

Supporting information

**Size effect of oxygen reduction reaction on nitrogen-doped graphene quantum  
dots**

Peng Zhang<sup>1</sup>, Qiang Hu<sup>1</sup>, Xuejing Yang<sup>1</sup>, Xiuli Hou<sup>\*,1</sup>, Jianli Mi<sup>1</sup>, Lei Liu<sup>1</sup>,  
Mingdong Dong<sup>\*,2</sup>

*1. Institute for Advanced Materials, School of Materials Science and Engineering,  
Jiangsu University, Zhenjiang 212013, China*

*2. Center for DNA Nanotechnology (CDNA), interdisciplinary Nanoscience Center  
(iNANO), Aarhus University, DK-8000 Aarhus, Denmark*

\*E-mail: houxiuli@ujs.edu.cn, dong@inano.au.dk

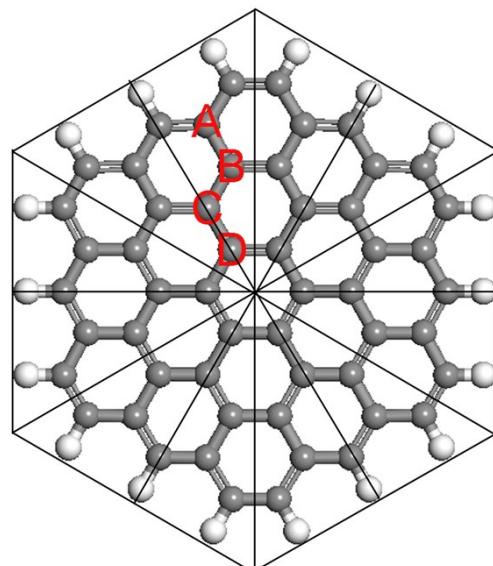


Figure S1. Schematic doped site of N atom in C53H18N.

Table S1. Relative energies of N-doped C<sub>53</sub>H<sub>18</sub>N with different doped sites.<sup>a</sup> All results are in unit of eV.

Site	A	B	C	D
Energ y	0	0.15	0.26	0.23

<sup>a</sup> Here, we take m C<sub>53</sub>H<sub>18</sub>N as an example to study the favorable N doped site. As shown in Figure S1, the hexagonal GQD can be divided into 12 regions and the C atoms between different regions are equivalent. Thus, we only need to study the doped sites in one region. From Table S1, it is clear that Site A shows the lowest DFT energy, suggesting that Site A is the most favorable N-doped site.

Table S2. The calculated electronic energies ( $E$ ), zero point energies ( $ZPE$ ), and entropy values ( $S$ ) of N-doped GQDs with adsorbed ORR intermediates. All results are in unit of eV.

	C23H12N	C23H12N +O <sub>2</sub>	C23H12N +OOH	C23H12N +O	C23H12N +OH
$E$	-25535.441	-29632.584	-29649.851	-27584.781	-27601.947
$ZPE$	7.392	7.615	7.901	7.526	7.838
$S$	0.694	0.986	0.928	0.877	0.807
	C53H18N	C53H18N +O <sub>2</sub>	C53H18N +OOH	C53H18N +O	C53H18N +OH
$E$	-56745.939	-60843.318	-60860.545	-58795.475	-58812.726
$ZPE$	14.232	14.458	14.768	14.340	14.697
$S$	1.580	1.578	1.622	1.543	1.528
	C95H24N	C95H24N +O <sub>2</sub>	C95H24N +OOH	C95H24N +O	C95H24N +OH
$E$	-100402.294	-104499.688	-104516.903	-102451.838	-102469.109
$ZPE$	23.107	23.333	23.635	23.205	23.561
$S$	2.478	2.595	2.667	2.553	2.557
	C94H24N2	C94H24N2 +O <sub>2</sub>	C94H24N2 +OOH	C94H24N2 +O	C94H24N2 +OH
$E$	-100855.052	-104952.742	-104969.685	-102904.963	-102921.880
$ZPE$	23.080	23.338	23.600	23.217	23.529
$S$	2.431	2.594	2.654	2.544	2.556
	C92H24N4	C92H24N4 +O <sub>2</sub>	C92H24N4 +OOH	C92H24N4 +O	C92H24N4 +OH
$E$	-101760.567	-105858.310	-105875.086	-103810.594	-103827.268
$ZPE$	22.941	23.209	23.476	23.123	23.406
$S$	2.302	2.559	2.590	2.566	2.514

Table S3. The bond length of O-C bond ( $d_{O-C}$ ) for adsorbed ORR intermediates on N-doped GQDs. All results are in unit of Å.

$d_{O-C}$	O <sub>2</sub>	OOH	O	OH	H <sub>2</sub> O
C23H12N	1.506	1.462	1.334	1.452	3.268
C53H18N	1.465	1.457	1.312	1.436	3.156
C95H24N	1.456	1.455	1.307	1.432	3.189
C94H24N2	1.434	1.456	1.307	1.434	3.221
C92H24N4	1.425	1.459	1.310	1.438	3.219