

Electronic Supplementary Information (ESI)

Template-free synthesis of nitrogen doped carbon materials from an organic ionic dye (Murexide) for supercapacitor application

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Table S1: Elemental composition of NCM_{(MDE)-X} by CHN microanalysis

Sample ID	C	H	N
NCM _{(MDE)-700}	61.64	1.78	19.51
NCM _{(MDE)-800}	61.29	1.62	13.75
NCM _{(MDE)-900}	63.34	1.37	7.17

Thermal studies

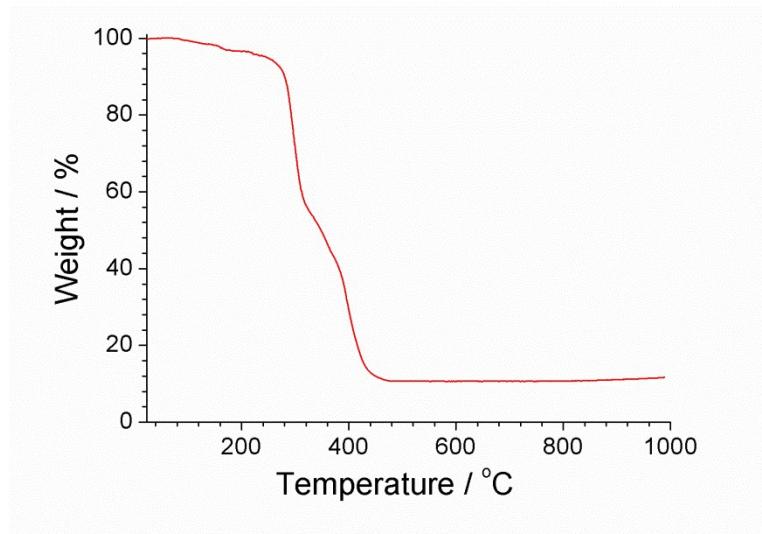


Figure S1: TGA of murexide under N₂ flow.

Morphology characterization

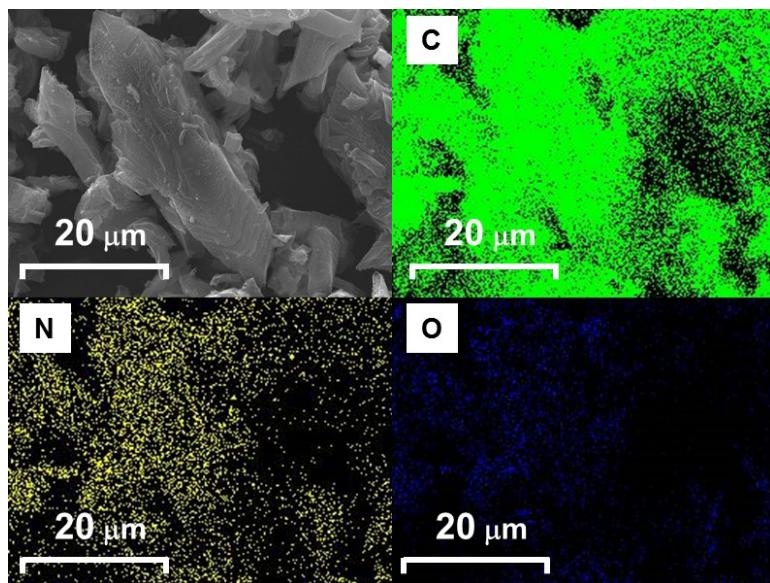


Figure S2: SEM image and EDS elemental maps of NCM_(MDE)-800.

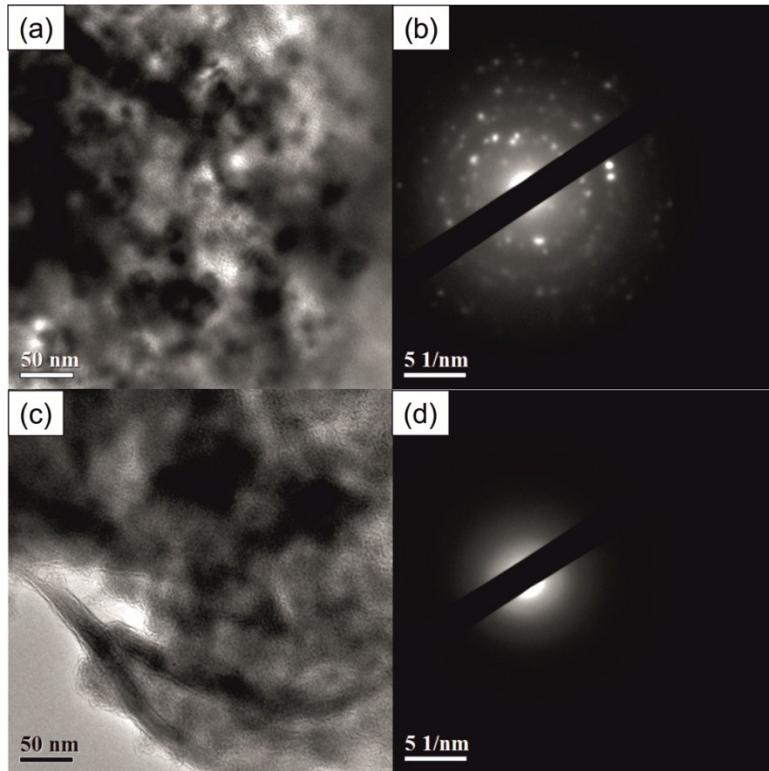


Figure S3: TEM images for NCM_(MDE)-800 and their corresponding SAED pattern in two different areas of the sample. Figure (b) shows the diffraction from the lattice in the crystalline part and (d) diffuse rings from the amorphous area.

BET Isotherms

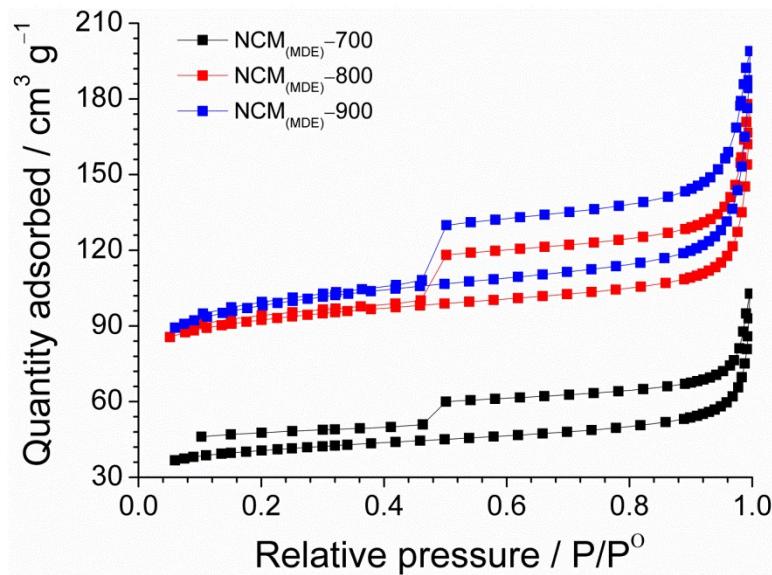


Figure S4: N₂ adsorption desorption Isotherms for NCM_(MDE)-X samples via Brunauer-Emmett-Teller (BET) characterization.

PSD curves

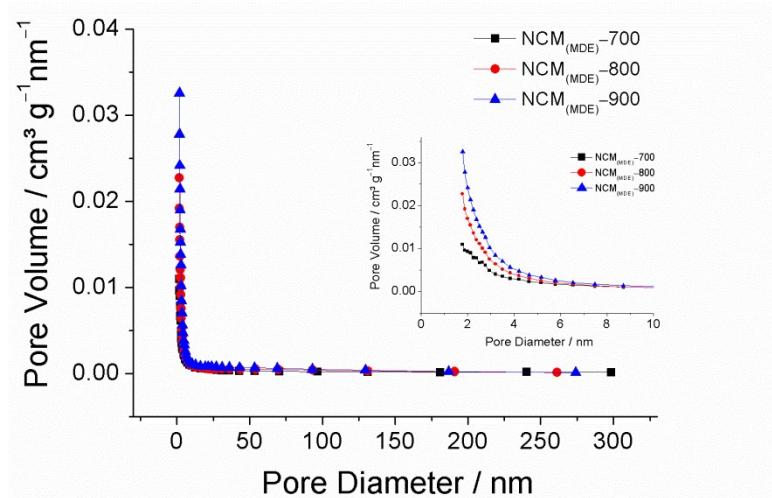


Figure S5: PSD curves for NCM_(MDE)-700, NCM_(MDE)-800 and NCM_(MDE)-900. The inset displays the pore size distribution data for these materials up to 10 nm.

Table S2: BET analysis data for NCM_(MDE)-700, NCM_(MDE)-800 and NCM_(MDE)-900.

BET Parameter	NCM _(MDE) -700	NCM _(MDE) -800	NCM _(MDE) -900
Surface Area (m ² g ⁻¹)	127.88	286.51	306.68
Average pore diameter (nm)	11.46	10.09	9.39
t-plot micropore volume (cm ³ g ⁻¹)	0.042	0.11	0.11
Total pore volume (cm ³ g ⁻¹)	0.094	0.13	0.16
Average particle size (nm)	46.92	20.94	19.56

Electrochemical Studies

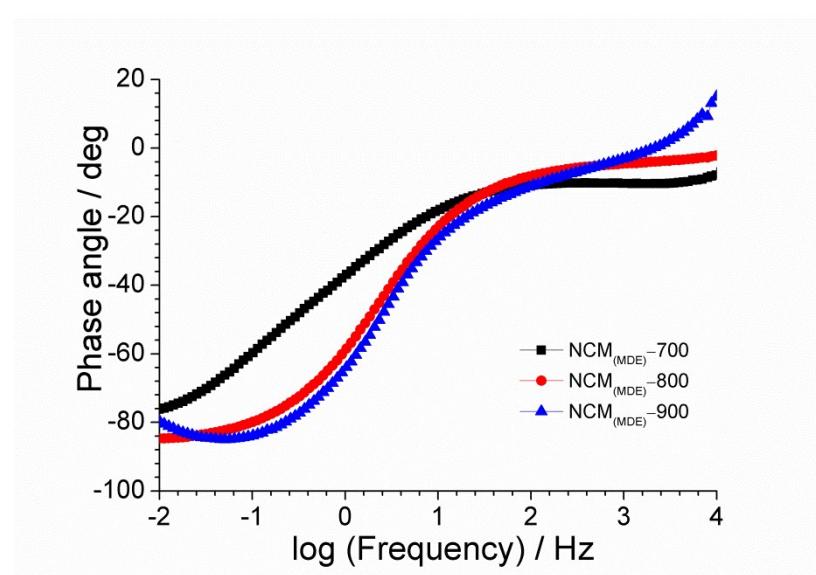


Figure S6: Bode plots for NCM_(MDE)-700, NCM_(MDE)-800 and NCM_(MDE)-900 electrodes

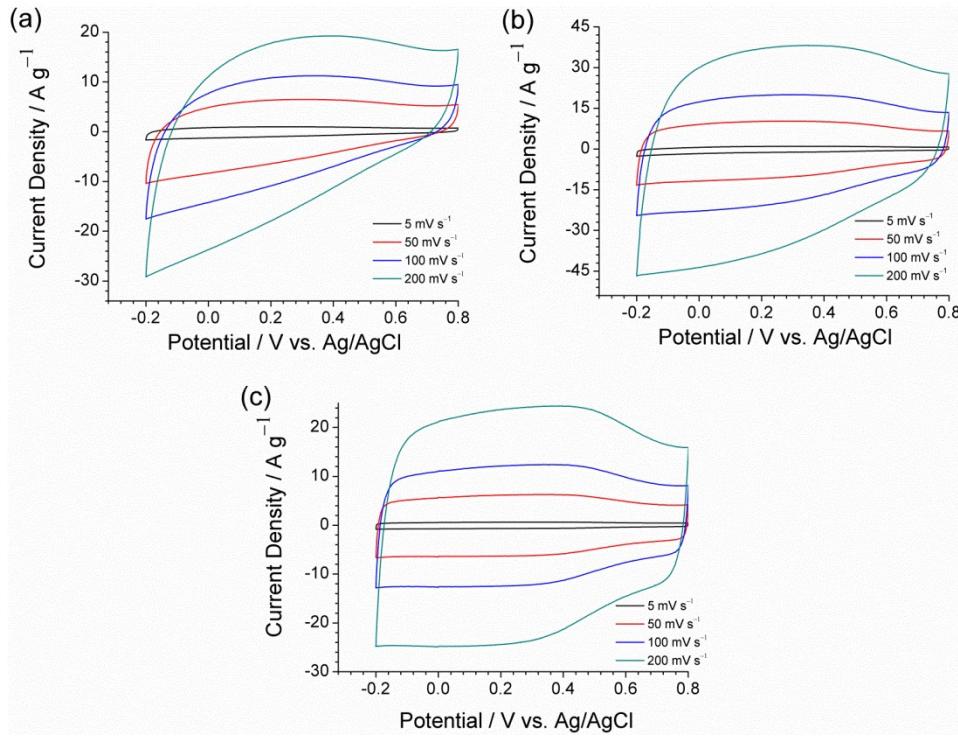


Figure S7: Cyclic voltammograms of (a) NCM_(MDE)-700, (b) NCM_(MDE)-800, and (c) NCM_(MDE)-900 at different scan rates in 1.0 M H₂SO₄.

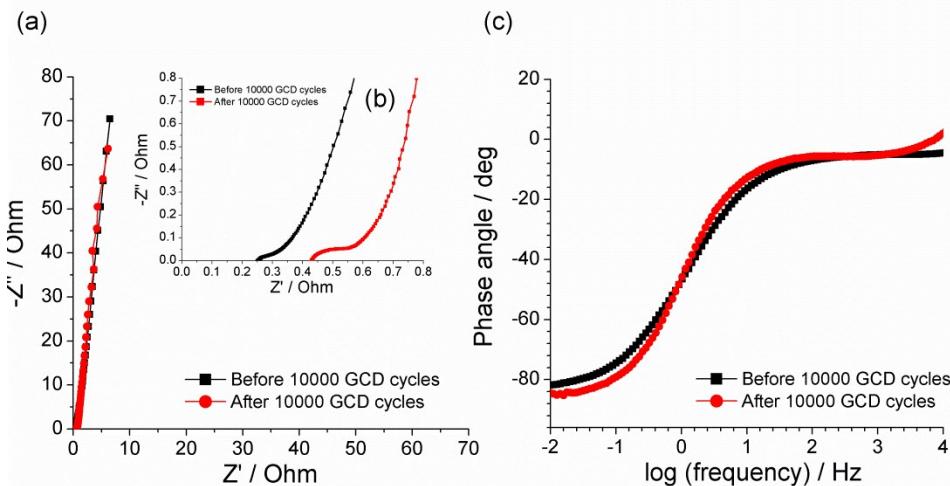


Figure S8: Electrochemical impedance spectra; (a) Nyquist plot, (b) Nyquist plot: magnified region 0-0.8Ω and (c) Bode plot of NDM_(MDE)-800 electrode show the effect of long CD cycling upto 10,000cycles at 10A g⁻¹.

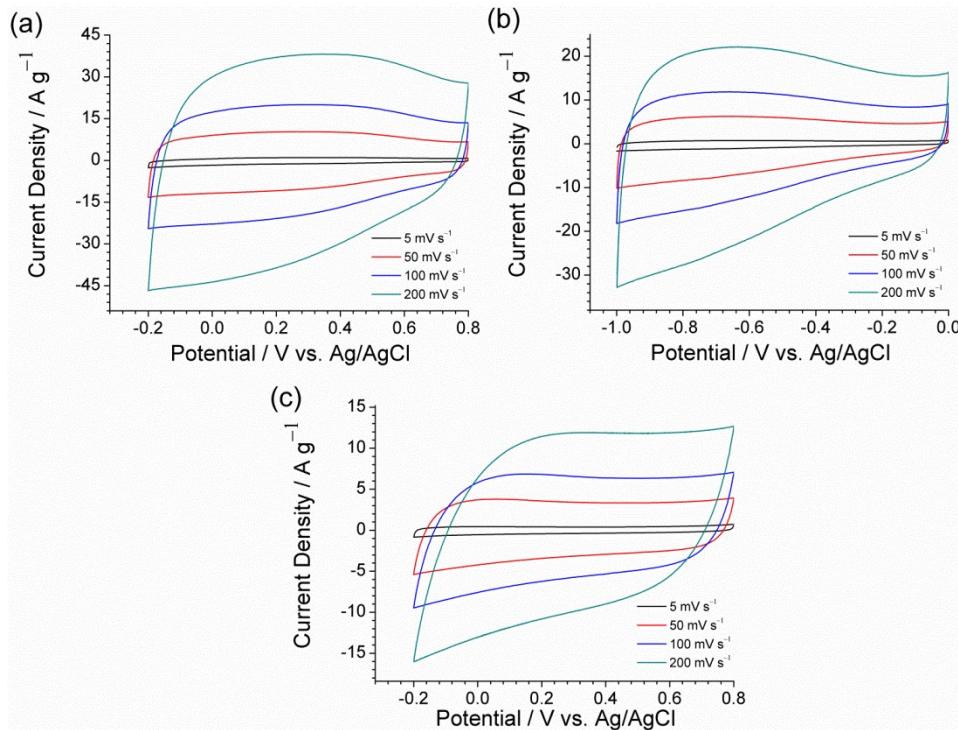


Figure S9: Cyclic voltammograms of $\text{NCM}_{(\text{MDE})}-800$ electrode at different scan rates in different electrolytes; **(a)** 1.0 M H_2SO_4 , **(b)** 6.0 M KOH and **(c)** 0.5 M Na_2SO_4

No major shape changes are observed in the CV curves recorded in 1 M $\text{H}_2\text{SO}_4_{(\text{aq})}$ and 6 M $\text{KOH}_{(\text{aq})}$ with increasing scan rate, particularly at high scan rate of 200mV s^{-1} Figure S9. This indicates that $\text{NCM}_{(\text{MDE})}-800$ has high rate capability in both acidic and basic electrolyte media. In contrast, the CV curves recorded in 0.5 M $\text{Na}_2\text{SO}_4_{(\text{aq})}$, clearly show a distortion in the shape as the scan rate increases, indicating a poor rate capability in this electrolyte system when compared with acid and basic media.