

Electronic supplementary information

Theoretical insights into the structural, relative stable, electronic, and gas sensing properties of Pb_nAu_n ($n=2-12$) clusters: a DFT study

Gaofeng Li,^{bc} Xiumin Chen,^{*abc} Zhiqiang Zhou,^{abc} Fei Wang,^{bc} Hongwei Yang,^{bc} Jia Yang,^{bc} Baoqiang Xu,^{bc}

Bin Yang,^{bc} and Dachun Liu^{bc}

^aState Key Laboratory of Complex Nonferrous Metal Resources Clear Utilization, Kunming University of Science and Technology, Kunming 650093, PR China

^bNational Engineering Laboratory for Vacuum Metallurgy, Kunming University of Science and Technology, Kunming 650093, PR China

^cYunnan Provincial Key Laboratory for Nonferrous Vacuum Metallurgy, Kunming University of Science and Technology, Kunming 650093, PR China

* Corresponding author.

E-mail addresses: chenxiumin9@hotmail.com (X. Chen)

Table S1 Calculated bond distances R (Å), vibrational frequencies ν (cm^{-1}), and vertical ionization potential energies VIPs (eV) of Au_2 dimer and Pb_2 dimer at different levels, and that of experimental data for Au_2 dimer and Pb_2 dimer.

System	Functional	Property	This work	Experiment
Au_2 dimer	GGA/PW91	R	2.49	2.47 ¹
	GGA/BPE		2.49	
	GGA/BLYP		2.53	
	GGA/PW91	ν	183.44	190.9 ¹
	GGA/BPE		181.92	
	GGA/BLYP		169.57	
	GGA/PW91	VIP	9.37	9.5 ²
	GGA/BPE		9.30	
	GGA/BLYP		9.19	
Pb_2 dimer	GGA/PW91	R	2.95	2.93 ³
	GGA/BPE		2.96	
	GGA/BLYP		2.99	
	GGA/PW91	ν	122.99	110 ³
	GGA/BPE		120.05	
	GGA/BLYP		117.76	
	GGA/PW91	VIP	6.55	6.2 ⁴
	GGA/BPE		6.56	
	GGA/BLYP		6.41	
PbAu cluster	GGA/PW91	R	2.67	
	GGA/BPE		2.67	
	GGA/BLYP		2.71	
	GGA/PW91	ν	149.10	
	GGA/BPE		148.99	
	GGA/BLYP		138.52	
	GGA/PW91	VIP	6.88	
	GGA/BPE		6.81	

	GGA/BLYP		6.83	
--	----------	--	------	--

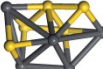
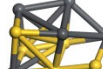
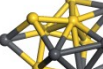
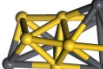
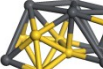
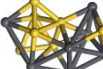
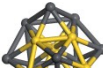
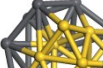
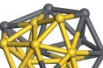
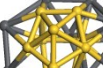
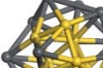
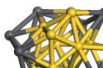
					
2a, $\Delta E=0$ eV	2b, $\Delta E=0.0001$ eV	3a, $\Delta E=0$ eV	3b, $\Delta E=0.0006$ eV	4a, $\Delta E=0$ eV	4b, $\Delta E=0.1175$ eV
					
5a, $\Delta E=0$ eV	5b, $\Delta E=0.0001$ eV	6a, $\Delta E=0$ eV	6b, $\Delta E=0.2106$ eV	7a, $\Delta E=0$ eV	7b, $\Delta E=0.0016$ eV
					
8a, $\Delta E=0$ eV	8b, $\Delta E=0.5032$ eV	9a, $\Delta E=0$ eV	9b, $\Delta E=0.0231$ eV	10a, $\Delta E=0$ eV	10b, $\Delta E=0.1173$ eV
					
11a, $\Delta E=0$ eV	11b, $\Delta E=0.0973$ eV	12a, $\Delta E=0$ eV	12b, $\Delta E=0.0286$ eV		

Figure S1 Low energy structures of Pb_nAu_n ($n=2-12$) clusters. Configuration a represents the lowest energy structure, and configuration b stand for the second lowest energy structure of the the corresponding Pb_nAu_n ($n=2-12$) clusters. The energy differences represent the energy differences between the lowest energy structures and the second lowest energy structures. The dark grey ball is Pb atom, and yellow ball is Au atom.

References

- 1 J. Ho, K. M. Ervin, W. C. Lineberger, Photoelectron Spectroscopy of Metal Cluster Anions: Cu_n^- , Ag_n^- , and Au_n^- , *J. Chem. Phys.*, 1990, **93**, 6987.
- 2 X. B. Li, H. Y. Wang, X. D. Yang, Z. H. Zhu, Y. J. Tang, Size Dependence of the Structures and Energetic and Electronic Properties of Gold Clusters, *J. Chem. Phys.*, 2007, **126**, 084505.
- 3 B. Wang, J. Zhao, X. Chen, , D. Shi, G. Wang, Atomic Structures and Covalent-to-Metallic Transition of Lead Clusters Pb_n ($n=2-22$). *Phys. Rev. A*, 2005, **71**(3), 033201.
- 4 S. Yahachi, Y. Kenzi, M. Kazuhiro, N. Tamotsu, Formation and Ionization Potentials of Lead Clusters, *Jpn. J. Appl. Phys.*, 1982, **21**, L396.