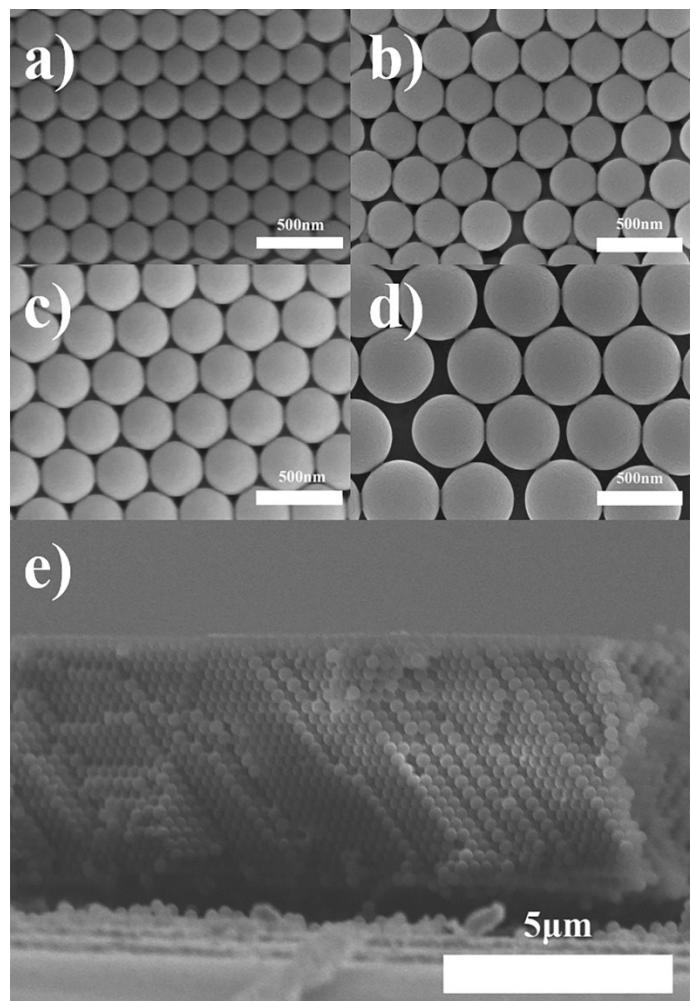


## Supporting Information

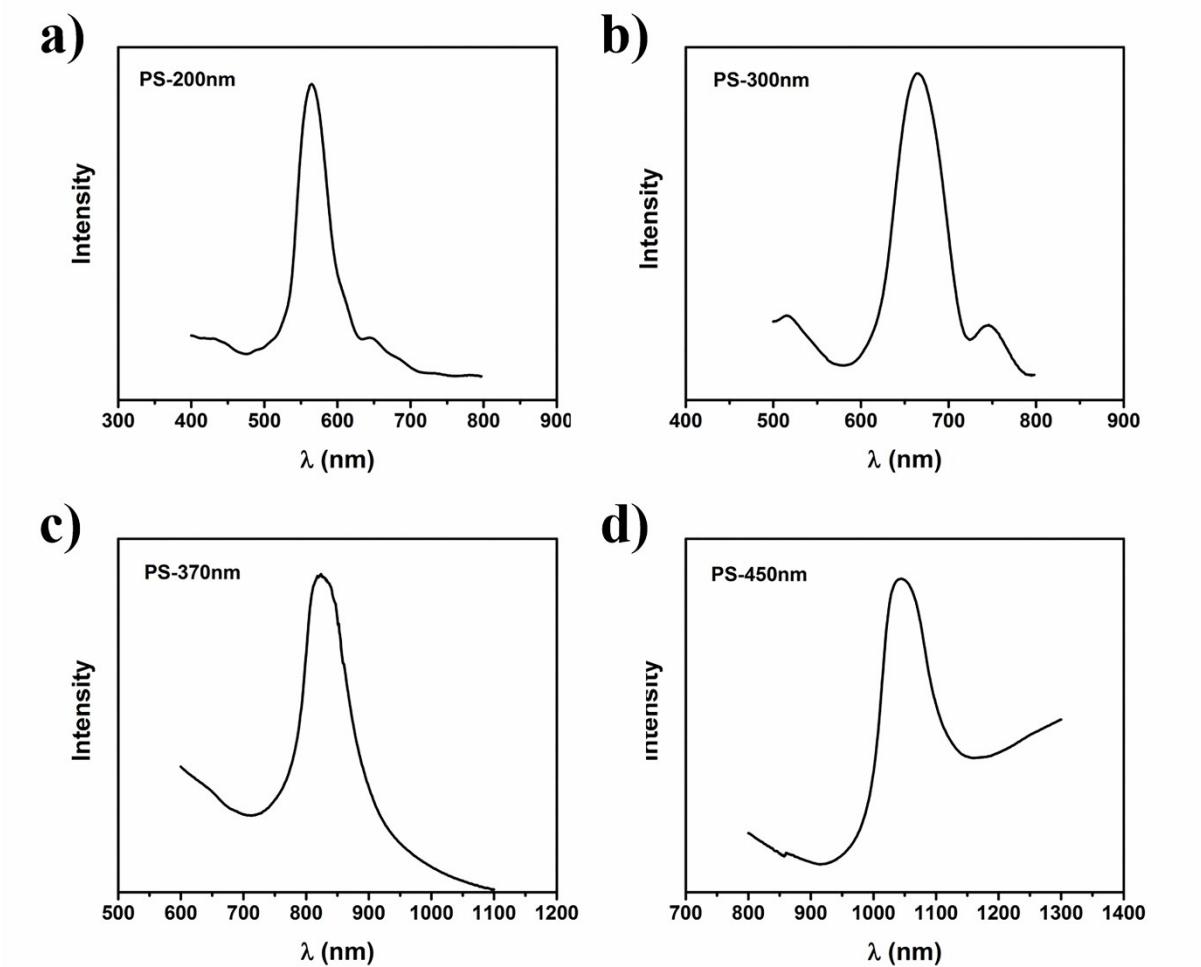
### Controllable Fabrication of Metallic Photonic Crystals for Ultra-Sensitive SERS and Photodetector

*Zihe Cai, Yang Yan, Lin Liu, Shengxuan Lin and Xiaobin Hu\**

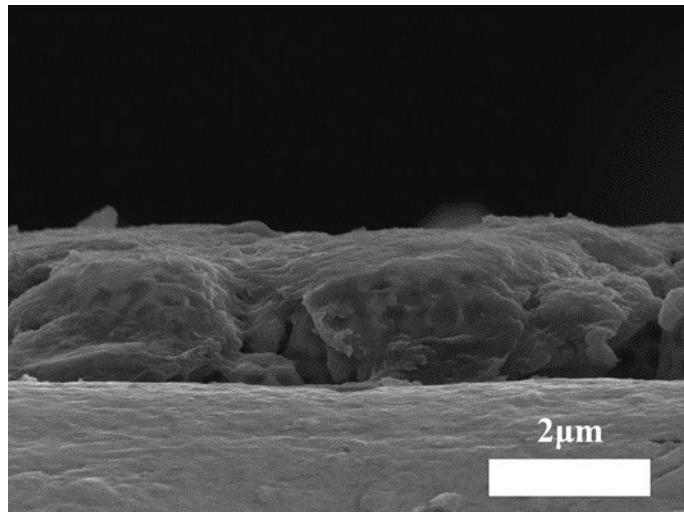
*State Key Laboratory of Metal Matrix Composites, Shanghai JiaoTong University, Shanghai 200240, People's Republic of China*



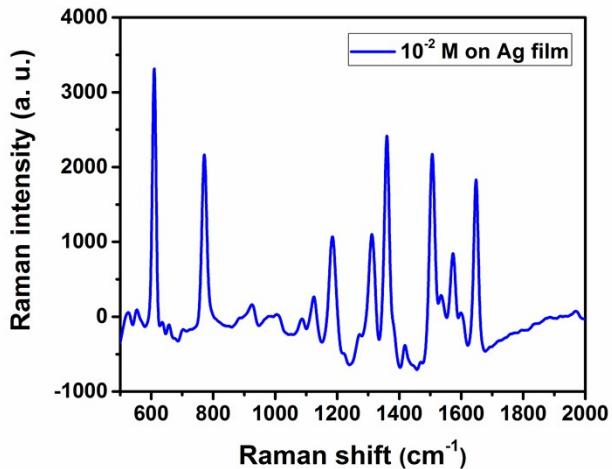
**Figure S1.** (a-e) SEM images of fabricated PS colloidal crystals templates on FTO glass slides after self-assembly, (a) 200 nm; (b) 300 nm; (c) 370 nm; (d) 450 nm; (e) top-view SEM image of PS template with diameter of 300nm.



**Figure S2.** (a-d) Reflection spectra of PS colloidal crystals templates on FTO glass slides, (a) 200 nm; (b) 300 nm; (c) 370 nm; (d) 450 nm.



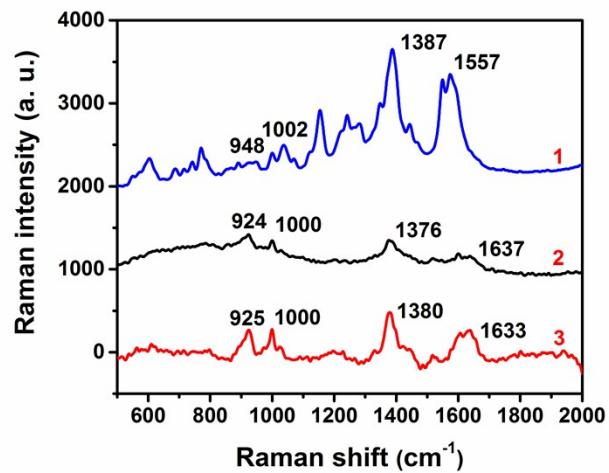
**Figure S3.** Cross-sectional view SEM image of Ag MPCs with diameter of 370nm.



**Figure S4.** Raman spectrum of  $10^{-2}$  M R6G collected from a normal Ag film substrate on FTO glass slides.

**Table S1.** A comparison between Metallic Inverse opal and recently reported noble-metal nanostructures in terms of the SERS enhancement factor of R6G.

Current results	Structure	EF
This work	Metallic Inverse Opal	$4.8 \times 10^9$
ACS Nano, 2011 <sup>1</sup>	Metallo-Dielectric Photonic Crystals	$0.76 \times 10^8$
ACS Nano, 2017 <sup>2</sup>	Au-Functionalized Si Nanorod Arrays	$3.3 \times 10^7$
ACS. Appl. Mater. Inter, 2012 <sup>3</sup>	Au/SiO <sub>2</sub> Nanocomposites	$5.4 \times 10^8$
Chem. Commun., 2011 <sup>4</sup>	Gold Nanoparticles	$5.8 \times 10^7$



**Figure S5.** Raman spectra of  $10^{-13}$  M R6G collected from one Ag-200nm sample where three different positions were chosen randomly for measurement.

**Table S2.** Detailed electrodeposition solutions and corresponding reaction conditions.

Materials	Deposition solutions	t (min)	T (°C)	Potential (V) or current density (mA/cm <sup>-2</sup> )
Ag	Silver nitrate (1.7g), EDTA (3.7g), boric acid (0.06g), potassium nitrate (1g), ammonium hydroxide, PH=9~10, deionized water (100ml)	5min	25°C	2.5 mA/cm <sup>-2</sup>
Cu	Copper sulfate (4g), trisodium citrate (4.7g), glucose (0.36g), SDS (0.02g), 4mol/L sodium hydroxide (0.5g), deionized water (100ml)	5min	25°C	3.5 mA/cm <sup>-2</sup>
Ni	Nickel sulfate (30g), nickel chloride (4.5g), boric acid (4g), glucose (1g), SDS (0.02g), deionized water (100ml)	10min	25°C	-0.90V
Co	Cobalt sulfate (30g), cobalt chloride (4.5g), boric acid (4g), glucose (1g), SDS (0.02g), deionized water (100ml)	10min	25°C	-0.95V

## Reference

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- 4 F. D. Mai, T. C. Hsu, Y. C. Liu, K. H. Yang and B. C. Chen, *Chem. Commun.*, 2011, **47**, 2958-2960.

