An Experiment-Based Model Quantifying Antimicrobial Activity of Silver Nanoparticles on Escherichia coli

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9 Supplementary Information

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Table S1. Dissolution of AgNPs in different media.

-	Solvent	AgNP (µg/mL)	Dissolved Ag ions (µg/L)	% Dissolved
	H ₂ O	5	135.7	2.7
	H ₂ O	10	173.8	1.7
	H ₂ O	20	221.2	1.1
	H ₂ O	40	252.2	0.6
	H ₂ O	80	287.8	0.4
_	LB	5	70.6	1.4
	LB	10	114.9	1.1
	LB	20	148.1	0.7
	LB	40	169.2	0.4
	LB	80	218.4	0.3

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Figure S1. (A) Relation between the CFU counts (per mL) and OD₆₀₀ of (diluted) liquid culture in
log-log scale. Inset: the same plot in linear scale. (B) CFU assay on *E. coli* treated with AgNPs
at various concentrations. Inset: the same plot in log-linear scale. (C) CFU assay on *E. coli*treated with AgNPs at 20 µg/mL for longer times. Inset: the same plot in log-linear scale. (D)
CFU assay on *E. coli* treated with AgNO₃ at various concentrations. Inset: the same plot in log-linear scale.



Figure S2. (A) Absorbance spectra of AgNPs at various concentrations in LB growth medium.
 The vertical dashed black line indicates the wavelength at 600 nm. (B) OD₆₀₀ of AgNPs in LB as
 a function of concentration.





Figure S3. Parabolic dependence of the lag time λ of bacterial growth on the concentration of AgNPs or AgNO₃.

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Figure S4. (A) Predicted growth curves of bacteria in the exponential phase from the SAD model. (B) Measured growth curves of bacteria with AgNO₃ added at the exponential phase (+: $10 \mu g/mL AgNO_3$ added; -: negative control).









- **Figure S7**. The dependence of the maximum specific growth rate (μ_m) and the lag time (λ) of *E. coli* on the initial concentration of bacteria (DD_{600}^0) when AgNPs (A and B) or AgNO₃ (C and D)

were added.









Figure S9. Predicted percentage of killed *E. coli* from the SAD model as a function of the concentration of AgNPs in the growth medium.