

Supporting Information

The study of perylene diimide-amino acids derivatives on fluorescent detecting of anions

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Section 1 ^1H NMR and ^{13}C NMR spectra of the H_2PDIAa s

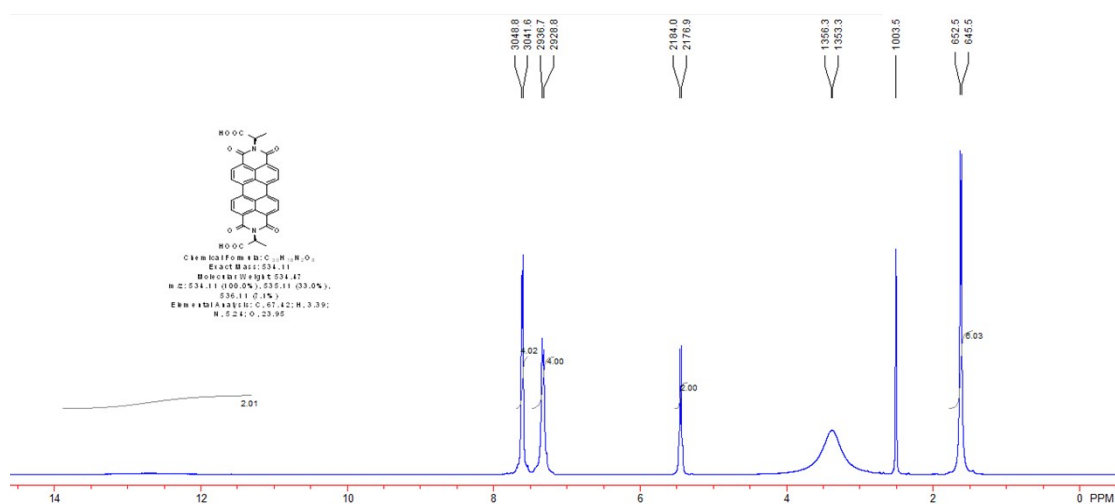


Fig.S1 Spectra of the H_2PDIAa in $\text{DMSO}-d^6$ solution.

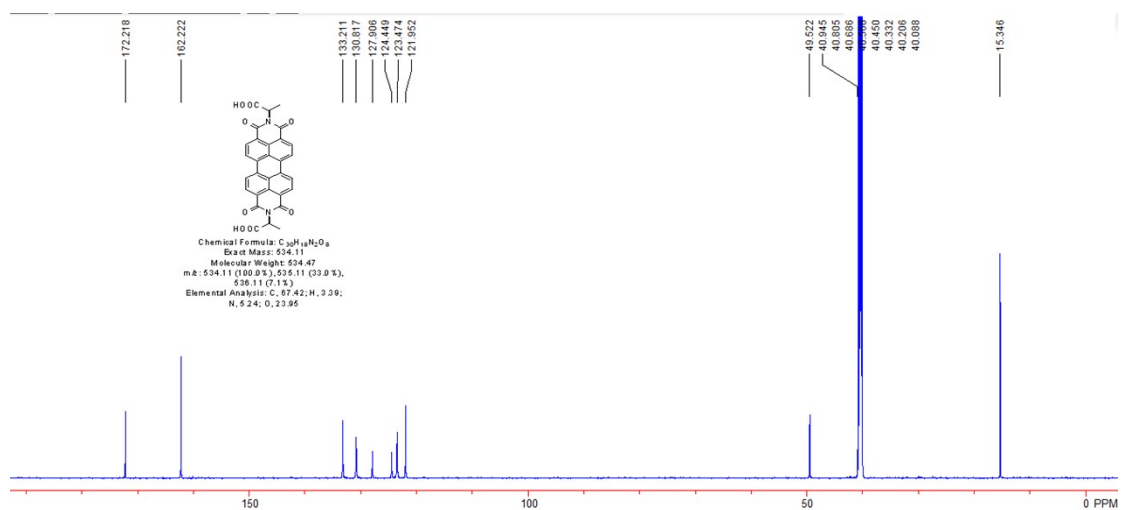


Fig.S2 ^{13}C NMR spectra of the H_2PDIAa in $\text{DMSO}-d^6$ solution.

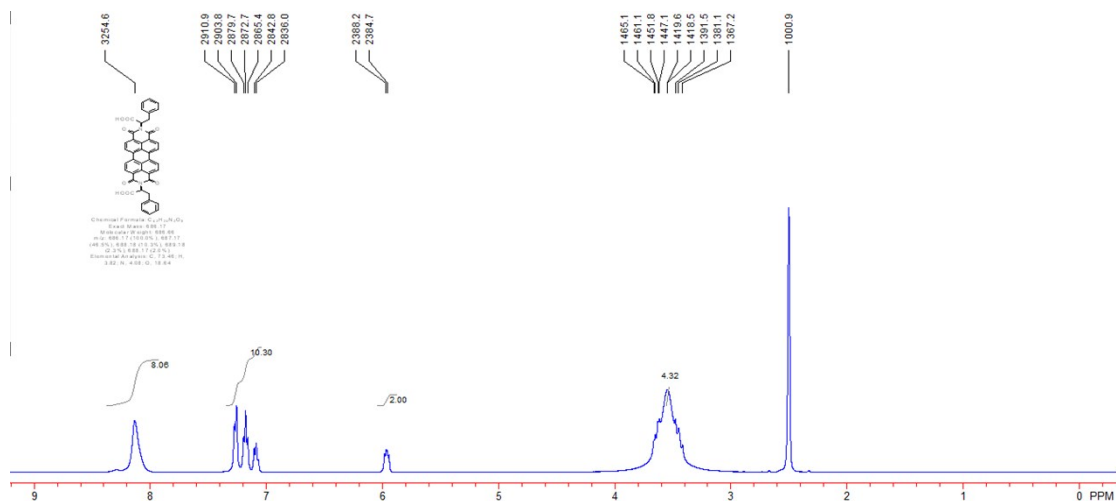


Fig.S5 1H NMR spectra of the $H_2PDIPhe$ in $DMSO-d_6$ solution.

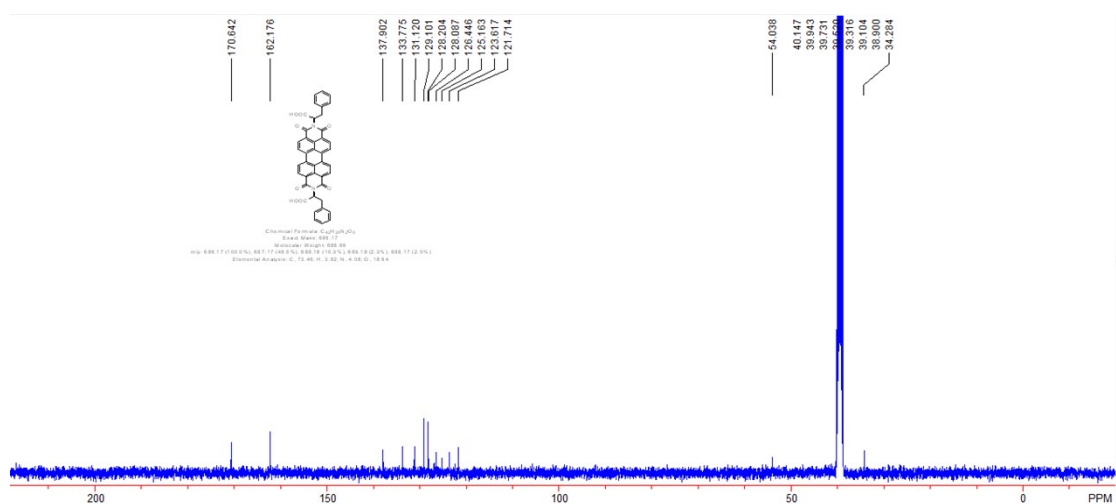


Fig.S6 ^{13}C NMR spectra of the $H_2PDIPhe$ in $DMSO-d_6$ solution.

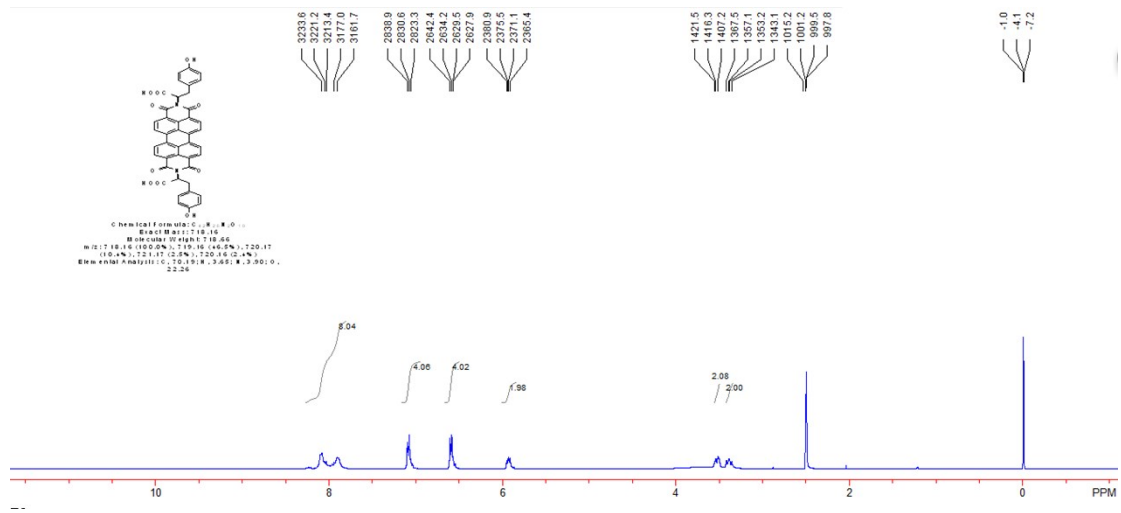


Fig.S7 ^1H NMR spectra of the H_2PDITyr in $\text{DMSO-}d^6$ solution.

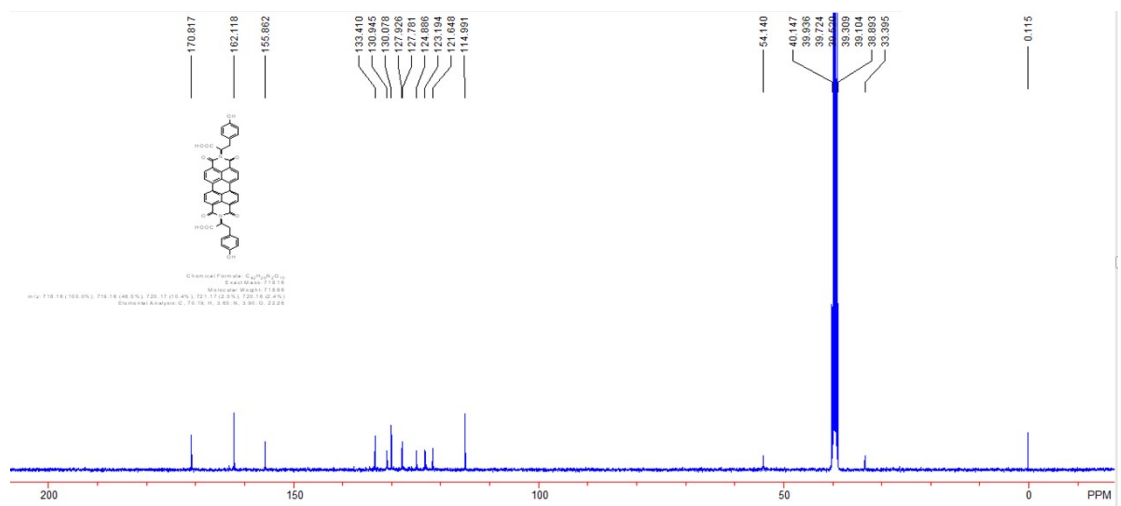


Fig.S8 ^{13}C NMR spectra of the H_2PDITyr in $\text{DMSO-}d^6$ solution.

Section 2 IR of the H₂PDIAAs

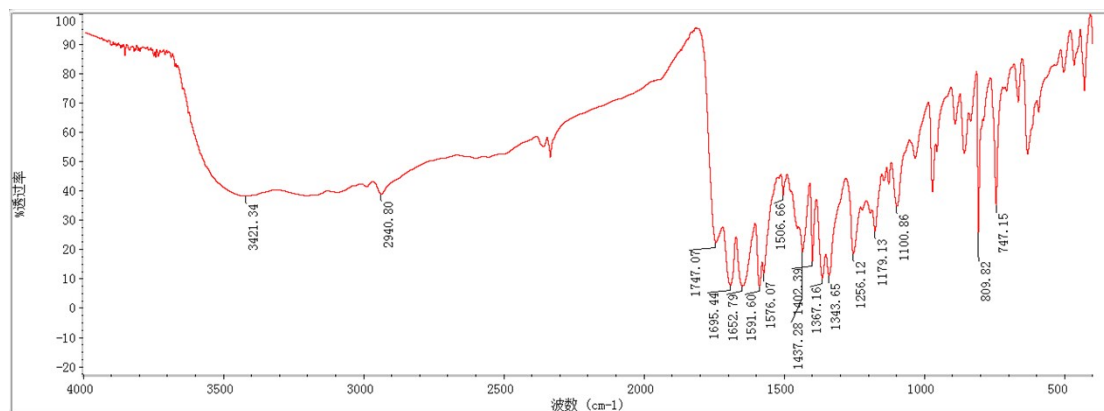


Fig.S9 IR of the H₂PDIAa.

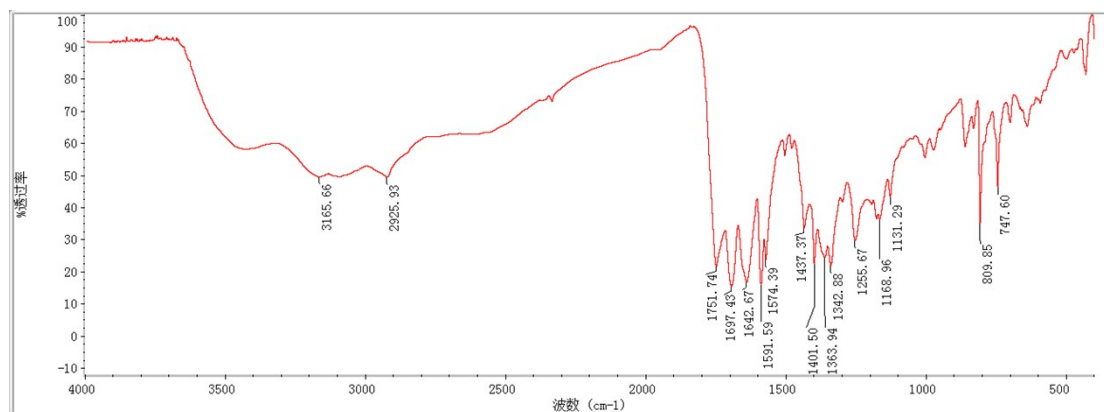


Fig.S10 IR of the H₂PDIGlu.

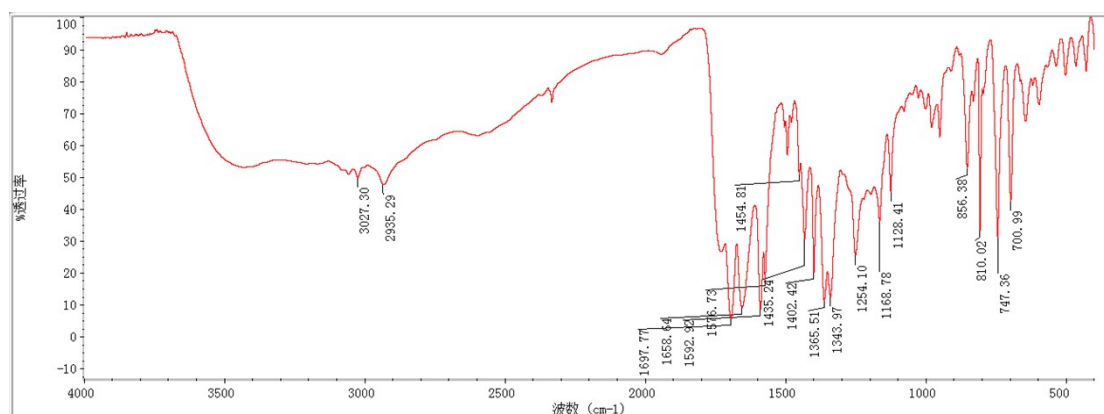


Fig.S11 IR of the H₂PDIPhe.

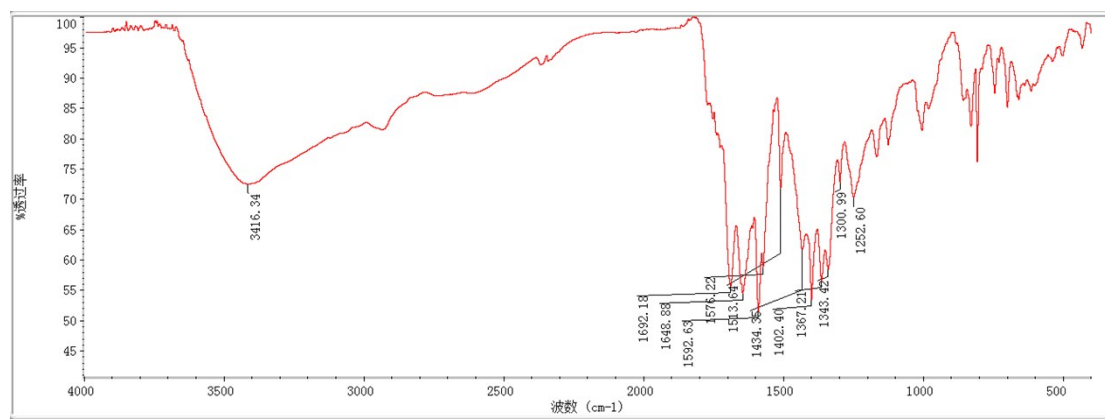


Fig.S12 IR of the H₂PDITyr.

Section 3 ESI-MS of the H₂PDIAAs

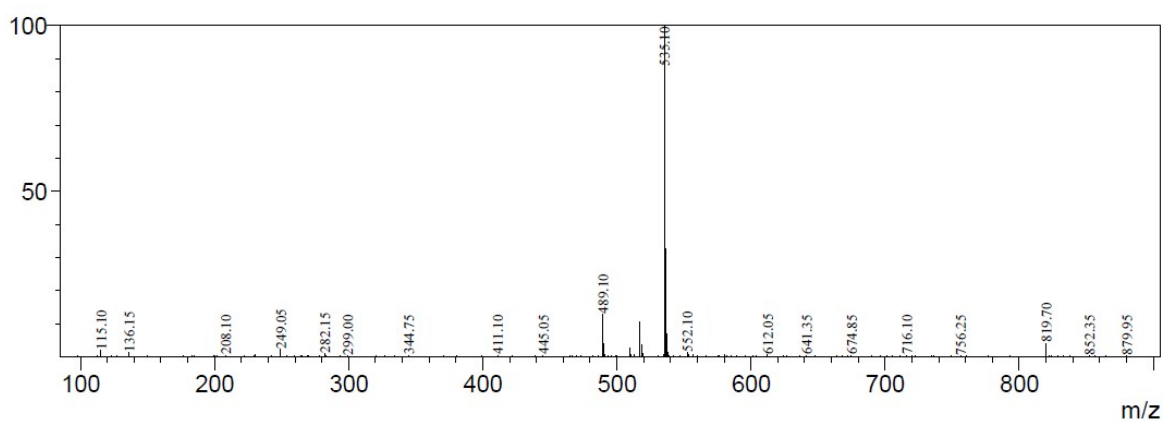


Fig.S13 ESI-MS of the H₂PDIAa.

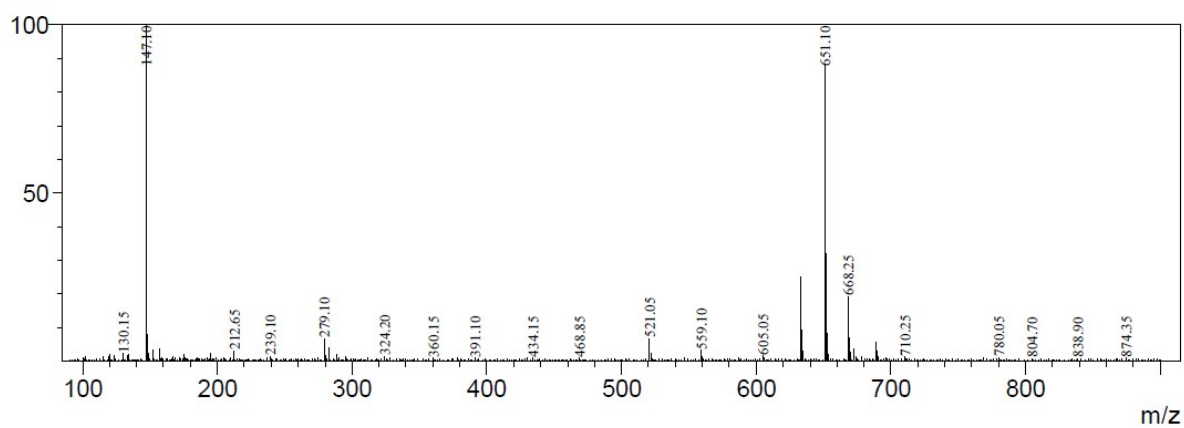


Fig.S14 ESI-MS of the H₂PDIGlu.

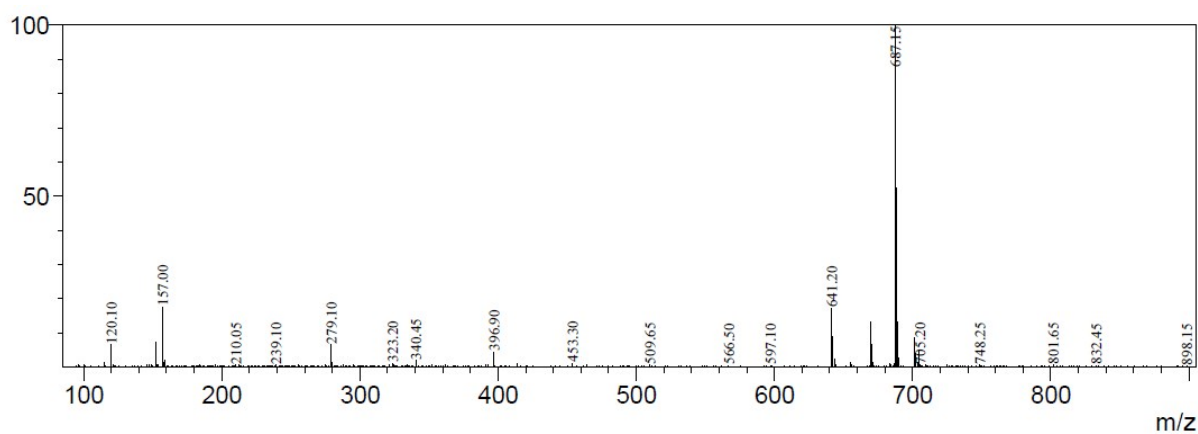


Fig.S15 ESI-MS of the H₂PDIPhe.

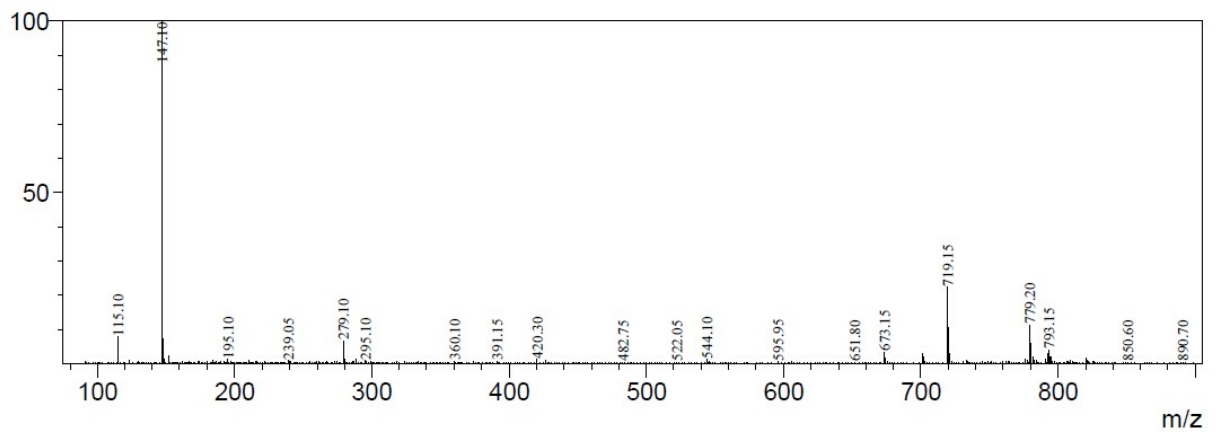


Fig.S16 ESI-MS of the H₂PDITyr

Section 4 UV-Vis absorbance of the H₂PDIAAs with anions

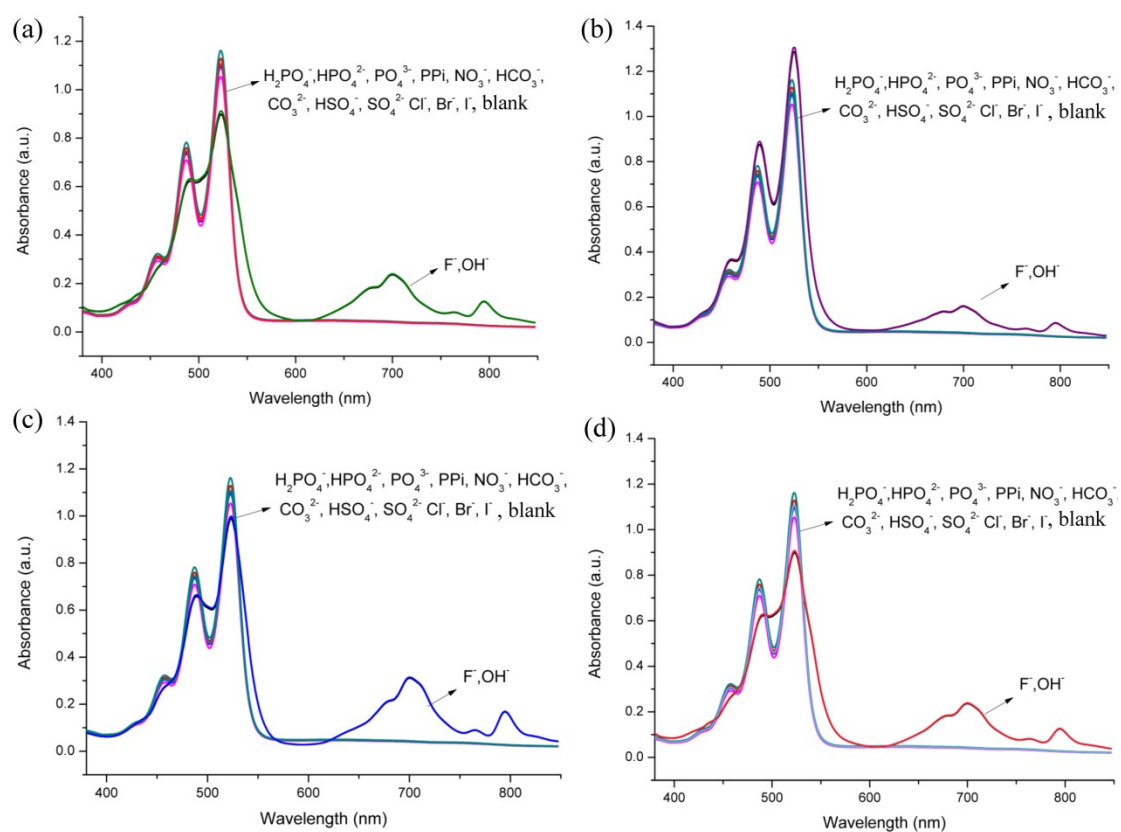


Fig.S17 UV-vis spectra of H₂PDIAAs (1 × 10⁻⁵ M) with the different anions(6 eq. each) in DMF, (a) H₂PDIAIa, (b) H₂PDIGlu, (c) H₂PDIPhe, (d) H₂PDITyr.

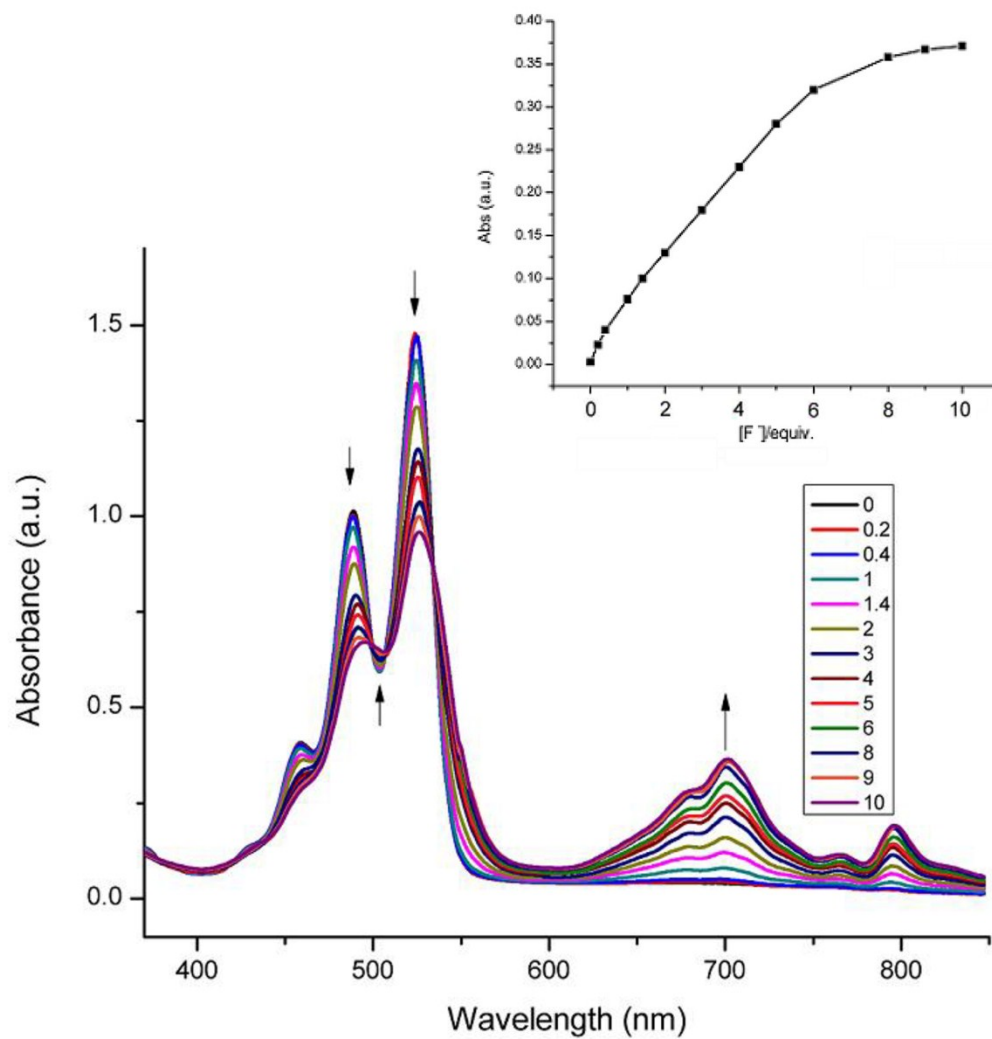


Fig.S18 Absorbance changes of $H_2PDIGlu$ ($1 \times 10^{-5} M$) in DMF on addition of 0-10 equivalents of TBAF. Inset: UV absorption intensity at 700 nm vs concentration of F^- .

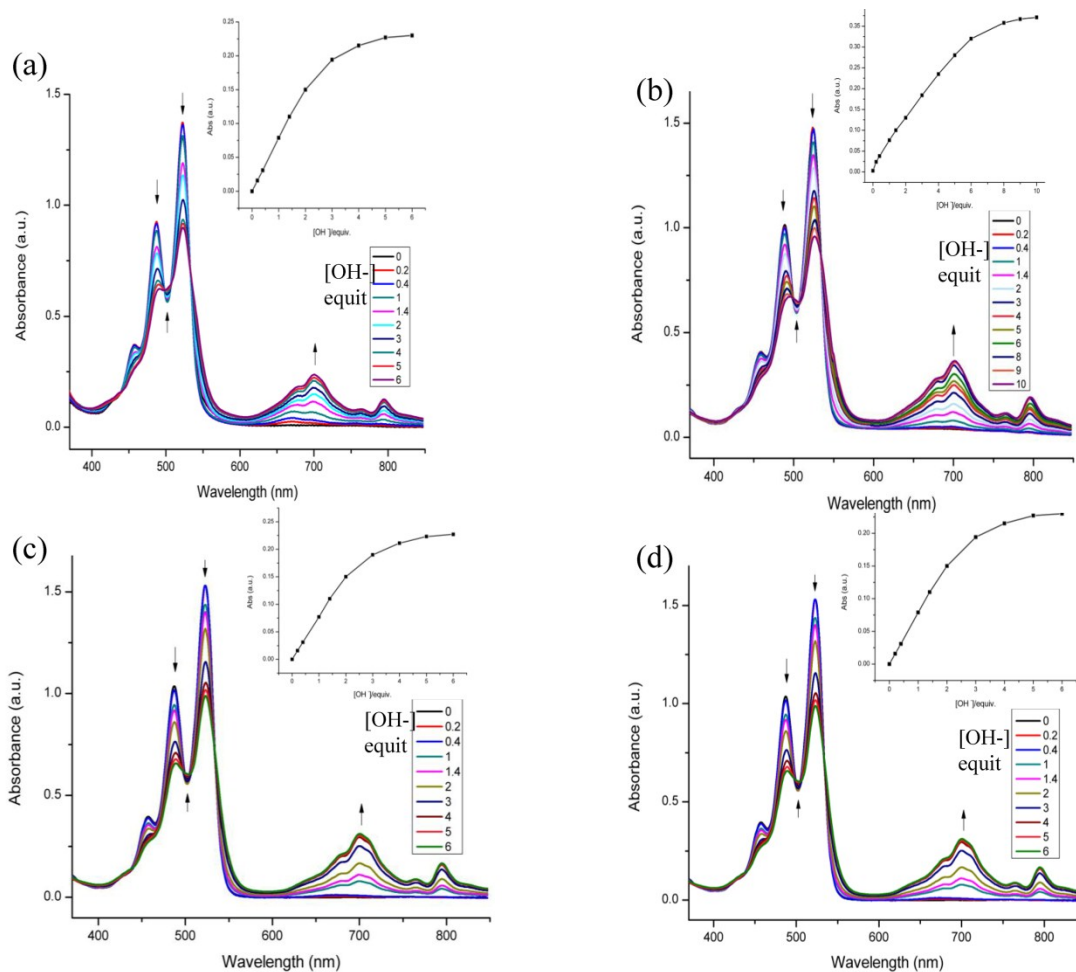


Fig.S19 UV-vis titration of H₂PDIAAs with TBOH in DMF (1 × 10⁻⁵ M), (a) H₂PDIAIa, (b) H₂PDIGlu, (c) H₂PDIPhe, (d) H₂PDITyr, Inset: UV absorbance intensity at 700 nm vs concentration of OH⁻ ion.

Section 5 Fluorescence of the H₂PDIAAs with anions

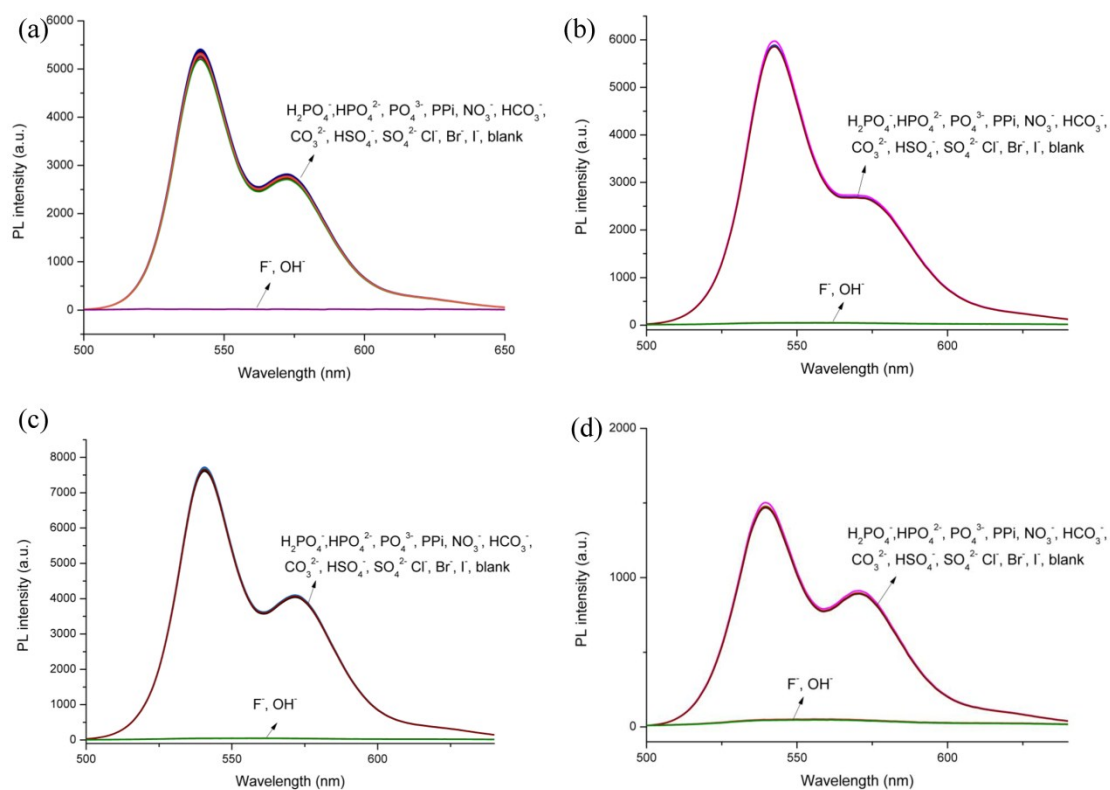


Fig.S20 Fluorescence spectra of H₂PDIAAs (1 × 10⁻⁵ M) with the different anions(6 eq, each) in DMF (λ_{ex} = 525 nm), (a) H₂PDIAIa, (b) H₂PDIGlu, (c) H₂PDIPhe, (d) H₂PDITyr.

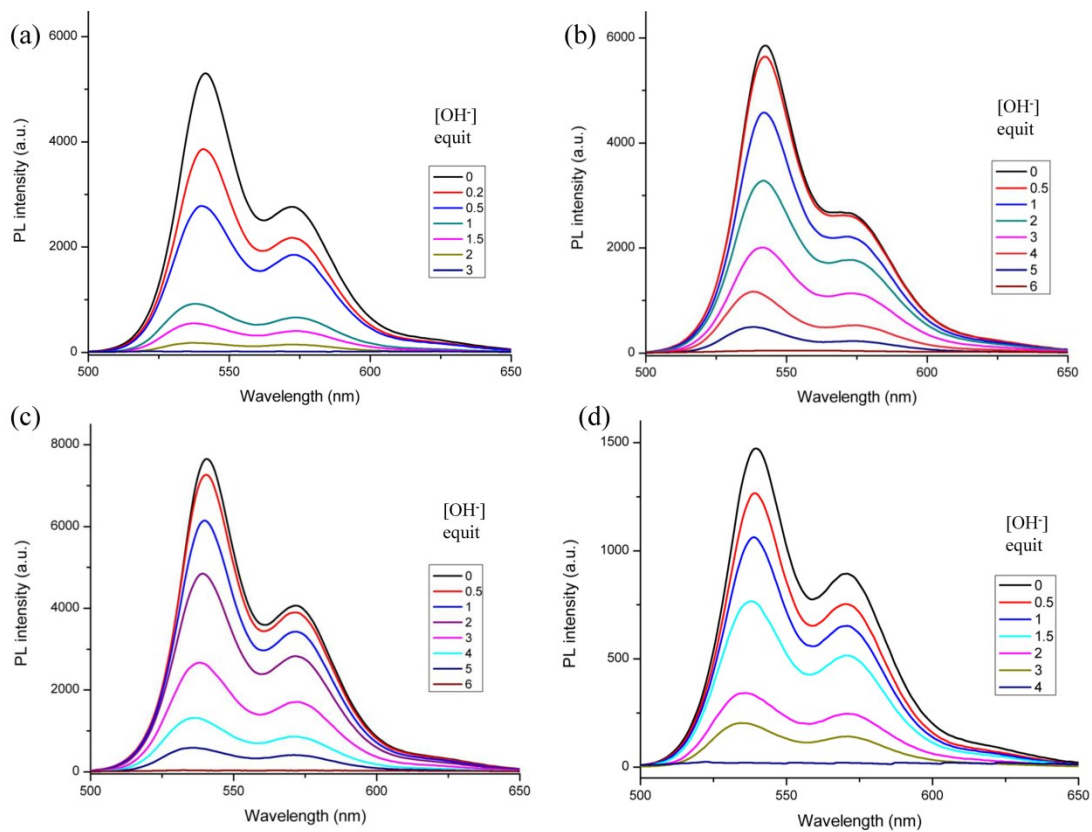


Fig.S21 Fluorescence emission of H₂PDIAAs with TBAOH titration in DMF (1 × 10⁻⁵ M) (λ_{ex} = 525 nm), (a) H₂PDIAIa, (b) H₂PDIGlu, (c) H₂PDIPhe, (d) H₂PDITyr.

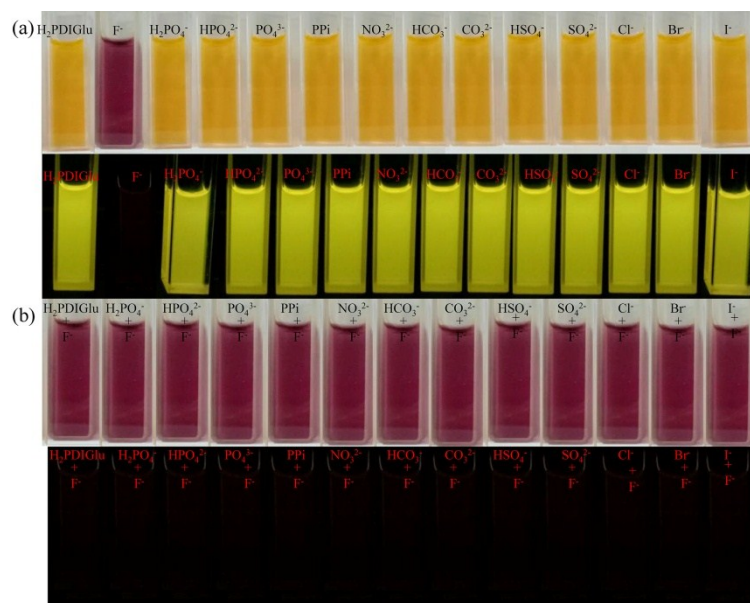


Fig.S22 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for H₂PDIGlu with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of H₂PDIGlu-F⁻ ion with various ions (6 eq, each).

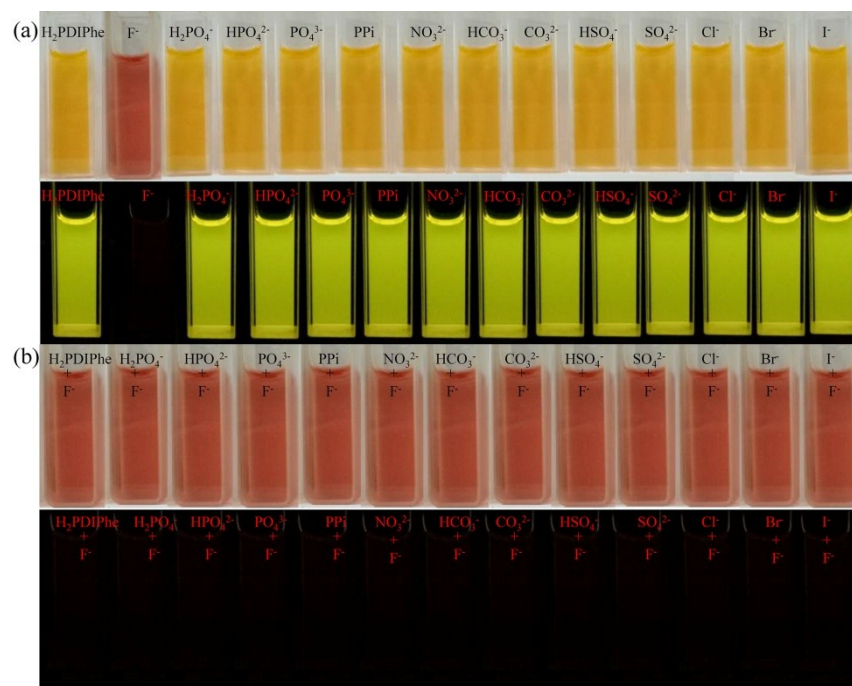


Fig.S23 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for H₂PDIPhe with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of H₂PDIPhe-F⁻ ion with various ions (6 eq, each).

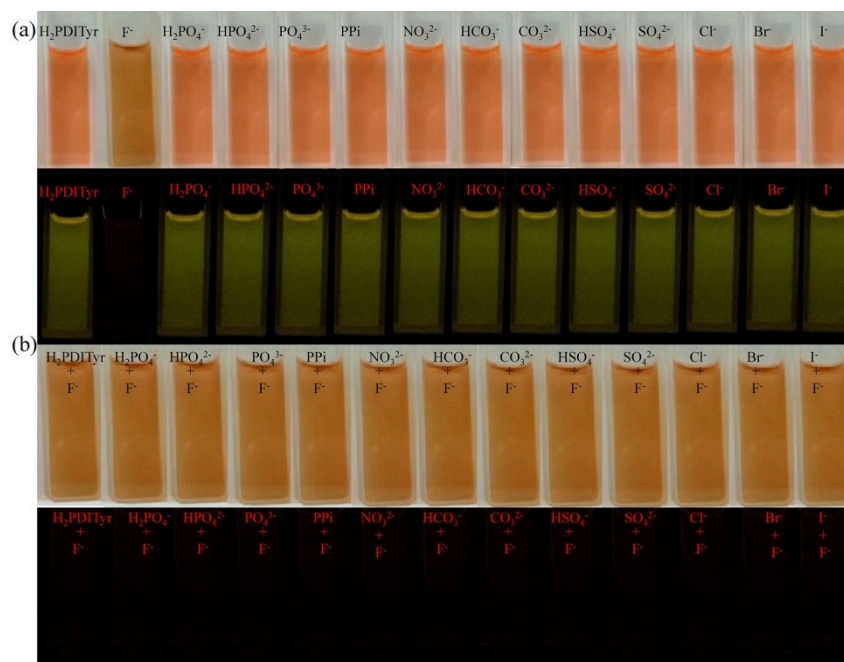


Fig.S24 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for $H_2PDITyr$ with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of $H_2PDITyr-F^-$ ion with various ions (6 eq, each).

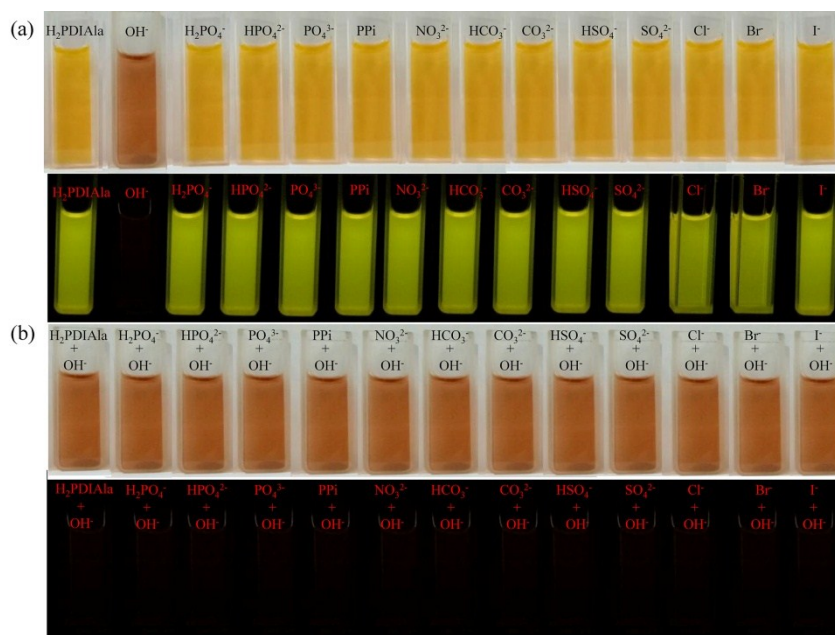


Fig.S25 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for H₂PDIALa with various anions(6 eq, each), The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of H₂PDIALa-OH ion with various ions (6 eq, each).

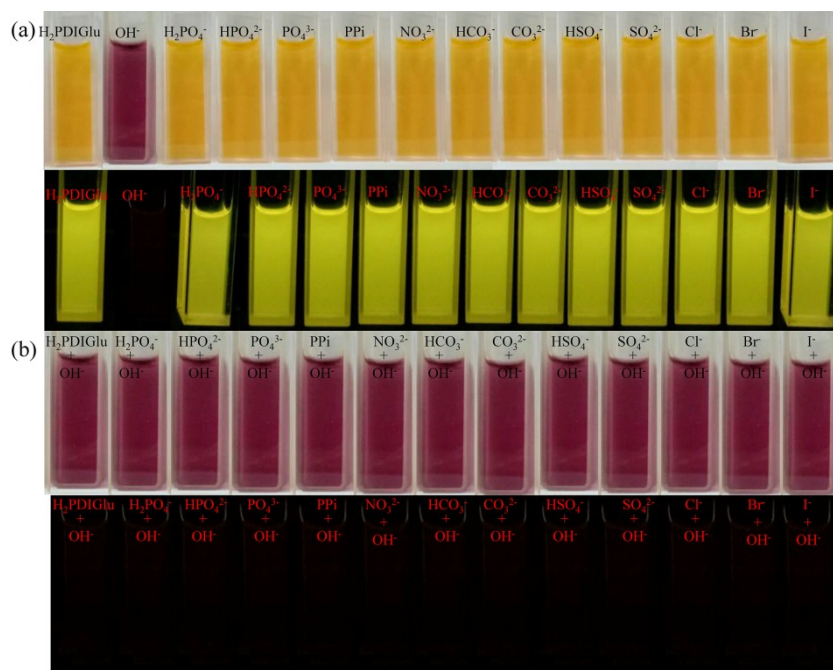


Fig.S26 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for H₂PDIGlu with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of H₂PDIGlu-OH⁻ ion with various ions (6 eq, each).

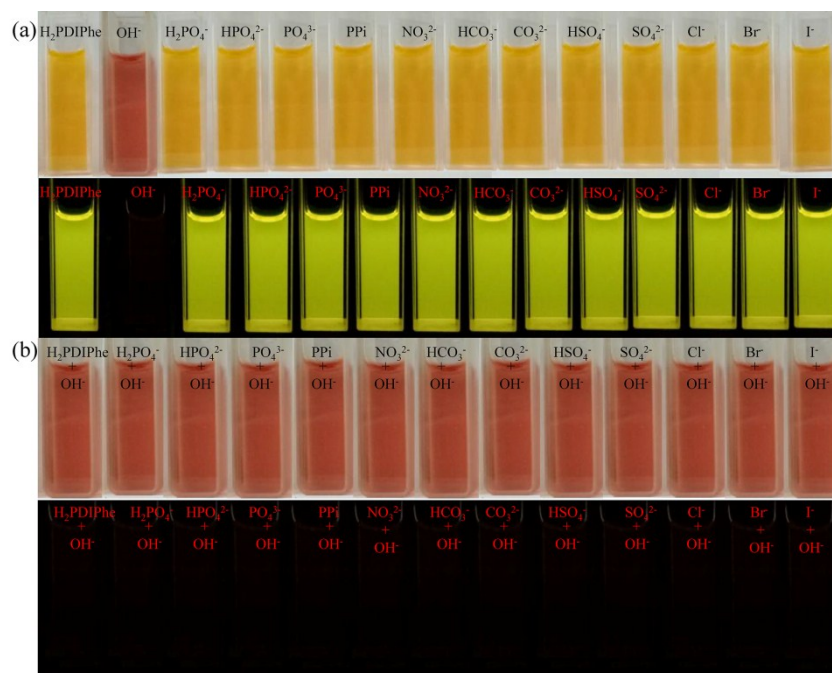


Fig.S27 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365 \text{ nm}$) for H₂PDIPhe with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365 \text{ nm}$) of H₂PDIPhe-OH⁻ ion with various ions (6 eq, each).

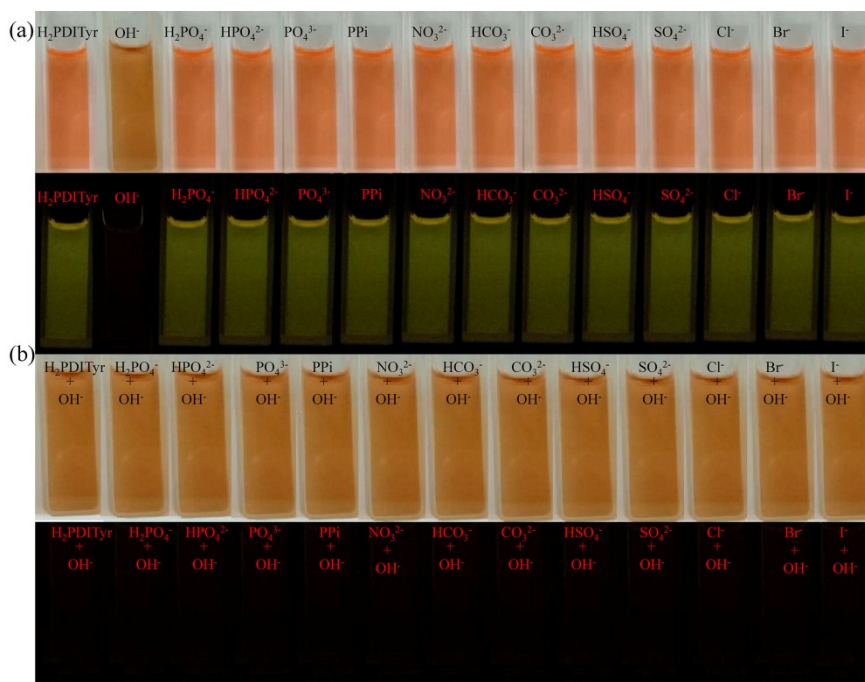


Fig.S28 (a) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) for H₂PDITyr with various anions(6 eq, each), (b) The photographs of colorimetric identification by naked eye and fluorescence emission ($\lambda = 365$ nm) of H₂PDITyr-OH⁻ ion with various ions (6 eq, each).

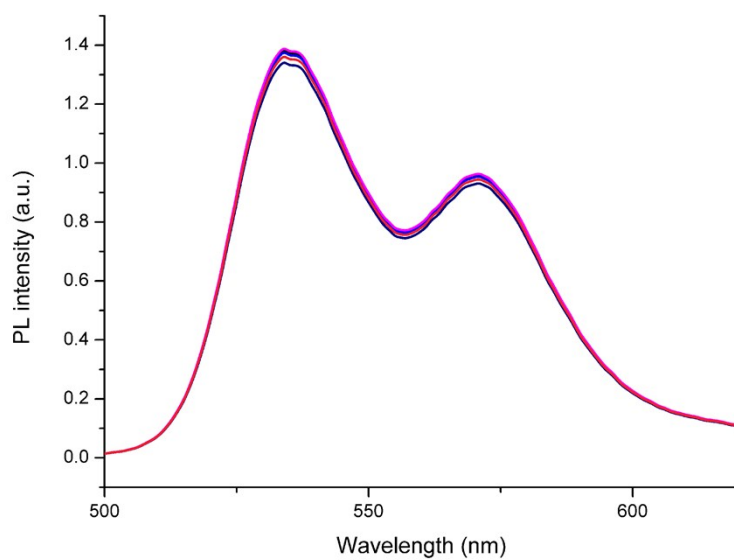


Fig.S29 Fluorescence spectra of H₂PDIAIa (1×10^{-5} M) added the TBAF, and the mixture solution with the different anions(6 eq, each) in DMF ($\lambda_{\text{ex}} = 525$ nm).

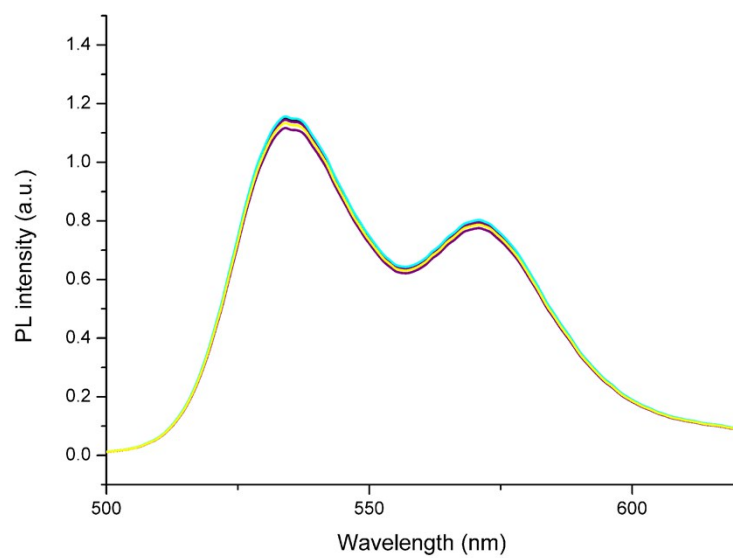


Fig.S30 Fluorescence spectra of H₂PDIAla (1×10^{-5} M) added the TBAOH, and the mixture solution with the different anions(6 eq, each) in DMF ($\lambda_{\text{ex}} = 525$ nm).

Section 6. Fluorescence lifetime decay of the H₂PDIAAs

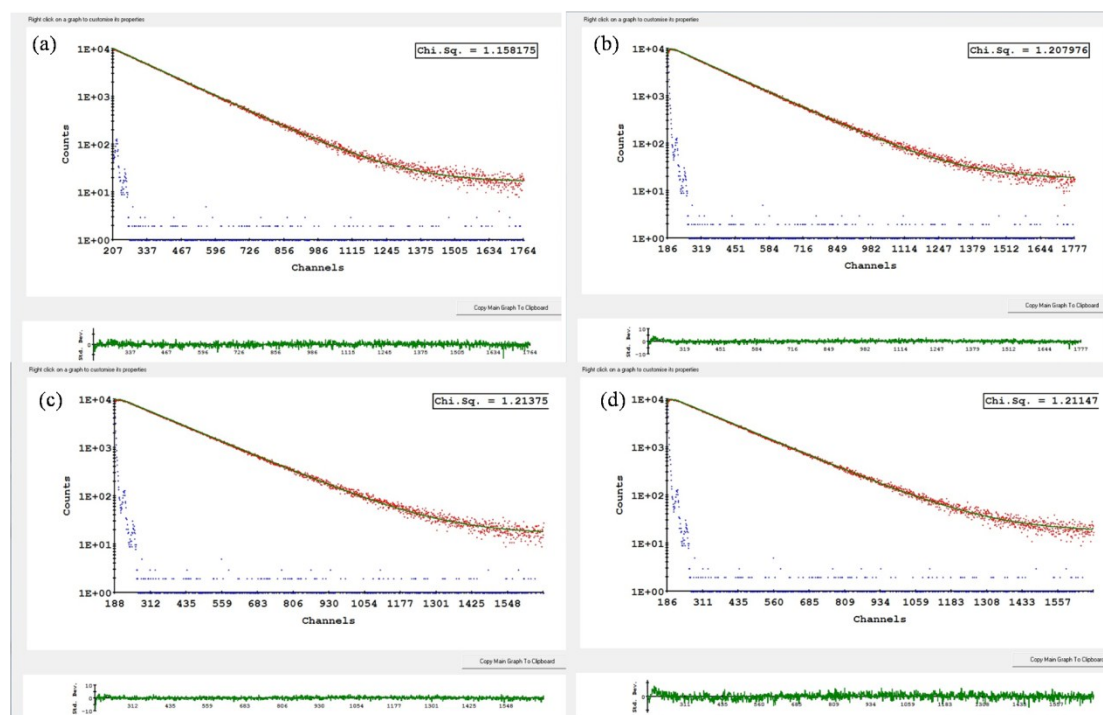


Fig.S31 Fluorescence lifetime decay of H₂PDIAAla (1×10^{-5} M) in DMF. (b) Fluorescence lifetime decay of H₂PDIGlu. (c) Fluorescence lifetime decay of H₂PDIPhe. (d) Fluorescence lifetime decay of H₂PDITyr.

Section 7. $^1\text{H-NMR}$ of the **H₂PDIAIa** with TBAOH

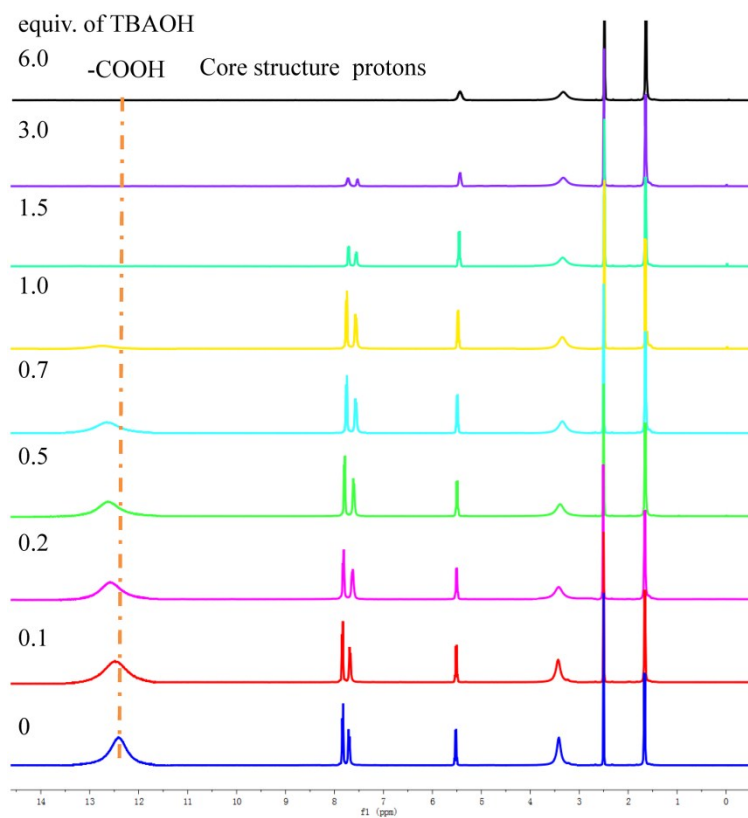


Fig.S32 The $^1\text{H-NMR}$ titrations of **H₂PDIAIa** with TBAOH in $\text{DMSO-}d_6$.