

Supplementary information for

Magnetic CoFe₂O₄-CNS nanocomposite as an efficient, recyclable catalyst for peroxymonosulfate activation and pollutant degradation

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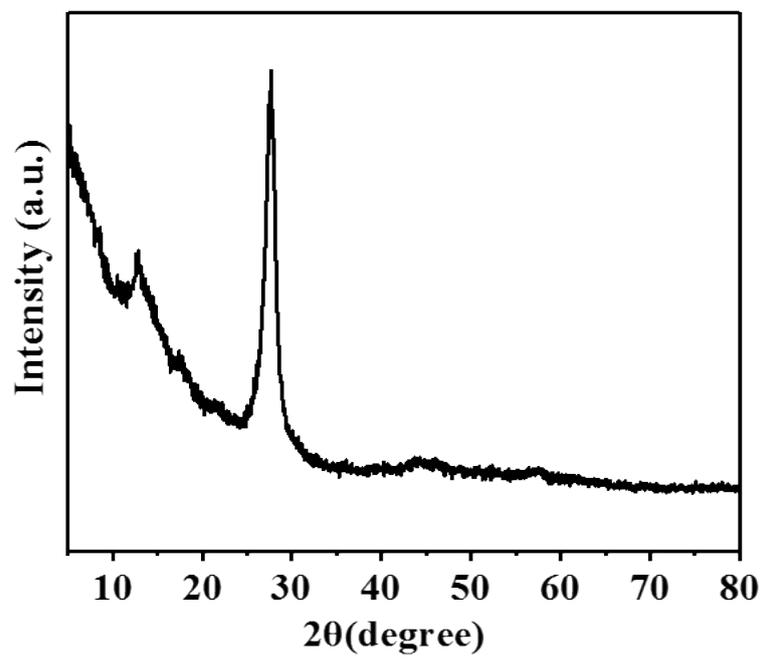


Figure S1. XRD patterns of g-C₃N₄

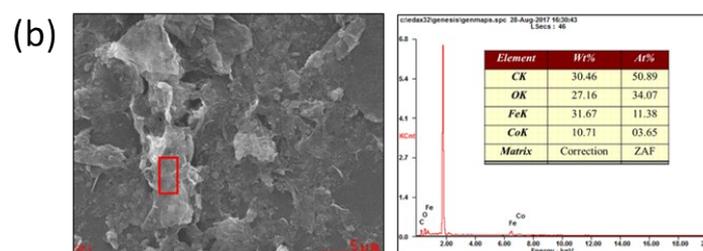
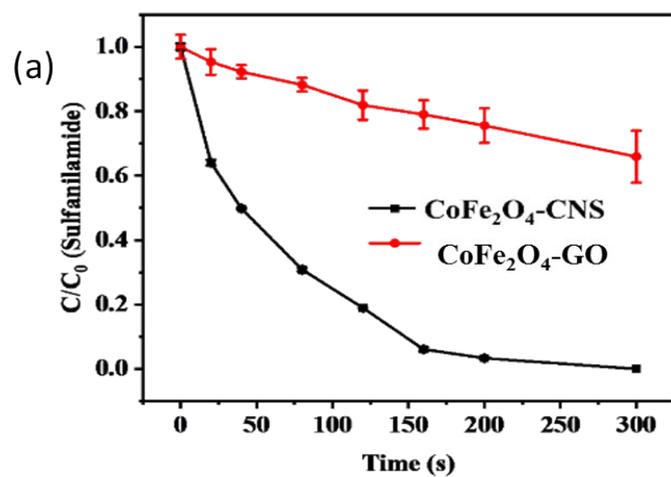


Figure S2. (a) Removal of SA in the reaction systems with CoFe₂O₄-CNS and CoFe₂O₄-GO catalysts; (b) SEM and EDX spectra of CoFe₂O₄-GO.

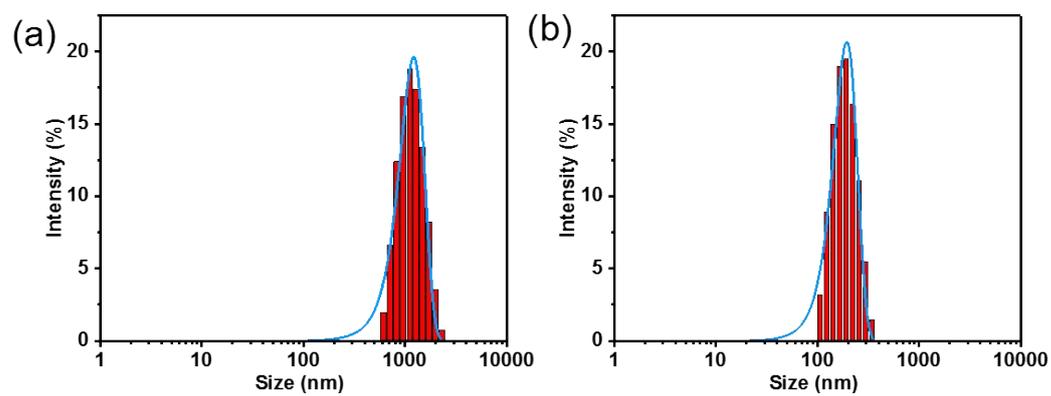


Figure S3. Size distribution histogram with Gaussian-fitting curve of (a) CoFe₂O₄-CNS and (b) CoFe₂O₄

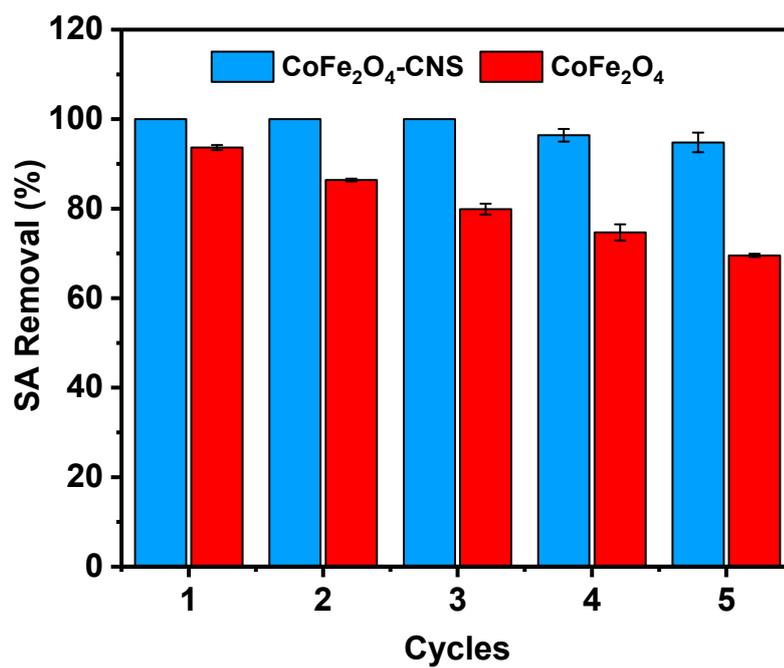


Figure S4. SA removal efficiencies within 30 min at five consecutive reaction cycles.

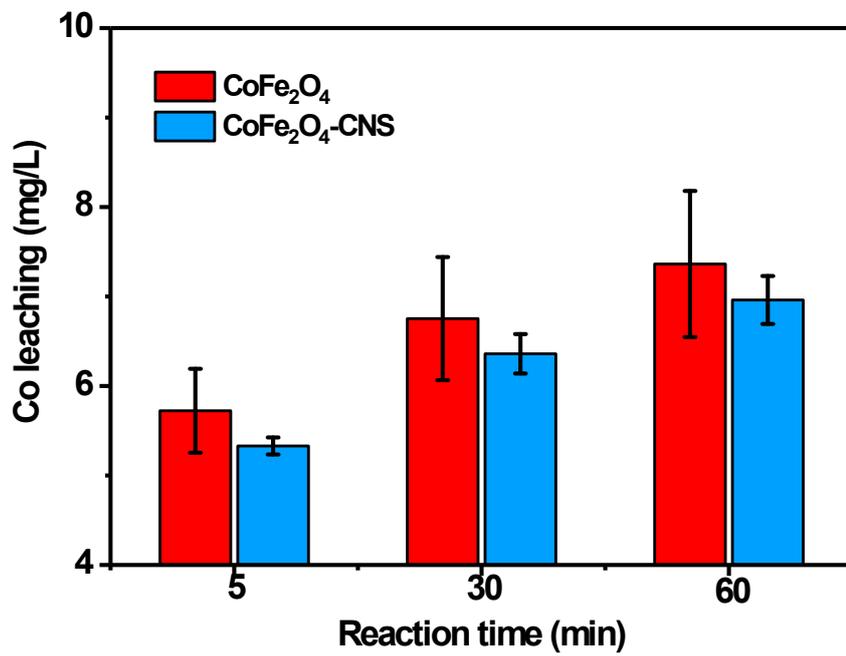


Figure S5. Accumulative concentrations of leached Co ions from $\text{CoFe}_2\text{O}_4\text{-CNS}$ and CoFe_2O_4 .

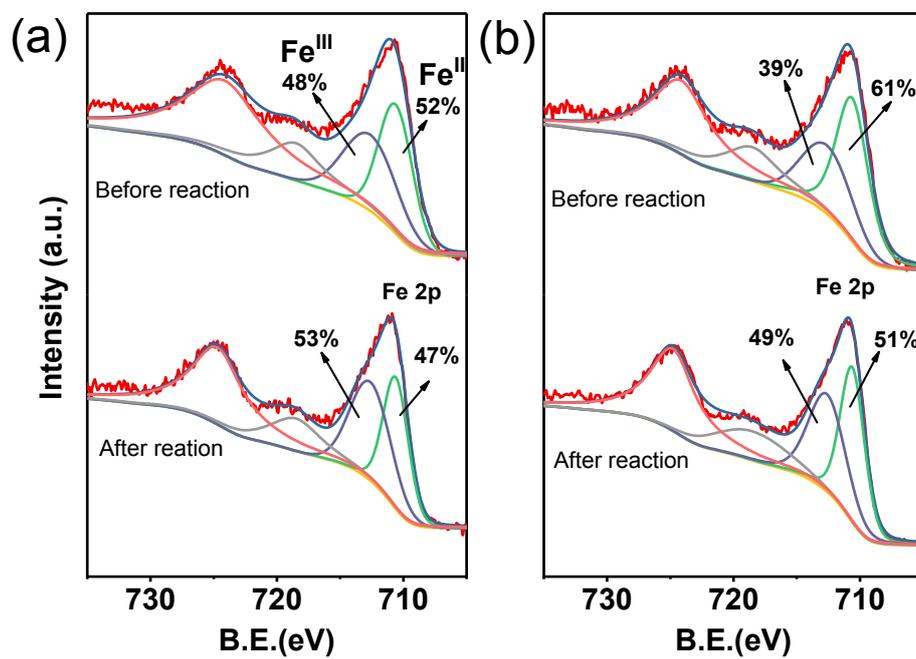


Figure S6. XPS of Fe 2p spectra of CoFe₂O₄-CNS and CoFe₂O₄ before and after the reaction.

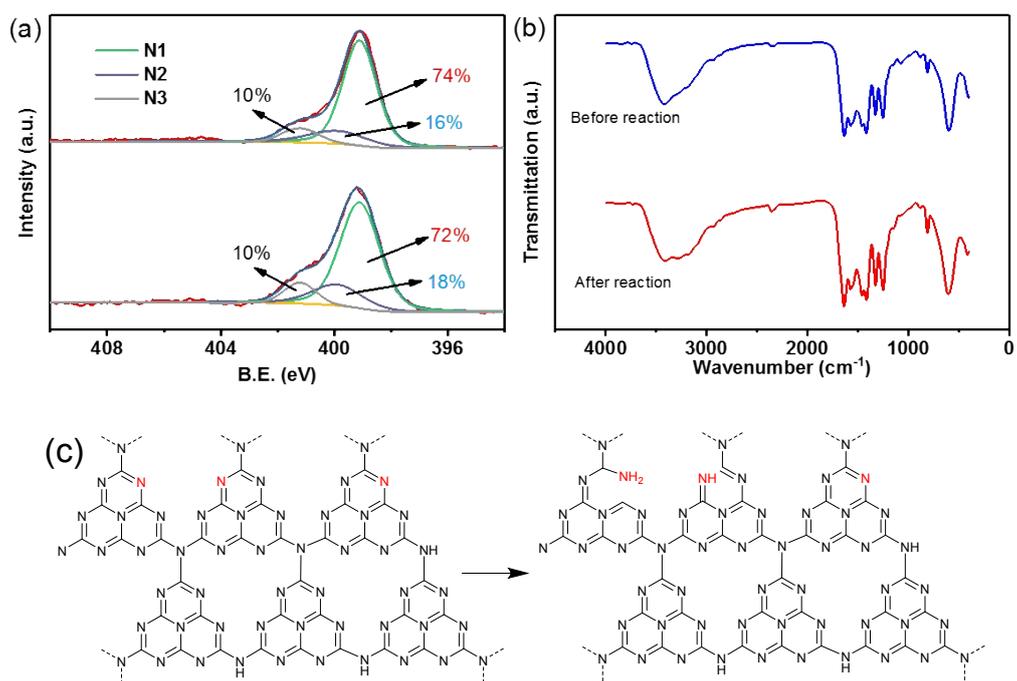


Figure S7. (a) High-resolution N 1s spectra of CoFe₂O₄-CNS. (b) FT-IR spectra of the CoFe₂O₄-CNS before and after the reaction with PMS. (c) The transformation of grapheme N on the zigzag edges of CNS.