

Silver NPs dispersed water extract of fly ash as green and efficient medium for oxidant-free dehydrogenation of benzyl alcohols

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TEM image of spent catalyst:

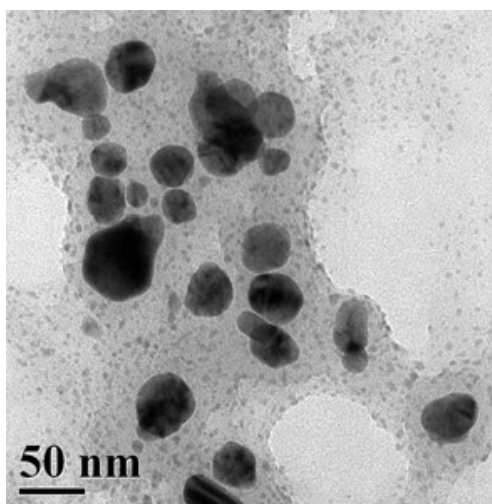


Fig.S1. TEM images of the spent Ag@WEFA catalyst

Spectral & analytical data of compounds:

(1a) Benzaldehyde: Colourless liquid; ^1H NMR (400 MHz, CDCl_3) δ 9.71 (s, 1H), 7.59 (m, $J=7.4$ Hz, 2H), 7.30 (t, 1H), 7.19 (t, 2H).

(2a) 4-Methylbenzaldehyde: Colourless liquid; ^1H NMR (400 MHz, CDCl_3) δ 9.57 (s, 1H), 7.37 (d, $J=7.2$ Hz, 2H), 6.87 (d, $J=7.2$ Hz, 2H), 1.96 (s, 3H).

(3a) 4-Chlorobenzaldehyde: Colourless solid; m.p. 50°C ; ^1H NMR (400 MHz, CDCl_3) δ 9.98 (s, 1H), 7.83 (d, $J=7$ Hz, 2H), 7.51 (d, $J=6.5$ Hz, 2H).

(4a) 4-Nitrobenzaldehyde: Pale yellow solid; m.p. 105°C ; ^1H NMR (400 MHz, CDCl_3) δ 10.18 (s, 1H), 8.41 (s, 1H), 8.12 (d, $J=6.5$ Hz, 1H), 8.09 (d, $J=6.5$ Hz, 1H), 7.30 (t, $J=6$ Hz, 1H).

(5a) 4-Hydroxybenzaldehyde: Colourless solid; m.p. 112°C ; ^1H (400 MHz, CDCl_3): δ =9.83 (s, 1H), 7.80 (d, $J=8.4$ Hz, 1H), 6.95 (d, $J=8.4$ Hz, 1H).

(6a) 4-Dimethylaminobenzaldehyde: Yellowish white powder; m.p. 72°C ; ^1H (400 MHz, CDCl_3): δ = 9.79 (s, 1H), 7.79 (d, $J=7.5$ Hz, 2H), 6.75 (d, $J=7.5$ Hz, 2H), 3.12 (s, 6H).

(7a) 4-methoxy benzaldehyde: Colourless liquid; ^1H (400 MHz, CDCl_3): δ 9.86 (s, 1H), 7.83 (d, $J=7.0$ Hz, 2H), 7.80 (d, $J=8.0$ Hz, 2H), 3.92 (s, 3H).

(8a) 4-Florobenzaldehyde: Colourless liquid; ^1H (400 MHz, CDCl_3): δ 7.19-7.24 (m, 2H), 7.90-7.94 (m, $J=5.8$ Hz, 2H), 9.97 (s, 1H).

(9a) 4-Bromobenzaldehyde: Colourless solid; m.p. $56-57^\circ\text{C}$; ^1H NMR (400 MHz, CDCl_3) δ 9.98 (s, 1H), 7.76 (d, $J=6.4$ Hz, 2H), 7.70 (d, $J=6.4$ Hz, 2H).

(10a) 3-Chlorobenzaldehyde: Colourless liquid; ^1H NMR (CDCl_3) δ 9.93 (s, 1H) 7.76 (s, 1H), 7.72 (d, $J=6.5$ Hz, 1H), 7.50 (d, $J=7.0$ Hz, 1H), 7.42 (d, $J=6.5$ Hz, 1H).

(11a) 3-Bromobenzaldehyde: Yellow liquid; $^1\text{H NMR}$ (400MHz, CDCl_3) $\delta=9.96$ (s, 1H), 7.61 (s, 1H), 7.50 (d, $J = 7.5$ Hz, 1H), 7.38 (d, $J = 7.0$ Hz, 1H), 7.11 (d, $J = 7.5$ Hz, 1H).

(12a) 3-Nitrobenzaldehyde: Yellow solid; m.p. 58°C ; $^1\text{H NMR}$ (CDCl_3) δ 10.13 (s, 1H), 8.71 (s, 1H), 8.49 (d, $J = 7.5$ Hz, 1H), 8.25 (d, $J = 7.5$ Hz, 1H), 7.80 (t, $J = 7.5$ Hz, 1H).

(13a) 2-Chlorobenzaldehyde: Colorless liquid; $^1\text{H NMR}$ (400MHz, CDCl_3) δ 10.33 (s, 1H), 7.78 (m, 1H), 7.75 (m, 1H), 7.34 (m, 1H), 7.27 (m, 1H).

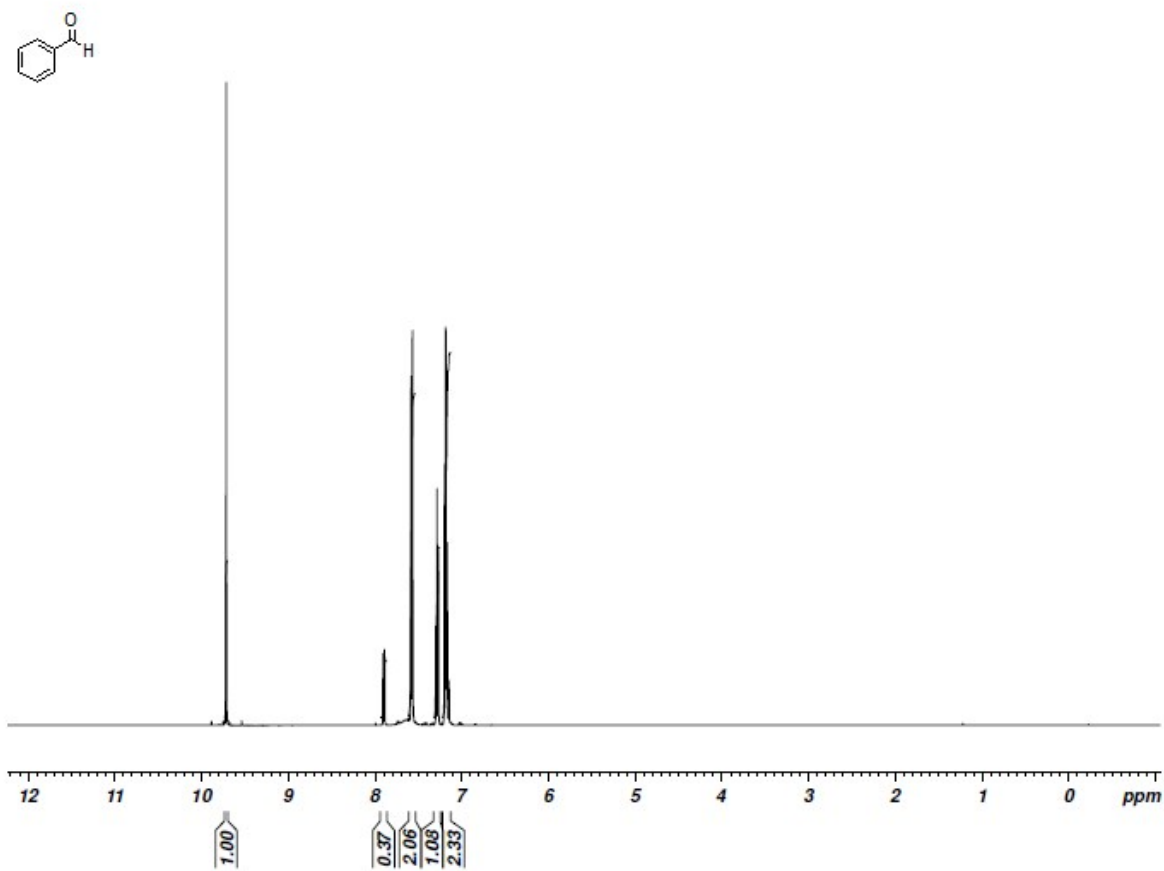


Fig.S2. $^1\text{H NMR}$ spectrum of compound (1a) in CDCl_3 .

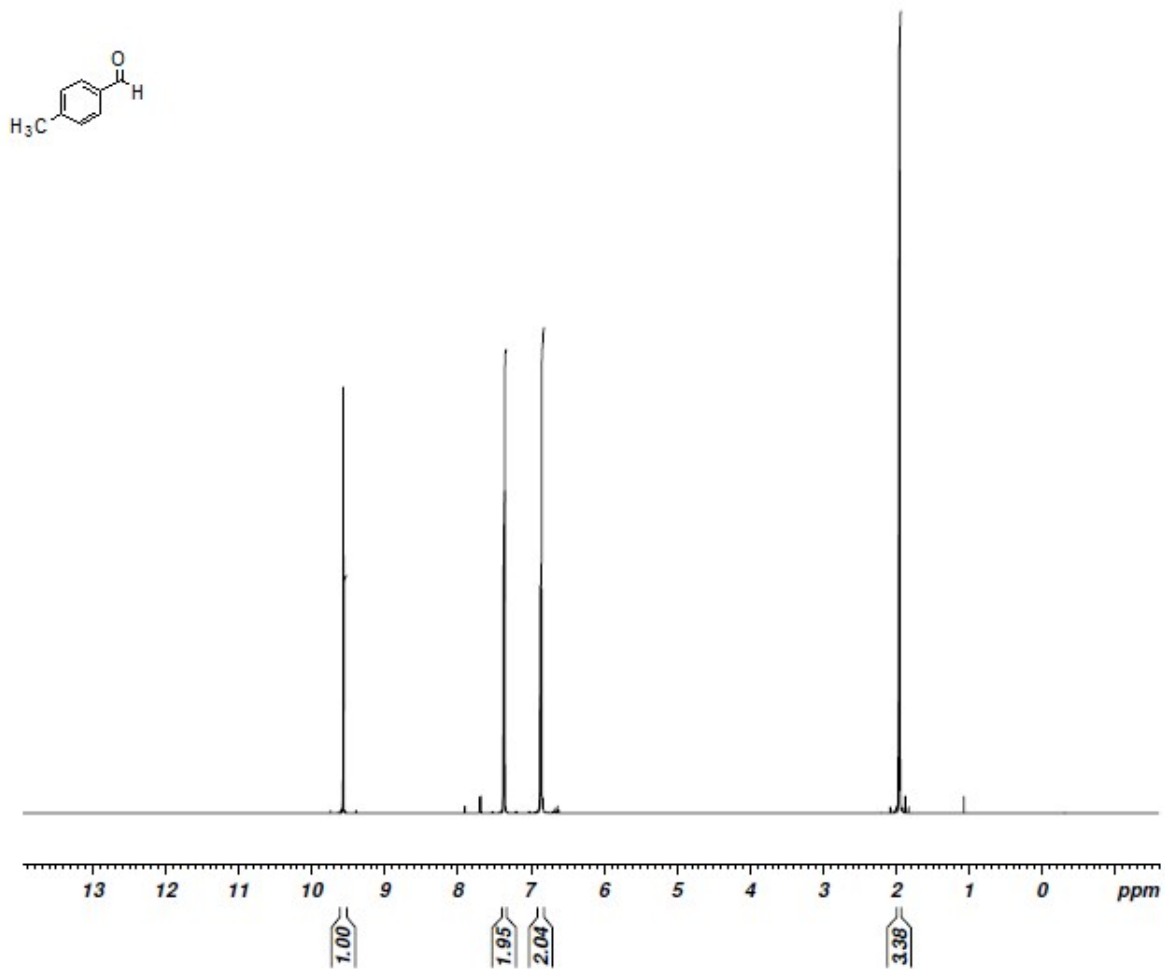


Fig.S3. ¹H NMR spectrum of compound (2a) in CDCl₃.

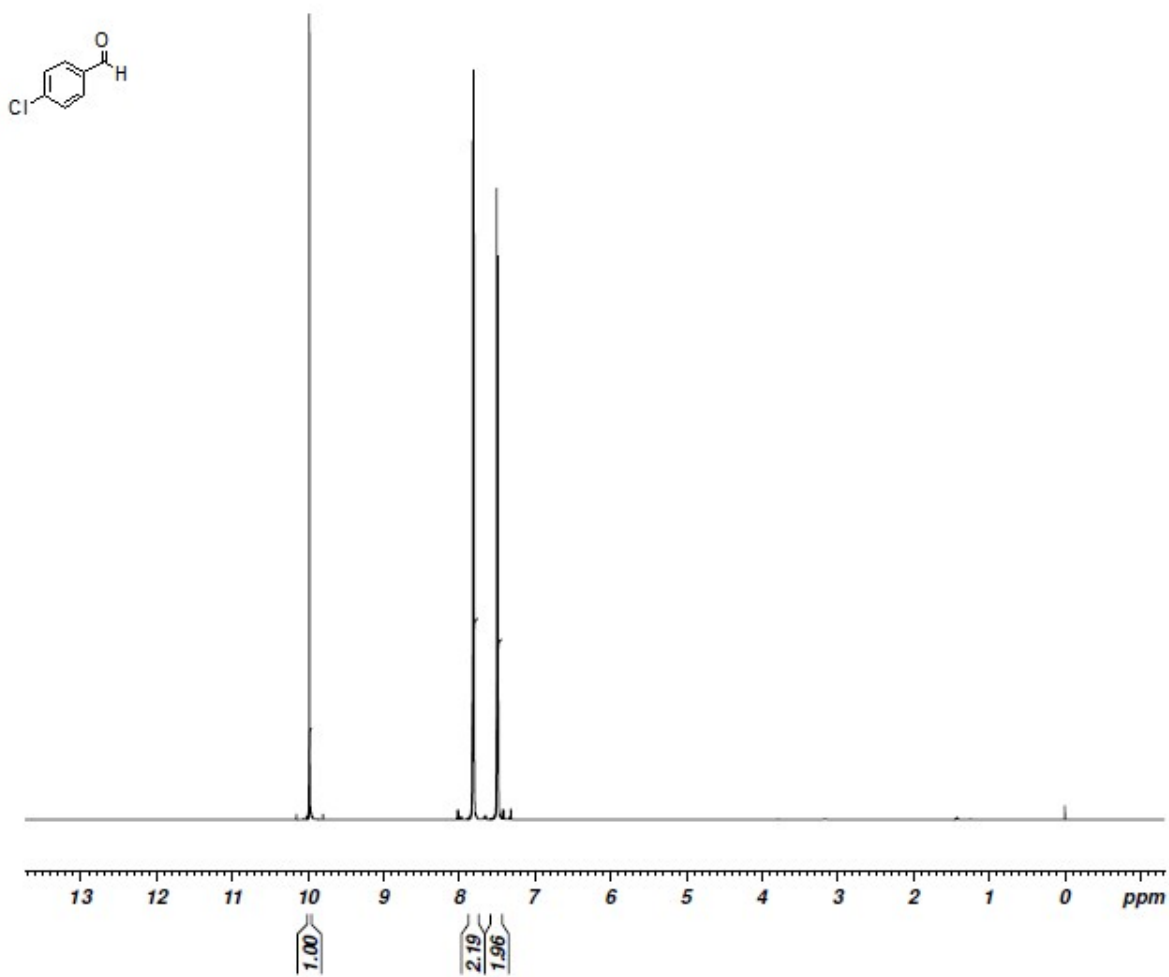


Fig.S4. ¹H NMR spectrum of compound (3a) in CDCl₃.

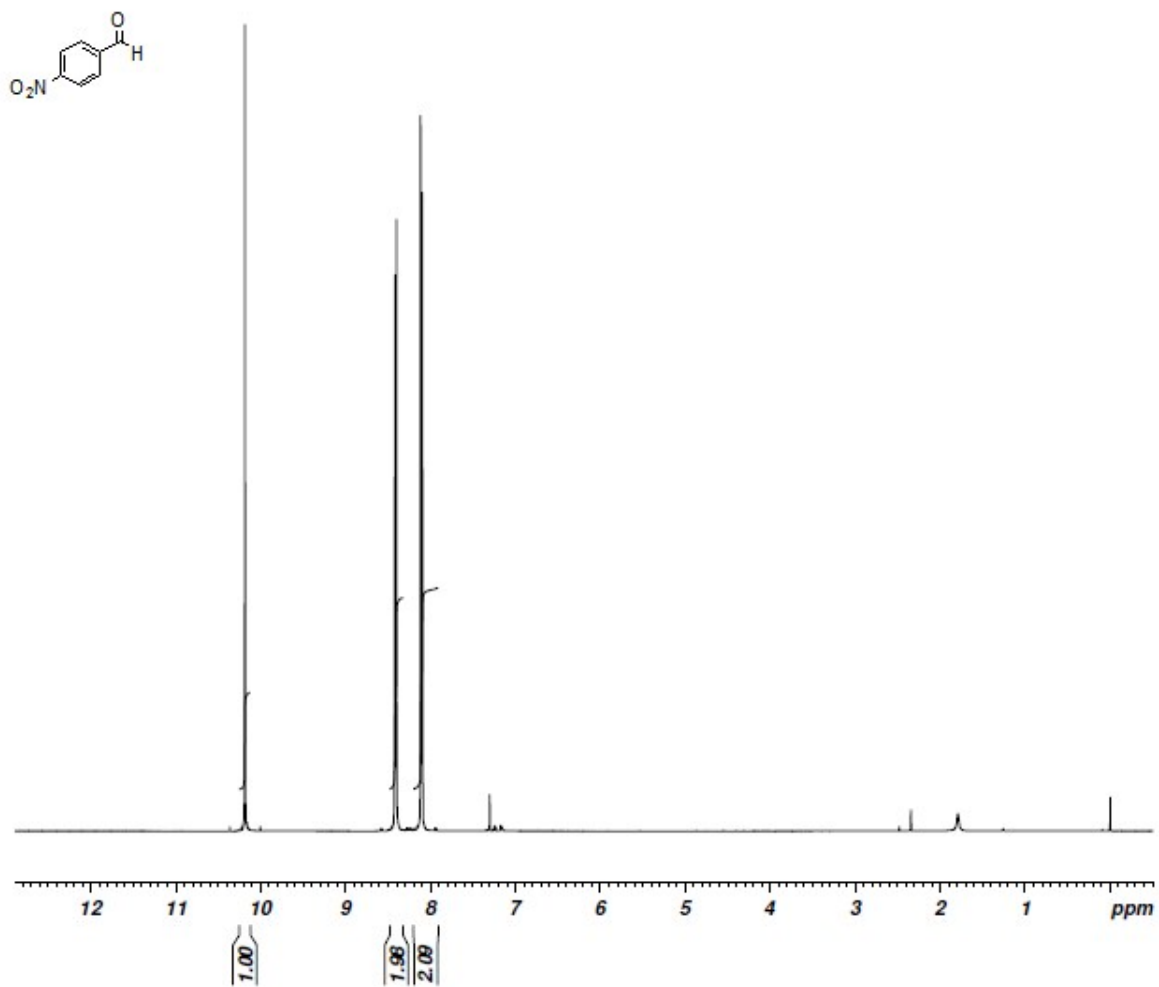


Fig.S5. ^1H NMR spectrum of compound (4a) in CDCl_3 .

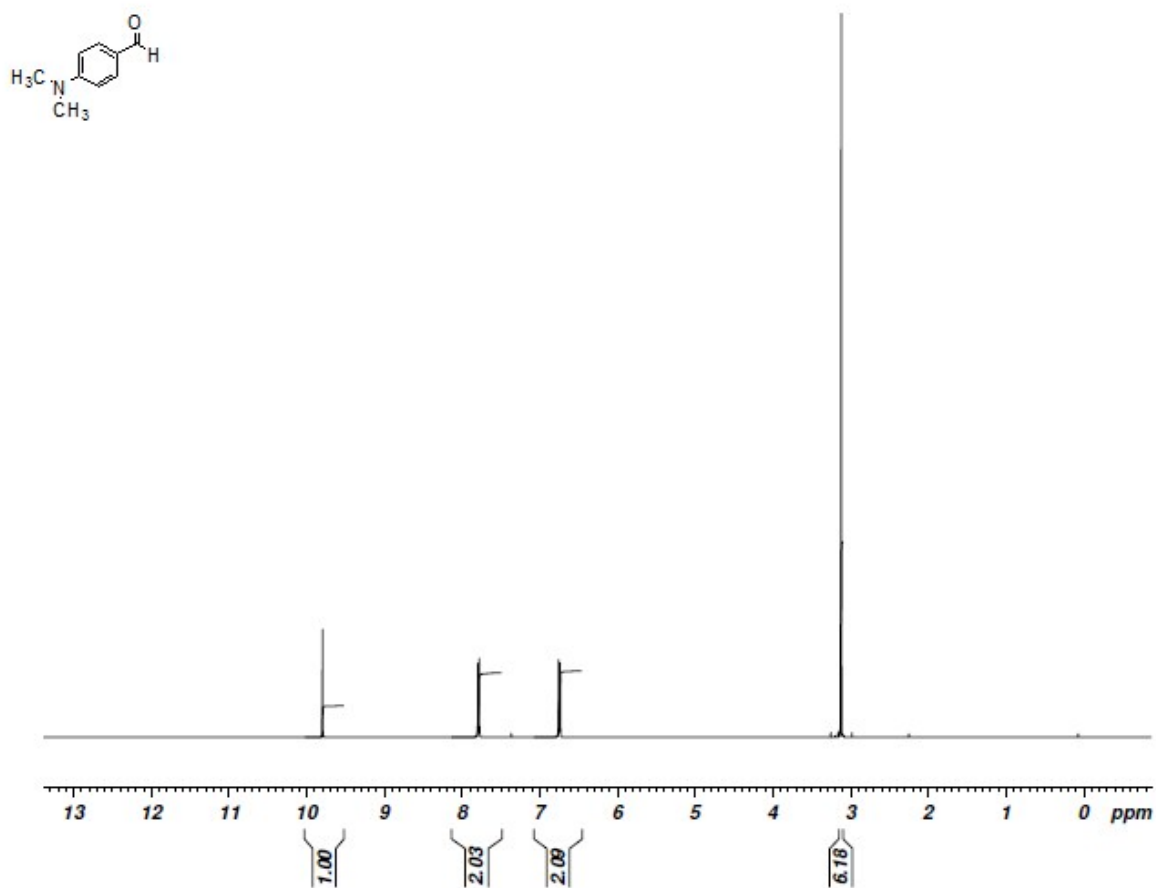


Fig.S6. ¹H NMR spectrum of compound (6a) in CDCl₃.

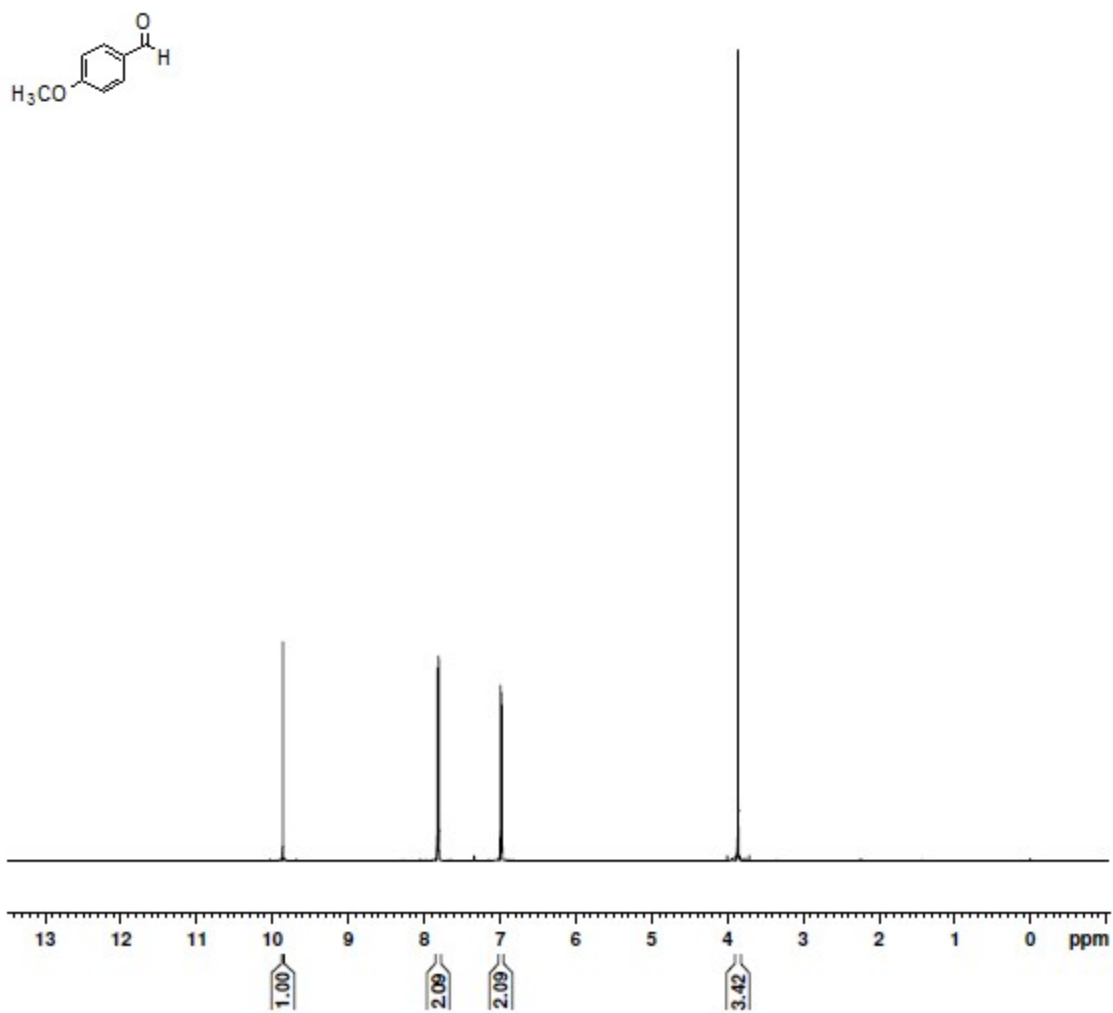


Fig.S7. ¹H NMR spectrum of compound (7a) in CDCl₃.

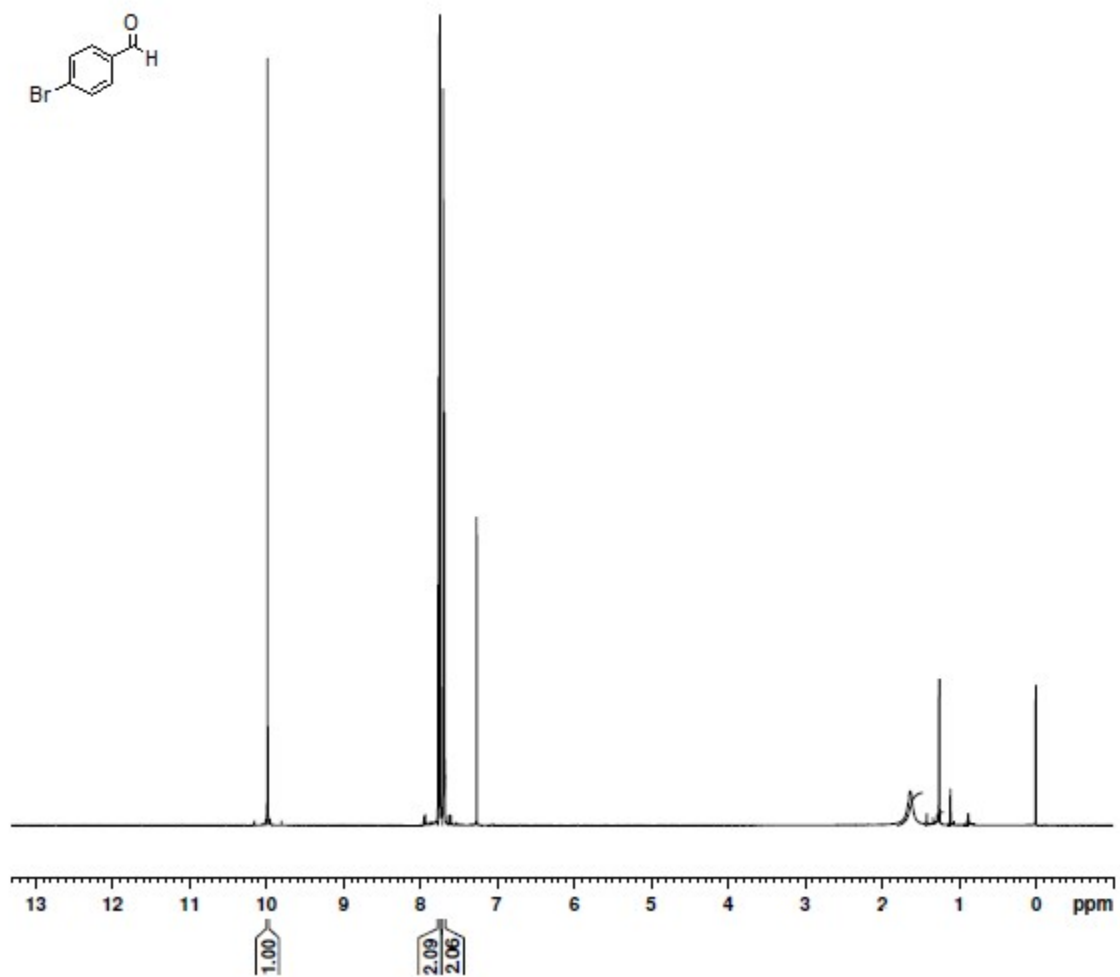


Fig.S8. ¹H NMR spectrum of compound (9a) in CDCl₃.

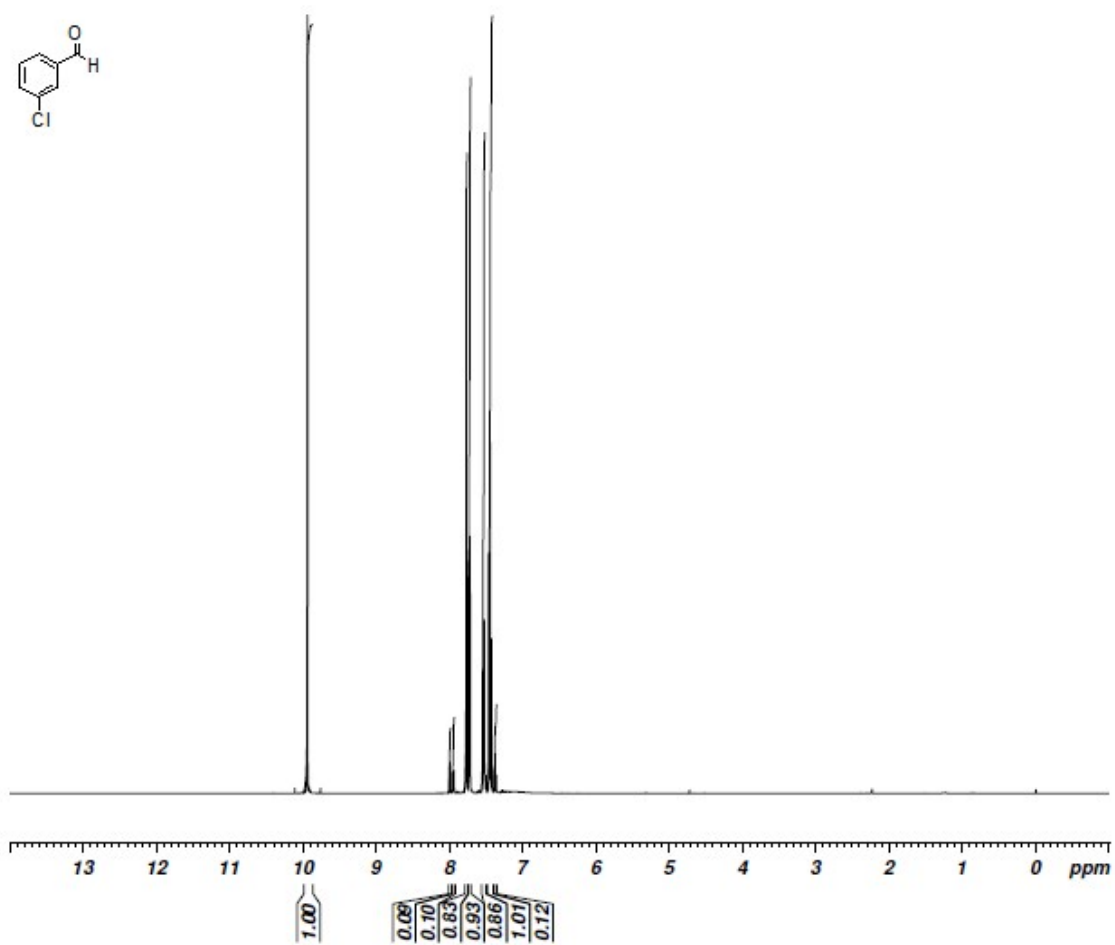


Fig.S9. ¹H NMR spectrum of compound (10a) in CDCl₃.

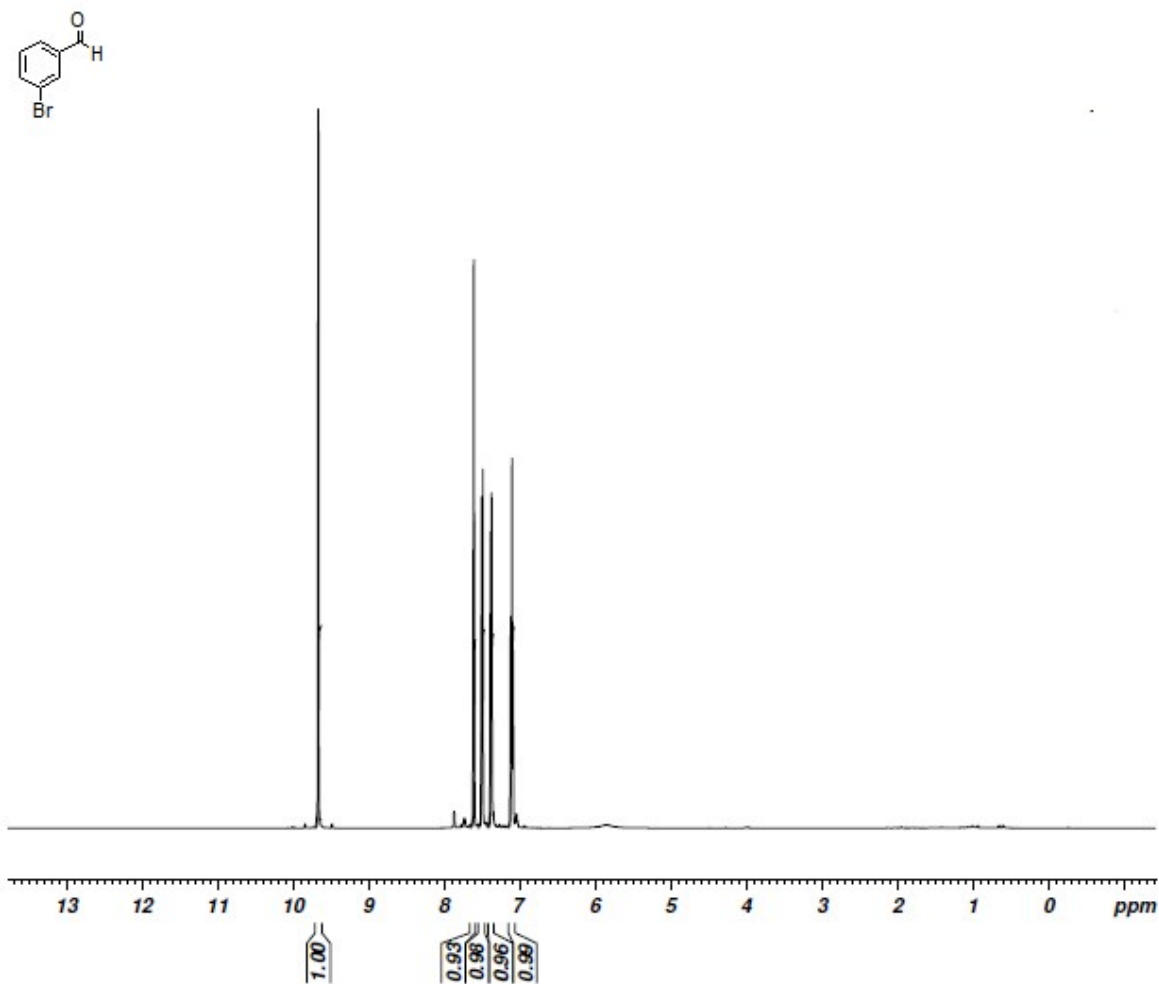


Fig.S10. ¹H NMR spectrum of compound (11a) in CDCl₃.

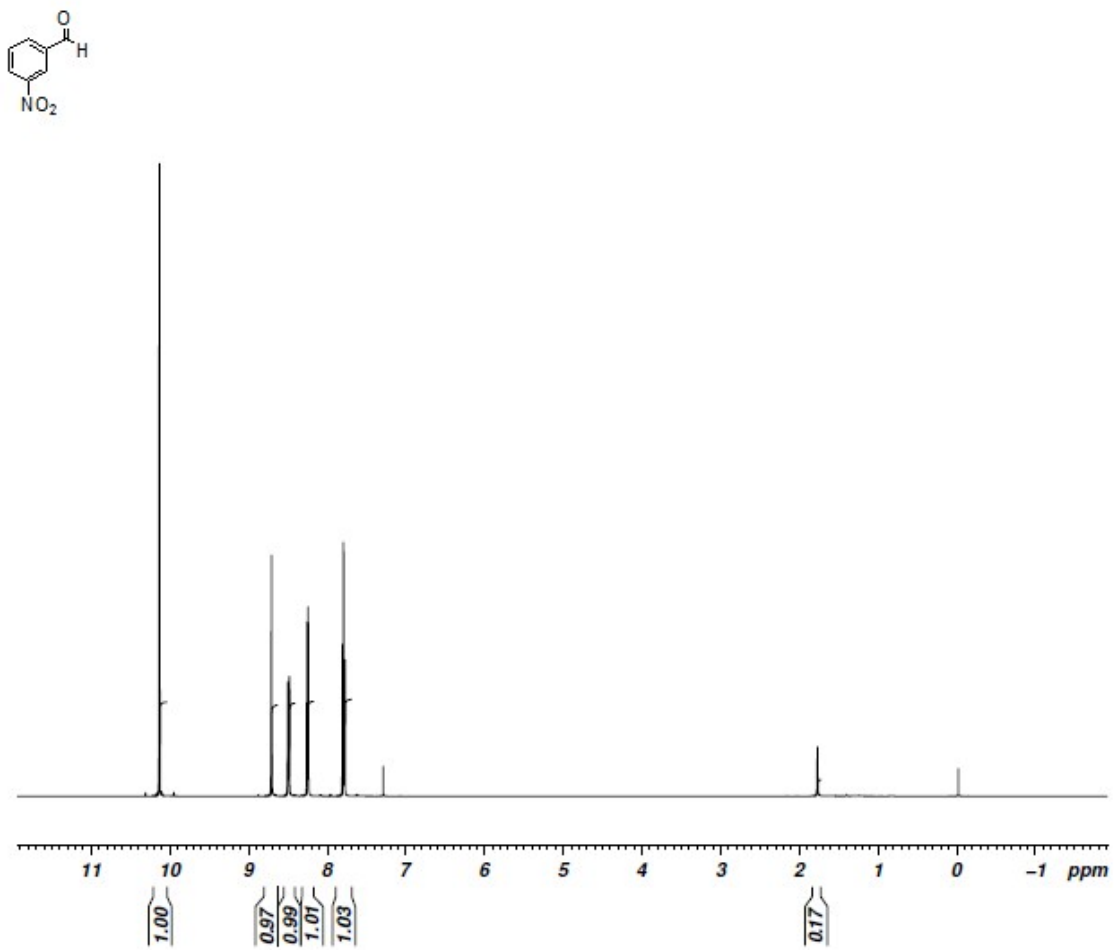


Fig.S11. ¹H NMR spectrum of compound (12a) in CDCl₃.

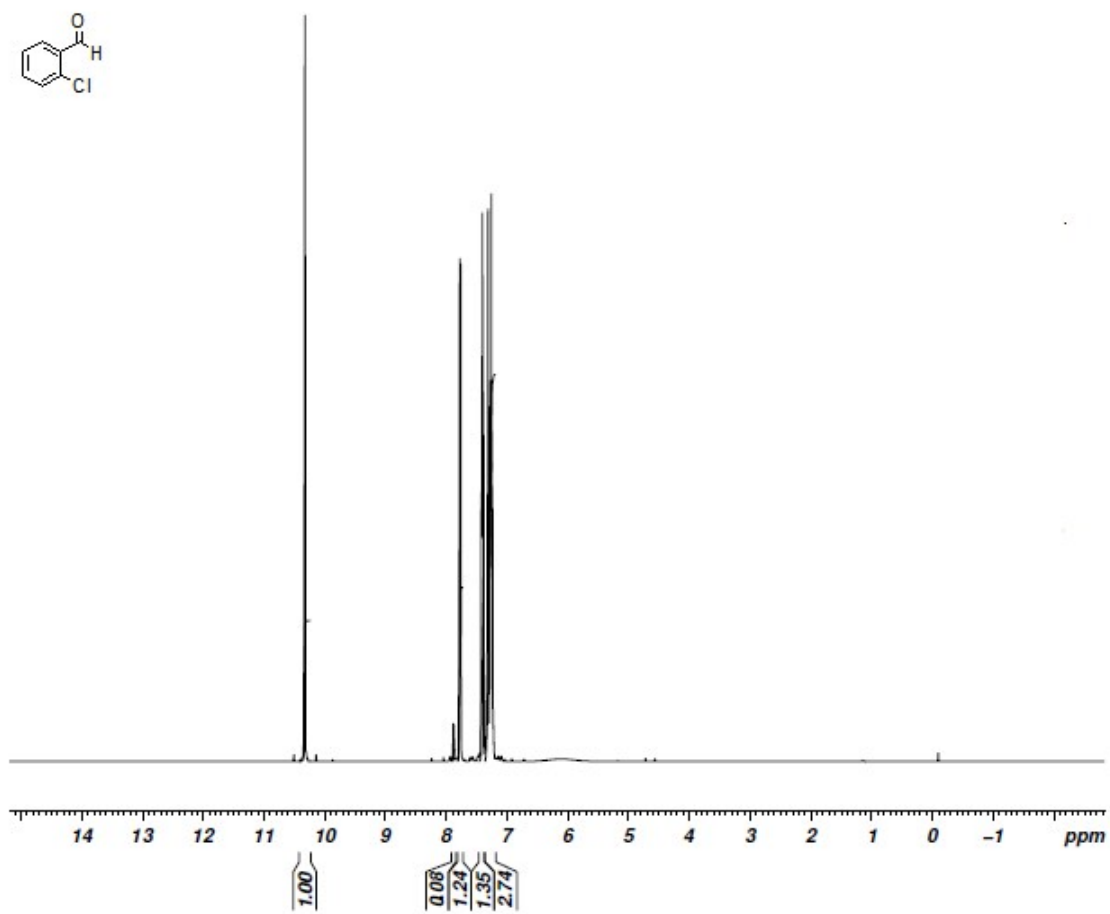


Fig.S12. ^1H NMR spectrum of compound (13a) in CDCl_3 .

Table.S1: Recyclability of the catalytic system comprising Ag@WEFA

Entry	Run	Time (h)	Yield (%)
1	1 st	3	96
2	2 nd	3	94
3	3 rd	3.5	93
4	4 th	3.5	92
5	5 th	4	90

Reaction conditions: 4-methylbenzyl alcohol (1 mmol), Ag@WEFA (3 mL), reaction temperature 70 °C, N_2 atmosphere