

Supporting Information

Spontaneously Grown Ni(OH)₂ on Iron Oxide nanoparticles with Enhanced Energy Storage Performance for Electrode of Asymmetric Supercapacitors

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KEYWORDS

Ni(OH)₂; Asymmetric Supercapacitors; Electrochemical energy; Intercalates; Iron
oxides; Specific capacitance

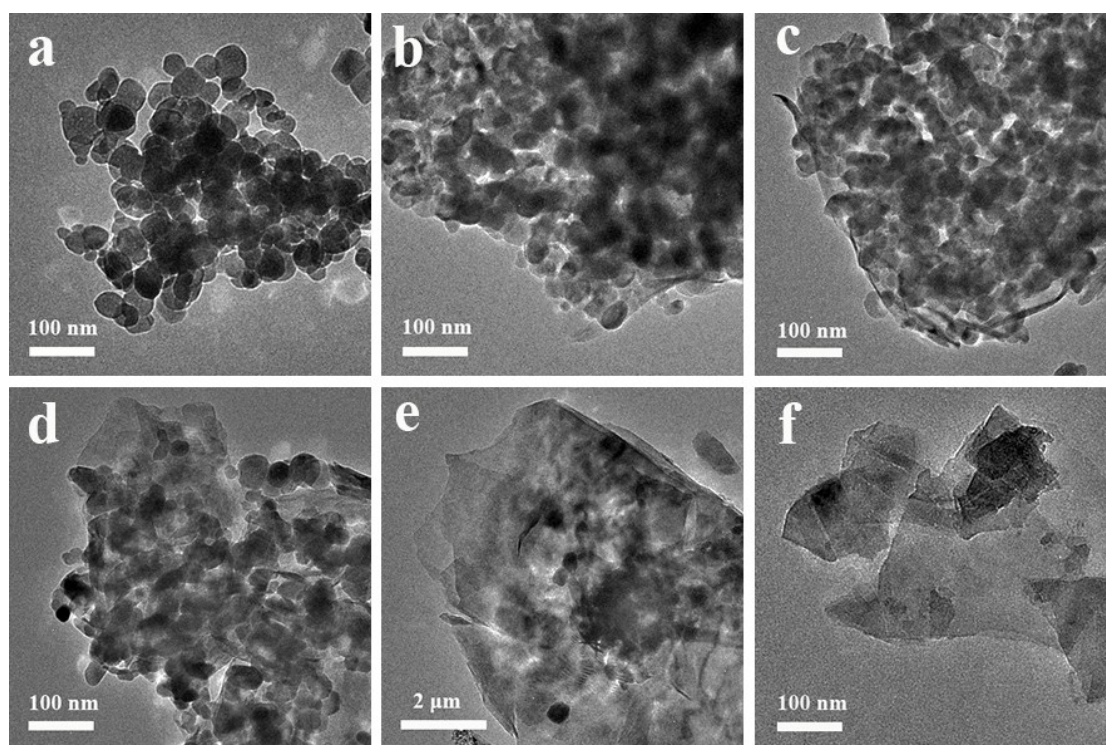


Figure S1. The TEM images of (a) Fe_2O_3 and (b) 5-NF, (c) 10-NF, (d) 20-NF, (e) 50-NF, and (f) $\text{Ni}(\text{OH})_2$.

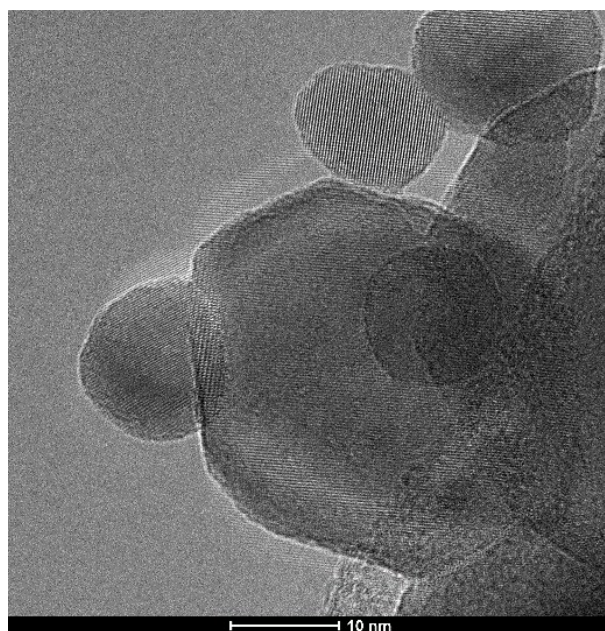


Figure S2. The HRTEM image of 10-NF.

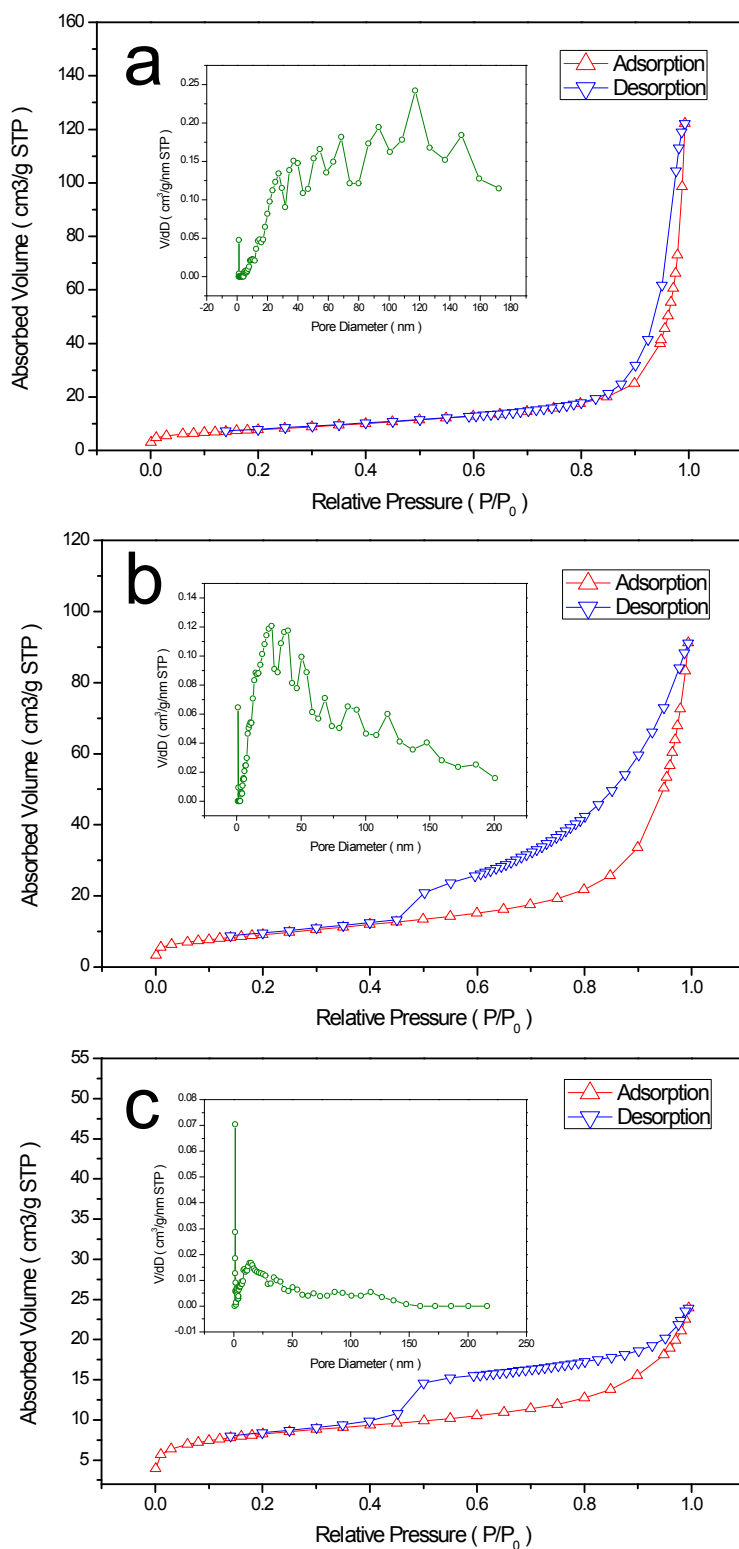


Figure S3. N_2 absorption and desorption isotherm plots of (a) $\alpha\text{-Fe}_2\text{O}_3$, (b) 50-NF, and (c) Ni(OH)_2 , the insets are the pore size distribution patterns for the corresponding samples.

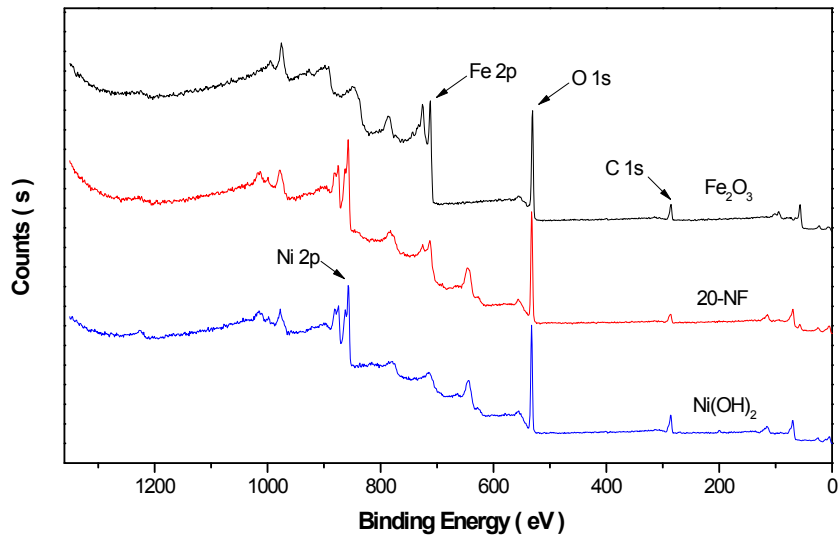


Figure S4. XPS survey spectra of Fe_2O_3 , 20-NF, $\text{Ni}(\text{OH})_2$

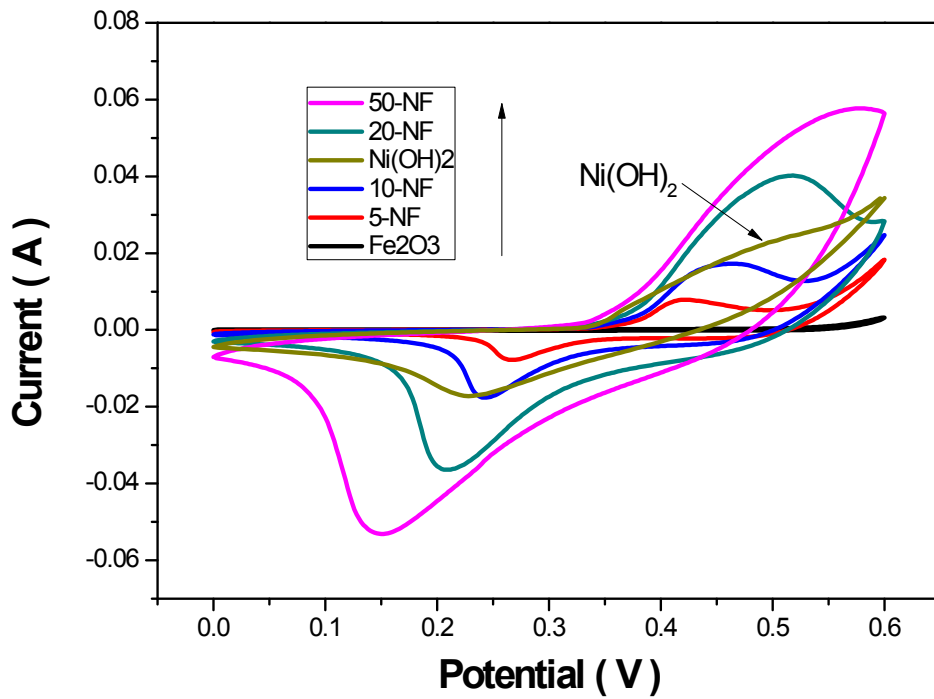


Figure S5. CV curves of $\alpha\text{-Fe}_2\text{O}_3$, 5-NF, 10-NF, 20-NF, 50-NF and $\text{Ni}(\text{OH})_2$ at the scan rate was 20 mV s^{-1} ;

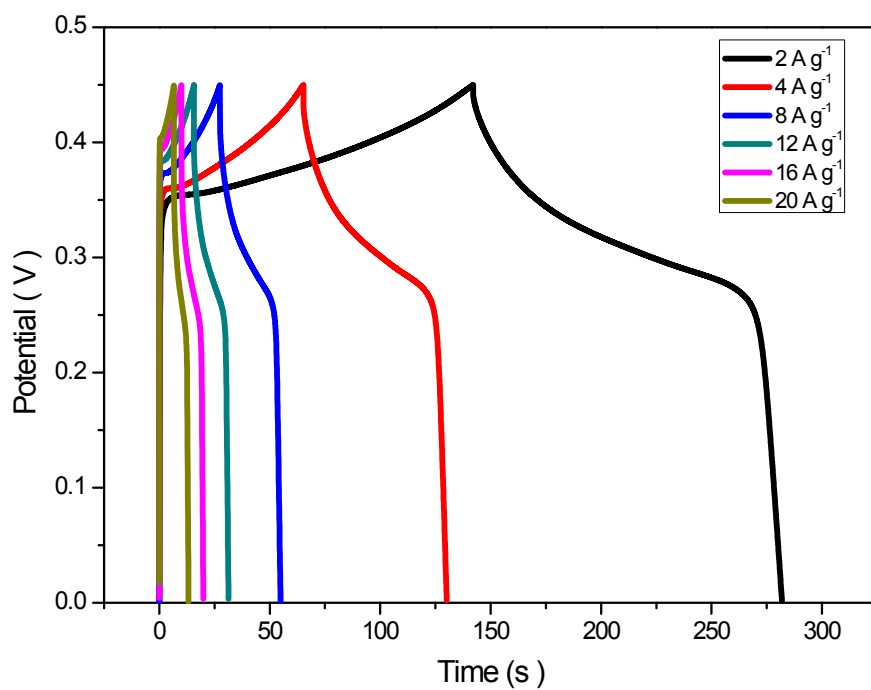


Figure S6. The GCD curves of 50-NF at different current density

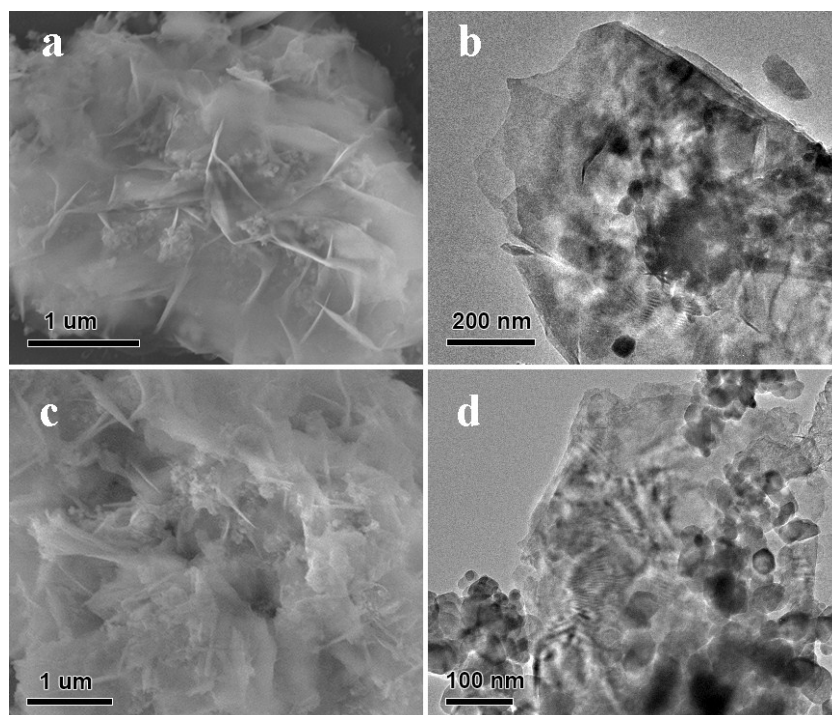


Figure S7. The SEM images of 50-NF (a) and (b) The TEM images of 50-NF, (c) The SEM images of 50-NF after 5000 charge/discharge cycle, (d) The TEM images of 50-NF after 5000 charge/discharge cycle

Table S1. Comparison of electrochemical performance with other electrodes materials comprised of Fe₂O₃- Ni(OH)₂.

Sample	Experiment method	Electrochemical test	Electrochemical performance	Capacitance retention (cycle numbers)	Ref.
α -Fe ₂ O ₃ @Ni(OH) ₂	hydrothermal 80°C	CV:0-0.6V ; GCD: 0-0.6V	356 F/g (16 A/g)	93.3% (500)	S1
NiO// α -Fe ₂ O ₃	hydrothermal 120°C	Fe ₂ O ₃ :-0.8-0V; NiO:0-0.8V; ASC: 0 - 1.25 V	1.3 F/cm ² (4 mA/cm ²); ASCs: energy densities 12.4 Wh /kg, power 951 W/kg	85% (10,000)	S2
α -Fe ₂ O ₃ @NiO	hydrothermal 120°C	CV:-0.2- 0.8V;GCD :-0.2- 0.8 V	557 mF/cm ² (1mA/cm ²)	96.2% (3000)	S3
Ni(OH) ₂ - Fe ₂ O ₃	solvothermal 140°C	CV:0-0.45V; GCD: 0-0.4V	1745.33 F/g (2 A /g)	84.28% (3000)	S4
Ni(OH) ₂ - Fe ₂ O ₃	hrdrothermal 85°C	CV:0-0.6V; GCD: 0-0.6V	390 F/g (54.6 A/g)	85.7%(5000)	S5
Ni(OH) ₂ - α -Fe ₂ O ₃	evaporation solvent method 60°C	CV:0-0.6V; GCD: 0-0.45V; ASCs : -1-1.4V	1107 F/g (20 mV/s); ASCs:energy densities 31.6 Wh /kg, power 474 W/kg	ASCs:89.6%(5000)	This work

- S1 H. Jiang, H. Ma, Y. Jin, L. Wang, F. Gao and Q. Lu, *Scientific Reports*, 2016, **6**.
- S2 S. Zhang, B. Yin, Z. Wang and F. Peter, *Chemical Engineering Journal*, 2016, **306**, 193-203.
- S3 Y. Jiao, Y. Liu, B. Yin, S. Zhang, F. Qu and X. Wu, *Nano Energy*, 2014, **10**, 90-98.
- S4 Z. Li, W. Zhang, Y. Su, Z. Li and J. Groeper, *Nanotechnology*, 2017, **28**, 045603.
- S5 W. Tian, X. Wang, C. Zhi, T. Zhai, D. Liu, C. Zhang, D. Golberg and Y. Bando, *Nano Energy*, 2013, **2**, 754-763.