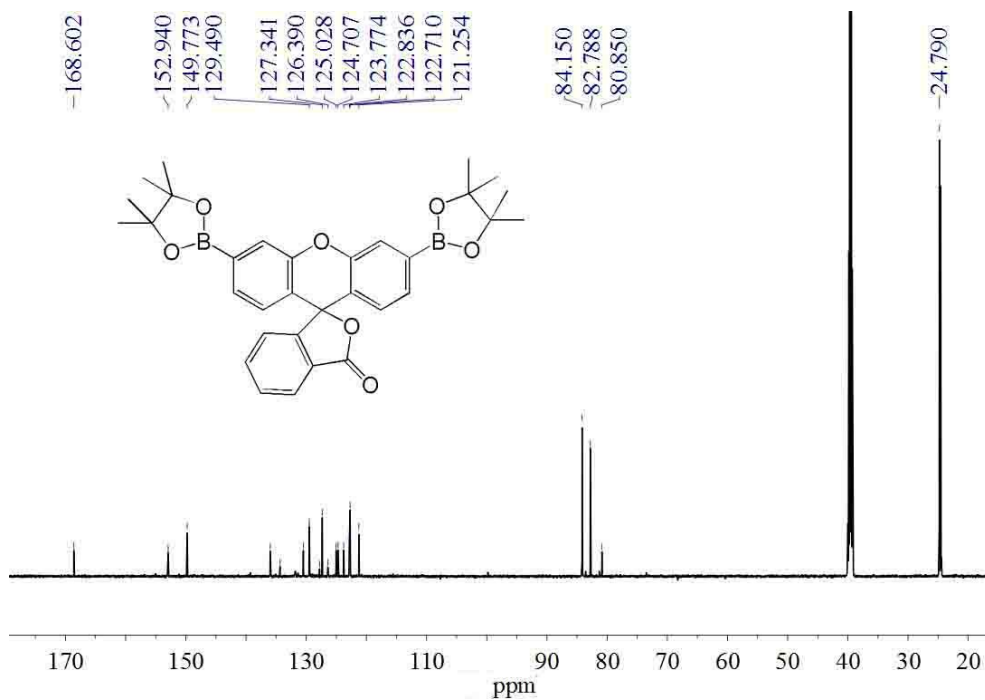
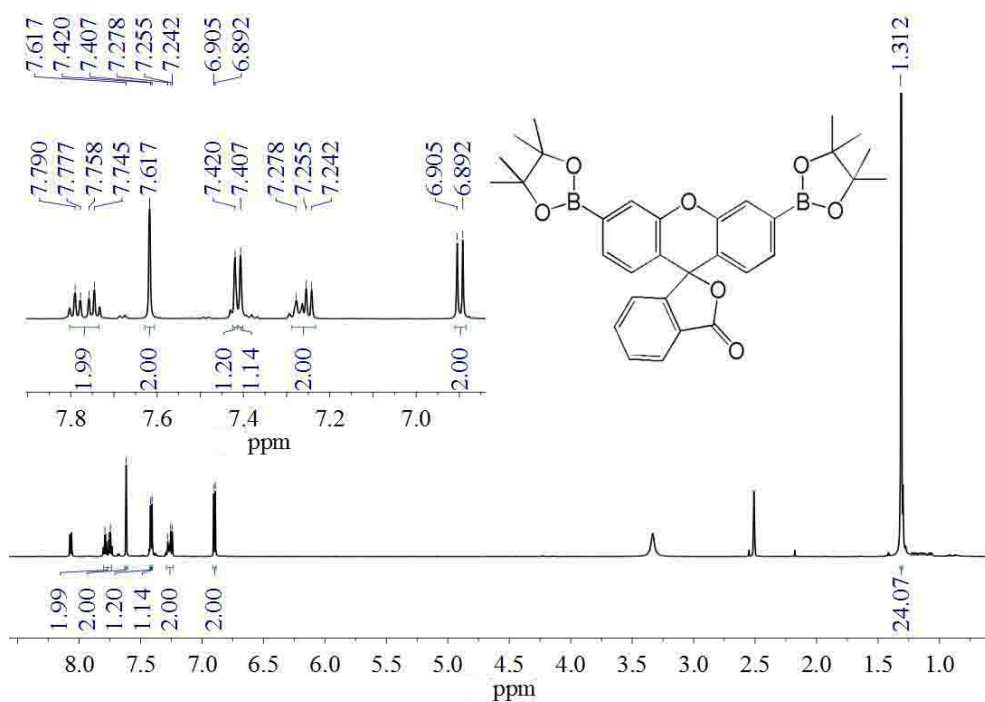


**Investigation of a Halloysite-Based Fluorescence Probe with
a Highly Selective and Sensitive “Turn-On” Response upon
Hydrogen Peroxide**

----- Supporting Information

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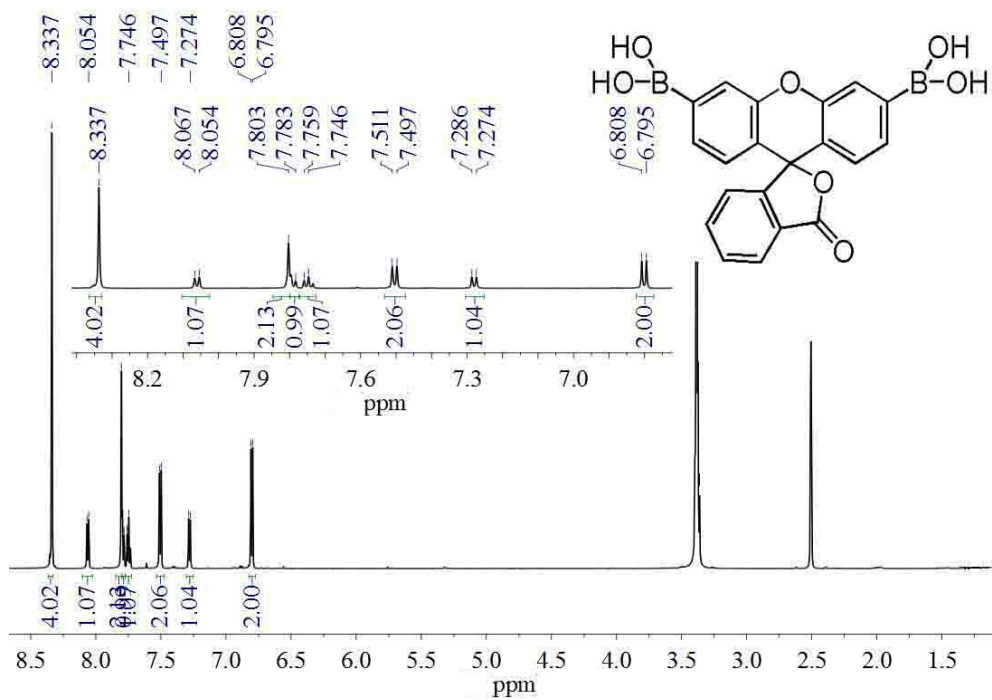


Figure S3. ¹H NMR spectrum of PA

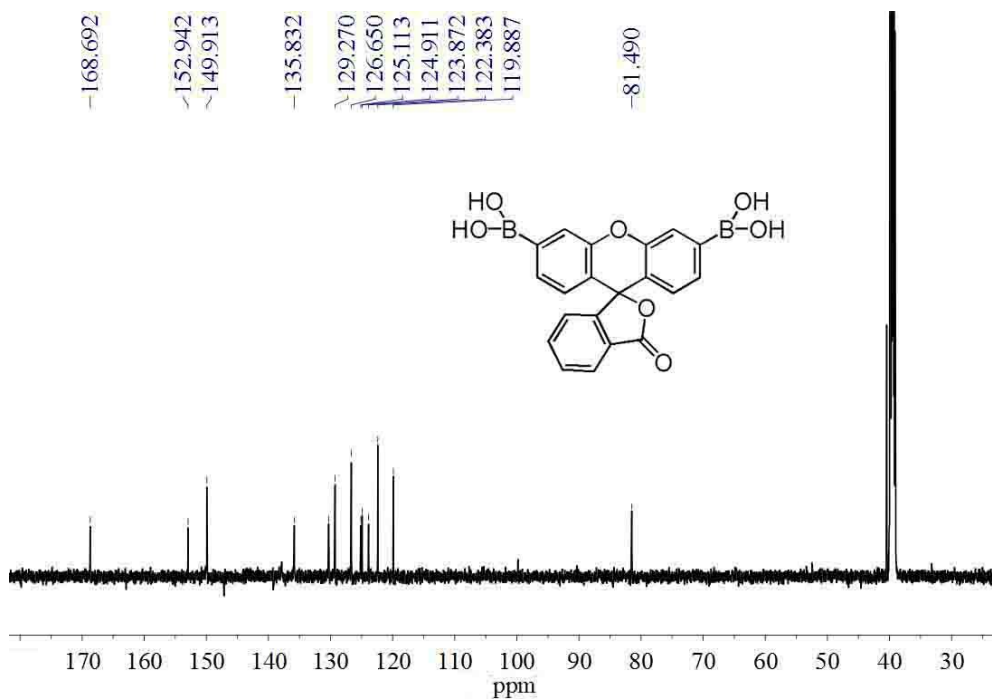


Figure S4. ¹³C NMR spectrum of PA

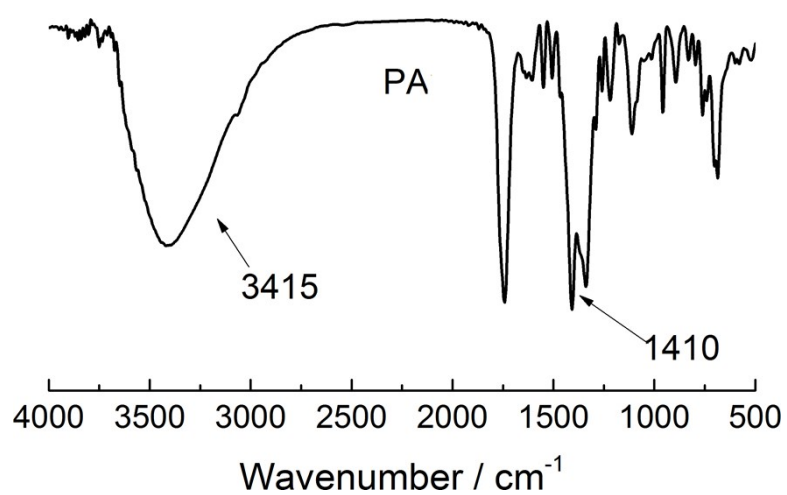


Figure S5. the FTIR spectrum of PA

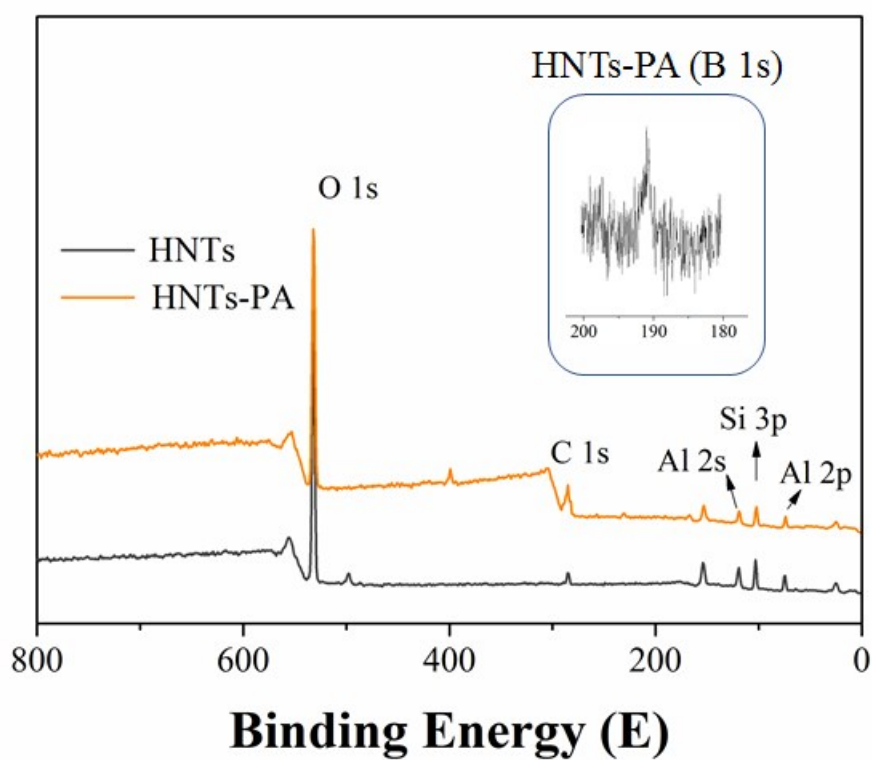


Figure S6. XPS spectra of HNTs and HNTs-PA

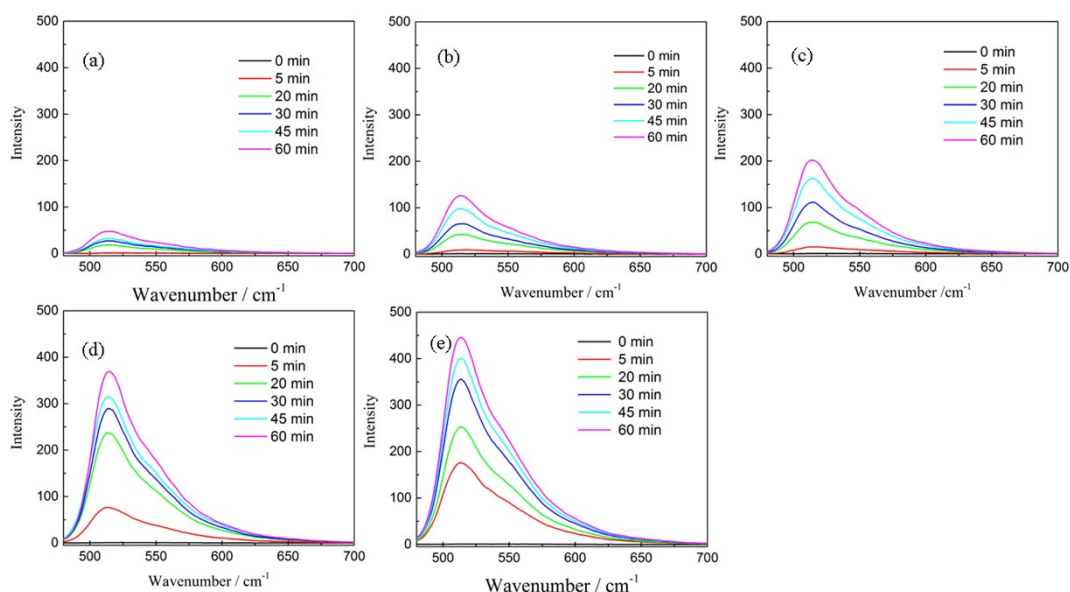


Figure S7. Fluorescence turn-on response of 0.1 mg/mL HNTs-PA. Data were acquired at 25 °C in 2 mL H₂O, with excitation at $\lambda=470$ nm and emission was collected between 480-700 nm. Time points represent 0-60 min after the addition of different times H₂O₂ (different times to the mole number of PA). (a) [H₂O₂] = 5×10^{-5} M (the addition amount of H₂O₂ is 1×10^{-7} mol); (b) [H₂O₂] = 2.5×10^{-4} M; (c) [H₂O₂] = 5×10^{-4} M; (d) [H₂O₂] = 2.5×10^{-3} M; (e) [H₂O₂] = 2.5×10^{-2} M. The turn-on response of each probe was completed at these time points.

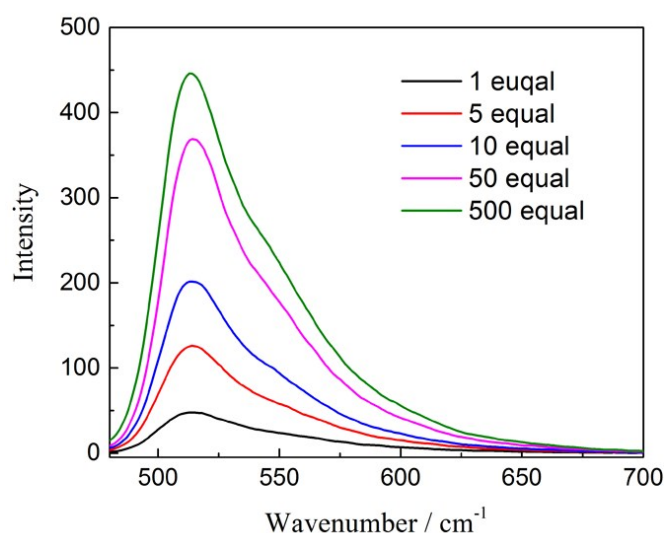


Figure S8. Fluorescence “turn-on” response of 0.1 mg/mL HNTs-PA. Data were acquired at 25 °C in 2 mL H₂O, with excitation at $\lambda=470$ nm and emission was collected between 480-700 nm for HNTs-PA. Time points represent 60 min after the addition of different times H₂O₂ (1 equal represents [H₂O₂] = 5×10^{-5} M; 5 equal represents [H₂O₂] = 2.5×10^{-4} M; 10 equal represents [H₂O₂] = 5×10^{-4} M; 50 equal represents [H₂O₂] = 2.5×10^{-3} M; 500 equal represents [H₂O₂] = 2.5×10^{-2} M).

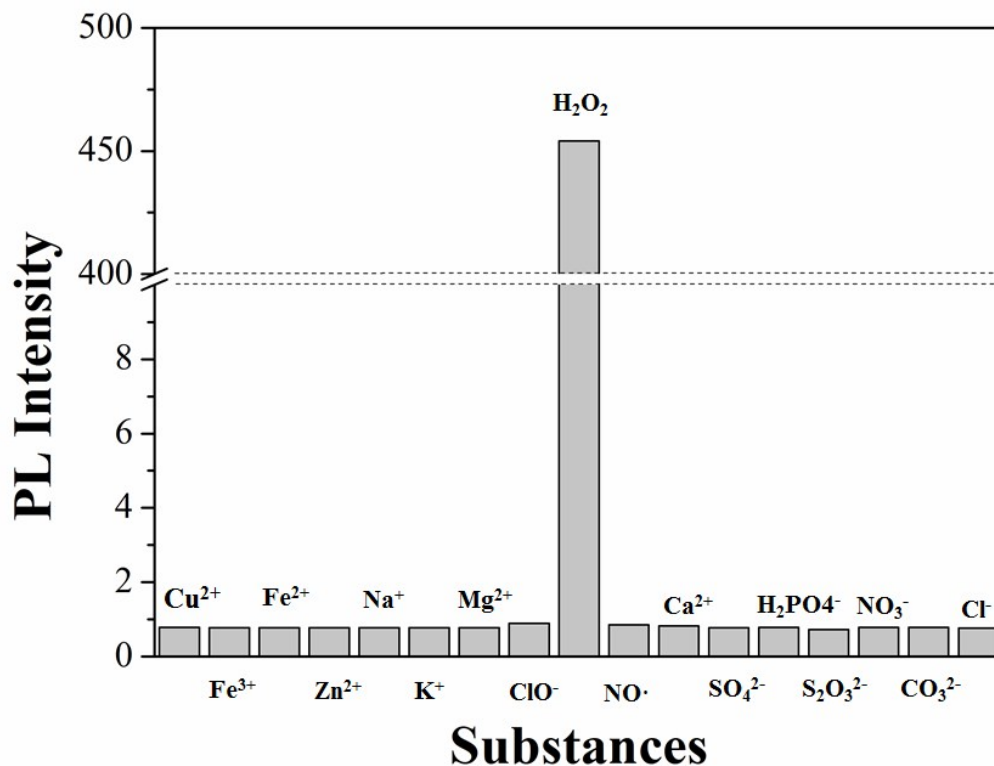


Figure S9. Fluorescence “turn-on” response of 0.1 mg/mL HNTs-PA upon different substances. Data were obtained at 25 °C in 2 mL H₂O, with excitation at $\lambda=470$ nm. The PL intensity represents the fluorescence intensity at the maximum emission wavelength.

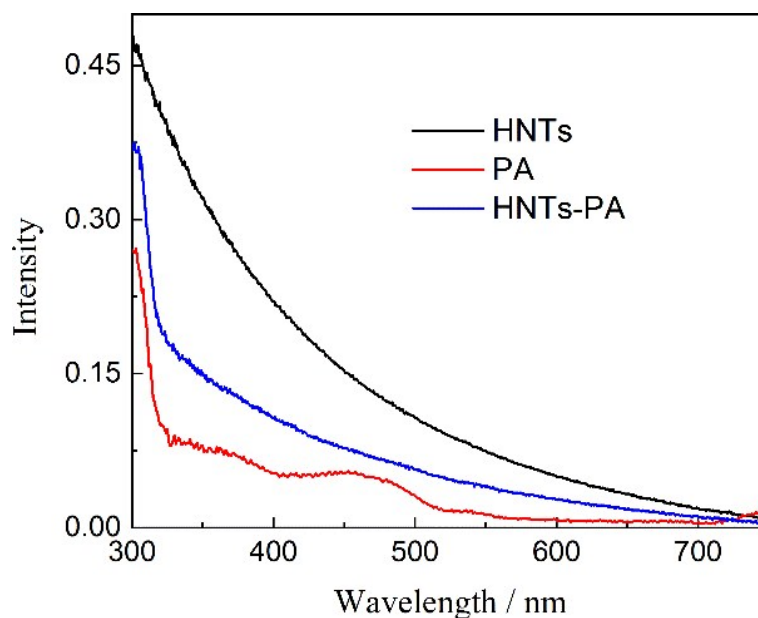


Figure S10. The UV-vis spectra of HNTs, PA and HNTs-PA

Spectrum processing :
No peaks omitted

Processing option : All elements analyzed (Normalised)
Number of iterations = 2

Standard :

C CaCO₃ 1-Jun-1999 12:00 AM
O SiO₂ 1-Jun-1999 12:00 AM
Al Al₂O₃ 1-Jun-1999 12:00 AM
Si SiO₂ 1-Jun-1999 12:00 AM

Element	Weight%	Atomic%
C K	6.26	11.25
O K	28.30	38.17
Al K	27.97	22.37
Si K	31.39	24.12

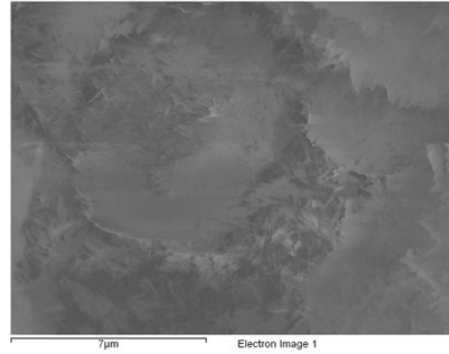


Figure S11. The energy dispersive spectra (EDS) results of HNTs-PA

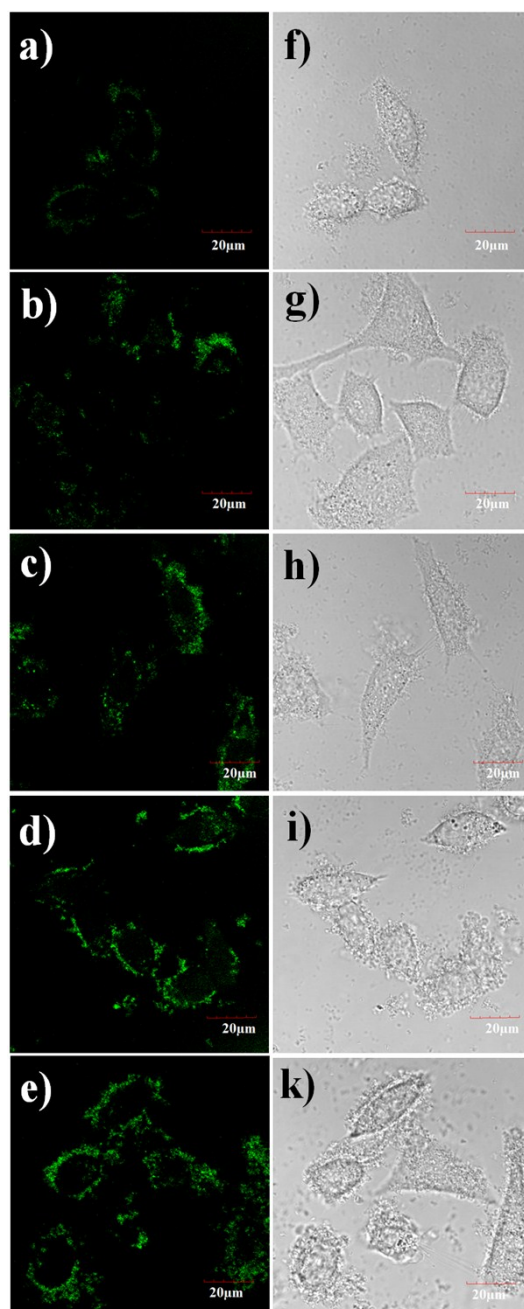


Figure S12. Confocal fluorescence images of living HeLa cells after treating with 0.5 mg/mL HNTs-PA for 1 (a and f), 2 (b and g), 4 (c and h), 8 (d and i) and 12 h (e and k) at 37 °C. The excitation wavelength was fixed at 488 nm and fluorescent signals were collected from 500 to 600 nm.

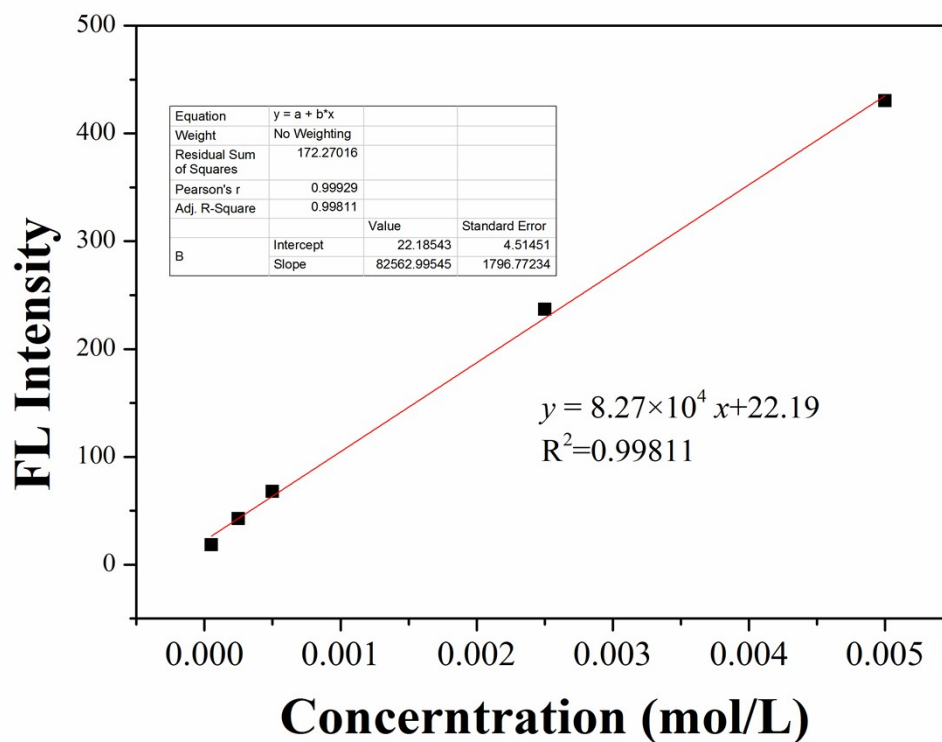


Figure S13. Plot of the fluorescence intensity of HNTs-PA solution vs the concentration of H₂O₂ (fluorescence intensity was recorded at 20 min after the addition of H₂O₂). Results of the linear regression for fluorescence intensity and concentration of H₂O₂ show that the linear correlation coefficient (R²) is greater than 0.99 which indicates that the fluorescence intensity of HNTs-PA solution and concentration of H₂O₂ shows a good linearity relationship in the range from 5 × 10⁻⁵ to 5 × 10⁻³ M.

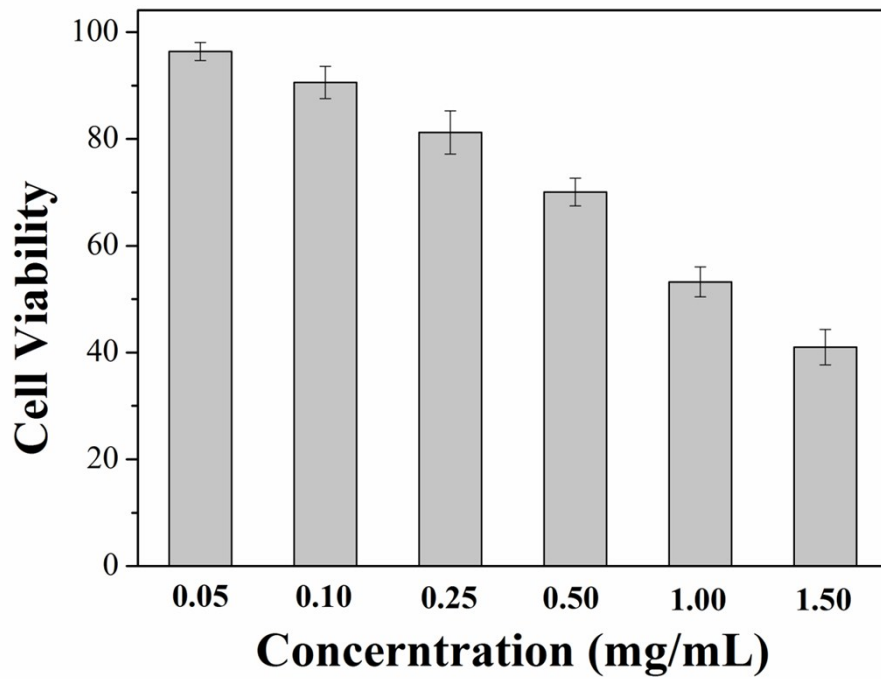


Figure S14. Cell viability of Hela cells at different concentrations of HNTs-PA. The IC_{50} value was calculated as 1.11 mg/mL