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Electronic supplementary information

Magnetic graphene solid-phase extraction in the determination of polycyclic aromatic hydrocarbons in water †

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1. Optimization of the extraction parameters

The detailed extraction conditions used in the experiments of optimization of the extraction parameters were as follows.

- 1.1 Effect of the amount of the sorbents on the chromatography peak areas of the PAHs. Extraction conditions: sample volume, 100mL; extraction time, 5min; organic modifier, 3% 2-propanol; desorption solvent, 100μL of composite solvent (toluene: acetone (v:v) 3:1); desorption time, 5min; concentration of the analytes, 5μg/L.
- 1.2 Effect of the extraction time on the chromatography peak areas of the PAHs. Extraction conditions: sample volume, 100mL; amount of the sorbents, 10mg; organic modifier, 3% 2-propanol; desorption solvent, 100μ L of composite solvent (toluene: acetone (v:v) 3:1); desorption time, 5min; concentration of the analytes, 5μ g/L.
- 1.3 Effect of the selection of desorption solvent on the chromatography peak areas of the PAHs. Extraction conditions: sample volume, 100mL; amount of the sorbents, 10mg; extraction time, 5min; organic modifier, 3%

2-propanol; volume of desorption solvent, 100 µL, desorption time, 5min; concentration of the analytes, 5µg/L.

- 1.4 Effect of the amount of desorption solvent on the chromatography peak areas of the PAHs. Extraction conditions: sample volume, 100mL; amount of the sorbents, 10mg; extraction time, 5min; organic modifier, 3% 2-propanol; desorption solvent, composite solvent (toluene: acetone (v:v) 3:1); desorption time, 5min; concentration of the analytes, 5µg/L.
- 1.5 Effect of the desorption time on the chromatography peak areas of the PAHs. Extraction conditions: sample volume, 100mL; amount of the sorbents, 10mg; extraction time, 5min; organic modifier, 3% 2-propanol; desorption solvent, 100μ L composite solvent (toluene: acetone (v:v) 3:1); concentration of the analytes, 5μ g/L.
- **2. Figure scheme 1:** The SEM image in Figure scheme 1 is same as Fig. 1e, where the image is clearer and the size is larger. In order to avoid needless duplication, we deleted the SEM image in Figure scheme 1. Finally, the figure scheme 1 is as follows.

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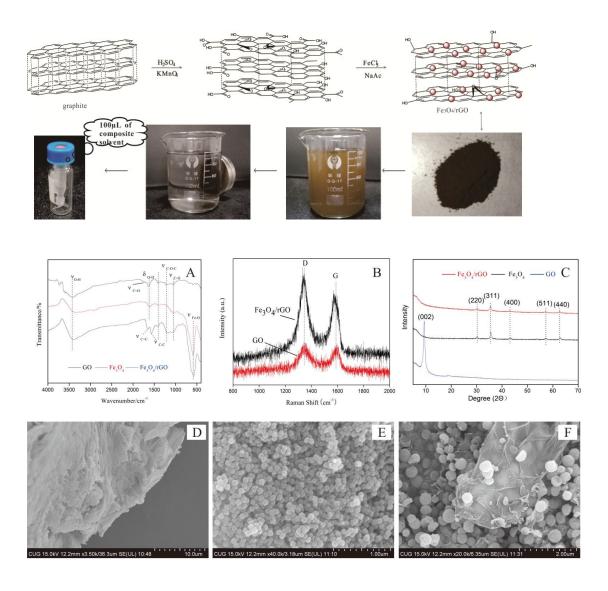


Fig. 1 (A) FTIR spectra of GO, Fe₃O₄ and Fe₃O₄/rGO; (B) Raman spectra of Fe₃O₄/rGO and GO; (C) X-ray diffraction patterns of GO, Fe₃O₄ and Fe₃O₄/rGO; SEM images of (D) GO, (E) Fe₃O₄ and (F) Fe₃O₄/rGO.

3. Fig. S1: Total adsorption of 10mg of Fe $_3$ O $_4$ /rGO for 100mL water containing 20mg/L PAHs

