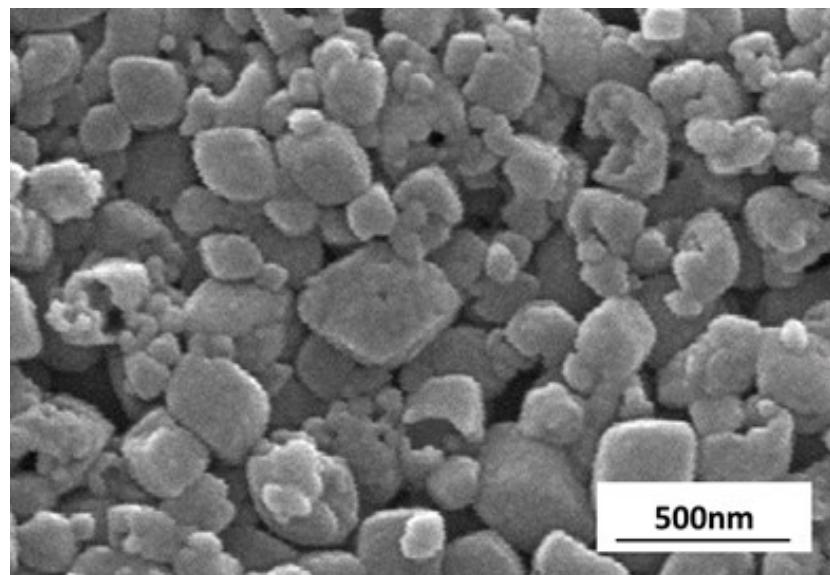


## Supporting Information for

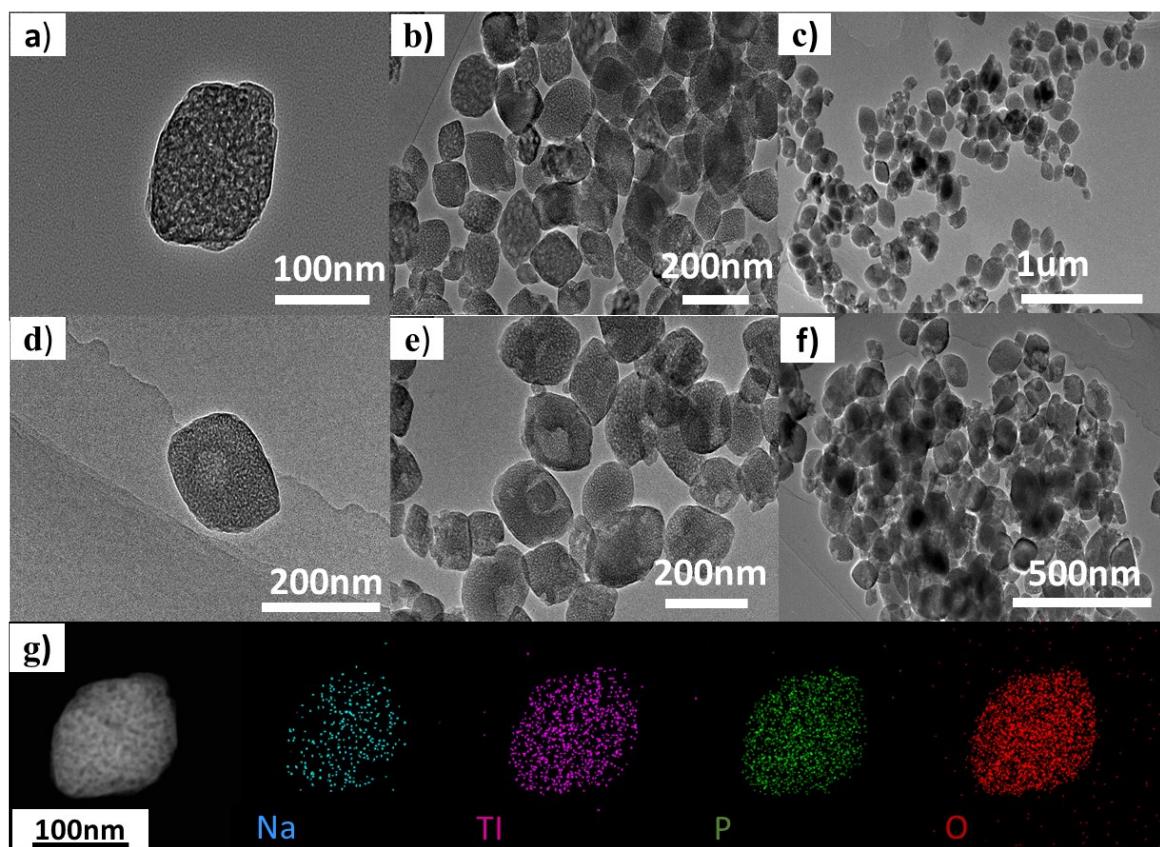
### **Self-generated Hollow NaTi<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> Nanocubes Decorated with Graphene as a Large Capacity and Long Lifetime Anode for Sodium-ion Batteries**

**Shaocheng Ye<sup>a</sup>, Zhihong Li<sup>a</sup>, Tianbing Song<sup>a</sup>, Danhong Cheng<sup>a</sup>,**

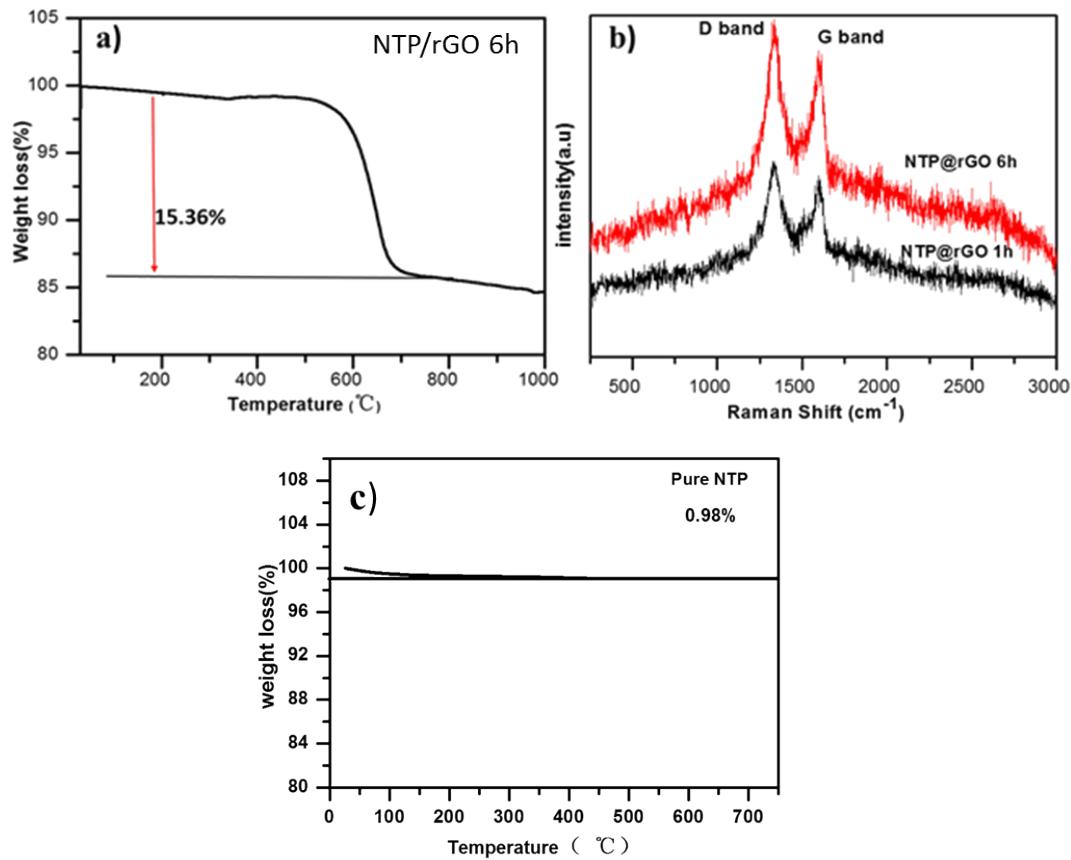
**Qunjie Xu<sup>a</sup> ,Haimei Liu <sup>a</sup>, Yonggang Wang<sup>b</sup>**



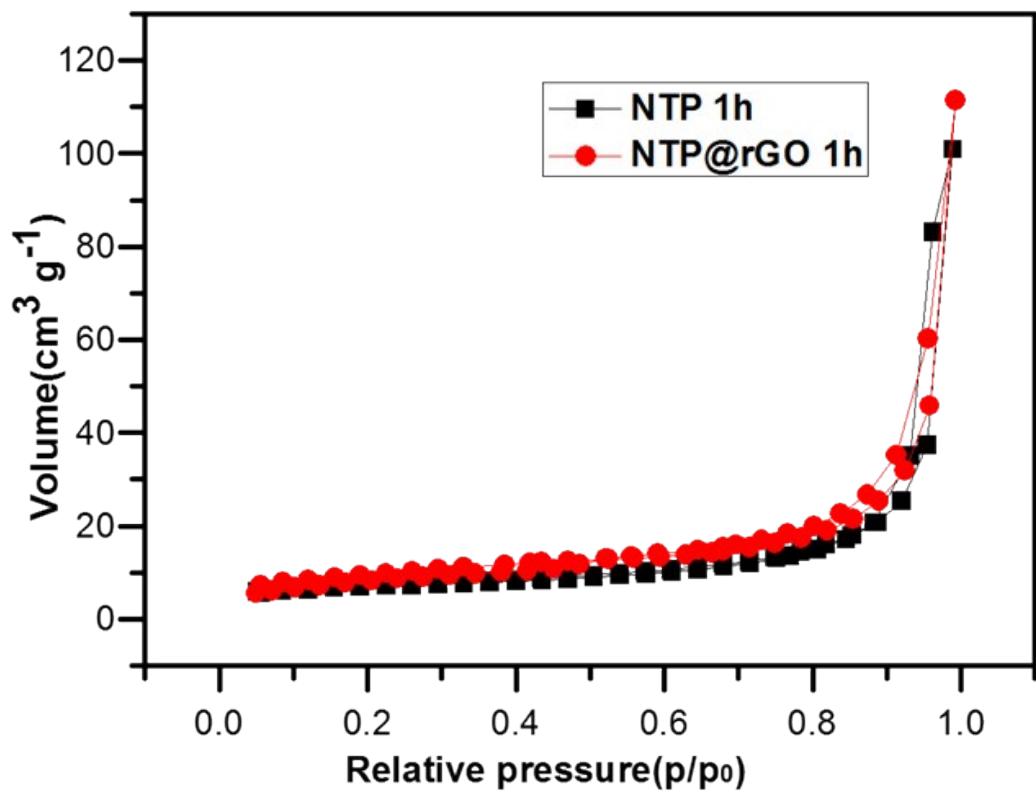
**Fig.S1. SEM images of pure NaTi<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> with hydrothermal time of 9hs.**



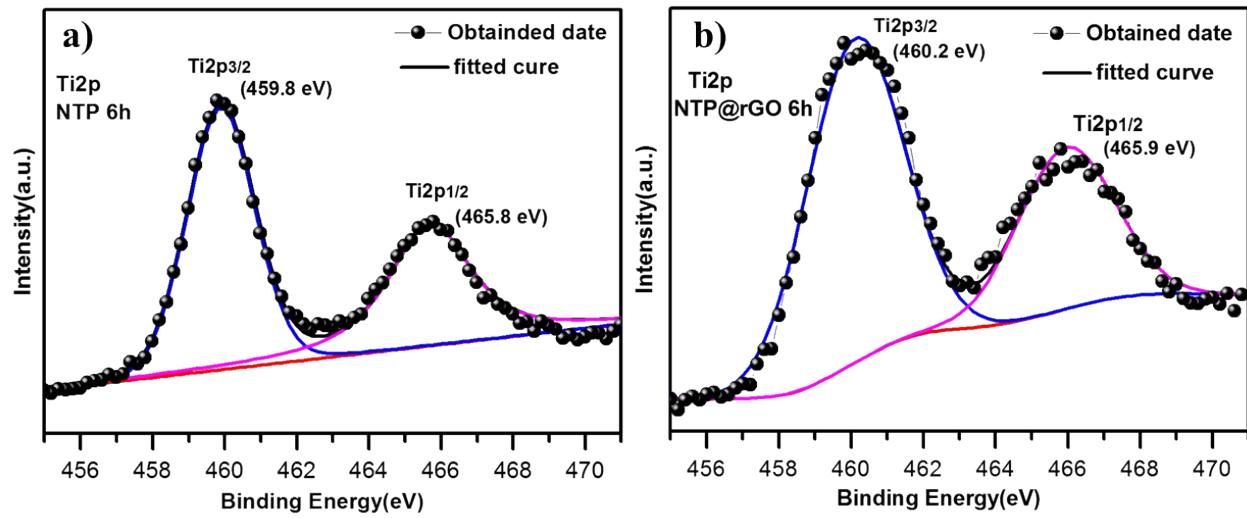
**Fig.S2.** a, b, c) TEM images of  $\text{NaTi}_2(\text{PO}_4)_3$  1h; d, e, f) showing the  $\text{NaTi}_2(\text{PO}_4)_3$  6h; g) TEM image of  $\text{NaTi}_2(\text{PO}_4)_3$  1h and the corresponding EDX mapping of Na (blue), Ti (purple), P (green), O (red).



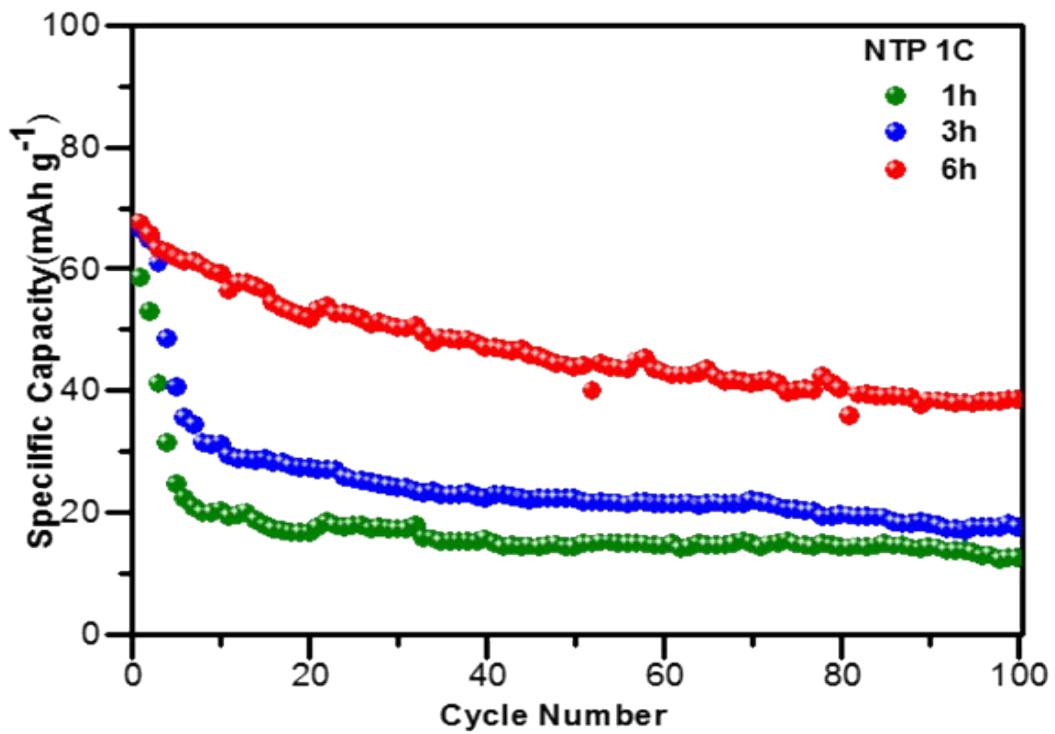
**Fig. S3.** a) The TG curve of the  $\text{NaTi}_2(\text{PO}_4)_3@\text{rGO}$  6h sample ; b) Raman spectra of the  $\text{NaTi}_2(\text{PO}_4)_3/\text{rGO}$  1h and  $\text{NaTi}_2(\text{PO}_4)_3/\text{rGO}$  6h composites; c) The TG curve of the pure  $\text{NaTi}_2(\text{PO}_4)_3$  sample.



**Fig.S4.** a)  $\text{N}_2$  absorption–desorption isotherms of the  $\text{NaTi}_2(\text{PO}_4)_3$  1h and  $\text{NaTi}_2(\text{PO}_4)_3/\text{rGO}$  1h.



**Fig.S5.** a) XPS survey spectra of the  $\text{NaTi}_2(\text{PO}_4)_3$  6h and  $\text{NaTi}_2(\text{PO}_4)_3@\text{rGO}$  6h; a, b) Deconvoluted  $\text{Ti}2\text{p}$  peaks of the  $\text{NaTi}_2(\text{PO}_4)_3$  6h and the  $\text{NaTi}_2(\text{PO}_4)_3@\text{rGO}$  6h composite.



**Fig.S6.** Cycling performance at 1C of pure  $\text{NaTi}_2(\text{PO}_4)_3$  prepared with hydrothermal times of 1h, 3h, and 6h .

**Table S1.** Comparison of electrochemical performance of different NTP materials

Electrode definition	Specific capacity (mAh g <sup>-1</sup> )	Cycle performance	References
H-NTP NC@rGO	128mAh g <sup>-1</sup> at 1C	120mAh g <sup>-1</sup> after 150 cycles at 1C	<b>This work</b>
	118 mAh g <sup>-1</sup> at 3C	103mAh g <sup>-1</sup> after 500 cycles at 3C	
High rate performance of NTP@rGO	112 mAh g <sup>-1</sup> at 1C	100 mAh g <sup>-1</sup> after 150 cycles at 1C	S1
Porous NTP@rGO	138 mAh g <sup>-1</sup> at 1C	101 mAh g <sup>-1</sup> after 200 cycles at 1C	S2
Self-assembled wafer-like porousNTP decorated with hierarchical carbon	114mAh g <sup>-1</sup> at 2C	92 mAh g <sup>-1</sup> after 300 cycles at 2C	S3
Facile solvothermal synthesis of NaTi <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> /C porous	85mAh g <sup>-1</sup> at 10C	70mAh g <sup>-1</sup> after 120 cycles at 10C	S4
NTP Nanocubes with Synergistic Coating of Carbon and Rutile TiO <sub>2</sub>	86 mAh g <sup>-1</sup> at 5C	77.8 mAh g <sup>-1</sup> after 2000 cycles at 5C	S5

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