

Supplementary Information

The diversities of magnetic behaviors appearing simultaneously in three isomorphism based on 1H-tetrazolyl- acetic acid ligands

Tong Yan,^{a†} Lin Du,^{*a†} Lin Sun,^b Xiao-Feng Zhang,^a Tao Wang,^a Jing Feng,^a Jie Zhou,^a and Qi-Hua Zhao ^{*a}

a. Key Laboratory of Medicinal Chemistry for Natural Resource Education Ministry, School of Chemical Science and Technology · Pharmacy, Yunnan University, Kunming, 650091 PR China.

E-mail: lindu@ynu.edu.cn; qhzhao@ynu.edu.cn; Tel: +86-871-5032929; Fax: +86-871-5032929.

b. Department of Chemistry, Northwest University, Xi'an, Shaanxi, 710069 P.R. China.

Table S1. Selected bond lengths (Å) and angle (°) for **1**.

Mn1–O1	2.1528(11)	Mn1–O2 ^{#3}	2.1659(10)
Mn1–O1 ^{#1}	2.1528(11)	Mn1–N1 ^{#4}	2.3015(12)
Mn1–O2 ^{#2}	2.1659(10)	Mn1–N1 ^{#5}	2.3015(12)
O1–Mn1–O1	188.12(6)	O2 ^{#2} –Mn1–N1 ^{#4}	88.62(4)
O1–Mn1–O2 ^{#2}	178.44(4)	O2 ^{#3} –Mn1–N1 ^{#4}	90.99(4)
O1 ^{#1} –Mn1–O2 ^{#2}	91.44(4)	O1–Mn1–N1 ^{#5}	87.54(5)
O1–Mn1–O2 ^{#3}	91.44(4)	O1 ^{#1} –Mn1–N1 ^{#5}	92.86(5)
O1 ^{#1} –Mn1–O2 ^{#3}	178.44(4)	O2 ^{#2} –Mn1–N1 ^{#5}	90.99(4)
O2 ^{#2} –Mn1–O2 ^{#3}	89.03(6)	O2 ^{#3} –Mn1–N1 ^{#5}	88.62(4)
O1–Mn1–N1 ^{#4}	92.86(4)	N1 ^{#4} –Mn1–N1 ^{#5}	179.45(6)
O1 ^{#1} –Mn1–N1 ^{#4}	87.54(5)	C3–O1–Mn ^{#1}	130.56(9)
C3–O2–Mn1 ^{#3}	131.39(8)	C1–N1–Mn1 ^{#5}	129.47(10)

Symmetry codes: #1, 2-x, +y, 3/2-z; #2, +x, -y, -1/2+z; #3, 2-x, -y, 2-z; #4, -1/2+x, 1/2-y, -1/2+z; #5, 5/2-x, 1/2-y, 2-z.

Table S2. Selected bond lengths (Å) and angle (°) for **2**.

Co1–O2	2.0736(10)	Co1–N4 ^{#4}	2.1731(13)
Co1–O2 ^{#1}	2.0736(10)	Co1–N4 ^{#5}	2.1731(13)
Co1–O1 ^{#2}	2.0881(11)	O1–Co1 ^{#3}	2.0881(11)
Co1–O1 ^{#3}	2.0881(11)	N4–Co1 ^{#6}	2.1731(13)
O2–Co1–O2 ^{#1}	91.69(6)	O1 ^{#3} –Co1–N4 ^{#4}	86.01(5)
O2–Co1–O1 ^{#2}	177.12(4)	O2 ^{#1} –Co1–N4 ^{#5}	88.40(4)
O2 ^{#1} –Co1–O1	290.54(4)	O1 ^{#2} –Co1–N4 ^{#5}	86.01(5)
O2–Co1–O1 ^{#3}	90.54(4)	O1 ^{#3} –Co1–N4 ^{#5}	93.34(5)
O2 ^{#1} –Co1–O1 ^{#3}	177.12(4)	O1 ^{#3} –Co1–N4 ^{#4}	86.01(5)
O1 ^{#2} –Co1–O1 ^{#3}	87.29(6)	O2 ^{#1} –Co1–N4 ^{#5}	88.40(4)
O2–Co1–N4 ^{#4}	88.40(4)	O1 ^{#2} –Co1–N4 ^{#5}	86.01(5)
O2 ^{#1} –Co1–N4 ^{#4}	92.23(4)	O1 ^{#3} –Co1–N4 ^{#5}	93.34(5)
O1 ^{#2} –Co1–N4 ^{#4}	93.34(4)	C1–N4–Co1 ^{#6}	129.45(10)

Symmetry codes:

#1,1-x,+y,3/2-z;#2,+x,-y,1/2+z;#3,1-x,-y,1-z;#4,-1/2+x,1/2+y,+z;#5,3/2-x,1/2+y,3/2-z;#6,1/2+x,-1/2+y,+z.

Table S3. Selected bond lengths (Å) and angle (°) for **3**.

Ni1–O1	2.0665(15)	Ni1–N4 ^{#4}	2.1082(17)
Ni1–O1 ^{#1}	2.0665(15)	Ni1–N4 ^{#5}	2.1082(17)
Ni1–O2 ^{#2}	2.0686(14)	O2–Ni1 ^{#3}	2.0686(14)
Ni1–O2 ^{#3}	2.0686(14)	N4–Ni1 ^{#4}	2.1082(17)

O1-Ni1-O1 ^{#1}	86.93(8)	O2 ^{#2} -Ni1-N4 ^{#4}	92.75(6)
O1-Ni1-O2 ^{#2}	176.85(5)	O2 ^{#3} -Ni1-N4 ^{#4}	87.80(6)
O1 ^{#1} -Ni1-O2 ^{#2}	90.18(6)	O1-Ni1-N4 ^{#5}	93.25(6)
O1-Ni1-O2 ^{#3}	90.18(6)	O1 ^{#1} -Ni1-N4 ^{#5}	86.17(6)
O11-Ni1-O2 ^{#3}	176.85(5)	C3-O2-Ni1 ^{#3}	132.24(12)
O22-Ni1-O2 ^{#3}	92.75(8)	C3-O1-Ni1	128.90(13)
O1-Ni1-N4 ^{#4}	86.17(6)	C1-N4-Ni1 ^{#4}	129.54(15)
O1 ^{#1} -Ni1-N4 ^{#4}	93.25(6)	N3-N4-Ni1 ^{#4}	124.51(14)

Symmetry codes: #1, 1-x, +y, 3/2-z; #2, +x, -y, -1/2+z; #3, 1-x, -y, 2-z; #4, 3/2-x, 1/2-y, 2-z; #5, -1/2+x, 1/2-y, -1/2+z.

Table S4. The θ Angles in complexes **1,2** and iso-**1,2**.

Compound	$\theta_{1-1} /$ °	$\theta_{1-2} /$ °	$\theta_{1-3} /$ °	$\theta_{1-4} /$ °	$\theta_2 /$ °	$\theta_3 /$ °	$\theta_4 /$ °	$\theta_5 /$ °	mode
1	87.54(5)	88.62(4)	90.99(4)	92.86(4)	179.45(6)	112.80(11)	84.486(32)	79.307(53)	I
Iso-1	81.90(5)	83.21(5)	96.79(5)	98.10(6)	180.00(1)	113.1(2)	73.202(29)	0.000(85)	II
2	86.01(5)	88.40(4)	92.23(4)	93.34(5)	179.11(6)	112.93(11)	86.176(34)	80.667(45)	I
Iso-2	82.48(10)	83.81(9)	96.19(9)	97.52(10)	180.000(1)	113.1(2)	76.814(46)	0.000(152)	II

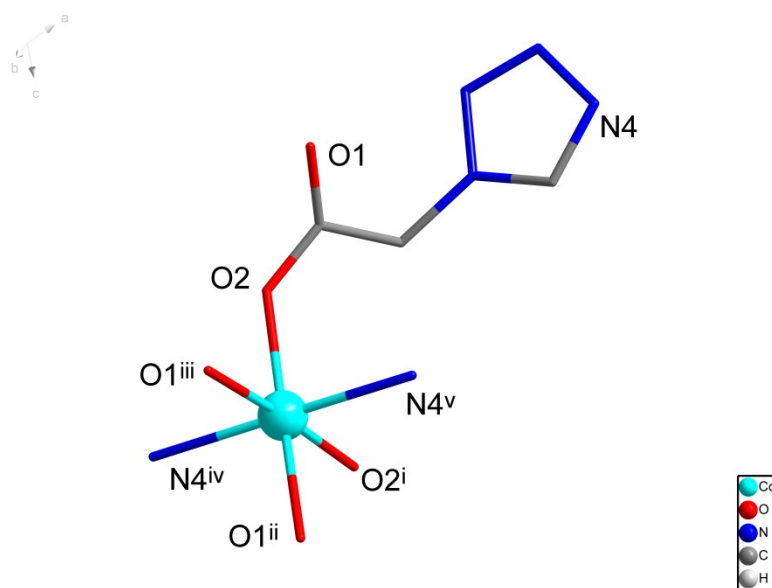


Figure S1. Coordination environment of the Co(II) ion in complex 2 with hydrogen atoms omitted for clarity. Symmetry codes: (i) $-x+1, y, -z+3/2$; (ii) $x, -y, z+1/2$; (iii) $-x+1, -y, -z+1$; (iv) $-x+3/2, y+1/2, -z+3/2$; (v) $x-1/2, y+1/2, z$; (vi) $x+1/2, y-1/2, z$.

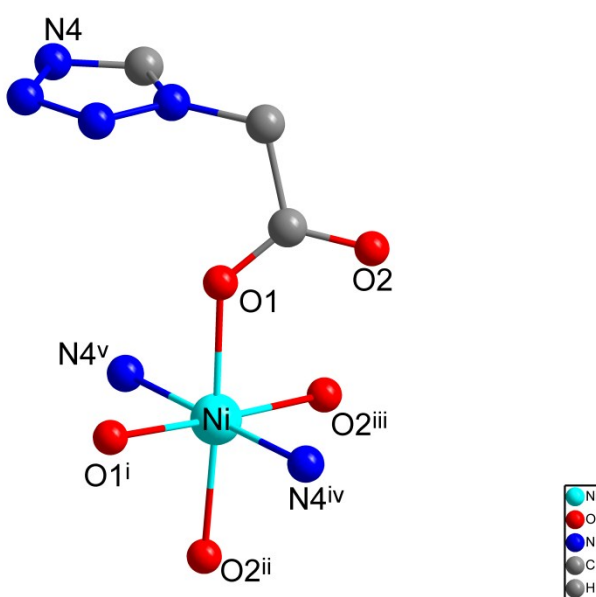


Figure S2. Coordination environment of the Ni(II) ion in complex 3 with hydrogen atoms omitted for clarity. Symmetry codes: (i) $-x+1, y, -z+3/2$;

(ii) $x, -y, z-1/2$; (iii) $-x+1, -y, -z+2$; (iv) $x-1/2, -y+1/2, z-1/2$; (v) $-x+3/2, -y+1/2, -z+2$.

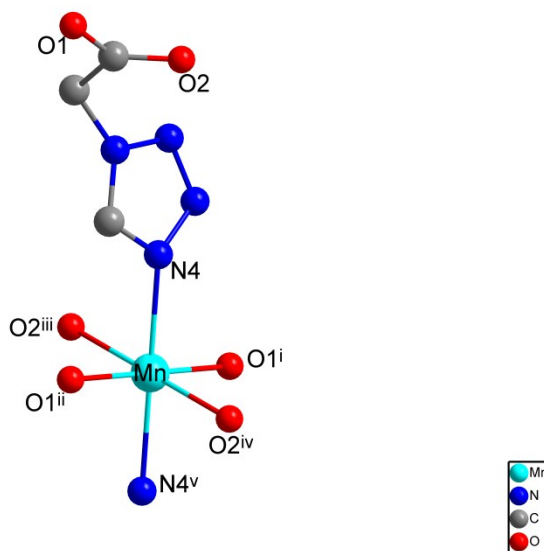


Figure S3. Coordination environment of the Mn (II) ion in iso-1 with hydrogen atoms omitted for clarity. Symmetry codes: (i) $x+1/2, -y+5/2, z-1/2$; (ii) $-x-1/2, y-1/2, -z+5/2$; (iii) $x-1/2, -y+5/2, z-1/2$; (iv) $-x+1/2, y-1/2, -z+5/2$; (v) $-x, -y+2, -z+2$.

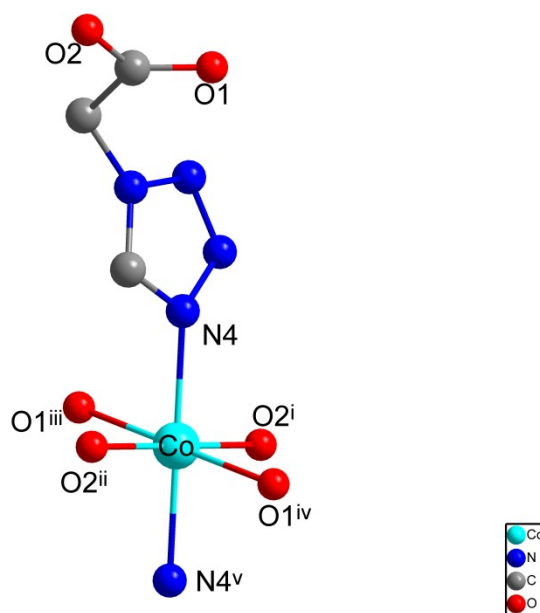


Figure S4. Coordination environment of the Co (II) ion in iso-1 with hydrogen atoms omitted for clarity. Symmetry codes: (i) $x+1/2, -y+5/2, z-1/2$; (ii) $-x-1/2, y-1/2, -z+5/2$; (iii) $x-1/2, -y+5/2, z-1/2$; (iv) $-x+1/2, y-1/2, -z+5/2$; (v) $-x, -y+2, -z+2$.

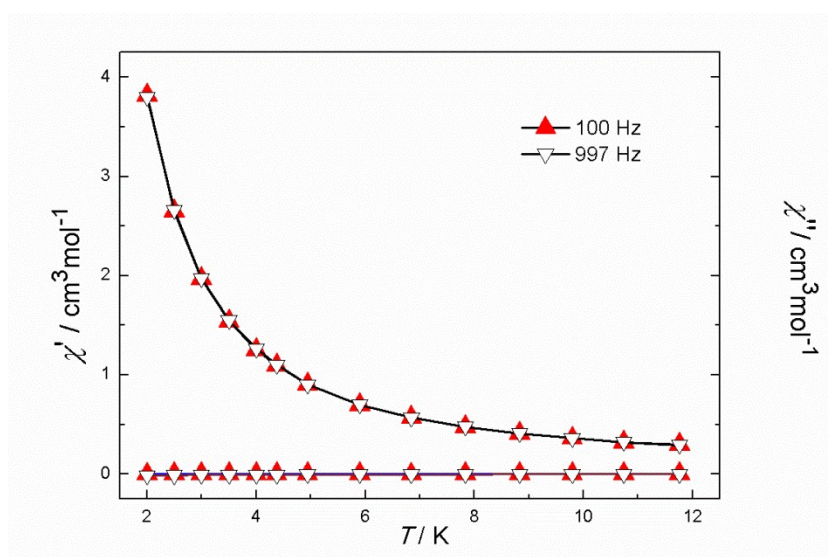


Figure S5. Temperature dependence of the in-phase (χ') and out-of-phase (χ'') ac susceptibilities at $H_{ac} = 2.5$ Oe for complex **2** measured without an applied dc field.

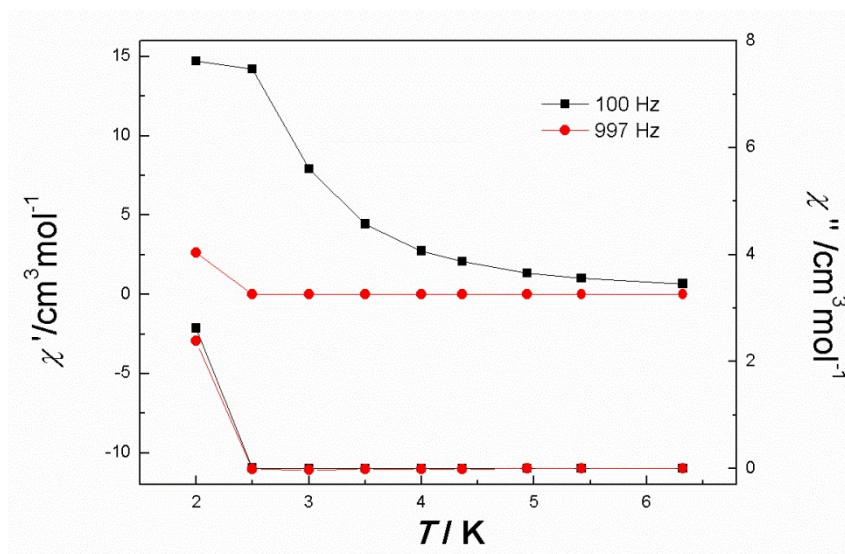


Figure S6. Temperature dependence of the in-phase (χ') and out-of-phase (χ'') ac susceptibilities at $H_{ac} = 2.5$ Oe for complex **3** measured without an applied dc field.

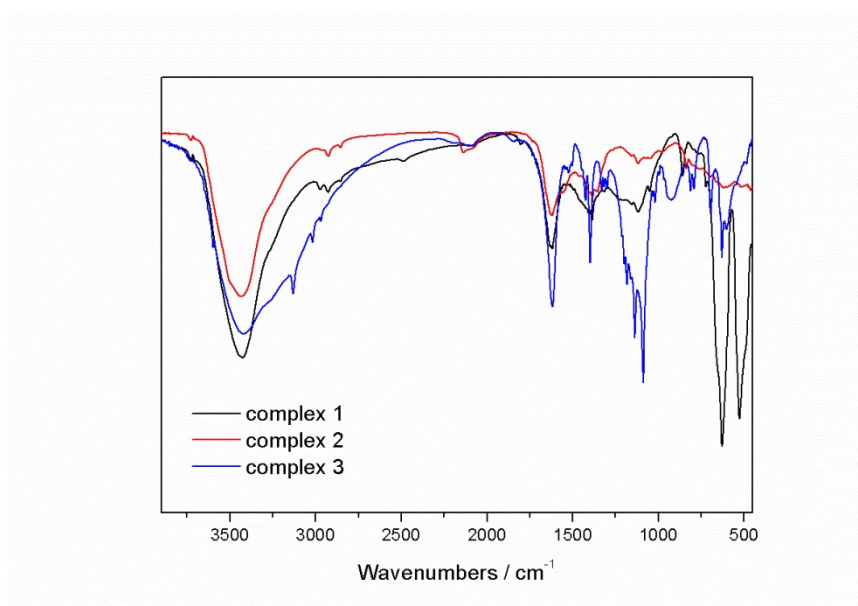


Figure S7. Infrared spectra of the complexes **1-3**.

In the spectra, characteristic bands at 3423, 2972, 2925, 2079, 1618, 1384, 1116, 857, 627 and 526 cm^{-1} for **1**, 3431, 2924, 2851, 2136, 1619, 1384, 1116, 836 and 617 cm^{-1} for **2**, 3421, 3130, 3016, 2962, 2079, 1617, 1400, 1137, 1087, 811 and 628 cm^{-1} for **3**.

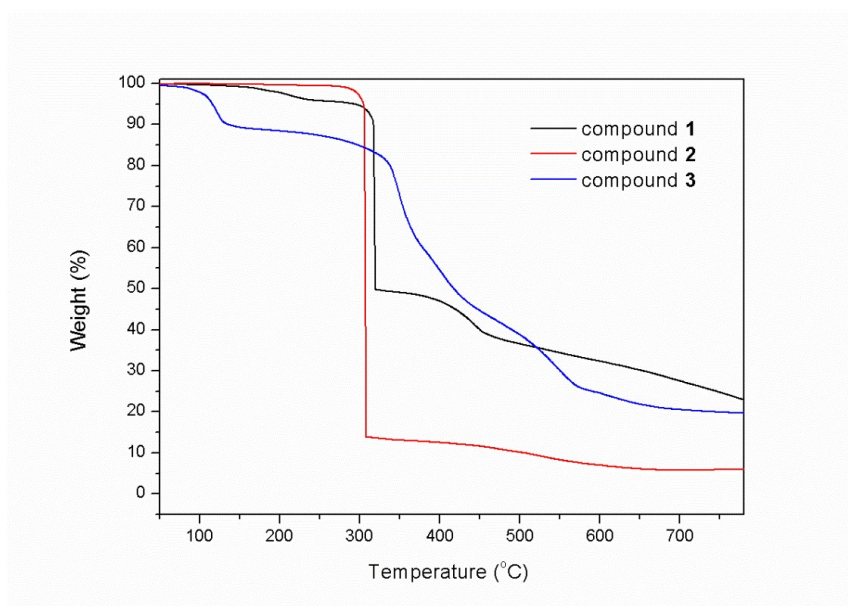


Figure S8. TGA plots of the complexes **1-3**