Electronic supporting information

Surface-active ionic liquids for Palladium-catalysed cross coupling in water: Effect of ionic liquid concentration on the catalytically active species

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| | CMC [mM] ^a | | |
|--------------|---------------------------|------------------------------|--|
| Ionic Liquid | Conductivity ^b | Surface tension ^c | |
| 1 | 14.53 [1] | 13.25 [1] | |
| 2 | 13.75 [2] | 10.33 [2] | |
| 3 | 2.69 [3] | 2.31 [3] | |
| 4 | 4.32 | 3.24 | |
| 5 | 5.52 | 4.33 | |
| 6 | 5.69 | 5.07 | |
| 7 | 5.04 | 4.91 | |
| 8 | 4.17 | 5.61 | |

1. Table S1: Critical micelle concentrations for all involved ionic liquids

^a Solutions were prepared with doubly-distilled Millipore Milli-Q water. Samples were equilibrated at 25 °C with a HAAKE K15 thermostat before measurements. ^b Conductivity measurement were performed on a Mettler Toledo SevenExcellence system. InLAB® 741-ISM electrode (cell constant k = 0.105). The conductimeter was calibrated with a standard KCI solution and measurements were performed in duplicate; ^c Surface tension was determined with the Du Noüy ring method on a Krüss tensiometer at RT. Each measurement was repeated 5 times.

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² A. Cognigni, S. Kampichler and K. Bica, J. Colloid Interface Sci., 2017, **492**, 136–145.

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2. Figure S1: Concentration dependence of the Heck reaction of ethyl acrylate 9 and iodobenzene 10 in aqueous solution of ionic liquids [C₁₂mim]Cl 1 using DBU as base.



3. Figure S2: Copy of ¹³C NMR spectroscopy showing the formation of a Pd-carbene at 177 ppm.



Conditions: 2 mL [C_{12} mimCl] **2** solution (50 mM) in D_2O , 0.02 mmol Pd_2 allyl₂Cl₂ and 0.02 mmol K_2CO_3 ; 30 min at 80 °C under air.



Conditions: $[C_{12}mimCl]$ **2** solution (50 mM) in D_2O