## The investigation of organic acids assisted sol-gel method for preparing monolithic zirconia aerogels

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Figure S1 Photos for the wet gel and the aerogel after supercritical drying, (a) LAA-4wet gel, (b) LAA-4-aerogel, (c) LMA-6-wet gel and (d) LMA-6-aerogel.



**Figure S2** Enlarged FT-IR spectra of LAA-4-xerogel from1800-1500 cm<sup>-1</sup> and 1400-1100 cm<sup>-1</sup>, and the possible peak positions for O=C-NH, amide II and amide III vibration bands are marked.



**Figure S3** Enlarged FT-IR spectra of MSA-8-aerogel from1800-1400 cm<sup>-1</sup> and 1300-1000 cm<sup>-1</sup>, and the possible peak positions for O=C-O are marked.



Figure S4 XPS spectra of LAA-4-aerogel, (a) C 1s, (b) O 1s and (c) N 1s.



Figure S5 XPS spectra of MSA-8-aerogel, (a) C 1s, (b) S 1s and (c) O 1s.



Figure S6 XPS spectra of in organic acid LMA and LMA-6-xerogel, (a) C 1sand (b) O

1s.



Figure S7 XPS spectra of organic acid MSA and MSA-8-xerogel, (a) C 1s, (b) S 2p and (c) O 1s.



Figure S8 XPS spectra of (a) C 1s and (b) O 1s in PO-8-aerogel.



Figure S9 XPS spectra of Zr 3d in LAA-4-aerogel and (b) PO-8-aerogel.



Figure S10 TEM images of MSA-8 aerogel (a) as-prepared and (b) after heat treatment at 1000 °C.



Figure S11 Temperature-dependent weight remaining and the corresponding derivatives of organic acid LAA.