

## Supporting Information

### **Electrochromism and electrochemical properties of complexes of transition metal ions with benzimidazole-based ligand**

Monika Wałęsa-Chorab,<sup>a\*</sup> Radosław Banasz,<sup>a</sup> Damian Marcinkowski,<sup>a,b</sup> Maciej Kubicki,<sup>a</sup> Violetta Patroniak<sup>a</sup>

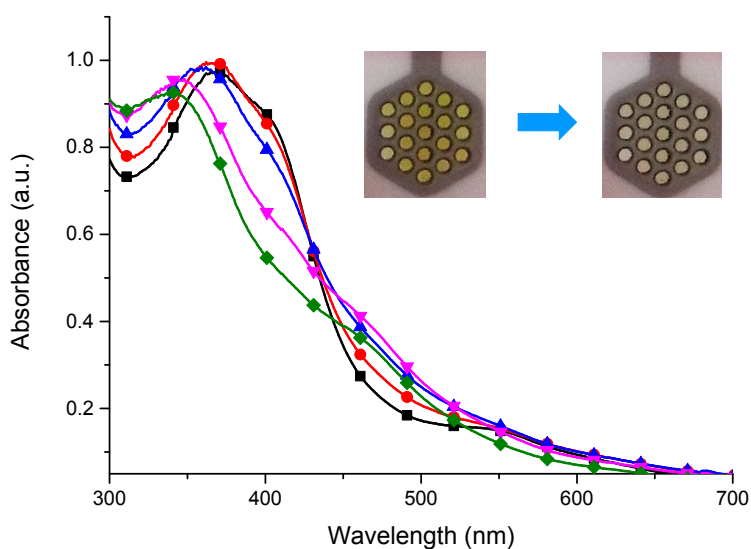
<sup>a</sup>*Faculty of Chemistry, Adam Mickiewicz University, Umultowska 89b, 61614 Poznań, Poland*

<sup>b</sup>*Current address: Institute of Technology and Life Sciences, Biskupińska 67, 60-463 Poznań, Poland*

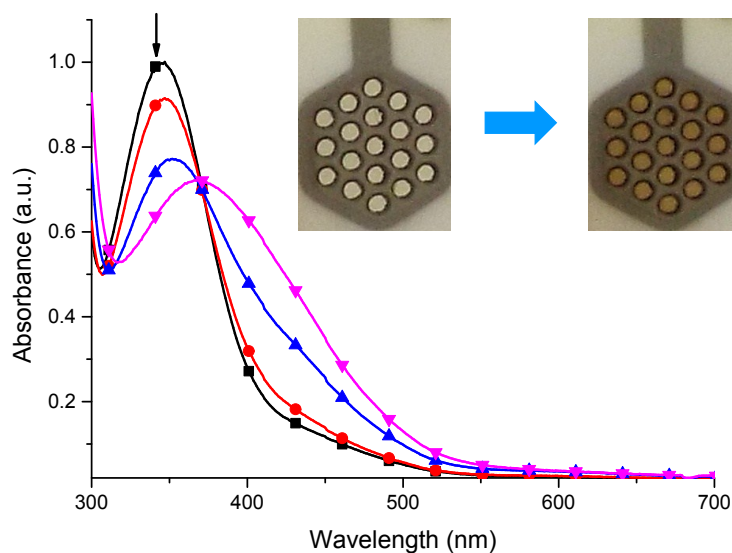
*\*Corresponding author: E-mail: mchorab@amu.edu.pl; Fax: +48 618291508; Tel: +48 618291772*

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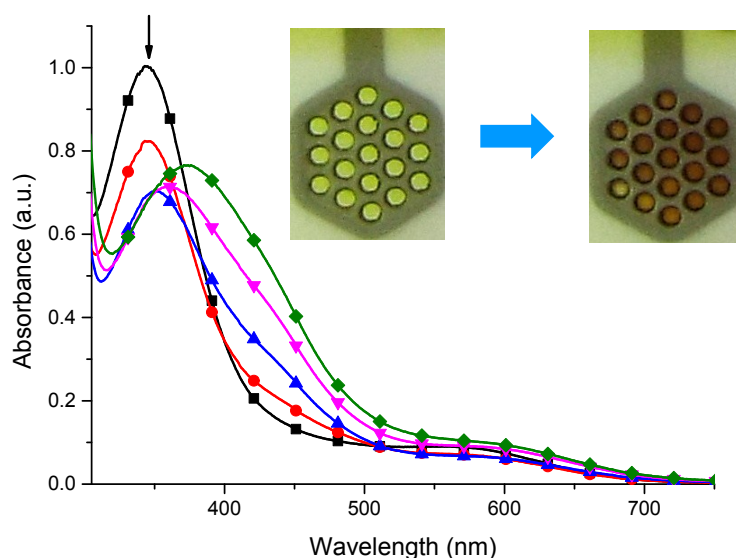
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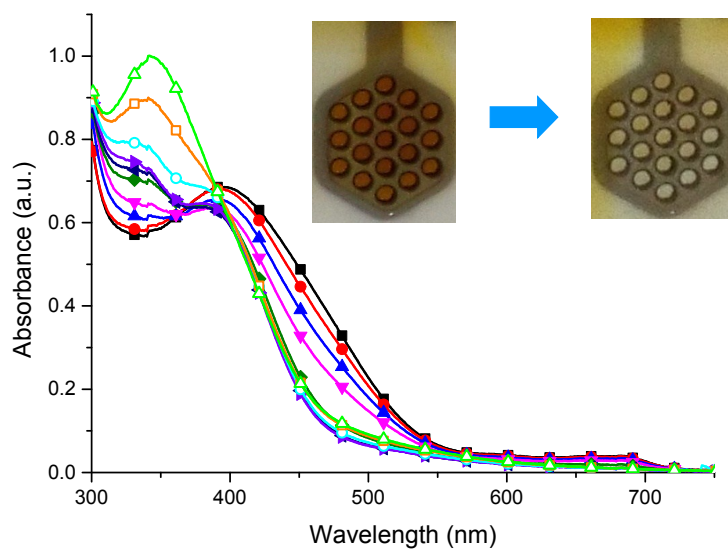
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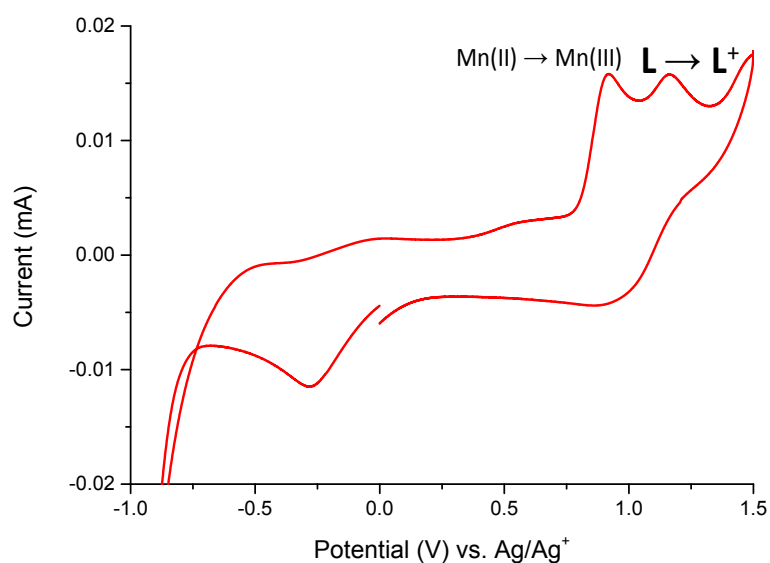
**Figure S2.** Spectral changes of **3** in dehydrated and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte by applying 0 (■), -500 (●), -600 (▲) and -700 mV (▼) potentials versus Ag/AgCl gel reference electrode held for 30 s per potential. Insert: photographs of the original (left) and electrochemically reduced (right) **3** by applying a potential for 1 min.



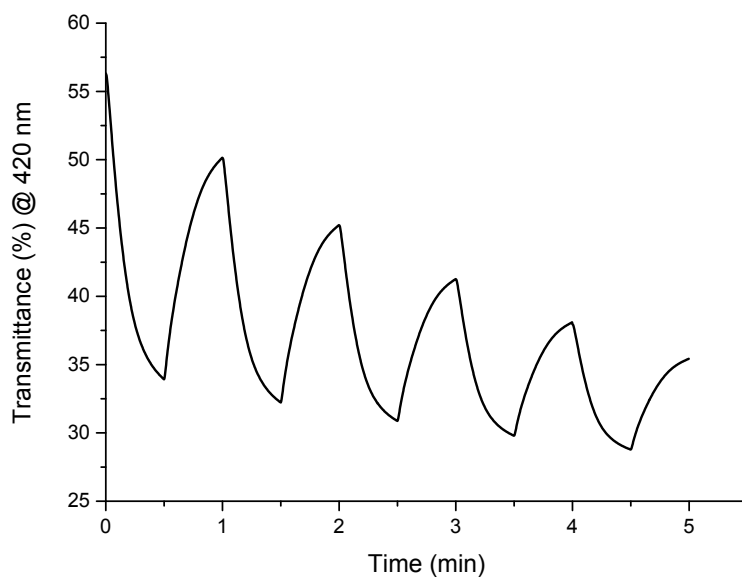
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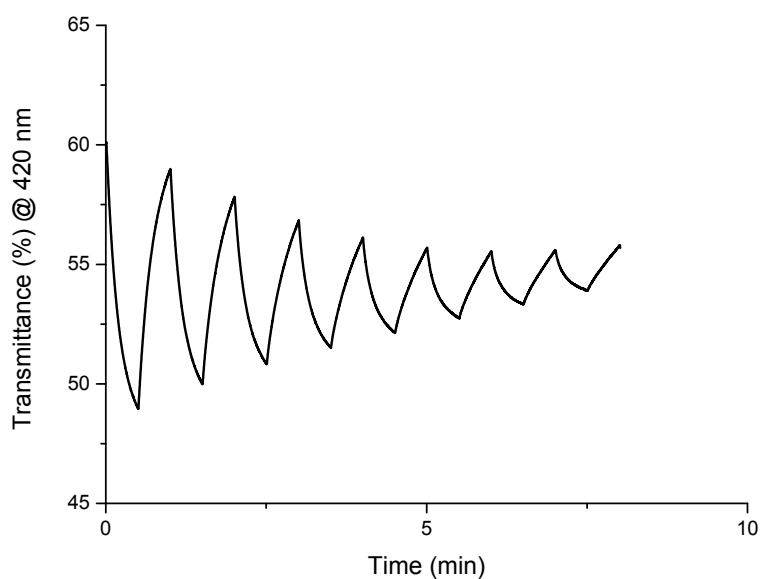
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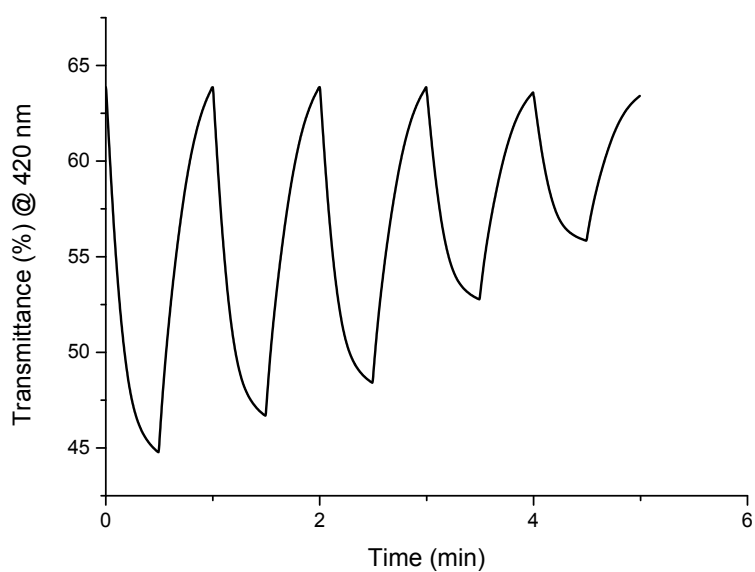
**Figure S5.** The cyclic voltammogram (2<sup>nd</sup> cycle) of Mn(II) complex **6** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte at a scan rate 100 mV/s scanned in the negative direction.



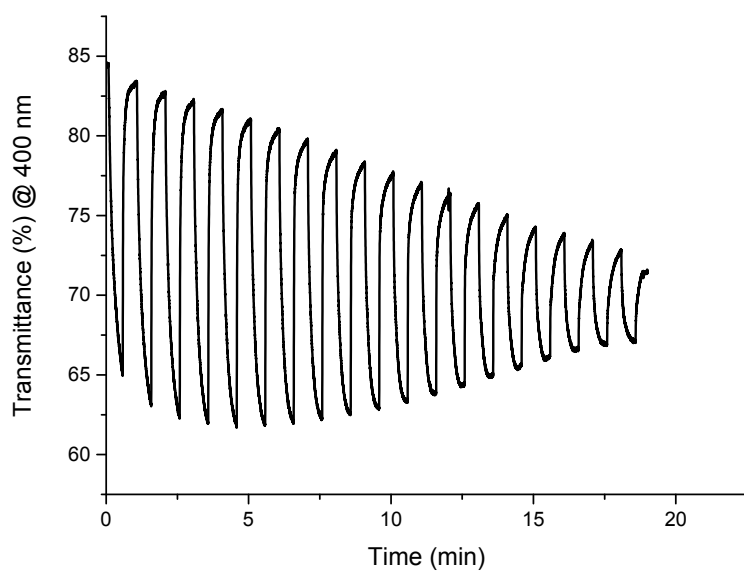
**Figure S6.** Changes in transmittance of Cu(II) complex **2** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte and monitored at 420 nm when switching between -400 mV and +200 mV potential at 60 s cycles.



**Figure S7.** Changes in transmittance of Cu(II) complex **3** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte and monitored at 400 nm when switching between -700 mV and +300 mV potential at 60 s cycles.



**Figure S8.** Changes in transmittance of Cu(II) complex **4** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte and monitored at 400 nm when switching between -800 mV and +200 mV potential at 60 s cycles.



**Figure S9.** Changes in transmittance of Co(II) complex **5** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAPF<sub>6</sub> as a supporting electrolyte and monitored at 400 nm when switching between -600 mV and +400 mV potential at 60 s cycles.

**Table S1.** CIE coordinates with D65 illuminant and 2° observer angle for the different states of ligand **L** and its complexes **1-5** in different states.

Compound		L*	a*	b*
Ligand <b>L</b>	Neutral	98.7	-5.5	20.1
	oxidized	98.7	-4.1	12.3
Complex <b>1</b>	Neutral	80.3	28.1	-7.5
	1 <sup>st</sup> oxidation state	83.4	2.8	45.3
	2 <sup>nd</sup> oxidation state	82.2	0.2	32.9
	Reduced	79.7	-19.6	33.3
Complex <b>2</b>	Neutral	85.6	-6.4	26.15
	Reduced	83.8	10.0	25.4
Complex <b>3</b>	Neutral	95.1	-2.7	10.2
	Reduced	93.2	5.8	9.5
Complex <b>4</b>	Neutral	80.1	-7.4	30.1
	Reduced	78.8	16.2	27.4
Complex <b>5</b>	Neutral	94.6	-5.2	29.9
	1 <sup>st</sup> oxidation state	91.6	15.1	48.5
	2 <sup>nd</sup> oxidation state	95.0	-4.2	16.3



**Table S2.** Crystal data, data collection and structure refinement.

Compound	1	2	3	4	5	6
Formula	C <sub>30</sub> H <sub>30</sub> FeN <sub>10</sub> ·2(CF <sub>3</sub> O <sub>3</sub> S)·CH <sub>3</sub> CN	C <sub>18</sub> H <sub>19</sub> CuF <sub>6</sub> N <sub>5</sub> O <sub>7</sub> S <sub>2</sub>	C <sub>16</sub> H <sub>19</sub> CuN <sub>6</sub> O <sub>4</sub> <sup>+</sup> ·NO <sub>3</sub> <sup>-</sup>	C <sub>15</sub> H <sub>15</sub> Br <sub>2</sub> CuN <sub>5</sub> ·CH <sub>3</sub> OH	C <sub>15</sub> H <sub>15</sub> Cl <sub>2</sub> CoN <sub>5</sub> ·C <sub>2</sub> H <sub>5</sub> OH	C <sub>15</sub> H <sub>15</sub> Br <sub>2</sub> MnN <sub>5</sub>
Formula weight	925.68	659.04	484.92	520.72	441.22	480.08
Crystal system	monoclinic	triclinic	monoclinic	triclinic	orthorhombic	triclinic
Space group	P2 <sub>1</sub> /n	P-1	P2 <sub>1</sub> /n	P-1	Pbca	P-1
a(Å)	9.7491(3)	8.0767(3)	9.6735(9)	9.6841(12)	11.59108(13)	7.6305(5)
b(Å)	16.1976(4)	11.3717(5)	17.4644(11)	10.0927(9)	15.95415(19)	8.9863(5)
c(Å)	24.3248(6)	13.9815(8)	12.1452(13)	10.4709(9)	20.3432(3)	13.5352(8)
α(°)	90	76.300(4)	90	91.231(7)	90	83.946(5)
β(°)	95.452(2)	82.265(4)	103.407(10)	92.328(8)	90	75.427(6)
γ(°)	90	85.663(3)	90	117.823(11)	90	67.854(6)
V(Å <sup>3</sup> )	3823.80(18)	1234.98(10)	1995.9(3)	903.40(18)	3761.98(8)	831.92(10)
Z	4	2	4	2	8	2
D <sub>x</sub> (g cm <sup>-3</sup> )	1.61	1.77	1.61	1.91	1.56	1.92
F(000)	1896	666	996	514	1816	470
μ(mm <sup>-1</sup> )	0.60	1.15	1.15	5.65	9.91	5.60
Reflections:						
collected	14830	8888	7123	5979	18640	5914
unique (R <sub>int</sub> )	7317 (0.019)	5079 (0.015)	3514 (0.021)	3675 (0.017)	3838 (0.057)	3403 (0.019)
with I>2σ(I)	6090	4552	2618	3322	3622	3123
R(F) [I>2σ(I)]	0.050	0.027	0.055	0.028	0.050	0.025
wR(F <sup>2</sup> ) [I>2σ(I)]	0.122	0.069	0.132	0.071	0.149	0.057
R(F) [all data]	0.061	0.032	0.079	0.033	0.052	0.029
wR(F <sup>2</sup> ) [all data]	0.130	0.071	0.142	0.074	0.152	0.059
Goodness of fit	1.04	1.04	1.05	1.05	1.05	1.08
max/min Δρ (e Å <sup>-3</sup> )	1.50/-0.84	0.39/-0.38	0.80/-0.46	0.8450/-1.02	1.68/-0.61	0.53/-0.49