









Figure S10. ¹H-¹H COSY spectrum of 2 in methanol-d4







Figure S12. HMBC spectrum of 2 in methanol-d4



Figure S13. NOSEY spectrum of 2 in DMSO-d6

Figure S14. HRESIMS spectrum of 2



Figure S15. ¹H NMR (600 MHz, methanol-d4) spectrum of 3









Figure S17. ¹H-¹H COSY spectrum of 3 in methanol-d4



Figure S18. HSQC spectrum of 3 in methanol-d4



Figure S19. HMBC spectrum of 3 in methanol-d4



Figure S20. NOSEY spectrum of 3 in DMSO-d6

Figure S21. HRESIMS spectrum of 3



Figure S22. ¹H NMR (600 MHz, methanol-d4) spectrum of 4









Figure S24. ¹H-¹H COSY spectrum of 4 in methanol-d4



Figure S25. HSQC spectrum of 4 in methanol-d4



Figure S26. HMBC spectrum of 4 in methanol-d4



Figure S27. NOSEY spectrum of 4 in DMSO-d6





Figure S29. ¹H NMR (600 MHz, methanol-d4) spectrum of 5



Figure S30. ¹³C NMR (150 MHz, methanol-d4) spectrum of 5





Figure S31. ¹H-¹H COSY spectrum of 5 in methanol-d4



Figure S32. HSQC spectrum of 5 in methanol-d4



Figure S33. HMBC spectrum of 5 in methanol-d4



Figure S34. NOSEY spectrum of 5 in DMSO-d6

Figure S35. HRESIMS spectrum of 5



Figure S36. ¹H NMR (600 MHz, chloroform-d) spectrum of 1a-R



Figure S37. ¹H NMR (600 MHz, chloroform-d) spectrum of 1b-S



Figure S38. ¹H NMR (600 MHz, chloroform-d) spectrum of 2a-R



Figure S39. ¹H NMR (600 MHz, chloroform-d) spectrum of 2b-S



Figure S40. ¹H NMR (600 MHz, chloroform-d) spectrum of 3a-R



Figure S41. ¹H NMR (600 MHz, chloroform-d) spectrum of 3b-S



Figure S42. ¹H NMR (600 MHz, chloroform-d) spectrum of 4a-R



Figure S43. ¹H NMR (600 MHz, chloroform-d) spectrum of 4b-S



Figure S44. ¹H NMR (600 MHz, chloroform-d) spectrum of 5a-R



Figure S45. ¹H NMR (600 MHz, chloroform-d) spectrum of 5b-S





Figure S46-50. IR spectrum of 1,2,3,4,5





Figure S51 The CD profiles of compounds **1**(A), **2**(B), **3**(C), **4**(D), **5**(E)







Figure S52 The plausible biosynthetic pathways of compounds 1-7

Compounds1-7 structures could be explained as being derived from poly- β -keto chains, formed by coupling of acetic acid (C2) units via condensation reactions. The formation of the poly- β -keto chain could be envisaged as a series of Claisen reactions. Two molecules of acetyl-CoA could participate in a Claisen condensation giving acetoacetyl-CoA, and this reaction could be repeated to generate a poly- β -keto ester of appropriate chain length. The conversion of acetyl-CoA into malonyl-CoA increased the acidity of the α -hydrogens, and thus provided a better nucleophile for the Claisen condensation. In the biosynthetic sequence, no acylated malonic acid derivatives were produced, and no label from [14C] bicarbonate was incorporated, so the carboxyl group introduced into malonyl-CoA was simultaneously lost by a decarboxylation reaction during the Claisen condensation.

Table S1 ¹ H NMR	Chemical Shift	Data for Diagnost	ic Signals from	the (S)- and (R)-

^a proton chemical shifts ($\Delta \delta_{\rm H} = \delta_{\rm S} - \delta_{\rm P}$)										
MTPA-ester	Н-2	Н-3	H-4	Н-5		Н-6		H-8		Н-9
R	5.8750	6.1645	5.2496	2.1117	2.554	9 2.11	17 2.2	.745 2	.2522	2.9325
S	6.1021	7.0267	4.5579	2.1095	5 2.516	2.10	95 2.2	668 2	.1963	2.8655
$\Delta \delta_{ m H}$	+0.2271	+0.8622		-0.002	2 -0.038	-0.00	-0.0	0077 -0	.0559	-0.0670
			^b proton	chemical	shifts ($\Delta \delta_{ m H}$	$=\delta_{\rm S}-\delta_{\rm R}$)				
MTPA-ester	H-2	H-3	H-4	H-5	H	-6	H-7	Н	-8	H-9
R	6.0950	7.0727	4.3880) 1.8790	1.7175	1.9981	5.1017	2.3442	1.5265	2.4436
S	6.1001	7.0909	4.3981	1.8760	1.7076	1.9982	5.1008	2.3446	1.5262	2.4463
$\Delta\delta_{ m H}$	+0.0051	+0.0182	2	-0.003	-0.0099					
			^c proton	chemical s	hifts ($\Delta \delta_{ m H}$ =	$= \delta_{\rm S} - \delta_{\rm R}$				
MTPA-ester	H-2	Н-3	H-4	H-5]	H-6	H-7	I	I- 8	H-9
R	6.1010	7.0959	4.3979	1.8771	1.7079	1.9976	5.1009	2.3440	1.5269	2.4473
S	6.0934	7.0712	4.3865	1.8774	1.7163	1.9966	5.1001	2.3431	1.5250	2.4470
$\Delta \delta_{ m H}$	-0.0076	-0.0247		+0.0055	+0.0084	Ļ				
			^d proton o	chemical s	hifts ($\Delta \delta_{ m H}$ =	$= \delta_{\rm S} - \delta_{\rm R}$				
MTPA-ester	H-2	Н-3	H-4	H-5	H	-6	H-7	Н	-8	H-9
R	6.0857	7.0970	4.3755	1.8024	1.6454	1.5872	4.3456	2.2524	1.3656	2.4202
S	6.0877	7.1006	4.3744	1.7994	1.6495	1.5904	4.3483	2.2481	1.3656	2.4143
$\Delta \delta_{ m H}$	+0.0020	+0.0036		-0.0030	+0.0041	+0.0032		-0.0043		-0.002
			^e proton cl	nemical sh	ifts ($\Delta \delta_{\rm H} =$	$\delta_{\rm S} - \delta_{\rm R})$				
MTPA-ester	H-2	Н-3	H-4	H-5	Н	-6	H - 7	H-	8	H-9
R	6.1121	7.0863	4.4173	1.7987	1.7255	1.5977	4.3249	2.2433	1.3800	2.4182
S	6.1148	7.0930	4.4199	1.7957	1.7073	1.5900	4.3360	2.2504	1.3800	2.4230
$\Delta \delta_{ m H}$	+0.0027	+0.0067		-0.0030	-0.0182	-0.0077		+0.0071		+0.0048

Ester Derivatives of 1^a , 2^b , 3^c , 4^d , 5^e (Measured at 600 MHz in CDCl₃)