

Xiehualing 1H 1# MeOD 20160108

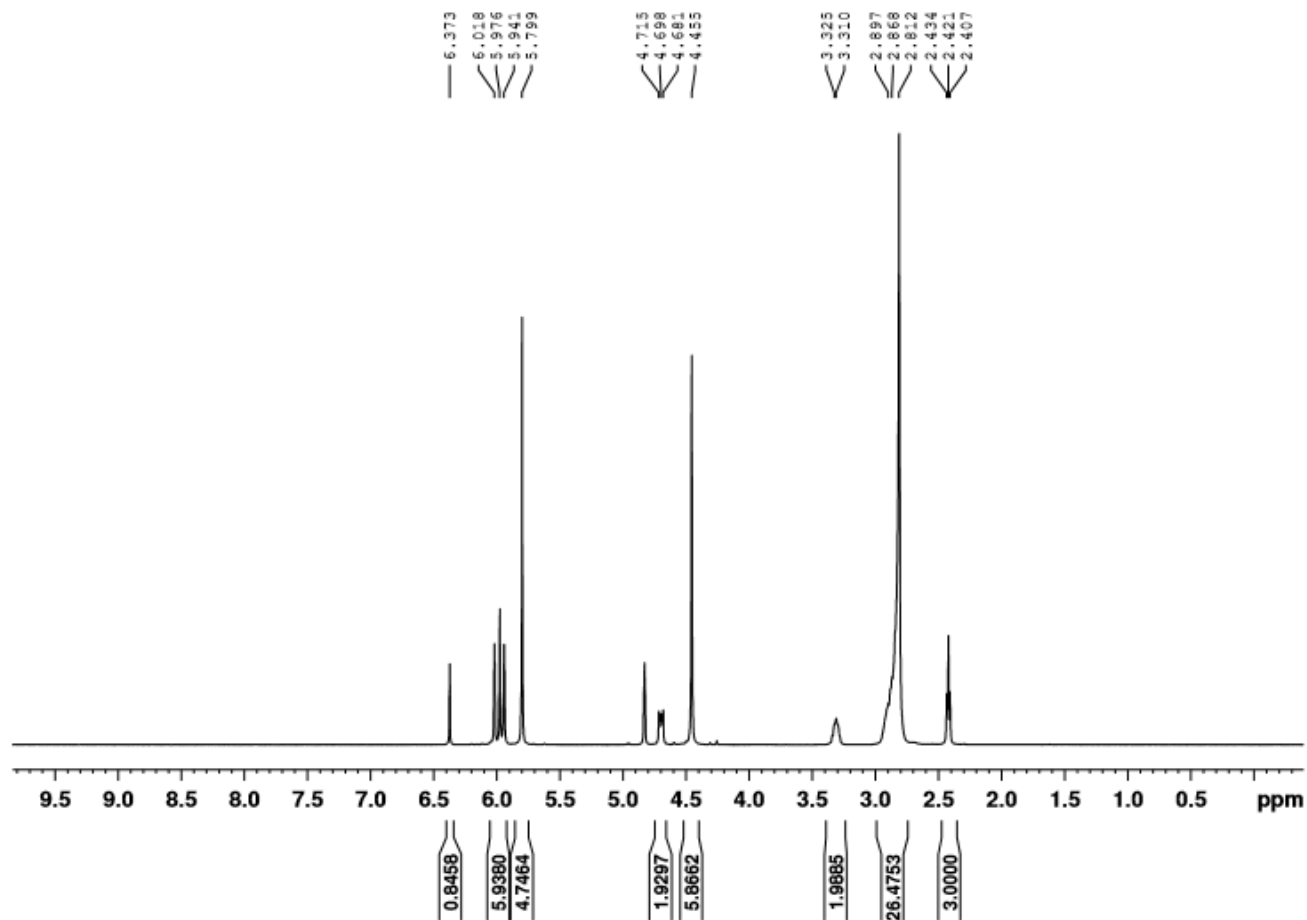


Fig. 1S Hydrogenous nuclear magnetic resonance spectroscopy

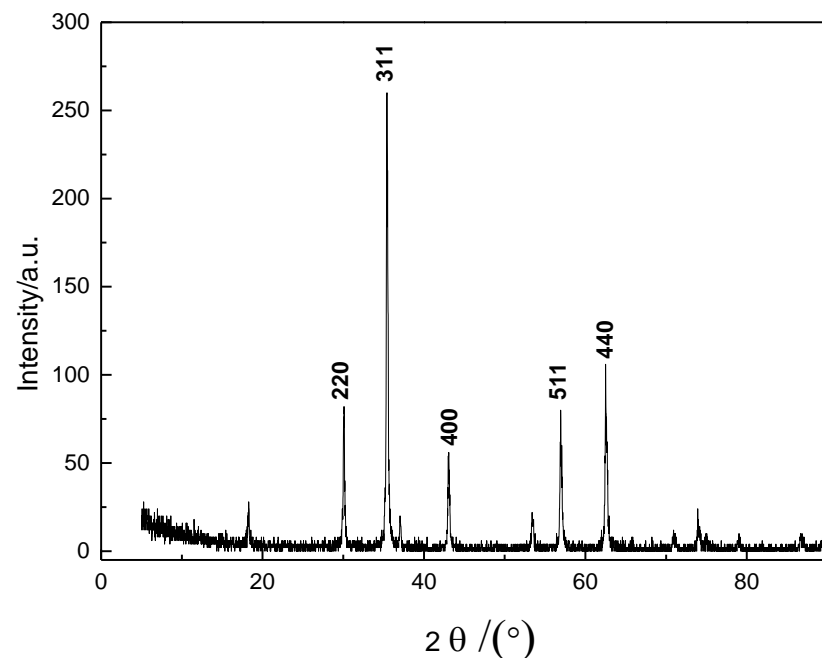


Fig. 2S wide-angle XRD patterns for Fe₃O₄.

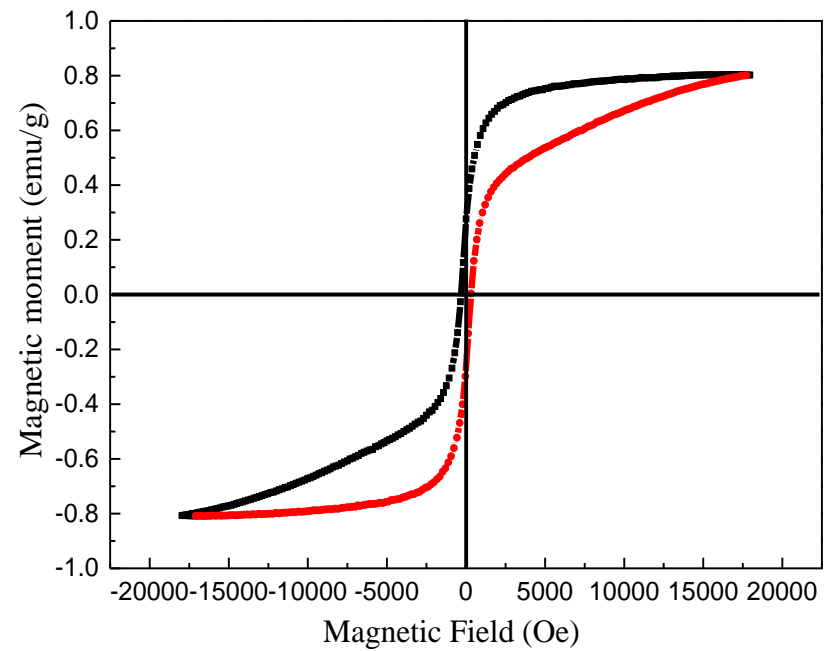


Fig. 3S Magnetic Hysteresis for MMS at 293K.

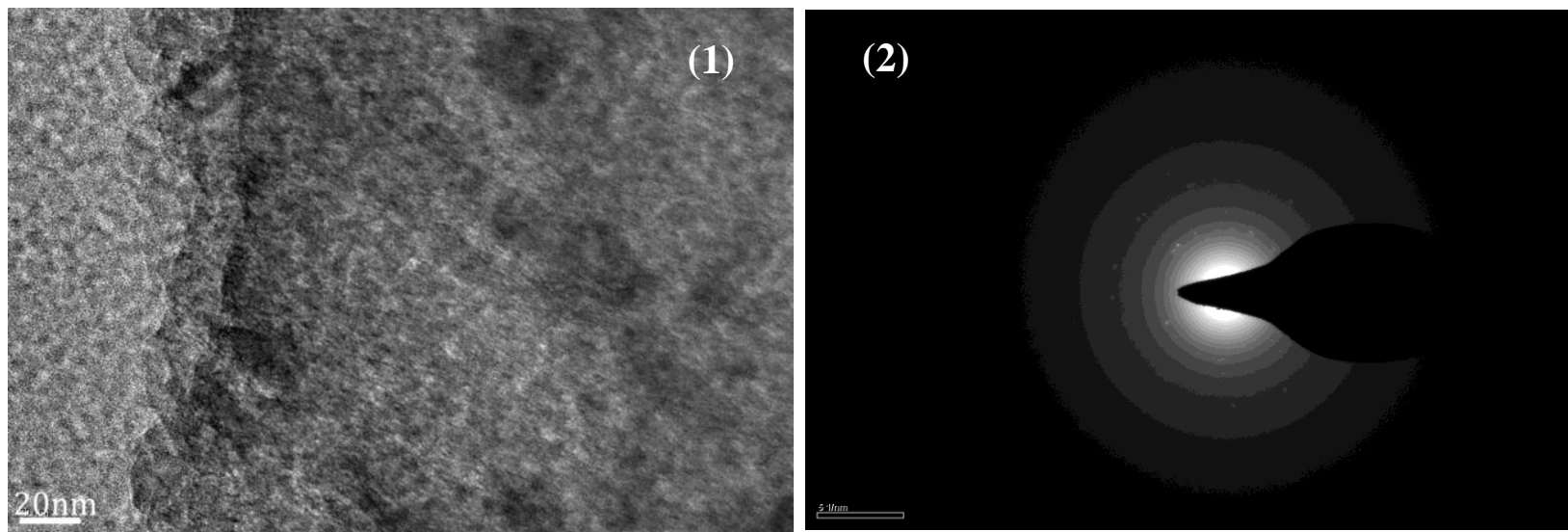


Fig. 4S TEM images of MMIOC for MMS.

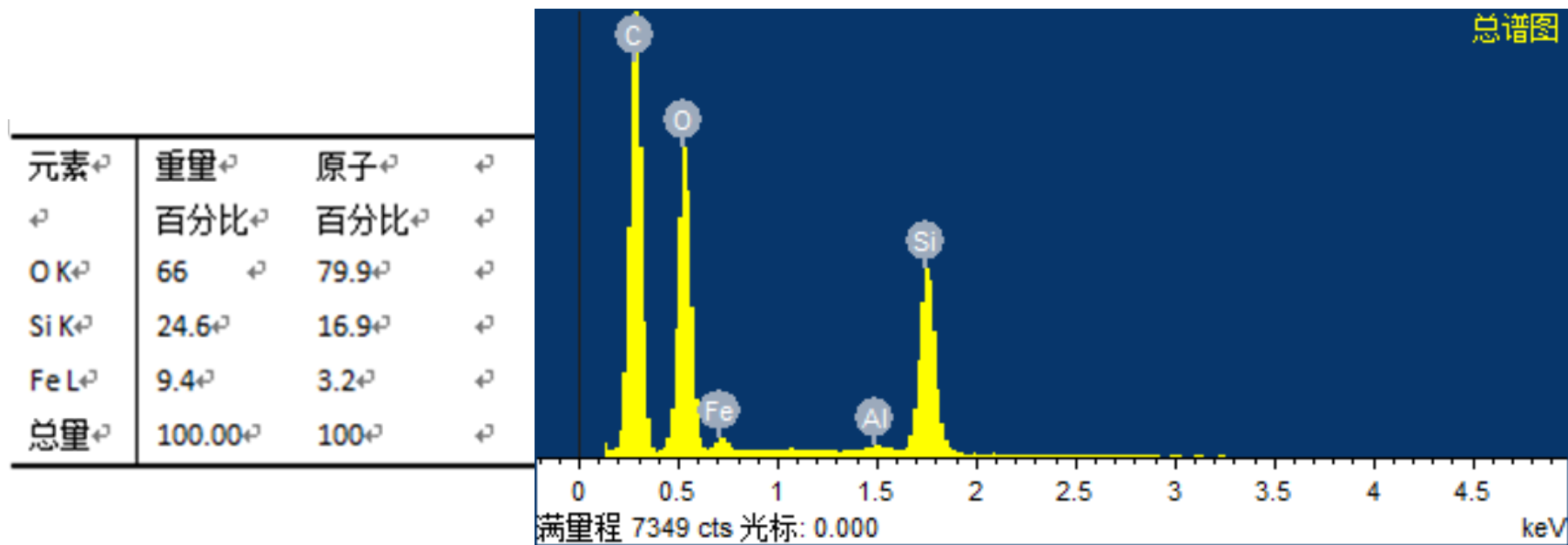


Fig. 5S EDS of MMIOC for MMS

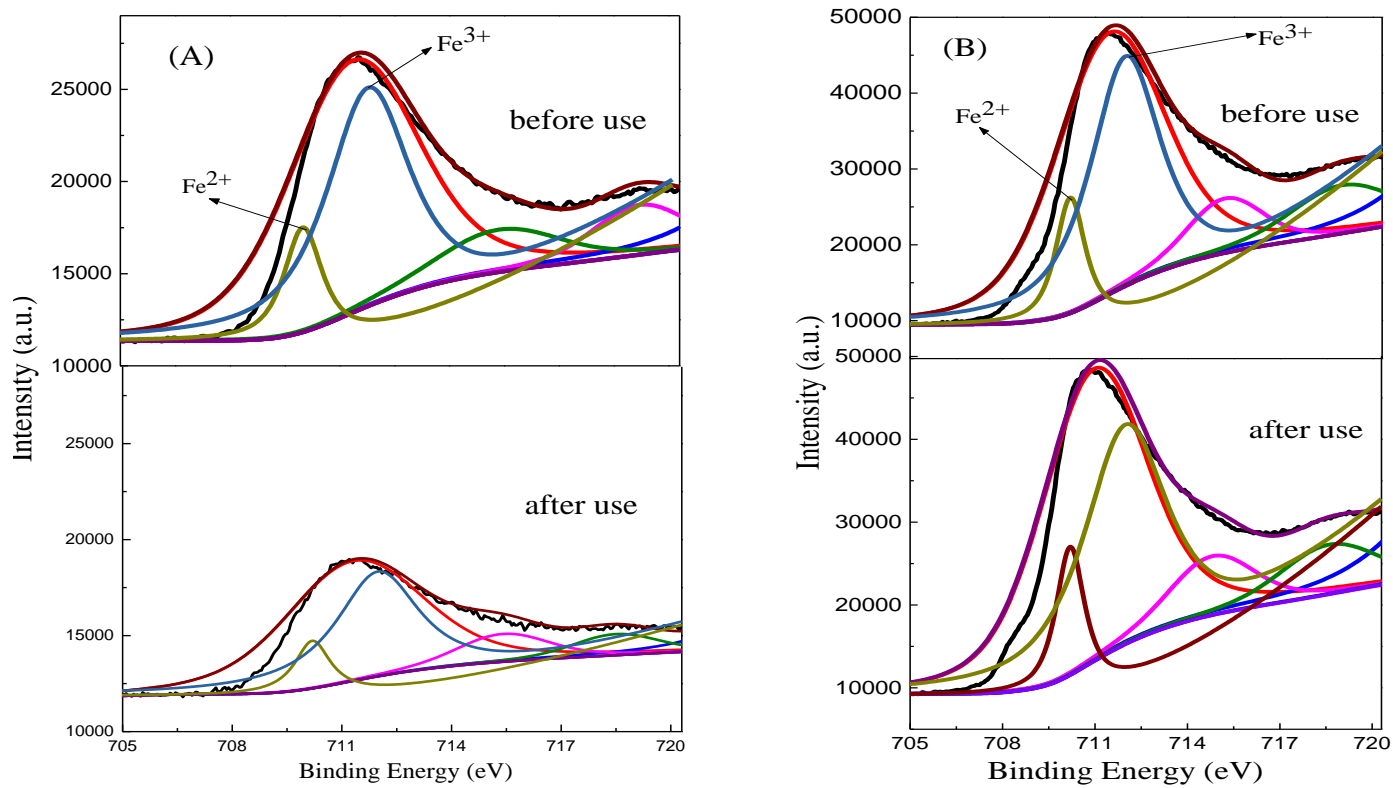


Fig. 6S Fe 2p_{3/2} XPS spectra for (A) MMS and (B) Fe₃O₄

Table 1S Binding energies (B.E.) in eV (± 0.1) for MMS and Fe₃O₄ before (B.U.) and after (A.U.) use

Catalysts	Si(2p)	O(1S)	Fe(2p _{3/2})			
			Fe(III)	%	Fe(II)	%
MMS(B.U.)	154	532.75	711.25	85%	710	15%
MMS(A.U.)	154	532.60	712.12	90%	710.39	10%
Fe ₃ O ₄ (B.U.)	~	530.5	714.66	57.81%	711.49	42.19%
Fe ₃ O ₄ (A.U.)	~	530.1	712.61	67%	710.25	33%

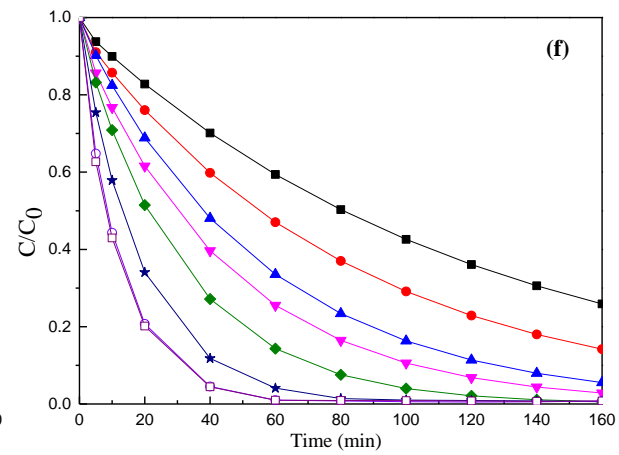
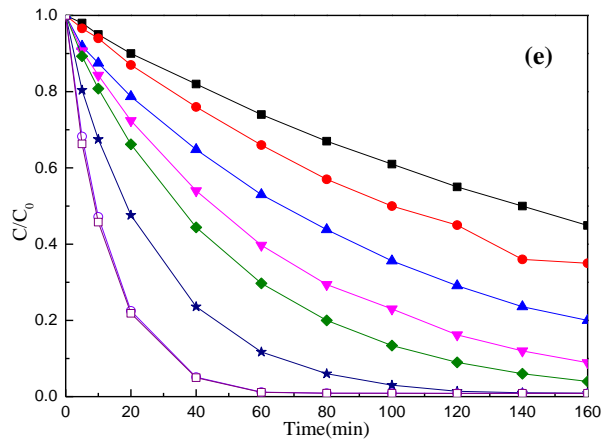
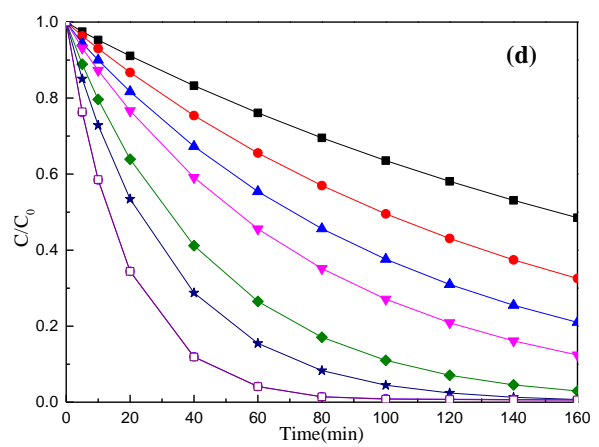
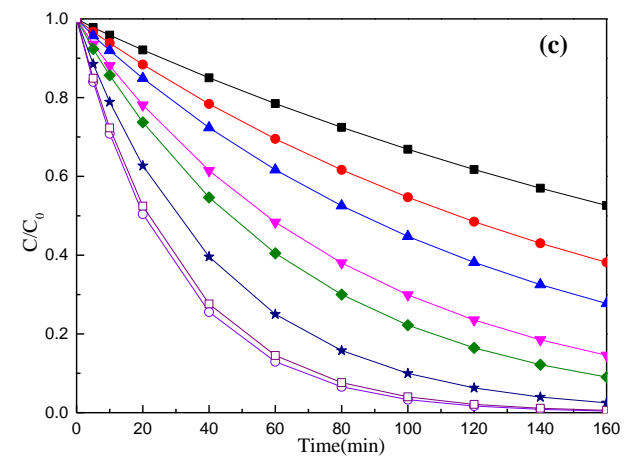
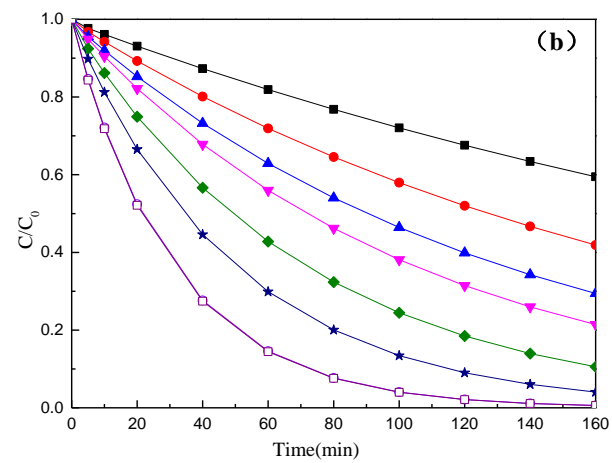
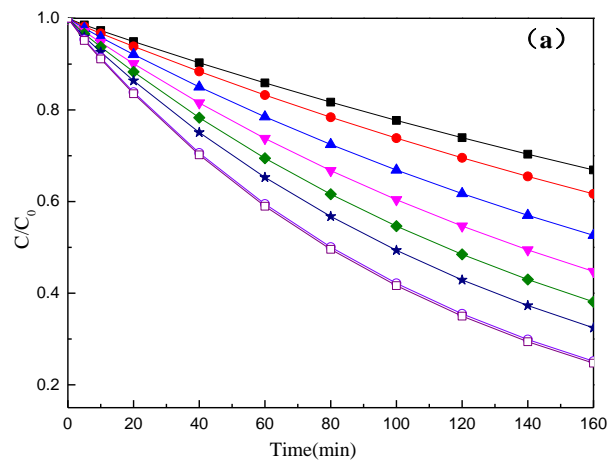
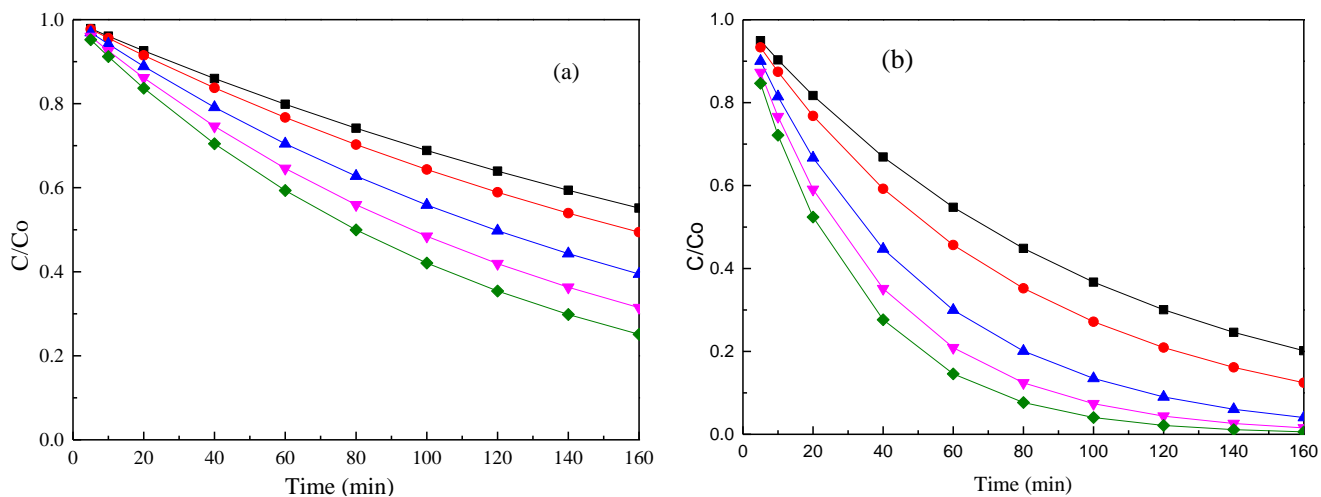


Fig. 7S Degradation effect of various catalysts: (a) Fe_3O_4 ; (b) P7/3; (c) F7/3; (d) P8/2; (e) F8/2; (f) MMS; (■) 0.05g L^{-1} ; (●) 0.1g L^{-1} ; (▲) 0.2g L^{-1} ; (▼) 0.5g L^{-1} ; (◆) 1g L^{-1} ; (★) 1.5g L^{-1} ; (○) 2g L^{-1} ; (□) 4g L^{-1} ; Expect investigated parameter, others fixed at $[\text{RhB}] = 1\text{ mM}$; $[\text{PS}] = 40\text{ mM}$; initial $\text{pH} = 7.0$; $T = 25^\circ\text{C}$.



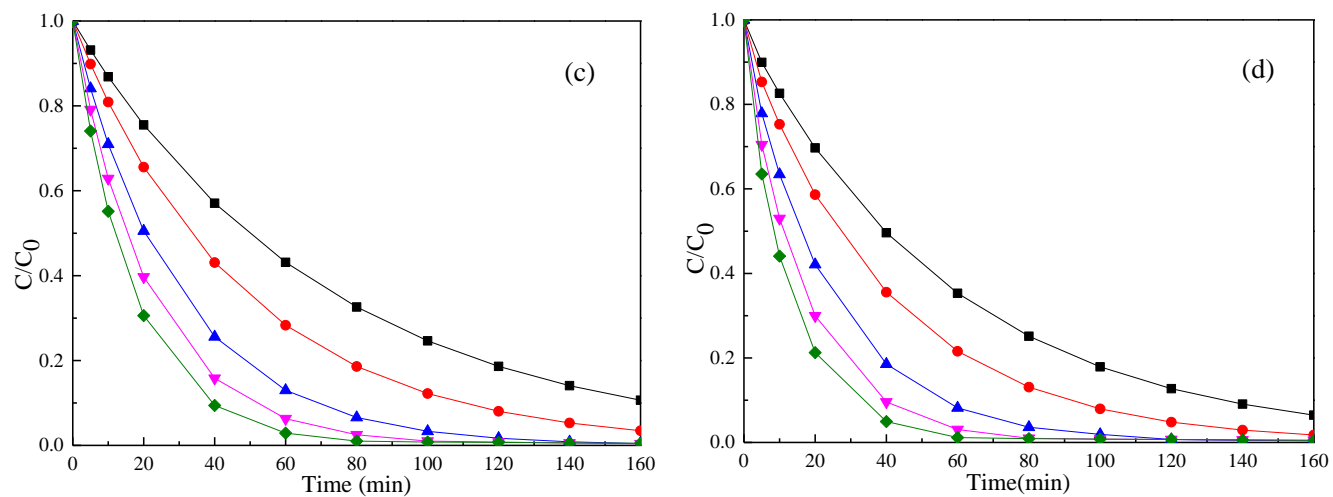


Fig. 8S Decolorization effect of different PS dosage: (a) Fe_3O_4 ; (b) P8/2; (c) F8/2; (d) MMS; (■)5mM; (●)10mM; (▲)20mM; (▼)30mM; (◆)40mM. Expect investigated parameter, others fixed at $[\text{RhB}] = 1 \text{ mM}$; $[\text{Catalyst}] = 2.0 \text{ g L}^{-1}$; initial pH = 7.0; T = 25°C.

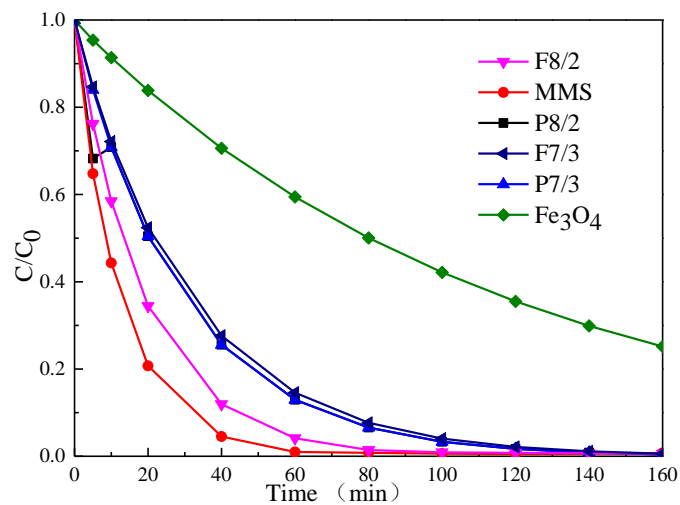


Figure 9S Effect of various catalysts on RhB degradation in different Fe_xO_y/PS systems. Expect investigated parameter, others fixed at $[RhB] = 1$

mM; $[Catalyst] = 2.0 \text{ g L}^{-1}$; $[PS] = 40 \text{ mM}$; initial pH = 7.0; $T = 25^\circ \text{C}$.