Supporting Information

Racemic Hemiacetals as Oxygen-Centered Pronucleophiles Triggering Cascade 1,4-Addition/Michael Reaction through Dynamic Kinetic Resolution under Iminium catalysis. Reaction Development and Mechanistic Insights

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1. General Methods and Materials

Monodimensional and/or bidimensional nuclear magnetic resonance (NMR) proton and carbon spectra (¹H-NMR and ¹³C-NMR) were acquired at 25°C on a Bruker AC-300 spectrometer (300 MHz for ¹H, 75.5 MHz for ¹³C and 283 MHz for ¹⁹F) and a Bruker AC-500 spectrometer (500 MHz for ¹H and and 125.7 MHz ¹³C). Chemical shifts (δ) are reported in ppm relative to residual solvent signals (CHCl₃, 7.26 ppm for ¹H NMR, CDCl₃, 77.0 ppm for ¹³C NMR) and coupling constants (J) in hertz (Hz). The following abbreviations are used to indicate the multiplicity in ¹H NMR spectra: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; bs, broad signal. ¹³C NMR spectra were acquired on a broad band decoupled mode using DEPT experiments (Distortionless Enhancement by Polarization Transfer) for assigning different types of carbon environment. Infrared spectra (IR) were measured in a Jasco FT/IR 4100, a Perkin-Elmer 1600 and a Perkin-Elmer Spectrum BX apparatus, in the interval between 4000 and 400 cm⁻¹ with a 4 cm⁻¹ resolution. Only characteristic bands are given in each case. Mass spectra (MS) were recorded on an Agilent 7890A gas chromatograph coupled to an Agilent 5975 mass spectrometer under electronic impact (EI) conditions at 70 eV. The obtained data is presented in mass units (m/z) and the values found in brackets belong to the relative intensities comparing to the base peak (100%). High-resolution mass spectra (HRMS) were recorded on a Micromass GCT spectrometer using chemical ionization techniques (CI+) or on an Acquity UPLC coupled to a QTOF mass spectrometer (SYNAPT G2 HDMS) using electrospray ionization (ESI). Melting points (M.p.) were measured in a Büchi B-540 apparatus in open capillary tubes and are uncorrected. High performance liquid chromatography (HPLC) on a chiral stationary phase was performed in a Waters 2695 chromatograph coupled to a Waters 2998 photodiode array detector. Daicel Chiralpak IA, ASH and Chiralcel OZ-3 columns (0.46 cm x 25 cm) were used; specific conditions are indicated for each case. Optical rotations (α value) were measured at 20°C on a Jasco P-2000 polarimeter with a sodium lamp at 589 nm and a path length of 1 dm. Solvent and concentration are specified in each case. X-ray data collections were performed in an Agilent Supernova diffractometer equipped with an Atlas CCD area detector, and a CuK α micro-focus source with multilayer optics ($\lambda = 1.54184$ Å, 250 µm FWHM beam size). The quality of the crystals was checked under a polarizing microscope, and a suitable crystal or fragment was mounted on a MitegenMicromountTM using Paratone-N inert oil and transferred to the diffractometer. Analytical grade solvents and commercially available reagents were used without further purification. Reactions were monitored using analytical thin layer chromatography (TLC), in pre-coated aluminium-backed plates (Merck Kieselgel 60 F254). These were visualized by ultraviolet irradiation, potassium permanganate or *p*-anisaldehyde dips.¹ For flash chromatography Silicycle 40-63, 230-400 mesh silicagel was used.² Anhydrous solvents were dried with activated molecular sieves prior to use. For the removal of solvents under reduced pressure Büchi R-210 rotary evaporators were used. For reactions carried out under inert conditions, the argon was previously dried through a column of P_2O_5 and a column of KOH and CaCl₂. All the glassware was dried for 12 hours prior to utilizing in an oven at 140°C, and allowed to cool under a dehumidified atmosphere.

¹ E. Stahl, *Thin Layer Chromatography*, Springer-Verlag, Berlin, 1969.

² W. C. Still, H. Kann, A. J. Mitra, *J. Org. Chem.*, 1978, **43**, 2923.

2. Experimental Procedures and Characterizations

2.1. Preparation of 2-substituted (2R,3S,3aR,7aS)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-carbaldehyde (4i-o) (R¹ = Alkyl; R² = H) and 2,2-disubstituted (3*S*,3a*R*,7a*S*)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-carbaldehyde (4p) (R¹ = R² = Alkyl)



Scheme ESI-1. General overview for the synthesis of 4i-p.

General procedure

An ordinary vial equipped with a magnetic stirring bar was charged with catalyst (*S*)-3c (0.04 mmol, 20 mol%), trichloroacetic acid (0.07 mmol, 40 mol%) and the corresponding α,β -unsaturated aldehyde **2i-p** (0.18 mmol) in chloroform (1.0 mL). After stirring the mixture for 15 minutes at 0°C, 6-hydroxy-2*H*-pyran-3(6*H*)-one **1** (0.36 mmol, 2eq.) was added and the reaction mixture was further stirred at room temperature until completion (tipically 18 h). The crude mixture was quenched with saturated NaHCO₃ aqueous solution (1.0 mL), extracted with CH₂Cl₂ (3 × 2.0 mL) and the combined organic fractions were collected, dried over Na₂SO₄, filtered and the solvent removed under reduced pressure. Pure aldehydes **4i-p** were isolated after flash column chromatography purification



(2*R*,3*S*,3*aR*,7*aS*)-5-Oxo-2-propylhexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4i). Following the general procedure 4i (29 mg, 0.14 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 76% yield as a yellow oil starting from (*E*)-hex-2-enal (18 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as

solvent. Measured d.r. (**4i**:**5i**) before purification: 8:1. ¹H NMR (300 MHz, CDCl₃) δ 9.76 (d, J = 2.3 Hz, 1H, CHO), 5.44 (d, J = 5.3 Hz, 1H, OCHO), 4.58 (dt, J = 8.1, 6.0 Hz, 1H, *n*-PrC**H**), 4.19 (d, J = 18.1 Hz, 1H, C**H**_aH_bO), 3.91 (d, J = 18.1 Hz, 1H, CH_aH_bO), 3.24-3.12 (m, 1H, CHCH₂CO), 2.88 (ddd, J = 9.8, 8.1, 2.3 Hz, 1H, CHCHO), 2.71 (dd, J = 15.6, 7.5 Hz, 1H, CHCH_aH_bCO), 2.53 (dd, J = 15.6, 7.5 Hz, 1H, CHCH_aH_bCO), 1.61-1.33 (m, 4H, CH₂CH₂), 0.94 (t, J = 7.2 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.1 (CO), 199.9 (CHO), 101.0 (OCHO), 78.8 (*n*-PrCH), 69.1 (CH₂O), 58.5 (CHCHO), 40.1 (CHCH₂CO), 36.6 (CHCH₂CO), 36.1 (CH₃CH₂CH₂), 19.1 (CH₃CH₂), 13.9 (CH₃). IR (ATR): 1731, 1720 (C=O) cm⁻¹. MS (70 eV) m/z (%): 212 (M⁺, 3), 169 (17), 111 (100), 81 (26), 55 (31). HRMS: Calculated for [C₁₁H₁₇O₄]⁺: 213.1127 [(M+H)⁺]; found: 213.1132. [α]_D^{rt}: +14.5, (*c* 0.3, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7i**.



(2R,3S,3aR,7aS)-2-Ethyl-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4j). Following the general procedure 4j (26 mg, 0.13 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 72% yield as a yellow oil starting from (*E*)-pent-2-enal (16 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as

solvent. Measured d.r. (**4j**:**5j**) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.78 (d, J = 2.2 Hz, 1H, CHO), 5.46 (d, J = 5.3 Hz, 1H, OCHO), 4.54 (dt, J = 12.9, 6.4 Hz, 1H, EtC**H**), 4.22 (d, J = 18.2 Hz, 1H, C**H**_aH_bO), 3.94 (d, J = 18.2 Hz, 1H, CH_aH_bO), 3.22-3.14 (m, 1H, C**H**CH₂CO), 2.98-2.86 (m, 1H, C**H**CHO), 2.73 (dd, J = 15.6, 7.3 Hz, 1H, CHC**H**_aH_bCO), 2.56 (dd, J = 15.6, 7.5 Hz, 1H, CHCH_aH_bCO), 1.78-1.56 (m, 2H, CH₃CH₂), 1.01 (t, J = 7.4 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.0 (CO), 199.8 (CHO), 101.0 (OCHO), 80.2 (EtCH), 69.2 (CH₂O), 58.0 (CHCHO), 40.2 (CHCH₂CO), 36.1 (CHCH₂CO), 27.4 (CH₃CH₂), 10.0 (CH₃). IR (ATR): 1732, 1726 (C=O), 1072, 1017 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 198 (M⁺, 3), 169 (22), 111 (25), 97 (100), 55 (30). HRMS: Calculated for [C₁₀H₁₅O₄]⁺: 199.0970 [(M+H)⁺]; found: 199.0978. [α]_D^{n⁺}: +9.4 (c 0.4, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7j**.



(2*R*,3*S*,3*aR*,7*aS*)-2-Butyl-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4k). Following the general procedure 4k (30 mg, 0.13 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 69% yield as a yellow oil starting from (*E*)-hept-2-enal (22 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1f (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as

solvent. Measured d.r. (**4**k:**5**k) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.77 (d, *J* = 2.1 Hz, 1H, CHO), 5.45 (d, *J* = 5.3 Hz, 1H, OCHO), 4.58 (dt, *J* = 13.7, 6.3 Hz, 1H, *n*-BuCH), 4.21 (d, *J* = 18.2 Hz, 1H, C**H**_aH_bO), 3.93 (d, *J* = 18.2 Hz, 1H, CH_aH_bO), 3.24-3.12 (m, 1H, C**H**CH₂CO), 2.93-2.96 (m, 1H, C**H**CHO), 2.72 (dd, *J* = 15.6, 7.4 Hz, 1H, CHCH_aH_bCO), 2.55 (dd, *J* = 15.6, 7.6 Hz, 1H, CHCH_aH_bCO), 1.74-1.23 (m, 6H, (CH₂)₃), 0.90 (t, *J* = 6.9 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.1 (CO), 199.8 (CHO), 101.0 (OCHO), 79.0 (*n*-BuCH), 69.2 (CH₂O), 58.5 (CHCHO), 40.2 (CHCH₂CO), 36.1 (CHCH₂CO), 34.2 (*n*-PrCH₂), 27.9 (EtCH₂), 22.5 (CH₃CH₂), 13.9 (CH₃). IR (ATR): 1735, 1720 (C=O), 1071, 1009 (OCHO) cm⁻¹. MS (70 eV) *m*/*z* (%): 226 (M⁺, 4), 169 (20), 125 (100), 81 (25), 55 (28). HRMS: Calculated for [C₁₂H₁₉O₄]⁺: 227.1283 [(M+H)⁺]; found: 227.1273. [α]_Dⁿ: +14.1 (*c* 0.3, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7k**.



(2R,3S,3aR,7aS)-5-Oxo-2-pentylhexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4l). Following the general procedure 4l (31 mg, 0.13 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 73% yield as a white solid starting from (*E*)-oct-2-enal (24 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as

solvent. Measured d.r. (**41:51**) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.77 (d, *J* = 2.3 Hz, 1H, CHO), 5.45 (d, *J* = 5.4 Hz, 1H, OCHO), 4.58 (dt, *J* = 8.1, 6.2 Hz, 1H, *n*-C₅H₁₁C**H**), 4.21 (d, *J* = 18.1 Hz, 1H, C**H**_aH_bO), 3.92 (d, *J* = 18.1 Hz, 1H, CH_aH_bO), 3.22-3.12 (m, 1H, CHCH₂CO), 2.88 (ddd, *J* = 9.8, 8.1, 2.3 Hz, 1H, C**H**CHO), 2.72 (dd, *J* = 15.6, 7.5 Hz, 1H, CHCH_aH_bCO), 1.75-1.16 (m, 8H, (CH₂)₄), 0.88 (t, *J* = 6.6 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.1 (CO), 199.8 (CHO), 101.0

(OCHO), 79.0 (n-C₅H₁₁CH), 69.1 (CH₂O), 58.5 (CHCHO), 40.2 (CHCH₂CO), 36.1 (CHCH₂CO), 34.5 (n-BuCH₂), 31.6 (n-PrCH₂), 25.5 (EtCH₂), 22.5 (CH₃CH₂), 13.9 (CH₃). IR (ATR): 1733 (C=O), 1073 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 240 (M⁺, 3), 169 (20), 139 (100), 111 (18), 81 (30), 55 (36). HRMS: Calculated for [C₁₃H₂₁O₄]⁺: 241.1440 [(M+H)⁺]; found: 241.1452. M.p. (hexanes/EtOAc): 118-120°C. [α]_D^{rt}: +8.6 (c 0.6, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **71**.



(2*R*,3*S*,3*aR*,7*aS*)-2-Hexyl-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4m). Following the general procedure 4m (28 mg, 0.11 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 61% yield as a white solid starting from *trans*-non-2-enal (26 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using

chloroform (1.0 mL) as solvent. Measured d.r. (**4m**:**5m**) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.77 (d, J = 2.1 Hz, 1H, CHO), 5.45 (d, J = 5.4 Hz, 1H, OCHO), 4.58 (dt, J = 13.9, 6.2 Hz, 1H, n-C₆H₁₃CH), 4.21 (d, J = 18.1 Hz, 1H, CH_aH_bO), 3.93 (d, J = 18.1 Hz, 1H, CH_aH_bO), 3.28-3.11 (m, 1H, CHCH₂CO), 3.02-2.81 (m, 1H, CHCHO), 2.72 (dd, J = 15.5, 7.4 Hz, 1H, CHCH_aH_bCO), 2.55 (dd, J = 15.5, 7.5 Hz, 1H, CHCH_aH_bCO), 1.69-1.20 (m, 10H, (CH₂)₅), 0.88 (t, J = 5.6 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.1 (CO), 199.8 (CHO), 101.0 (OCHO), 79.0 (n-C₆H₁₃CH), 69.1 (CH₂O), 58.5 (CHCHO), 40.2 (CHCH₂CO), 36.1 (CHCH₂CO), 34.5 (n-C₅H₁₁CH₂), 31.6 (n-BuCH₂), 29.1 (n-PrCH₂), 25.7 (EtCH₂), 22.5 (CH₃CH₂), 14.0 (CH₃). IR (ATR): 1732, 1719 (C=O), 1072, 1010 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 254 (M⁺, 3), 169 (23), 153 (100), 81 (38), 55 (39). HRMS: Calculated for [C₁₄H₂₃O₄]⁺: 255.1596 [(M+H)⁺]; found: 255.1587. [α]_Dⁿ: +10.3 (c 0.3, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7m.**



(2S,3S,3aR,7aS)-2-Isopropyl-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4n). Following the general procedure 4n (22 mg, 0.10 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 48 h in 59% yield as a yellow oil starting from (*E*)-4-methylpent-2-enal (19 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform

(1.0 mL) as solvent. Measured d.r. (**4n**:**5n**) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.76 (d, J = 2.7 Hz, 1H, CHO), 5.45 (d, J = 5.0 Hz, 1H, OCHO), 4.40 (dd, J = 13.3, 7.2 Hz, 1H, *i*-PrCH), 4.21 (d, J = 17.9 Hz, 1H, CH_aH_bO), 3.95 (d, J = 17.9 Hz, 1H, CH_aH_bO), 3.20-3.09 (m, 1H, CHCH₂CO), 2.96 (ddd, J = 9.9, 7.2, 2.7 Hz, 1H, CHCHO), 2.74 (dd, J = 16.1, 6.1 Hz, 1H, CHCH_aH_bCO), 2.56 (dd, J = 16.1, 7.8 Hz, 1H, CHCH_aH_bCO), 1.88-1.74 (m, 1H, (CH₃)₂CH), 0.99 (t, J = 6.7 Hz, 3H, CH₃), 0.91 (t, J = 6.8 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 207.5 (CO), 200.4 (CHO), 101.0 (OCHO), 84.4 (*i*-PrCH), 69.3 (CH₂O), 55.5 (CHCHO), 41.0 (CHCH₂CO), 36.0 (CHCH₂CO), 32.1 ((CH₃)₂CH), 18.6 (CH₃), 18.1 (CH₃). IR (ATR): 1727 (C=O), 1074 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 182 (7), 140 (46), 111 (100), 98 (49), 83 (44), 55 (37). HRMS: Calculated for $[C_{11}H_{17}O_4]^+$: 213.1127 $[(M+H)^+]$; found: 213.1139. $[\alpha]_D^{\pi}$: +13.7 (*c* 0.1, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7n**.



(2R,3S,3aR,7aS)-2-[(Z)-Hex-3-en-1-yl]-5-oxohexahydro-4H-furo[2,3b]pyran-3-carbaldehyde (4o). Following the general procedure 4o (27 mg, 0.11 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 60% yield as a yellow oil starting from (2*E*,6*Z*)-nona-2,6-dienal (25 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as solvent. Measured d.r. (40:50) before

purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.76 (d, *J* = 2.1 Hz, 1H, CHO), 5.46 (d, *J* = 5.3 Hz, 1H, OCHO), 5.45-5.20 (m, 2H, CH=CH), 4.58 (dt, *J* = 13.2, 7.0 Hz, 1H, CH₂CH₂CH), 4.20 (d, *J* = 18.2 Hz, 1H, CH_aH_bO), 3.93 (d, *J* = 18.2 Hz, 1H, CH_aH_bO), 3.37-3.08 (m, 1H, CHCH₂CO), 3.06-2.79 (m, 1H, CHCHO), 2.72 (dd, *J* = 15.6, 7.3 Hz, 1H, CHCH_aH_bCO), 2.56 (dd, *J* = 15.6, 7.4 Hz, 1H, CHCH_aH_bCO), 2.34-1.90 (m, 4H, CH₂CH₂), 1.63 (dt, *J* = 8.0, 7.5 Hz, 2H, CH₃CH₂), 0.96 (t, *J* = 7.5 Hz, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 208.0 (CO), 199.7 (CHO), 133.0 (EtCH=CH), 127.3 (EtCH), 101.0 (OCHO), 78.4 (CH₂CH₂CH), 69.1 (CH₂O), 58.4 (CHCHO), 40.2 (CHCH₂CO), 36.1 (CHCH₂CO), 34.5 (CH₂CH₂CH), 23.5 (CH₂CH₂CH), 20.5 (CH₃CH₂), 14.2 (CH₃). IR (ATR): 1732, 1723 (C=O), 1069, 1012 (OCHO) cm⁻¹. MS (70 eV) *m*/*z* (%): 252 (M⁺, 3), 140 (70), 111 (41), 95 (69), 81 (100), 55 (77). HRMS: Calculated for [C₁₄H₂₁O₄]⁺: 253.1440 [(M+H)⁺]; found: 253.1432. [α]_D^{n⁺}: +9.6 (*c* 0.2, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **70**.



(35,3aR,7aS)-2,2-Dimethyl-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (4p). Following the general procedure 4p (26 mg, 0.13 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18h in 73% yield as a yellow oil starting from 3-methylbut-2-enal (16 mg, 0.18 mmol), trichloroacetic acid (12 mg, 0.07 mmol) and pyranone 1 (41 mg, 0.36 mmol) in the presence of catalyst (*S*)-3c (19 mg, 0.04 mmol) and using chloroform (1.0 mL) as

solvent. Measured d.r. (**4p**:**5p**) before purification: 9:1. ¹H NMR (300 MHz, CDCl₃) δ 9.86 (d, *J* = 4.0 Hz, 1H, CHO), 5.39 (d, *J* = 5.5 Hz, 1H, OCHO), 4.25 (d, *J* = 18.0 Hz, 1H, C**H**_a**H**_bO), 3.92 (d, *J* = 18.0 Hz, 1H, CH_a**H**_bO), 3.34-3.05 (m, 1H, C**H**CH₂CO), 2.82 (dd, *J* = 9.1, 4.0 Hz, 1H, C**H**CHO), 2.68 (d, *J* = 7.6 Hz, 2H, CHC**H**₂CO), 1.51 (s, 3H, CH₃), 1.39 (s, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 209.8 (CO), 197.5 (CHO), 99.0 (OCHO), 80.7 (CH₃**C**), 67.0 (CH₂O), 64.7 (CHCHO), 38.2 (CHCH₂CO), 36.4 (CHCH₂CO), 28.3 (CH₃), 24.3 (CH₃). IR (ATR): 1720 (C=O), 1009 (OCHO) cm⁻¹. HRMS: Calculated for [C₁₀H₁₅O₄]⁺: 199.0970 [(M+H)⁺]; found: 199.0972. [α]_Dⁿ: +57.6 (*c* 1.4, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **7p**.

2.2. Preparation of 2-substituted (2S,3R,3aR,7aS)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-carbaldehyde (6a-h) (R = Aromatic or Heteroaromatic)



Scheme ESI-2. General overview for the synthesis of 6a-h.

General procedure

An ordinary vial equipped with a magnetic stirring bar was charged with catalyst (S)-3a (0.10 mmol, 20 mol%), 1,3-bis[3,5-bis(trifluoromethyl)phenyl]thiourea 15 (0.05 mmol, 10 mol%), the corresponding α,β -unsaturated aldehyde (0.50 mmol) and CHCl₃ (2.0 mL). After stirring the solution for 15 minutes at 0°C, 6-hydroxy-2*H*-pyran-3(6*H*)-one 1 (1.00 mmol, 2 eq.) was added and the reaction mixture was further stirred at room temperature until completion (tipically 18 h). Once the reaction was completed, aqueous HCl 4 *M* (2.0 mL) was added to the mixture and the crude mixture was stirred 4 h at room temperature, after which was extracted with CH₂Cl₂ (3 × 2.0 mL). The combined organic fractions were collected, dried over anhydrous Na₂SO₄, filtered and the solvent removed under reduced pressure. Pure aldehydes **6a-h** were isolated after flash column chromatography purification.



(2S,3R,3aR,7aS)-5-Oxo-2-phenylhexahydro-4*H*-furo[2,3-*b*]pyran-3carbaldehyde (6a). Following the general procedure 6a (36 mg, 0.15 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 74% yield as a yellow oil starting from (*E*)-cinnalmaldehyde (26 mg, 0.20 mmol), thiourea 7 (10 mg, 0.02 mmol) and pyranone 1 (46 mg, 0.40 mmol) in the presence of catalyst (S)-3a (13 mg, 0.04 mmol) and using chloroform (0.8 mL) as

solvent. Measured d.r. (**5a**:**6a**) before purification: 1:4. ¹H NMR (300 MHz, CDCl₃) δ 9.12 (d, *J* = 1.6 Hz, 1H, CHO), 7.53-7.26 (m, 5H, C_{arom}-H), 5.84 (d, *J* = 6.3 Hz, 1H, OCHO), 5.72 (d, *J* = 7.6 Hz, 1H, PhC**H**), 4.27 (d, *J* = 18.4 Hz, 1H, C**H**_aH_bO), 4.06 (d, *J* = 18.4 Hz, 1H, CH_aH_bO), 3.52-3.44 (m, 1H, C**H**CH₂CO), 3.17 (ddd, *J* = 7.6, 5.2, 1.6 Hz, 1H, C**H**CHO), 2.80 (dd, *J* = 15.6, 6.9 Hz, 1H, CHC**H**_aH_bCO), 2.58 (dd, *J* = 15.6, 5.9 Hz, 1H, CHCH_aH_bCO).¹³C NMR (75 MHz, CDCl₃) δ 209.3 (CO), 198.4 (CHO), 136.1(**C**_{arom}-C), 129.0, 128.6, 125.8 (C_{arom}-H), 101.5 (OCHO), 81.4 (PhCH), 68.8 (CH₂O), 61.0 (**C**HCHO), 38.8 (CHCH₂CO), 36.8 (**C**HCH₂CO). IR (ATR): 1720 (C=O), 1077, 1020 (OCHO) cm⁻¹. MS (70 eV) *m*/*z* (%): 140 (100), 131 (39), 111 (24), 91 (20), 77 (20). HRMS: Calculated for [C₁₄H₁₅O₄]⁺: 247.0970 [(M+H)⁺]; found:

247.0973. $[\alpha]_D^{\pi}$: -22.9 (*c* 0.4, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8a**.



(2S,3R,3aR,7aS)-2-(4-Methoxyphenyl)-5-oxohexahydro-4H-furo[2,3b]pyran-3-carbaldehyde (6b). Following the general procedure 6b (86 mg, 0.31 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 63% yield as a yellow oil starting from (*E*)-4-methoxycinnalmaldehyde (81 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol)

PMP in the presence of catalyst (*S*)-**3a** (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (**5b**:6**b**) before purification: 1:4. ¹H NMR (300 MHz, CDCl₃) δ 9.14 (d, J = 1.6 Hz, 1H, CHO), 7.17 (d, J = 8.8 Hz, 2H, C_{arom}-H), 6.89 (d, J = 8.8 Hz, 2H, C_{arom}-H), 5.81 (d, J = 6.2 Hz, 1H, OCHO), 5.68 (d, J = 7.5 Hz, 1H, PMPCH), 4.25 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.05 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.80 (s, 3H, CH₃), 3.54-3.34 (m, 1H, CHCH₂CO), 3.20-3.11 (m, 1H, CHCHO), 2.79 (dd, J = 15.6, 6.9 Hz, 1H, CHCH_aH_bCO), 2.56 (dd, J = 15.6, 5.9 Hz, 1H, CHCH_aH_bCO). ¹³C NMR (75 MHz, CDCl₃) δ 209.3 (CO), 198.5 (CHO), 159.7 (C_{arom}-C), 128.0 (C_{arom}-C), 127.1, 114.3 (C_{arom}-H), 101.4 (OCHO), 81.2 (PMPCH), 68.9 (CH₂O), 61.0 (CHCHO), 55.3 (CH₃), 38.8 (CHCH₂CO), 36.8 (CHCH₂CO). IR (ATR): 1722 (C=O), 1025 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 276 (M⁺, 6), 136 (100), 111 (15), 77 (15). HRMS: Calculated for [C₁₅H₁₇O₅]⁺: 277.1076 [(M+H)⁺]; found: 277.1079. [α]_D^{n^t}: 22.9 (c 0.2, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8b**.



(2*S*,3*R*,3a*R*,7a*S*)-2-(4-Fluorophenyl)-5-oxohexahydro-4*H*-furo[2,3*b*]pyran-3-carbaldehyde (6c). Following the general procedure 6c (97 mg, 0.37 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 73% yield as a yellow oil starting from (*E*)-4-fluorocinnalmaldehyde (75 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5c:6c) before purification: 1:2. ¹H NMR (300 MHz, CDCl₃) δ 9.14 (d, *J* = 1.6 Hz, 1H, CHO), 7.37-7.01 (m, 4H, C_{arom}-H), 5.82 (d, *J* = 6.2 Hz, 1H, OCHO), 5.70 (d, *J* = 7.3 Hz, 1H, 4-

FC₆H₄C**H**), 4.26 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.06 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.50-3.44 (m, 1H, CHCH₂CO), 3.17 (ddd, J = 7.3, 4.9, 1.6 Hz, 1H, CHCHO), 2.80 (dd, J = 15.5, 7.0 Hz, 1H, CHCH_aH_bCO), 2.58 (dd, J = 15.5, 6.1 Hz, 1H, CHCH_aH_bCO).¹³C NMR (75 MHz, CDCl₃) δ 209.0 (CO), 198.2 (CHO), 162.6 (d, J = 247.7 Hz, C_{arom}-F), 131.8 (d, J = 8.3 Hz, C_{arom}-C), 127.6 (d, ³J = 8.3 Hz, C_{arom}-H), 116.0 (d, ²J = 21.8 Hz, C_{arom}-H), 101.4 (OCHO), 80.8 (4-FC₆H₄CH), 68.9 (CH₂O), 61.0 (CHCHO), 38.8 (CHCH₂CO), 36.9 (CHCH₂CO). IR (ATR): 1732, 1723 (C=O), 1280 (C-F) cm⁻¹. MS (70 eV) *m*/*z* (%): 264 (M⁺, 3), 240 (36), 151 (100), 109 (49), 28 (37). HRMS: Calculated for [C₁₄H₁₄FO₄]⁺: 265.0876 [(M+H)⁺]; found: 265.0886. [α]_Dⁿ: -24.4 (*c* 0.8, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8c**.



(2*S*,3*R*,3*aR*,7*aS*)-2-(4-Nitrophenyl)-5-oxohexahydro-4*H*-furo[2,3*b*]pyran-3-carbaldehyde (6d). Following the general procedure 6d (113 mg, 0. 39 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 78% yield as a yellow oil starting from (*E*)-4-nitrocinnalmaldehyde (88 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5d:6d) before purification: 1:2. ¹H NMR (300 MHz, CDCl₃) δ 9.14 (d, *J* = 1.5 Hz, 1H, CHO), 8.25 (d, J = 8.7 Hz, 2H, C_{arom}-H), 7.50 (d, J = 8.7 Hz, 2H, C_{arom}-H), 5.85 (d, J = 6.2 Hz, 1H, OCHO), 5.79 (d, J = 7.2 Hz, 1H, 4-NO₂C₆H₄CH), 4.28 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.08 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.54-3.42 (m, 1H, CHCH₂CO), 3.39-3.11 (m, 1H, CHCHO), 2.82 (dd, J = 15.5, 7.0 Hz, 1H, CHCH_aH_bCO), 2.62 (dd, J = 15.5, 6.4 Hz, 1H, CHCH_aH_bCO).¹³C NMR (75 MHz, CDCl₃) δ 208.5 (CO), 197.3 (CHO), 148.0 (C_{arom}-N), 143.1 (C_{arom}-C), 126.8, 124.2 (C_{arom}-H), 101.6 (OCHO), 80.3 (4-NO₂C₆H₄CH), 69.0 (CH₂O), 61.1 (CHCHO), 38.7 (CHCH₂CO), 37.2 (CHCH₂CO). IR (ATR): 1727 (C=O), 1518, 1347 (NO₂) cm⁻¹. MS (70 eV) *m/z* (%): 261 (M⁺-CHO, 4), 190 (36), 98 (100), 77 (12). HRMS: Calculated for [C₁₄H₁₄NO₆]⁺: 292.0821 [(M+H)⁺]; found: 292.0807. [α]_D^{n⁺}: +10.8 (*c* 0.8, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8d**.



(2*S*,3*R*,3a*R*,7a*S*)-2-(4-Bromophenyl)-5-oxohexahydro-4*H*-furo[2,3*b*]pyran-3-carbaldehyde (6e). Following the general procedure 6e (140 mg, 0.43 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 87% yield as a yellow oil starting from (*E*)-4-bromocinnalmaldehyde (105 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5e:6e) before purification: 1:1.5. ¹H NMR (300 MHz, CDCl₃) δ 9.12 (d, *J* = 1.7 Hz, 1H, CHO), 7.50 (d, *J* = 8.5 Hz, 2H, C_{arom}-H), 7.15 (d, *J* = 8.5 Hz, 2H, C_{arom}-H), 5.80 (d, *J* = 6.2 Hz, 1H,

OCHO), 5.66 (d, J = 6.9 Hz, 1H, 4-BrC₆H₄CH), 4.25 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.05 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.47-3.43 (m, 1H, CHCH₂CO), 3.17 (ddd, J = 6.9, 4.8, 1.7 Hz, 1H, CHCHO), 2.78 (dd, J = 15.5, 7.0 Hz, 1H, CHCH_aH_bCO), 2.57 (dd, J = 15.5, 6.2 Hz, 1H, CHCH_aH_bCO).¹³C NMR (75 MHz, CDCl₃) δ 208.9 (CO), 198.0 (CHO), 135.0 (C_{arom}-Br), 132.2, 127.5 (C_{arom}-H), 122.6 (C_{arom}-C), 101.5 (OCHO), 80.7 (4-BrC₆H₄CH), 68.9 (CH₂O), 60.9 (CHCHO), 38.7 (CHCH₂CO), 36.9 (CHCH₂CO). IR (ATR): 1724 (C=O), 1070, 1009 (OCHO) cm⁻¹. MS (70 eV) m/z (%): 323 (M⁺, 1), 226 (85), 131 (77), 98 (100). HRMS: Calculated for [C₁₄H₁₄BrO₄]⁺: 325.0075 [(M+H)⁺]; found: 325.0091. [α]_D^{rt}: -12.1 (*c* 1.9, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8e**.



(2*S*,3*R*,3*aR*,7*aS*)-2-(2-Methoxyphenyl)-5-oxohexahydro-4*H*-furo[2,3*b*]pyran-3-carbaldehyde (6f). Following the general procedure 6f (84 mg, 0.30 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 61% yield as a yellow oil starting from (*E*)-2-methoxycinnalmaldehyde (81 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5f:6f) before purification: 1:3. ¹H NMR (300 MHz, CDCl₃) δ 9.08 (d, *J* = 1.8 Hz, 1H, CHO), 7.40-7.17 (m, 2H,

C_{arom}-H), 6.98 (dd , J = 7.5 Hz, 1H, C_{arom}-H), 6.88 (d, J = 8.2 Hz, 1H, C_{arom}-H), 5.88 (d, J = 7.6 Hz, 1H, OCHO), 5.84 (d, J = 6.4 Hz, 1H, 2-MeOC₆H₄CH), 4.26 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.05 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.84 (s, 3H, CH₃), 3.41-3.33 (m, 1H, CHCH₂CO), 3.29-3.24 (m, 1H, CHCHO), 2.76 (dd, J = 15.7, 6.7 Hz, 1H, CHCH_aH_bCO), 2.55 (dd, J = 15.7, 5.4 Hz, 1H, CHCH_aH_bCO). ¹³C NMR (75 MHz, CDCl₃) δ 209.4 (CO), 198.6 (CHO), 155.2 (C_{arom}-O), 129.4, 126.4 (C_{arom}-H), 124.9 (C_{arom}-C), 121.1, 110.1 (C_{arom}-H), 101.2 (OCHO), 77.4 (2-MeOC₆H₄CH), 68.7 (CH₂O), 59.1 (CHCHO), 55.2 (CH₃), 38.6 (CHCH₂CO), 36.7 (CHCH₂CO). IR (ATR): 1724 (C=O), 1010 (OCHO) cm⁻¹. MS (70 eV) *m/z* (%): 276 (M⁺, 25), 136 (100), 118 (45), 91 (35), 77 (24). HRMS: Calculated for [C₁₅H₁₇O₅]⁺: 277.1076 [(M+H)⁺];

found: 277.1067. $[\alpha]_D^{rt}$: -48.5 (*c* 0.5, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8f**.



(2*S*,3*R*,3*aR*,7*aS*)-2-(2-Nitrophenyl)-5-oxohexahydro-4*H*-furo[2,3*b*]pyran-3-carbaldehyde (6g). Following the general procedure 6g (100 mg, 0.34 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18 h in 71% yield as a yellow oil starting from (*E*)-2-nitrocinnalmaldehyde (88 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.10 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5g:6g) before purification: 1:3. ¹H NMR (300 MHz, CDCl₃) δ 9.21 (s, 1H, CHO), 8.12 (d, *J* = 8.2 Hz, 1H, C_{arom}-H),

7.83-7.44 (m, 3H, C_{arom} -H), 6.16 (d, J = 7.8 Hz, 1H, OCHO), 5.83 (d, J = 6.1 Hz, 1H, 2-NO₂C₆H₄CH), 4.26 (d, J = 18.3 Hz, 1H, CH_aH_bO), 4.07 (d, J = 18.3 Hz, 1H, CH_aH_bO), 3.87-3.62 (m, 1H, CHCH₂CO), 3.60-3.20 (m, 1H, CHCHO), 2.78 (dd, J = 15.8, 6.4 Hz, 1H, CHCH_aH_bCO), 2.64 (dd, J = 15.8, 6.4 Hz, 1H, CHCH_aH_bCO).¹³C NMR (75 MHz, CDCl₃) δ 208.2 (CO), 197.8 (CHO), 147.3 (C_{arom}-N), 134.4 (C_{arom}-H), 133.0 (C_{arom}-C), 129.5, 128.3, 125.1 (C_{arom}-H), 101.5 (OCHO), 78.2 (2-NO₂C₆H₄CH), 69.3 (CH₂O), 60.3 (CHCHO), 38.6 (CHCH₂CO), 36.8 (CHCH₂CO). IR (ATR): 1725 (C=O), 1526, 1350 (NO₂) cm⁻¹. MS (70 eV) m/z (%): 291 (M⁺, 1), 172 (43), 135 (100), 92 (87), 78 (93). HRMS: Calculated for [C₁₄H₁₄NO₆]⁺: 292.0821 [(M+H)⁺]; found: 292.0839. [α]_D^{rt}: +105.5 (*c* 0.3, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8g**.



(2*S*,3*R*,3a*R*,7a*S*)-2-(Furan-2-yl)-5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-carbaldehyde (6h). Following the general procedure 6h (76 mg, 0.32 mmol) was isolated by FC (hexanes/EtOAc 7:3) after 18h in 64% yield as a yellow oil starting from (*E*)-3-(furan-2-yl)acrylaldehyde (61 mg, 0.50 mmol), thiourea 7 (25 mg, 0.05 mmol) and pyranone 1 (114 mg, 1.0 mmol) in the presence of catalyst (*S*)-3a (32 mg, 0.1 mmol) and using chloroform (2.0 mL) as solvent. Measured d.r. (5h:6h) before purification: 1:10. ¹H NMR (300 MHz, CDCl₃) δ 9.30 (d, *J* = 1.4 Hz, 1H, CHO), 7.48-7.31 (m,

1H, C_{hetarom}-H), 6.38-6.29 (m, 2H, C_{hetarom}-H), 5.85 (d, J = 6.9 Hz, 1H, OCHO), 5.61 (d, J = 8.2 Hz, 1H, C_{hetarom}-CH), 4.18 (d, J = 18.4 Hz, 1H, CH_aH_bO), 4.02 (d, J = 18.4 Hz, 1H, CH_aH_bO), 3.75-3.56 (m, 1H, CHCH₂CO), 3.12-3.92 (m, 1H, CHCHO), 2.84 (dd, J = 16.1, 6.4 Hz, 1H, CHCH_aH_bCO), 2.51 (dd, J = 16.1, 3.4 Hz, 1H, CHCH_aH_bCO). ¹³C NMR (75 MHz, CDCl₃) δ 209.3 (CO), 196.5 (CHO), 150.7 (C_{hetarom}-C), 143.3, 110.5, 109.6 (C_{hetarom}-H), 101.0 (OCHO), 73.2 (C_{hetarom}-CH), 67.6 (CH₂O), 58.9 (CHCHO), 38.4 (CHCH₂CO), 36.4 (CHCH₂CO). IR (ATR): 1723 (C=O), 1131 (FuranC-O), 1006 (OCHO) cm⁻¹. MS (70 eV) *m/z* (%): 236 (M⁺, 9), 140 (100), 111 (30), 94 (41). HRMS: Calculated for [C₁₂H₁₃O₅]⁺: 237.0763 [(M+H)⁺]; found: 237.0748. [α]_D^{n^t}: -55.1 (*c* 0.2, CH₂Cl₂). The enantiomeric excess (e.e.) was measured after transformation into **8h**.

2.3. Preparation of 2-substituted (2R,3S,3aR,7aS)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-yl acrylates (7i-o) (R¹ = Alkyl; R² = H), 2,2-disubstituted (3*S*,3a*R*,7a*S*)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-yl acrylate (7p) (R¹ = R² = Me) and 2-substituted (2*R*,3*S*,3a*R*,7a*S*)-configured 5-oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-yl acrylates (8a-h) (R = Aromatic or Heteroaromatic).



Scheme ESI-3. General overview for the synthesis of 7i-p and 8a-h.

General procedure

An ordinary vial equipped with a magnetic stirring bar was charged with furopyrane **4** or **6**, THF (1.0 mL) and (carbethoxymethylene)triphenylphosphorane (2.0 eq.) was added. The crude mixture was stirred at room temperature until completion (30-60 min) and, once finished monitored by tlc, it was concentrated *in vacuo* and purified by flash column chromatography. *Note:* Racemic standards were prepared starting from racemic furopyranes **4** and **6**, respectively, which were previously synthesized using racemic catalyst (\pm)-**3a**.



Ethyl (*E*)-3-[(2*R*,3*S*,3*aR*,7*aS*)-5-oxo-2-propylhexahydro-4*H*furo[2,3-*b*]pyran-3-yl]acrylate (7i). Following the general procedure 7i (31 mg, 0.11 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 78% yield as a yellow oil starting from pure 4i (30 mg, 0.14 mmol) and (carbethoxymethylene)triphenylphosphorane (97 mg, 0.28 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃)

δ 6.77 (dd, J = 15.6, 9.3 Hz, 1H, CH=CHCO), 5.91 (d, J = 15.6 Hz, 1H, CHCO), 5.48 (d, J = 5.7 Hz, 1H, OCHO), 4.28-4.16 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.16-4.06 (m, 1H, *n*-PrCH), 3.89 (d, J = 18.4 Hz, 1H, OCH_aH_bCO), 2.99-2.87 (m, 1H, CHCH₂CO), 2.85-2.57 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.41 (dd, J = 14.7, 7.1 Hz, 1H, CHCH_aH_bCO), 1.56-1.40 (m, 4H, CH₂CH₂), 1.30 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 0.93 (t, J = 6.3 Hz, 3H, CH₃CH₂CH₂). ¹³C NMR (75 MHz, CDCl₃) δ 210.2 (CH₂CO), 165.5 (COO), 142.4 (COCH=CH), 125.6 (COCH=CH), 101.5 (OCHO), 81.4 (*n*-PrCH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 50.8 (CHCH=CH), 42.4 (CHCH₂CO), 36.1 (CHCH₂CO), 35.5 (EtCH₂), 19.3 (CH₃CH₂CH₂), 14.2 (CH₃CH₂CH₂), 14.0

(CH₃CH₂O). IR (ATR): 1717, 1654 (C=O) cm⁻¹. MS (70 eV) m/z (%): 282 (M⁺, 3), 237 (16), 210 (76), 136 (56), 79 (100). HRMS: Calculated for $[C_{15}H_{23}O_5]^+$: 283.1545 $[(M+H)^+]$; found: 283.1556. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane /*i*-PrOH (95:5)]; flow rate 1.00 mL/min; $\tau_{major} = 29.82 \text{ min}$, $\tau_{minor} = 22.11 \text{ min}$ (93% e.e.). $[\alpha]_D^{rt}$: +124.8 (*c* 0.2, CH₂Cl₂).



Ethyl (*E*)-3-[(*2R*,3*S*,3*aR*,7*aS*)-2-ethyl-5-oxohexahydro-4*H*furo[2,3-*b*]pyran-3-yl]acrylate (7j). Following the general procedure 7j (15 mg, 0.05 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 75% yield as a yellow oil starting from pure 4j (15 mg, 0.07 mmol) and (carbethoxymethylene)triphenylphosphorane (53 mg, 0.15 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300

MHz, CDCl₃) δ 6.78 (dd, J = 15.6, 9.2 Hz, 1H, CH=CHCO), 5.91 (d, J = 15.6 Hz, 1H, CHCO), 5.49 (d, J = 5.7 Hz, 1H, OCHO), 4.32-4.15 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.13-4.00 (m, 1H, EtCH), 3.90 (d, J = 18.4 Hz, 1H, OCH_aH_bCO), 3.05-2.88 (m, 1H, CHCH₂CO), 2.98-2.77 (m, 1H, CHCH=CH), 2.66 (dd, J = 14.7, 8.8 Hz, 1H, CHCH_aH_bCO), 2.41 (dd, J = 14.7, 7.1 Hz, 1H, CHCH_aH_bCO), 1.76-1.42 (m, 2H, CH₃CH₂CH), 1.30 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.01 (t, J =7.4 Hz, 3H, CH₂CH₂CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 210.2 (CH₂CO), 165.5 (COO), 142.4 (COCH=CH), 125.6 (COCH=CH), 101.5 (OCHO), 82.7 (EtCH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 50.2 (CHCH=CH), 42.5 (CHCH₂CO), 36.1 (CHCH₂CO), 26.3 (CH₃CH₂CH), 14.2 (CH₃CH₂O), 10.1 (CH₃CH₂CH₂). IR (ATR): 1716, 1654 (C=O) cm⁻¹. MS (70 eV) *m/z* (%): 268 (M⁺, 1), 210 (31), 136 (35), 79 (100). HRMS: Calculated for [C₁₄H₂₁O₅]⁺: 269.1389 [(M+H)⁺]; found: 269.1382. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane /*i*-PrOH (95:5)]; flow rate 1.00 mL/min; $\tau_{major} = 34.04$ min, $\tau_{minor} = 26.85$ min (92% e.e.). [α]_Dⁿ: +138.3 (*c* 0.1, CH₂Cl₂).



(E)-3-[(2R,3S,3aR,7aS)-2-butyl-5-oxohexahydro-4H-Ethyl furo[2,3-*b*]pyran-3-yl]acrylate (7k). Following the general procedure 7k (24 mg, 0.08 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 1 h in 87% yield as a yellow oil starting from pure 4k (21)mg, 0.09 mmol) and

n-Bù (carbethoxymethylene)triphenylphosphorane (65 mg, 0.18 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 6.78 (dd, J = 15.6, 9.2 Hz, 1H, C**H**=CHCO), 5.91 (d, J = 15.6 Hz, 1H, C**H**CO), 5.49 (d, J = 5.7 Hz, 1H, OCHO), 4.37-4.13 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.13-4.01 (m, 1H, *n*-BuCH), 3.90 (d, J = 18.4 Hz, 1H, OCH_aH_bCO), 3.02-2.87 (m, 1H, CHCH₂CO), 2.99-2.87 (m, CHCH=CH), 2.66 (dd, J = 14.7, 8.8 Hz, 1H, CHCH_aH_bCO), 2.41 (dd, J = 14.7, 7.1 Hz, 1H, CHCH_aH_bCO), 1.73-1.41 (m, 6H, (CH₂)₃), 1.30 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.01 (t, J = 7.4 Hz, 3H, CH₂CH₂CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 210.2 (CH₂CO), 165.5 (COO), 142.4 (COCH=CH), 125.6 (COCH=CH), 101.5 (OCHO), 81.6 (*n*-BuCH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 50.7 (CHCH=CH), 42.4 (CHCH₂CO), 36.1 (CHCH₂CO), 33.1 (*n*-PrCH₂), 28.2 (EtCH₂), 22.6 (CH₃CH₂CH₂), 14.2 (CH₃CH₂O), 13.9 (CH₃CH₂CH₂). IR (ATR): 1718, 1653 (C=O) cm⁻¹. MS (70 eV) *m/z* (%): 296 (M⁺, 1), 210 (73), 136 (66), 79 (100). HRMS: Calculated for [C₁₆H₂₅O₅]⁺: 297.1702 [(M+H)⁺]; found: 297.1695. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane /*i*-PrOH (95:5)]; flow rate 1.00 mL/min; τ_{major} = 34.63 min, τ_{minor} = 22.60 min (94% e.e.). [α]_Dⁿ: +98.2 (*c* 0.5, CH₂Cl₂).



Ethyl (E)-3-[(2R,3S,3aR,7aS)-5-oxo-2-pentylhexahydro-4Hfuro[2,3-b]pyran-3-yl]acrylate (71). Following the general procedure 71 (20 mg, 0.06 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 70% yield as a yellow oil starting from pure 41 (22 mg, 0.09 mmol) and (carbethoxymethylene)triphenylphosphorane (64 mg, 0.18 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃)

5.7 Hz, 1H, OCHO), 4.33-4.16 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.16-4.02 (m, 1H, n- $C_{5}H_{11}CH$), 3.89 (d, J = 18.4 Hz, 1H, OCH₂H_bCO), 2.98-2.88 (m, 1H, CHCH₂CO), 2.84-2.72 (m, 1H, CHCH=CH), 2.66 (dd, J = 14.7, 8.9 Hz, CHCH_aH_bCO), 2.41 (dd, J = 14.7, 7.1 Hz, 1H, CHCH_a \mathbf{H}_{b} CO), 1.56-1.28 (m, 11H, (CH₂)₄, OCH₂C \mathbf{H}_{3}), 0.88 (t, J = 6.6 Hz, 3H, CH₂CH₂CH₃).¹³C NMR (75 MHz, CDCl₃) δ 210.2 (CH₂CO), 165.5 (COO), 142.4 (COCH=CH), 125.6 (COCH=CH), 101.5 (OCHO), 81.6 (n-C₅H₁₁CH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 50.7 (CHCH=CH), 42.4 (CHCH₂CO), 36.1 (CHCH₂CO), 33.4 (*n*-BuCH₂), 31.7 (n-PrCH₂), 25.7 (EtCH₂), 22.5 (CH₃CH₂CH₂), 14.2 (CH₃CH₂O), 13.9 (CH₃CH₂CH₂). IR (ATR): 1733, 1717 (C=O) cm⁻¹. MS (70 eV) m/z (%): 310 (M⁺, 1), 210 (77), 136 (65), 79 (100), 55 (25). HRMS: Calculated for $[C_{17}H_{27}O_5]^+$: 311.1858 $[(M+H)^+]$; found: 311.1855. The e.e. was determined by HPLC using a Chiralpak IA column [n-hexane/i-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{\text{major}} = 20.65 \text{ min}$, $\tau_{\text{minor}} = 10.83 \text{ min}$ (95% e.e.). $[\alpha]_{D}^{\text{rt}} + 79.8$ (*c* 0.1, CH₂Cl₂).



(E)-3-[(2R,3S,3aR,7aS)-2-hexyl-5-oxohexahydro-4H-Ethvl furo[2,3-*b*]pyran-3-yl]acrylate (7m). Following the general procedure 7m (15 mg, 0.05 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 69% yield as a yellow oil starting from pure **4**m mg, 0.07 mmol) (17)and (carbethoxymethylene)triphenylphosphorane (49 mg, 0.14 mmol),

using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 6.77 (dd, J = 15.6, 9.3 Hz, 1H, CH=CHCO), 5.90 (d, J = 15.6 Hz, 1H, CHCO), 5.48 (d, J = 5.7 Hz, 1H, OCHO), 4.33-4.15 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.15-4.01 (m, 1H, n-C₆H₁₃CH), 3.89 (d, J = 18.4 Hz, 1H, $OCH_{a}H_{b}CO$, 2.93-2.86 (m, 1H, CHCH₂CO), 2.84-2.72 (m, 1H, CHCH=CH), 2.66 (dd, J =14.7, 8.8 Hz, 1H, CHCH_aH_bCO), 2.40 (dd, J = 14.7, 7.1 Hz, 1H, CHCH_aH_bCO), 1.63-1.21 (m, 13H, (CH₂)₅, CH₃CH₂O), 0.87 (t, J = 6.5 Hz, 3H, CH₂CH₂CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 210.2 (CH₂CO), 165.5 (COO), 142.4 (COCH=CH), 125.6 (COCH=CH), 101.5 (OCHO), 81.6 (n-C₆H₁₃CH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 50.7 (CHCH=CH), 42.4 (CHCH₂CO), 36.0 (CHCH₂CO), 33.4 (*n*-C₅H₁₁CH₂), 31.7 (*n*-BuCH₂), 29.2 (*n*-PrCH₂), 26.0 (EtCH₂), 22.5 (CH₃CH₂CH₂), 14.2 (CH₃CH₂O), 14.0 (CH₃CH₂CH₂). IR (ATR): 1719 (C=O) cm⁻¹. MS (70 eV) m/z (%): 324 (M⁺, 1), 210 (78), 164 (51), 136 (72), 79 (100). HRMS: Calculated for $[C_{18}H_{29}O_5]^+$: 325.2015 $[(M+H)^+]$; found: 325.2008. The e.e. was determined by HPLC using a Chiralpak AS-H column [*n*-hexane/*i*-PrOH (95:5)]; flow rate 1.00 mL/min; $\tau_{major} = 27.80$ min, $\tau_{\text{minor}} = 16.12 \text{ min (96\% e.e.). } [\alpha]_{\text{D}}^{\text{rt}} + 112.9 (c \ 0.2, \text{CH}_2\text{Cl}_2).$



Ethyl (E)-3-[(2R,3S,3aR,7aS)-2-isopropyl-5-oxohexahydro-4Hfuro[2,3-b]pyran-3-yl]acrylate (7**n**). Following the general procedure 7n (13 mg, 0.04 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 1 h 67% yield as a yellow oil starting from pure 4n (14)mg, 0.07 mmol) and (carbethoxymethylene)triphenylphosphorane (46 mg, 0.13 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 6.80 (dd, J = 15.6, 9.2 Hz, 1H, CH=CHCO), 5.90 (d, *J* = 15.6 Hz, 1H, CHCO), 5.46 (d, *J* = 5.1 Hz, 1H, OCHO), 4.33-4.12 (m, 3H, OCH_aH_bCO, CH₃CH₂O), 4.05-3.77 (m, 2H, *i*-PrCH, OCH_aH_bCO), 3.03-2.81 (m, 2H, CH=CHCH, CHCH₂CO), 2.64 (dd, *J* = 15.1, 7.2 Hz, 1H, CHCH_aH_bCO), 2.42 (dd, *J* = 15.1, 6.8 Hz, 1H, CHCH_aH_bCO), 1.86-1.70 (m, 1H, (CH₃)₂CH), 1.30 (t, *J* = 7.1 Hz, 3H, OCH₂CH₃), 0.97 (d, *J* = 6.5 Hz, 3H, (CH₃)₂CH), 0.95 (d, *J* = 6.5 Hz, 3H, (CH₃)₂CH). ¹³C NMR (75 MHz, CDCl₃) δ 209.8 (CH₂CO), 165.5 (COO), 143.8 (COCH=CH), 125.1 (COCH=CH), 101.3 (OCHO), 86.6 (*i*-PrCH), 68.9 (COCH₂O), 60.7 (CH₃CH₂O), 47.5 (CHCH=CH), 42.9 (CHCH₂CO), 36.0 (CHCH₂CO), 30.9 (CH₃)₂C), 19.4 (CH₃CH), 17.2 (CH₃CH), 14.2 (CH₃CH₂). IR (ATR): 1718 (C=O) cm⁻¹. MS (70 eV) *m*/*z* (%): 282 (M⁺, 1), 239 (13), 210 (43), 136 (34), 79 (100). HRMS: Calculated for [C₁₅H₂₃O₅]⁺ : 283.1545 [(M+H)⁺]; found: 283.1539. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane/*i*-PrOH (95:5)]; flow rate 1.00 mL/min; τ_{major} = 18.55 min, τ_{minor} = 17.75 min (84% e.e.). [α]_D^{n⁺} : +30.8 (*c* 0.08, CH₂Cl₂).



(E)-3-{(2R,3S,3aR,7aS)-2-[(Z)-hex-3-en-1-yl]-5-Ethyl oxohexahydro-4H-furo[2,3-b]pyran-3-yl]}acrylate (70) Following the general procedure 70 (23 mg, 0.07 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 74% yield as a yellow oil starting from pure 40 (24 mg, 0.09 mmol) and (carbethoxymethylene)triphenylphosphorane (66 mg, 0.19 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 6.77

(dd, J = 15.6, 9.2 Hz, 1H, CH=CHCO), 5.91 (d, J = 15.6 Hz, 1H, CHCO), 5.50 (d, J = 5.7 Hz, 1H, OCHO), 5.48-5.21 (m, 2H, CH₂CH=CH), 4.25-4.18 (m, 3H, OCH₃H_bCO, CH₃CH₂O), 4.11 $(dt, J = 9.1, 3.5 \text{ Hz}, 1\text{H}, CH_2CH_2CH), 3.90 (d, J = 18.4 \text{ Hz}, 1\text{H}, OCH_aH_bCO), 3.00-2.87 (m, J)$ 1H, CHCH₂CO), 2.86-2.74 (m, 1H, CHCH=CH), 2.66 (dd, J = 14.8, 8.6 Hz, CHCH_aH_bCO), 2.42 (dd, J = 14.8, 7.1 Hz, 1H, CHCH_aH_bCO), 2.11-1.91 (m, 4H, CH₂CH₂), 1.67-1.43 (m, 2H, CH₃CH₂CH), 1.30 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 0.96 (t, J = 7.5 Hz, 3H, CHCH₂CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 210.1 (CH₂CO), 165.4 (COO), 142.4 (COCH=CH), 132.8 (EtCH=CH), 127.6 (EtCH=CH), 125.7 (COCH=CH), 101.5 (OCHO), 80.9 (CH=CHCH₂CH₂CH), 68.8 (COCH₂O), 60.7 (CH₃CH₂O), 50.7 (CHCH=CH), 42.4 (CHCH₂CO), 36.1 (CHCH₂CO), 33.5 (CH=CHCH₂CH₂CH), 23.7 (CH=CHCH₂CH₂CH), 20.5 (CH₃CH₂CH), 14.3 (CH₃CH₂CO), 14.0 (CH₃CH₂CH). IR (ATR): 1732, 1718 (C=O) cm⁻¹. MS (70 eV) m/z (%): 322 (M⁺, 2), 210 (61), 136 (48), 79 (100). HRMS: Calculated for $[C_{18}H_{27}O_5]^+$: 323.1858 [(M+H)⁺]; found: 323.1866. The e.e. was determined by HPLC using a Chiralpak IA column [hexanes/*i*-PrOH (95:5)]; flow rate 1.00 mL/min; $\tau_{maior} = 45.91$ min, $\tau_{minor} = 18.02$ min (96% e.e.). $[\alpha]_D^{\text{rt}}$: +25.7 (*c* 0.5, CH₂Cl₂).



Ethyl (E)-3-[(3S,3aR,7aS)-2,2-dimethyl-5-oxohexahydro-4Hfuro[2,3-b]pyran-3-yl]acrylate (7p). Following the general procedure 7p (29 mg, 0.11 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 80% yield as a yellow oil 0.14 starting from pure **4**p (27 mg, mmol) and (carbethoxymethylene)triphenylphosphorane (94 mg, 0.27 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 6.66

(dd, J = 15.5, 9.3 Hz, 1H, CH=CHCO), 5.89 (d, J = 15.5 Hz, 1H, CHCO), 5.62 (d, J = 7.1 Hz, 1H, OCHO), 4.39-4.09 (m, 3H, OCH_aH_bCO, CH₃CH₂), 3.98 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 2.97-2.85 (m, 1H, CHCH₂CO), 2.71 (dd, J = 16.1, 6.2 Hz, 1H, CHCH_aH_bCO), 2.34 (dd, J = 16.1, 2.3 Hz, 1H, CHCH_aH_bCO), 2.17-2.09 (m, 1H, CHCH=CH), 1.33-1.25 (m, 6H, OCH₂CH₃, CH₃), 1.16 (s, 3H, CH₃).¹³C NMR (75 MHz, CDCl₃) δ 209.9 (CH₂CO), 165.6 (COO), 143.5

(COCH=CH), 125.2 (COCH=CH), 99.0 (OCHO), 82.9 ((CH₃)₂C), 66.9 (COCH₂O), 60.7 (CH₃CH₂O), 56.7 (CHCH=CH), 41.7 (CHCH₂CO), 37.3 (CHCH₂CO), 26.7 (CH₃), 23.3 (CH₃), 14.2 (CH₃CH₂). IR (ATR): 1718 (C=O) cm⁻¹. HRMS: Calculated for $[C_{14}H_{21}O_5]^+$: 269.1389 $[(M+H)^+]$; found: 269.1388. The e.e. was determined by HPLC using a Chiralpak AS-H column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 15.05$ min, $\tau_{minor} = 27.06$ min (77% e.e.). $[\alpha]_D^{rt}$: +45.4 (*c* 0.3, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-5-oxo-2-phenylhexahydro-4Hfuro[2,3-*b*]pyran-3-yl]acrylate (8a). Following the general procedure **8a** (21 mg, 0.06 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 79% as a yellow oil starting 6a from pure (20)mg, 0.08 mmol) and (carbethoxymethylene)triphenylphosphorane (56 mg, 0.16 mmol),

using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.54-6.90 (m, 5H, C_{arom}-H), 6.17 (dd, J = 15.7, 9.6 Hz, 1H, CH=CHCO), 6.00 (d, J = 6.1 Hz, 1H, OCHO), 5.79 (d, J = 15.7 Hz, 1H, CH=CHCO), 5.52 (d, J = 7.4 Hz, PhCH), 4.27 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 4.18-3.97 (m, 3H, CH₃CH₂O, OCH_aH_bCO), 2.99-2.91 (m, 1H, CHCH₂CO), 2.85-2.70 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.47 (dd, J = 15.6, 3.5 Hz, 1H, CHCH_aH_bCO), 1.22 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 209.0 (CH₂CO), 165.3 (COO), 144.1 (COCH=CH), 137.8 (C_{arom}-C), 128.5, 128.0, 125.8 (C_{arom}-H), 124.2 (COCH=CH), 101.4 (OCHO), 83.7 (PhCH), 68.1 (COCH₂O), 60.5 (CH₃CH₂O), 52.1 (CHCH=CH), 41.2 (CHCH₂CO), 37.4 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1736, 1720 (C=O) cm⁻¹. MS (70 eV) m/z (%): 271 (M⁺-CH₃CH₂O, 2), 210 (100), 136 (43), 79 (83). HRMS: Calculated for [C₁₈H₂₁O₅]⁺ : 317.1389 [(M+H)⁺]; found: 317.1392. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane/*i*-PrOH (95:5)]; flow rate 1.00 mL/min; $\tau_{major} = 30.99$ min, $\tau_{minor} = 68.43$ min (95% e.e.). [α]_D^{n⁺} : 47.9 (*c* 0.2, CH₂Cl₂).



Ethyl (*E*)-3-[(2*S*,3*R*,3a*R*,7a*S*)-2-(4-methoxyphenyl)-5oxohexahydro-4*H*-furo[2,3-*b*]pyran-3-yl]acrylate (8b). Following the general procedure 8b (16 mg, 0.046 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 83% yield as a yellow oil starting from pure 6b (15 mg, 0.054 mmol) and

PMP (carbethoxymethylene)triphenylphosphorane (38 mg, 0.11 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.05 (d, J = 8.7 Hz, 2H, C_{arom}-H), 6.87 (d, J = 8.7 Hz, 2H, C_{arom}-H), 6.22 (dd, J = 15.6, 9.5 Hz, 1H, CH=CHCO), 5.97 (d, J = 6.1Hz, 1H, OCHO), 5.79 (d, J = 15.6 Hz, 1H, CH=CHCO), 5.48 (d, J = 7.2 Hz, PMPCH), 4.26 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 4.21-3.95 (m, 3H, CH₃CH₂O, OCH_aH_bCO), 3.80 (s, 3H, CH₃O), 2.96-2.70 (m, 3H, CHCH₂CO, CHCH=CH, CHCH_aH_bCO), 2.47 (dd, J = 15.6, 3.5 Hz, 1H, CHCH_aH_bCO), 1.23 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 209.1 (CH₂CO), 165.4 (COO), 159.2 (COCH=CH), 144.4 (COCH=CH), 129.9 (C_{arom}-C), 127.2 (C_{arom}-H), 127.1 (C_{arom}-C), 124.0 (C_{arom}-H), 101.3 (OCHO), 83.4 (PMPCH), 68.1 (COCH₂O), 60.4 (CH₃CH₂), 55.2 (CH₃O), 52.1 (CHCH=CH), 41.3 (CHCH₂CO), 37.4 (CHCH₂CO), 14.1 (CH₃CH₂). IR (ATR): 1713 (C=O) cm⁻¹. MS (70 eV) m/z (%): 346 (M⁺, 1), 210 (100), 137 (95), 79 (68). HRMS: Calculated for [C₁₉H₂₃O₆]⁺: 347.1495 [(M+H)⁺]; found: 347.1479. The e.e. was determined by HPLC using a Chiralcel OZ-3 column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 66.91$ min, $\tau_{minor} = 55.59$ min (86% e.e.). [α]_D^{rt}: +45.8 (*c* 0.2, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-2-(4-fluorophenyl)-5-oxohexahydro-4H-furo[2,3-b]pyran-3-yl]acrylate (8c). Following the general procedure 8c (24 mg, 0.07 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 82% yield as a yellow oil 0.09 starting from pure 6c (23 mg, mmol) and (carbethoxymethylene)triphenylphosphorane (60 mg, 0.17 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.22-6.82 (m, 4H, C_{arom}-H), 6.17 (dd, J = 15.6, 9.6 Hz, 1H, CH=CHCO), 5.97 (d, J = 6.0 Hz, 1H, OCHO), 5.77 (d, J = 15.6 Hz, 1H,

CH=CHCO), 5.50 (d, J = 7.3 Hz, 4-FC₆H₄CH), 4.25 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 4.19-3.88 (m, 3H, CH₃CH₂, OCH_aH_bCO), 2.99-2.90 (m, 1H, CHCH₂CO), 2.83-2.70 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.52-2.46 (m, 1H, CHCH_aH_bCO), 1.22 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 208.8 (CH₂CO), 165.3 (COO), 162.3 (d, ¹J_{CF} = 246.6 Hz, C_{arom}-F), 143.9 (COCH=CH), 133.5 (d, ⁴J_{CF} = 3.2 Hz, C_{arom}-C), 127.5 (d, ³J_{CF} = 8.2 Hz, C_{arom}-H), 124.3 (COCH=CH), 115.5 (d, ²J_{CF} = 21.6 Hz, C_{arom}-H), 101.4 (OCHO), 83.0 (4-FC₆H₄CH), 68.3 (COCH₂O), 60.6 (CH₃CH₂), 52.1 (CHCH=CH), 41.5 (CHCH₂CO), 37.5 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1716 (C=O) cm⁻¹. MS (70 eV) *m*/*z* (%): 289 (M⁺-OEt, 2), 210 (96), 136 (52), 79 (100). HRMS: Calculated for [C₁₈H₂₀FO₅]⁺: 335.1295 [(M+H)⁺]; found: 335.1287. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{maior} = 20.09$ min, $\tau_{minor} = 47.81$ min (92% e.e.). [α]_D^{rt}: +37.9 (*c* 0.9, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-2-(4-nitrophenyl)-5-oxohexahydro-4H-furo[2,3-b]pyran-3-yl]acrylate (8d). Following the general procedure 8d (48 mg, 0.13 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 77% yield as a yellow oil starting from pure **6d** (50 mg, 0.17 mmol) and (carbethoxymethylene)triphenylphosphorane (119 mg, 0.34 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) & 8.22 (d, J = 8.7 Hz, 2H, C_{arom}-H), 7.35 (d, J = 8.7 Hz, 2H, C_{arom}-H), 6.15 (dd, J = 15.5, 9.6 Hz, 1H, CH=CHCO), 5.97 (d, J = 5.8 Hz, 1H,

OCHO), 5.80 (d, J = 15.5 Hz, 1H, CH=CHCO), 5.60 (d, J = 7.3 Hz, 4-NO₂C₆H₄CH), 4.27 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 4.16-4.05 (m, 3H, CH₃CH₂O, OCH_aH_bCO), 3.12-3.00 (m, 1H, CHCH₂CO), 2.84-2.73 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.58-2.50 (m, 1H, CHCH_aH_bCO), 1.22 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 208.2 (CH₂CO), 165.0 (COO), 147.6 (C_{arom}-N), 145.0 (C_{arom}-C), 142.9 (COCH=CH), 126.8 (C_{arom}-H), 124.9 (COCH=CH), 123.8 (C_{arom}-H), 101.5 (OCHO), 82.4 (4-NO₂C₆H₄CH), 68.5 (COCH₂O), 60.7 (CH₃CH₂), 52.5 (CHCH=CH), 42.1 (CHCH₂CO), 37.6 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1715, 1519 (C=O), 1345 (NO₂) cm⁻¹. HRMS: Calculated for [C₁₈H₂₀NO₇]⁺: 362.1240 [(M+H)⁺]; found: 362.1237. The e.e. was determined by HPLC using a Chiralcel OZ-3 column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 203.84$ min, $\tau_{minor} = 187.35$ min (92% e.e.). [α]_Dⁿ: +72.2 (*c* 0.2, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-2-(4-bromophenyl)-5oxohexahydro-4H-furo[2,3-b]pyran-3-yl]acrylate (8e). Following the general procedure 8e (73 mg, 0.19 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 85% yield as a yellow oil starting from 0.22 pure **6e** (70 mg, mmol) and (carbethoxymethylene)triphenylphosphorane (150 mg, 0.43 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.47 (d, J = 8.4 Hz, 2H, C_{arom}-H), 7.02 (d, J = 8.4 Hz, 2H, C_{arom}-H), 6.17 (dd, J = 15.5, 9.7 Hz, 1H, CH=CHCO), 5.95 (d, J = 6.0 Hz, 1H,

OCHO), 5.78 (d, J = 15.5 Hz, 1H, CH=CHCO), 5.47 (d, J = 7.3 Hz, 4-BrC₆H₄CH), 4.25 (d, J = 18.2 Hz, 1H, OCH_aH_bCO), 4.20-3.99 (m, 3H, CH₃CH₂O, OCH_aH_bCO), 3.01-2.90 (m, 1H, CHCH₂CO), 2.87-2.63 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.49 (dd, J = 13.2, 5.4 Hz, 1H, CHCH_aH_bCO), 1.23 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 208.7 (CH₂CO), 165.2 (COO), 143.6 (COCH=CH), 136.8 (C_{arom}-C), 131.7, 127.5 (C_{arom}-H), 124.5 (COCH=CH), 121.9 (C_{arom}-Br), 101.4 (OCHO), 82.9 (4-BrC₆H₄CH), 68.3 (COCH₂O), 60.6 (CH₃CH₂), 52.0 (CHCH=CH), 41.5 (CHCH₂CO), 37.4 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1713 (C=O) cm⁻¹. MS (70 eV) m/z (%): 210 (100), 136 (68), 79 (89). HRMS: Calculated for $[C_{18}H_{20}BrO_5]^+$: 395.0494 [(M+H)⁺]; found: 395.0483. The e.e. was determined by HPLC using a Chiralpak IA column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 59.37$ min, $\tau_{minor} = 23.37$ min (88% e.e.). $[\alpha]_D^{-n}$: +51.9 (*c* 1.9, CH₂Cl₂).



Ethvl (E)-3-[(2S,3R,3aR,7aS)-2-(2-methoxyphenyl)-5oxohexahydro-4H-furo[2,3-b]pyran-3-yl]acrylate (8f). Following the general procedure 8f (54 mg, 0.16 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 86% yield as a yellow oil starting from pure **6f** (50 mg, 0.18 mmol) and (carbethoxymethylene)triphenylphosphorane (125 mg, 0.36 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.31-7.21 (m, 2H, C_{arom} -H), 7.02-6.91 (m, 1H, C_{arom} -H), 6.82 (d, J = 8.5

Hz, 1H, C_{arom}-H), 6.25 (dd, J = 15.5, 9.6 Hz, 1H, CH=CHCO), 5.95 (d, J = 6.0 Hz, 1H, OCHO), 5.79 (d, J = 7.3 Hz, 2-MeOC₆H₄CH), 5.71 (d, J = 15.5 Hz, CH=CHCO), 4.27 (d, J = 18.3 Hz, 1H, OCH_aH_bCO), 4.19-3.90 (m, 3H, CH₃CH₂, OCH_aH_bCO), 3.73 (s, 3H, OCH₃), 3.06-2.94 (m, 1H, CHCH₂CO), 2.78-2.65 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.50 (dd, J = 17.9, 6.5 Hz, 1H, CHCH_aH_bCO), 1.21 (t, J = 7.1 Hz, 3H, CH₂CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 209.4 (CH₂CO), 165.7 (COO), 155.6 (C_{arom}-O), 145.1 (COCH=CH), 129.0 (C_{arom}-H), 126.4 (C_{arom}-C), 126.1, 122.7 (C_{arom}-H), 120.6 (COCH=CH), 110.1 (C_{arom}-H), 101.1 (OCHO), 79.7 (2-MeOC₆H₄CH), 68.3 (COCH₂O), 60.3 (CH₃CH₂), 55.0 (CHCH=CH), 51.2 (OCH₃), 41.5 (CHCH₂CO), 37.6 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1733, 1715 (C=O) cm⁻¹. MS (70 eV) m/z (%): 346 (M⁺, 1), 210 (100), 137 (84), 79 (70). HRMS: Calculated for [C₁₉H₂₃O₆]⁺: 347.1495 [(M+H)⁺]; found: 347.1500. The e.e. was determined by HPLC using a Chiralcel OD-3 column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 19.01$ min, $\tau_{minor} = 22.91$ min (94% e.e.). [α]_D^{n⁺}: -14.4 (*c* 0.4, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-2-(2-nitrophenyl)-5-oxohexahydro-4H-furo[2,3-b]pyran-3-yl]acrylate (8g). Following the general procedure 8g (44 mg, 0.12 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 79% yield as a yellow oil starting from pure 6g (50 mg, 0.15 mmol) and (carbethoxymethylene)triphenylphosphorane (107 mg, 0.31 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 8.10 (dd, J = 8.2, 1.0 Hz, 1H, Carom-H), 7.76-7.64 (m, 2H, Carom-H), 7.56-

7.38 (m, 1H, C_{arom}-H), 6.33-6.14 (m, 2H, COCH=CH, 2-NO₂C₆H₄CH), 5.89 (d, J = 5.4 Hz, 1H, OCHO), 5.71 (d, J = 15.5 Hz, COCH=CH), 4.28 (d, J = 18.2 Hz, 1H, OCH_aH_bCO), 4.19-3.94 (m, 3H, CH₃CH₂, OCH_aH_bCO), 3.42-3.30 (m, 1H, CHCH₂CO), 2.86-2.58 (m, 3H, CHCH=CH, CHCH₂CO), 1.19 (t, J = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 208.2 (CH₂CO), 165.4 (COO), 147.0 (C_{arom}-N), 143.4 (COCH=CH), 134.1 (C_{arom}-H), 134.0 (C_{arom}-C), 128.9, 127.9, 125.3 (C_{arom}-H), 124.1 (COCH=CH), 101.4 (OCHO), 80.2 (2-NO₂C₆H₄CH), 69.2 (COCH₂O), 60.5 (CH₃CH₂), 51.9 (CHCH=CH), 43.0 (CHCH₂CO), 37.9 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1715, 1525 (C=O) cm⁻¹. MS (70 eV) *m*/*z* (%): 210 (63), 152 (31), 136 (45), 79 (100). HRMS: Calculated for [C₁₈H₂₀NO₇]⁺: 362.1240 [(M+H)⁺]; found: 362.1225. The e.e. was determined by HPLC using a Chiralcel OZ-3 column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 98.25$ min, $\tau_{minor} = 123.10$ min (92% e.e.). [α]_D^{n⁺}: +35.3 (*c* 0.4, CH₂Cl₂).



Ethyl (E)-3-[(2S,3R,3aR,7aS)-2-(furan-2-yl)-5-oxohexahydro-4Hfuro[2,3-*b*]pyran-3-yl]acrylate (8h). Following the general procedure 8h (22 mg, 0.07 mmol) was isolated by FC (hexanes/EtOAc 1:1) after 30 min in 75% yield as a yellow oil starting from 0.09 pure 6h (22 mg, mmol) and (carbethoxymethylene)triphenylphosphorane (63 mg, 0.19 mmol), using THF (1.0 mL) as solvent. ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.38 (m, 1H, $C_{hetarom}$ -H), 6.35-6.21 (m, 3H, 2 \times $C_{hetarom}$ -H,

CH=CHCO), 5.95 (d, *J* = 7.1 Hz, 1H, OCHO), 5.86 (d, *J* = 15.6 Hz, CH=CHCO), 5.28 (d, *J* = 7.6 Hz, C_{hetarom}-CH), 4.23 (d, *J* = 18.4 Hz, 1H, OCH_aH_bCO), 4.19-3.96 (m, 3H, CH₃CH₂, OCH_aH_bCO), 3.36-3.13 (m, 1H, CHCH₂CO), 2.95-2.64 (m, 2H, CHCH=CH, CHCH_aH_bCO), 2.43 (dd, *J* = 16.2, 2.6 Hz, 1H, CHCH_aH_bCO), 1.24 (t, *J* = 7.1 Hz, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 209.5 (CH₂CO), 165.4 (COO), 151.8 (C_{hetarom}-C), 143.0 (COCH=CH), 142.5 (C_{hetarom}-H), 125.1 (COCH=CH), 110.4, 109.4 (C_{hetarom}-H), 101.0 (OCHO), 76.15 (C_{hetarom}-CH), 67.2 (COCH₂O), 60.5 (CH₃CH₂), 55.7 (CHCH=CH), 40.9 (CHCH₂CO), 37.2 (CHCH₂CO), 14.1 (CH₃). IR (ATR): 1716 (C=O) cm⁻¹. MS (70 eV) *m*/*z* (%): 210 (90), 136 (44), 91 (33), 79 (100). HRMS: Calculated for $[C_{16}H_{19}O_6]^+$: 307.1182 $[(M+H)^+]$; found: 307.1202. The e.e. was determined by HPLC using a Chiralcel OZ-3 column [*n*-hexane/*i*-PrOH (90:10)]; flow rate 1.00 mL/min; $\tau_{major} = 34.63$ min, $\tau_{minor} = 31.09$ min (97% e.e.). $[\alpha]_D^{rt}$: +47.3 (*c* 0.3, CH₂Cl₂).



(2*S*,3*S*,3a*S*,7a*R*)-3-(1,3-dioxolan-2-yl)-2-phenyltetrahydro-4*H*,6*H*-spiro[furo[2,3-b]pyran-5,2'-[1,3]dioxolane] (9a).

A solution of **5a** (20 mg, 0.081 mmol) and *p*-TSA (7 mg, 0.04 mmol) was stirred in 2-ethyl-2-methyl-1,3-dioxolane (1.0 mL) at room temperature for 18h. After completion of the reaction compound **9a** (20 mg, 0.06 mmol) was isolated by FC (hexanes/EtOAc 1:1) in 74% yield as a white solid. ¹H NMR (300 MHz, CDCl₃) δ 7.52 (d, *J* = 7.1 Hz, 2H), 7.39-7.24 (m, 3H), 5.27 (d, *J* = 4.1 Hz, 1H), 4.97 (d, *J* = 2.1 Hz, 2H), 4.95 (d, 1.9

Hz, 1H), 4.10-3.75 (m, 5H), 3.49 (d, J = 12.0 Hz, 1H), 2.92 (td, J = 8.8, 3.9 Hz, 1H), 2.66- 2.50 (m, 1H), 2.11 (ddd, J = 10.7, 4.3, 2.1 Hz, 1H), 1.96 (dd, J = 14.1, 6.5 Hz, 1H). ¹³C NMR (75 MHz, CDCl₃) δ 142.1, 128.2, 127.4, 126.9, 104.1, 104.0, 101.2, 82.5, 66.5, 65.0, 64.9, 64.6, 64.3, 50.6, 39.8, 32.5. IR (ATR): 1155, 1065 (O-C-O) cm⁻¹. MS (70 eV) m/z (%): 334 (1), 289 (2), 231 (2), 201 (68), 159 (5), 142 (6), 128 (9), 115 (8), 103 (4), 87 (27), 73 (100), 55 (3). HRMS: Calculated for $[C_{18}H_{23}O_6]^+$: 335.1495 $[(M+H)^+]$; found: 335.1487. $[\alpha]_D^{rt}$: +17.8 (*c* 0.61, CH₂Cl₂).



(2*S*,3*R*,3a*R*,7a*S*)-3-(1,3-dioxolan-2-yl)-2-phenyltetrahydro-4*H*,6*H*-spiro[furo[2,3-b]pyran-5,2'-[1,3]dioxolane] (10a).

A solution of **6a** (52 mg, 0.211 mmol) and *p*-TSA (20 mg, 0.12 mmol) was stirred in 2-ethyl-2-methyl-1,3-dioxolane (2.0 mL) at room temperature for 18h. After completion of the reaction compound **10a** (46 mg, 0.14 mmol) was isolated by FC (hexanes/EtOAc 1:1) in 65% yield as a white solid. ¹³C NMR (75 MHz, CDCl₃) δ 138.8, 127.7, 127.3, 127.2, 104.5, 103.9, 101.7, 81.8, 67.0, 64.8, 64.6, 64.3, 64.2, 47.8, 38.2, 32.3. IR

(ATR): 1070 (O-C-O) cm⁻¹. MS (70 eV) m/z (%): 334 (1), 232 (5), 201 (41), 191 (11), 159 (4), 128 (5), 115 (7), 105 (7), 87 (15), 73 (100), 55 (3). HRMS: Calculated for $[C_{18}H_{23}O_6]^+$: 335.1495 [(M+H)⁺]; found: 335.1490. $[\alpha]_D^{rt}$: -44.8 (*c* 0.15, CH₂Cl₂).



[(1S,3S,3aR,4S,8aR)-1,4-Isopropoxy-3-phenyltetrahydro-1H,3H-furo[3,4-c]oxepin-7(6H)-one (11a).

A solution of **9a** (45 mg, 0.18 mmol) and *p*-TSA (70 mg, 0.39 mmol) was stirred in *i*-PrOH/DCM (1.0/2.0 mL) at room temperature for 18h. After completion of the reaction compound **11a** (34 mg, 0.10 mmol) was isolated by FC (hexanes/EtOAc gradient from 9:1 to 7:3) in 54% yield as a yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.20 (m, 4H, C_{arom}-H), 5.78 (d, *J* = 6.5 Hz, 1H, *i*-PrOCHOCH), 5.39 (d, *J* = 7.1 Hz, 1H, *i*-

PrOCHOCH₂), 4.25 (d, J = 18.4 Hz, 1H, OCH_aH_bCO), 4.07-3.98 (m, 2H, OCH_aH_bCO, CHPh), 3.60-3.47 (m, 2H, CH_a(CH₃)₂, CH_b(CH₃)₂), 3.44-3.13 (m, 1H, COCH₂CH), 2.88-2.53 (COCH₂CH), 2.53-2.26 (m, 1H, CHCHPh), 1.07 (d, J = 6.2 Hz, 6H, CH_{3ab}), 0.98 (d, J = 6.1 Hz, 3H, CH_{3c}), 0.73 (d, J = 6.1 Hz, 3H, CH_{3d}). ¹³C NMR (75 MHz, CDCl₃) 211.6 (CO), 138.3 (C_{arom}-C), 128.1, 127.4, 126.5 (C_{arom}-H), 102.2 (*i*-PrOCHOCH), 97.6 (*i*-PrOCHOCH₂), 81.8 (CHPh), 68.5 (*i*-Pr_a), 68.3 (OCH₂CO), 68.2 (*i*-Pr_b), 54.3 (CHCHPh), 40.6 (COCH₂CH), 38.1 (COCH₂CH), 23.3 (CH_{3a}), 22.9 (CH_{3b}), 22.8 (CH_{3c}), 20.9 (CH_{3d}). IR (ATR): 1734 cm⁻¹. MS (EI) m/z (%): 289 (1), 210 (33), 141 (92), 128 (52), 115 (59), 89 (100). HRMS: Calculated for [C₁₄H₁₃O₃]⁺: 229.0844 [(M-(O*i*-Pr)- (O*i*-Pr)⁺]; found: 229.0896. [α]_D^{rt}: -6.8 (c = 1.2, CH₂Cl₂).



(2R,3S,3aR,5S,7aS)-5-hydroxy-2-propylhexahydro-4*H*-3,5-methanofuro[2,3-b]pyran-8-one (12k).

A solution of **4k** (48 mg, 0.226 mmol) was dissolved in CH_2Cl_2 (1 mL) and cooled to 0°C. Then a solution of 1,3-dimethyl-1*H*-imidazol-3-ium bromide (*N*-heterocyclic carbene, 17 mg, 0.05 mmol) and Et₃N (38 mL, 0.276 mmol) in CH_2Cl_2 (1 mL) was added dropewise and the reaction mixture was further

stirred at 0°C for 18h. After completion of the reaction the crude was directly charged onto silicagel column from where compound **12k** (46 mg, 0.14 mmol) was isolated (hexanes/EtOAc 7:3) in 58% yield. ¹H NMR (500 MHz, CDCl₃) δ 5.67 (dd, J = 11.1, 5.0 Hz, 1H), 4.41-4.30 (m, 1H), 3.72 (d, J = 12.6 Hz, 2H), 3.70 (d, J = 11.6 Hz, 2H), 3.16 (dt, J = 7.1, 5.4 Hz, 1H), 3.04 (s, 1H), 2.59 (dd, J = 8.3, 1.0 Hz, 1H), 2.39 (d, J = 11.6 Hz, 1H), 1.85 (dd, J = 11.6, 4.7 Hz, 1H), 1.54-1.15 (m, 6H), 0.94 (t, J = 7.1 Hz, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 102.8, 83.2, 78.2, 68.4, 51.5, 39.6, 39.3, 29.9, 18.6, 13.7. IR (ATR): 1733 (C=O) cm⁻¹. [α]_D^{rt}: -69.3 (*c* 0.18, CH₂Cl₂).

3. Determination of the Absolute Configuration of 4l, 6a and 9a (obtained from 5a) **3.1** Determination of the absolute configuration of 4l



Figure ESI-1. ORTEP diagram for compound (2R,3S,3aR,7aS)-4l

Crystal data for C₁₃H₂₀O₄ (M = 240.29 g/mol): CCDC 1525188, orthorhombic, space group P2₁2₁2₁, a = 5.56040(10) Å, b = 6.68950(10) Å, c = 33.0826(5) Å, $a = \beta = \gamma = 90^{\circ}$, V = 1230.55(3) Å³, Z = 4, T = 100(2) K, μ (CuK α) = 0.779 mm⁻¹, *Dcalc* = 1.297 g/cm³, 22597 reflections measured (5.34° $\leq 2\Theta \leq 142^{\circ}$), 2369 unique ($R_{int} = 0.0418$, $R_{sigma} = 0.0191$) which were used in all calculations. The final R_1 was 0.0306 (I > 2 σ (I)) and wR_2 was 0.0711 (all data).

Table ESI-1.	Crystal	data and	l structure	refinement	for 41 .
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Empirical formula	$C_{13}H_{20}O_4$
Formula weight	240.29
Temperature/K	100(2)
Crystal system	
Space group	$P2_{1}2_{1}2_{1}$
a/Å	5.56040(10)
b/Å	6.68950(10)
c/Å	33.0826(5)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	1230.55(3)
Z	4
$\rho_{calc}g/cm^3$	1.297
μ/mm^{-1}	0.779
F(000)	520.0
Crystal size/mm ³	$0.1821 \times 0.0405 \times 0.0323$
Radiation	$CuK\alpha (\lambda = 1.54184)$
2Θ range for data collection/°	5.34 to 142
Index ranges	$-6 \le h \le 6, -8 \le k \le 8, -40 \le l \le 40$
Reflections collected	22597
Independent reflections	2369 [$R_{int} = 0.0418$, $R_{sigma} = 0.0191$]
Data/restraints/parameters	2369/0/155
Goodness-of-fit on F ²	1.088
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0306, wR_2 = 0.0702$
Final R indexes [all data]	$R_1 = 0.0327, wR_2 = 0.0711$
Largest diff. peak/hole / e $Å^{-3}$	0.18/-0.16
Flack parameter	-0.13(18)

Atom	x	у	z	U(eq)
01	8012(2)	8866(2)	2526.1(3)	33.7(3)
O2	9841(2)	5722.5(14)	1088.7(3)	18.6(2)
O3	11219.7(19)	7200.1(16)	1660.0(3)	20.9(2)
O4	5758(2)	1311.0(16)	1672.2(4)	30.1(3)
C1	8578(3)	7984(2)	2219.7(4)	21.6(3)
C2	7245(3)	6226(2)	2047.4(4)	19.7(3)
C3	6891(3)	6497(2)	1589.1(4)	15.4(3)
C4	6267(3)	4526(2)	1365.5(4)	15.7(3)
C5	8680(3)	3863(2)	1193.5(4)	16.4(3)
C6	9270(3)	7146(2)	1382.7(4)	16.2(3)
C7	10757(3)	8602(2)	1976.1(5)	24.7(3)
C8	5031(3)	2997(2)	1626.5(4)	19.1(3)
C9	8631(3)	2551(2)	822.2(4)	18.6(3)
C10	11130(3)	1847(2)	694.7(4)	19.9(3)
C11	11114(3)	409(2)	336.3(4)	20.7(3)
C12	13603(3)	-477(2)	248.7(4)	21.4(3)
C13	14370(3)	-2023(2)	558.7(5)	25.0(3)

Table ESI-2. Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for **4l**. U_{eq} is defined as 1/3 of of the trace of the orthogonalised U_{IJ} tensor.

3.2 Determination of the absolute configuration of 6a



Figure ESI-2. ORTEP diagram for compound (2S,3R,3aR,7aS)-6a

Crystal data for C₁₄H₁₄O₄ (M = 246.25 g/mol): CCDC 1525189, monoclynic, space group P2₁ (no.4), a = 9.16074(6) Å, b = 6.01447(4) Å, c = 10.6243(1) Å, $\gamma = 92.0730(10)^{\circ}$, V = 584.984(10) Å³, Z = 2, T = 100(2) K, μ (CuK α) = 0.849 mm⁻¹, *Dcalc* = 1.398 g/cm³, 21603 reflections measured ($8.32^{\circ} \le 2\Theta \le 143.96^{\circ}$), 2281 unique ($R_{int} = 0.0556$, $R_{sigma} = 0.0187$) which were used in all calculations. The final R_1 was 0.0266 (I > 2 σ (I)) and wR_2 was 0.0716 (all data).

Table ESI-3. Crystal data and structure refinement for Ua	Table ESI-3.	Crystal	data and	structure	refinement	for 6a .
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Empirical formula	$C_{14}H_{14}O_4$
Formula weight	246.25
Temperature/K	100(2)
Crystal system	
Space group	P2 ₁
a/Å	9.16074(6)
b/Å	6.01447(4)
c/Å	10.6243(1)
$\alpha/^{\circ}$	90
β/°	92.0730(10)
$\gamma/^{\circ}$	90
Volume/Å ³	584.984(8)
Z	2
$\rho_{calc}g/cm^3$	1.398
μ/mm^{-1}	0.849
F(000)	260.0
Crystal size/mm ³	0.234 imes 0.1168 imes 0.0601
Radiation	$CuK\alpha \ (\lambda = 1.54184)$
20 range for data collection/°	8.32 to 143.96
Index ranges	$-11 \le h \le 11, -7 \le k \le 7, -12 \le l \le 13$
Reflections collected	21603
Independent reflections	2281 [$R_{int} = 0.0556$, $R_{sigma} = 0.0187$]
Data/restraints/parameters	2281/1/163
Goodness-of-fit on F ²	1.073
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0266, wR_2 = 0.0714$
Final R indexes [all data]	$R_1 = 0.0267, wR_2 = 0.0716$
Largest diff. peak/hole / e Å ⁻³	0.21/-0.17
Flack parameter	-0.03(13)

Atom	x	у	z	U(eq)
O2	1409.5(9)	3845.0(14)	2425.7(8)	19.66(19)
01	5804.7(9)	-1290.5(18)	4098.4(9)	29.7(2)
03	3780.0(9)	2754.4(16)	2329.3(8)	23.1(2)
O4	-595.6(10)	1779.4(15)	4877.0(9)	26.9(2)
C11	-2851.2(12)	2055(2)	-99.4(11)	21.9(3)
C6	2595.5(12)	3050.0(19)	3160.1(11)	17.7(2)
C10	-1528.8(12)	1380(2)	468.3(11)	18.9(2)
C14	-1486.9(13)	4664(2)	1757.8(12)	20.8(2)
C13	-2800.1(14)	5350(2)	1172.7(13)	24.1(3)
C8	-648.4(12)	751.2(19)	3903.3(11)	19.2(2)
C9	-836.7(12)	2687(2)	1392.7(10)	16.4(2)
C1	4799.9(12)	-230(2)	3619.4(11)	19.8(2)
C5	623.2(12)	1937.9(19)	1943.8(11)	16.1(2)
C3	2130.3(11)	826.3(18)	3747.0(11)	14.6(2)
C4	639.3(12)	326.9(17)	3086.7(10)	15.0(2)
C2	3244.8(12)	-1032(2)	3546.4(11)	18.9(2)
C12	-3476.0(13)	4060(2)	244.2(12)	23.7(3)
C7	5053.0(12)	1995(2)	3005.7(12)	22.6(3)

Table ESI-4. Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for **6a**. U_{eq} is defined as 1/3 of of the trace of the orthogonalised U_{IJ} tensor.

3.2 Determination of the absolute configuration of 9a (obtained from 5a)



Figure ESI-3. ORTEP diagram for compound (2S,3S,3aS,7aR)-9a

Crystal data for C₁₄H₁₄O₄ (M = 246.25 g/mol): CCDC 1525190, monoclynic, space group P2₁ (no.4), a = 9.16074(6) Å, b = 6.01447(4) Å, c = 10.6243(1) Å, $\gamma = 92.0730(10)^{\circ}$, V = 584.984(10) Å³, Z = 2, T = 100(2) K, μ (CuK α) = 0.849 mm⁻¹, *Dcalc* = 1.398 g/cm³, 21603 reflections measured ($8.32^{\circ} \le 2\Theta \le 143.96^{\circ}$), 2281 unique ($R_{int} = 0.0556$, $R_{sigma} = 0.0187$) which were used in all calculations. The final R_1 was 0.0266 (I > 2 σ (I)) and wR_2 was 0.0716 (all data).

Table ESI-5.	Crystal	data and	structure	refinement	for 11 .
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Empirical formula	$C_{18}H_{22}O_{6}$
Formula weight	334.37
Temperature/K	100.15
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
a/Å	8.96150(9)
b/Å	10.02283(10)
c/Å	17.71277(19)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	1590.95(3)
Z	4
$\rho_{calc}g/cm^3$	1.3959
μ/mm^{-1}	0.869
F(000)	714.6
Radiation	Cu Ka ($\lambda = 1.54184$)
2Θ range for data collection/°	9.98 to 144.88
Index ranges	$-11 \le h \le 11, -12 \le k \le 12, -22 \le l \le 22$
Reflections collected	31492
Independent reflections	3160 [$R_{int} = 0.0406$, $R_{sigma} = 0.0167$]
Data/restraints/parameters	3160/0/216
Goodness-of-fit on F ²	1.063
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0246, wR_2 = 0.0631$
Final R indexes [all data]	$R_1 = 0.0255, wR_2 = 0.0638$
Largest diff. peak/hole / e Å ⁻³	0.16/-0.19
Flack parameter	0.06(11)

Atom	x	у	Z	U(eq)
01	-5087.7(10)	-6750.8(8)	-2084.2(5)	22.56(18)
O4	-5669.8(9)	-7465.3(8)	-3261.3(4)	20.22(18)
O12	-10414.8(9)	-8928.2(8)	-2939.8(4)	19.72(18)
015	-10756.4(10)	-5542.3(9)	-3983.2(5)	22.97(19)
08	-7985.3(9)	-9191.8(8)	-2588.4(5)	18.85(17)
016	-9504.7(10)	-6530.9(8)	-4944.9(4)	20.27(17)
C20	-9402.6(14)	-10750.2(11)	-4041.7(6)	20.4(2)
C5	-6739.3(14)	-8653.4(12)	-2188.6(7)	20.4(2)
C17	-10123.7(12)	-9575.9(11)	-4253.6(6)	16.8(2)
C11	-9062.9(12)	-7358.0(11)	-3691.3(6)	15.4(2)
C13	-9381.5(13)	-6155.3(11)	-4179.8(6)	17.4(2)
C21	-10709.7(13)	-9477.4(12)	-4980.1(7)	21.6(2)
C9	-8988.8(12)	-6998.8(11)	-2845.6(6)	16.0(2)
C22	-9293.5(15)	-11811.5(12)	-4544.8(7)	23.0(2)
C1BA	-7563.0(12)	-6353.1(11)	-2544.4(6)	17.0(2)
C24	-9878.5(14)	-11707.1(12)	-5268.5(7)	22.3(2)
C10	-9277.3(13)	-8363.7(11)	-2504.7(7)	18.0(2)
C2	-6259.7(12)	-7311.0(11)	-2524.5(6)	17.4(2)
C14	-10311.2(12)	-8423.4(11)	-3707.9(6)	17.2(2)
C3	-3715.2(13)	-7251.3(13)	-2392.1(8)	24.9(3)
C18	-11437.5(15)	-5089.4(14)	-4670.9(7)	26.2(3)
C23	-10574.2(14)	-10532.7(13)	-5484.8(7)	23.7(2)
C7	-4096.7(14)	-7681.7(17)	-3190.7(8)	33.0(3)
C19	-10318.3(14)	-5453.7(12)	-5280.2(7)	24.0(3)

Table ESI-6. Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for **11**. U_{eq} is defined as 1/3 of of the trace of the orthogonalised U_{IJ} tensor.

4. Computational Methods

All of the calculations were performed using the Gaussian09 program.³ Molecular geometries were optimized with the M06-2X functional,⁴ in conjunction with cc-pVTZ basis set, a triplequality set.⁵ This combination of functional and basis has demonstrated to give acceptable geometries and energies quantitatively consistent with higher levels estimating that errors due to BSSE (basis set superposition error) differences between stereoisomers can be of the order of 0.2 kcal/mol for TZVP basis.⁶ Therefore is strongly recommended for calculating energy differences between alternative transition states in stereoselective organocatalytic reactions.⁷ Moreover, M06-2X functional performs well for systems with dispersion and ionic hydrogenbonding interactions,⁸ and in conjunction with minimally augmented basis sets such as ccpVTZ.All calculations have been carried out considering solvent effects (CHCl₃) with the PCM model.⁹ Analytical second derivatives of the energy were calculated to classify the nature of every stationary point, to determine the harmonic vibrational frequencies, and to provide zeropoint vibrational energy corrections. The thermal and entropic contributions to the free energies were also obtained from the vibrational frequency calculations, using the unscaled frequencies. All discussions are based on values of free energies (G). However, several of the individual reactions involved on the study are bimolecular processes. In order to avoid errors due to entropic effects when comparing all stationery points in an only energy diagram, we used corrected free energy (G_{corr}) values following Sasaki's model.¹⁰ Translational and rotational degrees of freedom in solution are highly suppressed owing to the interactions with solvent molecules and these interactions are not well-estimated by continuum solvent models like PCM; in consequence, thermodynamic corrections to potential energies calculated by using continuum solvation models overestimate the contributions of translational and rotational degrees of freedom to the entropy.¹¹ According to Sasaki's model only vibrational contributions to entropy must be considered calculating free energy as follows:

 $\Delta G_{\rm corr} = \Delta H - T \cdot \Delta S_{\rm vib}$

 ³ M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. J. A. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, Gaussian, Inc., Wallingford CT,: 2009.

⁴ Y. Zhao, D. G. Truhlar, Acc. Chem. Res., 2008, **41**, 157-167.

 ⁵ (a) T. H. Dunning Jr., J. Chem. Phys., 1989, 90, 1007-1023. (b) R. A. Kendall, T. H. Dunning Jr., R. J. Harris, J. Chem. Phys., 1992, 96, 6796-6806. (c) D. E. Woon, T. H. Dunning Jr., J. Chem. Phys., 1993, 98, 1358-1371.

⁶ A. Armstrong, R. A. Boto, P. Dingwall, J. Contreras-Garcia, M. J. Harvey, N. J. Masona, H. S. Rzepa, *Chem. Sci.*, 2014, **4**, 2057-2071.

⁷ S. Schenker, C. Schneider, S. B. Tsogoeva, T. Clark, J. Chem. Theory Comp., 2011, 7, 3586-3595

⁸ M. Walker, A. J. A. Harvey, A. Sen, C. E. H. Dessent, J. Phys. Chem. A, 2013, **117**, 12590

 ⁹ (a) V. Barone, M. Cossi, J. Tomasi, J. Comput. Chem., 1998, 19, 404-417. (b) M. Cossi, G. Scalmani, N. Rega, V. Barone, J. Chem. Phys., 2002, 117, 43-54. (c) J. Tomasi, M. Persico, Chem. Rev., 1994, 94, 2027-2094.

 ¹⁰ (a) M. Sumimoto, N. Iwane, T. Takahama, S. Sakaki, J. Am. Chem. Soc., 2004, **126**, 10457-10471. (b)
H. Tamura, H. Yamazaki, H. Sato, S. Sakaki, J. Am. Chem. Soc., 2003, **125**, 16114-16126.
¹¹ F. Bungel, P. A. Steire, Schwart effects in Chemistry, 2nd ed Wiley, Hebelen, 2016, pp. 135–136.

E. Buncel, R. A. Stairs, Solvent effects in Chemistry. 2nd ed. Wiley: Hoboken, 2016. pp.135-136.

It has been demonstrated that ΔG_{corr} is closer to the experimentally derived ΔG that the uncorrected calculated free energy.¹² All transition structures were characterized by one imaginary frequency. All the located TSs were confirmed to connect to reactants and products by intrinsic reaction coordinate (IRC) calculations.¹³ The IRC paths were traced using the Hratchian-Schlegel algorithm.¹⁴ NCI (non-covalent interactions) were computed using the methodology previously described.¹⁵ Data were obtained with the NCIPLOT program.¹⁶ A density cutoff of ρ =0.2 a.u. was applied and the pictures were created for an isosurface value of s=0.5 and colored in the [-0.3,0.3] a.u. sign(λ_2) ρ range using VMD software.¹⁷ Molecular graphics have been performed with CYLview 1.0 software.¹⁸

¹² C. T. Liu, C. I. Maxwell, D. R. Edwards, A. A. Neverov, N. J. Mosey, R. S. Brown, J. Am. Chem. Soc., 2010, **132**, 16599-16609.

¹³ (a) K. Fukui, J. Phys. Chem., 1970, 74, 4161-4163. (b) K. Fukui, Acc. Chem. Res., 1981, 14, 363-368.

¹⁴ H. P. Hratchian, H. B. Schlegel, J. Phys. Chem. A, 2002, **106**, 165-169.

¹⁵ (a) E. R. Johnson, S. Keinan, P. Mori-Sanchez, J. Contreras-Garcia, A. J. Cohen, W. Yang, J. Am. Chem. Soc., 2010, **132**, 6498-6506. (b) J. R. Lane, J. Contreras-Garcia, J.-P. Piquemal, B. J. Miller, H. G. Kjaergaard, J. Chem. Theory Comput., 2013, **9**, 3263-3266.

 ¹⁶ J. Contreras-Garcia, E. R. Johnson, S. Keinan, R. Chaudret, J.-P. Piquemal, D. N. Beratan, W. Yang, J. Chem. Theory Comput., 2011, 7, 625-632.

¹⁷ W. Humphrey, A. Dalke, K. Schulten, J. Mol. Graph., 1996, 14.

¹⁸ C. Y. Legault, Université de Sherbrooke 2009, http://www.cylview.org.

4.1. Preliminary Studies with racemic and achiral systems

A series of preliminary studies with achiral systems (using pyrrolidine as a catalyst) were carried out in order to determine the exact mechanism of the reaction between **1** and **2a**, and the required approaches to reach the real catalytic system (Scheme ESI-4).



Scheme ESI-4. Reaction between alcohol 1 and aldehyde 2, catalyzed by pyrrolidine.

Several different pathways are possible for the first step of the reaction (Michael addition) leading to intermediate enamine ENa which finally evolves to the final products P1 (Scheme ESI-5). The uncatalyzed reaction leads to enol MA-enol- This enol, after tautomerization can interact with the catalyst to afford the intermediate enamine EN. Path A involves participation of the catalyst by forming encounter complex EC-s1 but without formation of iminium salt, also leading to MA-enol through the corresponding transition structure TS-s2. A similar situation has been described by Lattanzi, Peluso and co-workers for the epoxidation of enones catalyzed by α, α -L-diarylprolinols.¹⁹ Finally, path B illustrates a different encounter complex EC-s2 in which formation of the hemiaminal is facilitated by the alcohol. The formation of iminium salt **IN-s2** also required the assistance of the alcohol and formation of enamine **ENa** takes place in two steps through TS-s4 and TS-s5a,b. A direct transition structure from IN-s1 to ENa was not found after exhaustive exploration of PES. It should have been taken into consideration that there are two possible transition structures **TS-s5** since enantiomeric alcohols can generate diastereomeric enamines ENa and ENA'. After formation of enamine EN the second step of the reaction consists of a second intramolecular Michael addition, through TS-s6a,b (also two diastereotopic attacks are possible), leading to IN-s3a,b. Further evolution of these intermediates furnish the final adducts P1a,b. Next, computational analyses of each path is described.

¹⁹ A. Capobianco, A. Russo, A. Lattanzi, A. Peluso, *Adv. Synth. Catal.*, 2012, **354**, 2789-2796.



Scheme ESI-5. Different pathways for the reaction between 1 and 2a.

4.2.1. Uncatalyzed reaction

The uncatalyzed reaction between **1** and **2a** was calculated (Scheme ESI-6). The reaction showed a typical concerted transfer of proton with a free energy barrier of 20.0 kcal/mol (Table ESI-7, Figure ESI-4).



Scheme ESI-6. Uncatalyzed reaction between alcohol 1 and aldehyde 2a.

Table ESI-7. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) absolute (hartrees) and relative (kcal/mol) energies of the stationary points corresponding to the reaction between alcohol 1 and aldehyde 2a.

-					
	E_0	ΔE_0^{a}	G	ΔG^{a}	im. freq
1	-419.671437		-419.668541		
2a	-422.804905		-422.803286		
TS-s1	-842.438973	23.4	-842.439994	20.0	-1022.8
MA-enol	-842.468764	4.8	-842.469374	1.5	
MA	-842.476680	-0.2	-842.478156	-4.0	
-					

^a Referred to isolated starting materials (1+2a)



Figure ESI-4. Uncatalyzed reaction between **1** and **2a**. Distances are given in ångstrøms. Relative energies are given in kcal/mol

4.2.2. Non-covalent activation of reactants (path A)

Alcohol **1** and aldehyde **2a** form encounter complex **EC-s1** with the catalyst, without barrier and located at 25.9 kcal/mol below the ground state in agreement with that suggested by Lattanzi, Peluso and co-workers for the epoxidation of enones catalyzed by diarylprolinols.¹⁹ That activated encounter complex leads to adduct **MA-enol-cat** through **TS-s2**, located 3 kcal/mol below the ground state. (Scheme ESI-7). This process does not involve the formation of an iminium intermediate but the non-covalent activation of the reactants. The corresponding transition structure showed an energy barrier of 20.0 kcal/mol (Table ESI-8, Figure ESI-5).



Scheme ESI-7. Catalyzed reaction between alcohol 1 and aldehyde 2a through non-covalent activation of the reactants.

Table ESI-8. Calculated (M06-2X/cc-pVTZ/PCM=CHCl ₃) absolute (hartrees) and relative
(kcal/mol) energies of the stationary points corresponding to the reaction between alcohol $\boldsymbol{1}$ and
aldehyde 2a through non-covalent activation of the reactants.

	E_0	ΔE_0^{a}	G	ΔG^{a}	im. freq
1	-419.671437		-419.668541		
2a	-422.804905		-422.803286		
Cat.	-212.427811		-212.425118		
EC-s1	-1054.929601	-16.0	-1054.938165	-25.9	
TS-s2	-1054.896396	4.9	-1054.901734	-3.0	-318.0
MA-enol-cat	-1054.916319	-7.6	-1054.921275	-15.3	
MA-cat	-1054.919012	-9.3	-1054.929534	-20.4	

^a Referred to isolated starting materials (1+2a). A molecule of catalyst has been added when necessary.



Figure ESI-5. Catalyzed reaction between alcohol 1 and aldehyde 2a through non-covalent activation of the reactants. Distances are given in ångstrøms. Relative energies are given in kcal/mol.

Given the importance of non-covalent interactions in this approach we also carried out a NCI analysis of **EC-s1** and **TS-s2** (Figure ESI-6). The topological analysis clearly shows H-bond interactions (blue discs) between aldehyde, alcohol and catalyst. Interestingly, important London interactions (π , π -stacking) between the unsaturated systems of alcohol and aldehyde are also evidenced (green surfaces). These interactions are observed in encounter complex **EC-s1** and they are maintained in **TS-s2**. In the transition state the forming bond is also evidenced (blue-red disc).



Figure ESI-6. NCI analysis of TS-s2.

4.2.3. Hemiaminal route (path B)

Alcohol 1 and aldehyde 2a form encounter complex EC-s2 with the catalyst, without barrier and located at 22.3 kcal/mol below the ground state. That activated encounter complex leads to hemiaminal in complex with 1 IN-s1 through TS-s3, located 18.5 kcal/mol below the ground state. (Scheme ESI-8). Complex IN-s1 further evolves to the intermediate iminium salt IN-s2 via TS-s4 (10.3 kcal/mol below the ground state). Thus, alcohol can act as an co-catalyst²⁰ to promote formation of the iminium salt by i) forming the initial hemiaminal²¹ and ii) facilitating elimination of water. In fact, such elimination of water is crucial in the following rate-limiting step in which the Michael addition takes place through TS-s5 to give enamine ENa.



Scheme ESI-8. Catalyzed reaction between alcohol 1 and aldehyde 2a via formation of iminium salt.

The addition of two enantiomeric alcohols (R)-1 and (S)-1 by one face of the iminium ion (the *trans*-configuration is the most stable one) gives rise to two diastereomers. Consequently, this step is determinant for the diastereoselectivity of the reaction involving a kinetic resolution of the two alcohols. Since both alcohols can racemize (see below) the process is, actually a DKR (Scheme ESI-9).

²¹ We have also calculated alternative transition structures for initial formation of hemiaminal in IN-s1 promoted by a molecule of water (TS-s3w) or by a second molecule of catalyst (pyrrolidine) (TS-s3p). However, these transition structures are either less favored or further decoordination of water/pyrrolidine and recoordination of alcohol to furnish coomplex IN-s1 requires more energy.



²⁰ It has been reported and computationally studied that protic co-catalysts can lower the barriers significantly. See: M. P. Patil, R. B. Sunoj, *Chem. Asian J.*, 2009, **4**, 714-724. In our case, compound **1** can also assume the role of a co-catalyst.



Scheme ESI-9. DKR in the reaction between alcohols (*R*)-1 and (*S*)-1and iminium salt derived from 2a.

After exhaustive exploration of the PES we determined the most stable transition structures for each approach, resulting in a difference of 0.9 kcal/mol in favor of **TS-s5b-(S)** (Figure ESI-7).²² Optimized geometries of both transition structures are given in Figure S4. Since we are studying the reactivity in achiral series, we will continue the study with the most stable **TS-s5b-(S)**.²³



Figure ESI-7. Optimized geometries (M06-2X/cc-pVTZ/PCM=CHCl₃) of transition structures corresponding to the DKR in the formation of intermediate enamine.

²² Although the difference (below 1.0 kcal/mol) is in the limit of error for the used level of theory we can predict a d.r. of ca. 4:1 in favor of **ENa**-(R,R). During PES exploration we also located up to 3 different transition structures per approach. A Bolztmann's distribution of the totality of transition structures afforded the same d.r value

²³ At this point, it has no sense to evaluate any prediction concerning absolute configurations since the whole system, including chiral catalyst, is required. The achiral study serves to determine the preferred path, relative diastereoselectivity (although it might change) and evaluate the catalytic cycle. Sereoselectivity (both diastereo and eantio) should also be studied with the full chiral model representing the real system studied experimentally.

A diagram energy for the formation of intermediate enamine **ENa-**(*R*,*R*), including paths A and B for the purpose of comparison, is given in Figure ESI-8.



Figure ESI-8. Energy diagram of the reactions between alcohol 1 and aldehyde 2a to form intermediate enamine. Red: uncatalyzed reaction; magenta: via non-covalent activation of reactants; black: via iminium ion. Relative energies are given in kcal/mol.

Actually, **ENa**-(R,R) is a complex between enamine and water that after re-coordination of the water moelecule leads to the productive complex **ENb** (Scheme ESI-10, Table ESI-9). This complex is activated towards the intramolecular Michael addition which can take place by two diastereotopic faces of the α , β -unsaturated system. Thus, two channels are possible, through **TS-s6a** -leading to **P1**- being the preferred one. Final products **P1** and **P2** are produced from intermediate iminium salts **IN-s3a,b** through hydrolysis via intermediate tautomeric hemiaminals **IN-s4a,b** and **IN-s5a,b** (Scheme ESI-10). The rate limiting for the whole process is the oxa-Michael reaction to the iminium salt **IN-s2** (Scheme ESI-9). The energy barrier corresponding to such rate-limiting **TS-s5** is 8.5 kcal/mol below the ground state, the lowest one of the above studied paths so, it is confirmed that path C is the preferred one. Optimized geometries for all stationary points are given in Table ESI-9. A energy diagram for this part of the reaction is given in Figure ESI-10.


Scheme ESI-10. Second step of the reaction between 1 and 2a catalyzed by pyrrolidine.



Figure ESI-9. Optimized geometries (M06-2X/cc-pVTZ/PCM=CHCl₃) of transition structures corresponding to the second step of the reaction between 1 and 2a.

	E_{0}	ΔE_0^{a}	G	ΔG^{a}	im. freq
1	-419.671437		-419.668541		
2a	-422.804905		-422.803286		
Cat.	-212.427811		-212.425118		
EC-s2	-1054.923628	-12.2	-1054.932476	-22.3	
TS-s3	-1054.919995	-9.9	-1054.926400	-18.5	-792.4
IN-s1	-1054.928430	-15.2	-1054.935423	-24.1	
TS-s4	-1054.905854	-1.1	-1054.913418	-10.3	-569.8
IN-s2	-1054.912618	-5.3	-1054.920665	-14.9	
TS-s5- (<i>S</i>)	-1054.905280	-0.7	-1054.910526	-8.5	-247.6
ENa-(R,R)	-1054.928780	-15.5	-1054.934501	-23.6	
ENb	-1054.926833	-14.2	-1054.935265	-24.0	
TS-s6a	-1054.917528	-8.4	-1054.921280	-15.3	-400.3
IN-s3a	-1054.924498	-12.8	-1054.928802	-20.0	
IN-s4a	-1054.948838	-28.0	-1054.952177	-34.7	
IN-s5a	-1054.962341	-36.5	-1054.965070	-42.7	
TS-s6b	-1054.913834	-6.1	-1054.918550	-13.6	-386.1
IN-s3b	-1054.920223	-10.1	-1054.925904	-18.2	
IN-s4b	-1054.953299	-30.8	-1054.956664	-37.5	
IN-s5b	-1054.962955	-36.9	-1054.966801	-43.8	
P1a	-842.508892	-20.4	-842.509948	-23.9	
P1b	-842.510111	-21.2	-842.511516	-24.9	

Table ESI-9. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) absolute (hartrees) and relative (kcal/mol) energies of the stationary points corresponding to the reaction between alcohol **1** and aldehyde **2a** via formation of iminium salt.

^a Referred to isolated starting materials (1+2a+cat). A molecule of catalyst has been added when necessary.



Figure ESI-10. Energy diagram of the second step of the reaction between alcohol 1 and aldehyde 2a. Relative energies are given in kcal/mol.

From the study of the achiral system it can be concluded that the preferred diastereoselectivity will consist on a relative 2R*3R*3aS*7aR* configuration, in other words: i) a *trans*-disposition between phenyl and aldehyde groups located at 2- and 3-position, respectively, ii) a *cis* fusion between the six-and five-membered cycles (3a- and 7a-positions), iii) a *cis* disposition between the aldehyde at C-3 and the contiguous bridgehead atom C-3a and iv) a *trans* disposition between the phenyl group at C-3 and the C-7a bridgehead atom. (Figure ESI-11)



Figure ESI-11. Numbering of final products. Note that absolute configuration can change depending on R

4.3. Real System (path C, chiral catalyst)

We studied two different orientations of the unsaturated chain leading to attacks by different stereofaces of the double bond,²⁴ the attack of the two enantiomeric alcohols to evaluate DKR and North and South conformations of the pyrrolidine ring²⁵ Thus, a total of 8 transition structures were located and characterized. Those corresponding to a North conformation of the pyrrolidine ring are always less stable than the corresponding South conformers. Consequently, it remain 4 different approaches to be considered (Figure ESI-12).

The absolute and relative energies as well as the relative abundance of these approaches is collected in Table ESI-10 (South conformations have also been included for comparison). According to the values of free energies it is predicted that essentially only 4 and 5 will be obtained in a ca. 91:9 diastereomeric ratio in qualitative agreement with experimental results since it is necessary to take into account that compound 4 epimerizes to compound 6 (see below). The optimized geometries for the located transition structures corresponding to South conformations of the pyrrolidine ring are given in Figure ESI-13.



Figure ESI-12. Approaches studied for the real system

²⁴ Irrelevant in the achiral analysis but necessary with the chiral catalyst

²⁵ The pyrrolidine ring can adopt two different and almost equivalent conformations, i.e. ${}^{3}T_{2}$ (called North or N) and ${}^{2}T_{3}$ (called South or S). The nomenclature is given by analogy with that used in conformational studies of furanoses. See: H. A. Taha, M. R. Richards, T. L. Lowary, *Chem. Rev.*, 2012, **102**, 1851-1876.

aluellyue 2a vi	allenyue 2a via formation of mininum sat.						
	E_0	$\Delta \Delta E_0^{b}$	G	$\Delta\Delta G^{b}$	im. freq	rel. ab. % ^c	
TS-s5-r1-S	-2040.596414	0.3	-2039.926213	1.6	-225.6	6.6	
TS-s5-r2-S	-2040.595722	0.0	-2039.928691	0.0	-250.3	90.7	
TS-s5-r3-S	-2040.594003	2.4	-2039.921494	4.5	-245.0	0.0	
TS-s5-r4-S	-2040.591843	3.5	-2039.921176	4.7	-235.5	0.0	
TS-s5-r1-N	-2039.909511	1.8	-2039.925163	2.2	-230.8	2.2	
TS-s5-r2-N	-2039.909708	1.6	-2039.922972	3.6	-242.1	0.2	
TS-s5-r3-N	-2039.909127	2.0	-2039.922934	3.6	-255.5	0.2	
TS-s5-r4-N	-2039.906767	3.5	-2039.921477	4.5	-220.2	0.0	

Table ESI-10. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) absolute (hartrees) and relative (kcal/mol) energies of the stationary points corresponding to the reaction between alcohol **1** and aldehyde **2a** via formation of iminium salt.^a

^a**N** and **S** series refer to North and South conformations of the pyrrolidine ring, respectively. ^b Referred to the most stable **TS-s5-r2-S**. ^c According to Boltzmann's distribution: a ratio 5:4:NO1:NO2 of 8.8:90.9:0.2:0 is predicted.



TS5-r2

Figure ESI-13. Optimized geometries (M06-2X/cc-pVTZ/PCM=CHCl₃) of transition structures corresponding to the rate-limitng step of the reaction between **1** and **2a** catalyzed by **3a**.

Regarding the orientation of the attack, if only the attack of the oxygen atom is exclusively considered, both models present an almost identical geometrical disposition, showing trajectory angles of 104.1° for **TS5-r1** and 104.9° for **TS5-r2**, which is also in good agreement with the Bürgi-Dunitz trajectory.46 Under this situation, (*R*)-1 (**TS5-r2**), orienting the pyrane ring outside, should present less unfavorable steric interactions with the pyrrolidine ring than (*S*)-1 in (**TS5-r1**), Practically the same disposition is appreciated for both TS, with rather similar

weak electrostatic interactions, H-bond and forming bond. However, whereas a weak repulsive steric interaction appears in the former, the latter presents an additional attractive London interaction between the methylene group of the heterocyclic ring and the aromatic ring. Thus, calculations predict a kinetic resolution in favor of the attack of (R)-1 to give compound 4 (which might epimerize, see below) as the major isomer.

4.3.1. Racemization of compound 1.

After elimination of DKR intermediate II and DYKAT intermediate III by means of experiments with deuterated compounds (see main text), two processes remain possible for the racemization of hydroxypyranone 1: i) racemization through intermediate open-chain aldehyde IN-r1 (Scheme ESI-11, right) and ii) formation of oxonium cation IN-r2 which could be estabilized by enolization to give aromatic species IN-r3 (Scheme ESI-11, left). In the case of ionic mechanism enolization could also take place over the initial complex C-r2 to give C-r3 and then aromatic IN-r3.



Scheme ESI-11. Racemization of hydroxypyranone 1 through neutral and ionic mechanisms

Essentially, the difference between the two mechanisms starts with the initial formation of different complexes C-r2 and C-r3 between compound 1 and the promoter of racemization, initially modeled in our case by formic acid.

Actually, racemization of compounds **1** can be considered as the typical process of mutarotation observed in carbohydrates. The process has been extensively studied experimental and computationally, and all those studies concluded that the process take place through the open-

chain aldehyde through a neutral mechanism.²⁶ Moreover, these studies were conducted in water as a solvent the most favorable scenario for a ionic mechanism. In any case, the use of chloroform as a solvent and the possibility of aromatization of intermediate **IN-r3** prompted us to calculate both mechanisms.

The energy data of the stationary points are given in Table ESI-11. The results are summarized in the energy diagram illustrated in Scheme ESI-12. Only the formation of intermediates has been studied since further transformation into *ent*-1 corresponds to an enantiomeric situation.

	-				
	E_0	ΔE_0^{a}	G	ΔG^{a}	im. freq
1+HCOOH	-609.411294	0.0	-609.452317	0.0	
C-r1	-609.423358	-7.6	-609.461968	-6.1	
TS-r1	-609.396487	9.3	-609.433282	11.9	-671.1
IN-r1	-609.407469	2.4	-609.446754	3.5	
C-r2	-609.418010	-4.2	-609.458572	-3.9	
TS-r2	-609.382349	18.2	-609.418378	21.3	-1489.4
C-r3	-609.405120	3.9	-609.444095	5.2	
C-r4	-609.405303	3.8	-609.445886	4.0	
IN-r2	-609.379616	19.9	-609.419653	20.5	
C-r5	-609.419103	-4.9	-609.459477	-4.5	
IN-r3	-609.342066	43.4	-609.384111	42.8	
IN-r4	-609.339173	45.3	-609.380686	44.9	

Table ESI-11. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) absolute (hartrees) and relative (kcal/mol) energies of the stationary points corresponding to the racemization of 1.

^a Referred to $\mathbf{1} + \text{HCO}_2\text{H}$

According to this data the preferred path corresponds to the formation of intermediate aldehyde **IN-r1** as in the case of carbohydrates. The barrier for this process is that of the rate-limiting step, i.e. 11.9 kcal/mol. The formation of aromatic intermediate **IN-r2** can be discarded since the corresponding rate-limiting step is 9.4 kcal/mol above. The possibility of formation of ionic species like **IN-r4** or **IN-r3** is negligible due to the high values of energy of such stationary points.

²⁶ (a) A. M. Silva, E. C. da Silva, C. O. da Silva, *Carbohydr. Res.*, 2006, **341**, 1029-1040 and references cited therein. (b) S.; Morpugo, A. Grandi, C. Zazza, M. Bossa, *J. Mol. Struct. TEOCHEM*, 2005, **79**, 71-82. (c) X. J. Qian, *Phys. Chem. B*, 2013, **117**, 11460-11465. For a definitive computational demonstration by using molecular simulations based on the combination of DFT methodology with the molecular dynamics formalism see: (d) W.; Plazinski, A. Plazinska, M. Drach, *Phys. Chem. Chem. Phys.*, 2015, **17**, 21622-21629.



Scheme ESI-12. Energy diagram (M06-2X/cc-pVTZ/PCM= CHCl₃) for racemization of **1**. Relative energies are given in kcal/mol

The geometries of representative points of the preferred route are given in Figure ESI-14.



Figure ESI-14 Optimized (M06-2X/cc-pVTZ/PCM=CHCl₃) geometries of stationary points corresponding to the racemization of alcohol **1** catalyzed by formic acid. Distances are given in angstroms.

Since the formation of ionic intermediates has been discarded even in the presence of an acid, the most favorable situation, and considering that the process takes place in the presence of water and mediated by thiourea **7**, we also calculate the corresponding transition structures (Figures ESI-15 and ESI-16). In the case of water, two molecules were considered on the basis of previous accurate studies carried out with D-glucose.^{10d} Barriers of 21.1 and 28.0 kcal/mol were found for water- and thiourea-mediated processes, respectively (Table ESI-12).



Figure ESI-15. Optimized (M06-2X/cc-pVTZ/PCM=CHCl₃) geometries of stationary points corresponding to the racemization of alcohol **1** catalyzed by water. Distances are given in angstroms.

Table ESI-12. Calculated (M06-2X/cc-pVTZ/PCM=CHCl ₃) absolute (hartrees) and relative
(kcal/mol) energies of the stationary points corresponding to the racemization of ${\bf 1}$ mediated by
water and thiourea.

	E_0	ΔE_0^{a}	G	ΔG^{a}	im. freq
C-r1w	-572.321248	0.0	-572.360029	0.0	
TS-r1w	-572.290079	23.5	-572.326396	21.1	-587.5
IN-rw1	-572.305199	11.4	-572.345296	9.2	
C-r1t	-2777.316000	0.0	-2777.391839	0.0	
TS-r1t	-2777.272036	27.6	-2777.347265	28.0	-391.9
IN-rwt	-2777.280878	22.0	-2777.358396	21.0	

^a Referred to the corresp;ponding complex C-r1w or C-r1t.



Figure ESI-16. Optimized (M06-2X/cc-pVTZ/PCM=CHCl₃) geometries of stationary points corresponding to the racemization of alcohol 1 catalyzed by thiourea 7. Distances are given in angstroms.

4.3.2. Epimerization of compounds 4.

Conformational analysis

In order to gain insight into the available conformational space we carried out a exhaustive conformational analysis of **4a,b** and **6a,b** using Macromodel (v. 11.1) software.²⁷ A total of 5000 conformations were generated and minimized using the OPLS_2005 force field²⁸ with chloroform solvation. Multiple searches were carried out using different starting geometries as well as different ring-closure bond choices. After minimization, a conformer was only saved if its steric energy was within 7 kcal/mol of the instant global minimum (lowest energy structure known during an incomplete conformational search) and did not duplicate a previously stored conformer. After completion of the search, the output files from the conformational searches were then analyzed using by XCluster program in the MacroModel. Finally, structures with energy over 5.0 kcal/mol higher than the current global minimum were optimized at DFT level (M062X/cc-pVDZ/PCM=chloroform). For both aryl and alkyl substituents compounds **6** resulted to be more stable by ca. 3-5 kcal/mol). Consequently, it can be concluded that *epimerization of* **4** *into* **6** *is thermodynamically favored independently of the nature of the substituent*.

²⁷ Schrödinger Release 2016-1: MacroModel, version 11.1, Schrödinger, LLC, New York, NY, 2016. ²⁸ (a) W. L. Jorganson, D. S. Maxwell, J. Tirado Piyos, *LAm. Cham. Soc.*, 1996, **118**, 11225, 11236.

 ⁸ (a) W. L. Jorgensen, D. S. Maxwell, J. Tirado-Rives, J. Am. Chem. Soc., 1996, 118, 11225-11236. (b)
W. L. Jorgensen, J. Tirado-Rives, J. Am. Chem. Soc., 1988, 110, 1657-1666.

Kinetic analysis

Epimerization of compounds **4** in acidic media should take place through the formation of the intermediate enol. Enol **EN-e1** has two possible configurations, i.e. E and Z; consequently, two different transition structures **TS-e1** are possible for the transformation of compounds **4** into **EN-e1**. Similarly, two transition structures **TS-e2** are also possible for the transformation of **EN-e1** into compounds **6**. We have studied all the possible paths for the transformation using formic acid as model acid and assuming the formation of initial complexes between **4a/6a** and formic acid²⁹ (Scheme ESI-13).



Scheme ESI-13. Transformation of compounds 4a,b into 6a,b.

The study showed that the preferred pathway was in both **a** and **b** series that involving the formation of (*E*)-enolates **EN-e1a-E** and **EN-e1b-E**. Absolute and relative energies are collected in Table ESI-13. The geometries of the transition structures are given in Figure S16. The rate limiting step was the transformation of the intermediate enolate into the most stable compound **6**. When compared the energy barriers of **a** and **b** series, the former (22.9 kcal/mol) - corresponding to R = Ph- was found to be 1.5 kcal/mol lower than that corresponding to epimerization of **4b** (24.4 kcal/mol). (Figure ESI-17 and Figure ESI-18).

²⁹ By considering transformations between unique entities it can be assumed the processes as unimolecular and estimation of free energy is correct (introduction of error due to translational entropy is avoided)

R		E ₀	ΔE_0^{a}	G	ΔG^{a}	im. freq
	4 -	-	0.0	1022 207020	0.0	
	48	1052.230997	0.0	-1052.507029	0.0	1508 3
	TS-e1a-E	1032.224107	20.6	-1032.271673	22.2	-1508.5
		-				-1510.3
	TS-e1a-Z	1032.220821	22.7	-1032.267457	24.8	
DI	EN-e1a-E	- 1032.248376	5.4	-1032.298601	5.3	
Ph		-				
	EN-e1a-Z	1032.246400	6.6	-1032.295595	7.2	
		-				-1510.7
	TS-e2a-E	1032.223282	21.2	-1032.270602	22.9	1415 0
	TS-020-7	- 1032 222453	21.7	-1032 270032	23.2	-1415.8
	15- C 2a-L	-	21.7	-1052.270052	23.2	
	6a	1032.259206	-1.4	-1032.310468	-2.2	
	4 b	-879.862352	0.0	-879.910067	0.0	
	TS-e1b-E	-879.829820	20.4	-879.873682	22.8	-1466.8
	TS-e1b-Z	-879.824474	23.8	-879.868891	25.8	-1636.2
Et	EN-e1b-E	-879.853790	5.4	-879.900298	6.1	
	EN-e1b-Z	-879.851488	6.8	-879.899670	6.5	
	TS-e2b-E	-879.827785	21.7	-879.871186	24.4	-1507.0
	TS-e2b-Z	-879.827965	21.6	-879.872235	23.7	-1455.1
	6b	-879.865394	-1.9	-879.912423	-1.5	

Table ESI-13. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) absolute (hartrees) and relative (kcal/mol) energies of the stationary points corresponding to the transformation of **4a,b** into **6a,b**.

^a Referred to **4a** or **4b**



Figure ESI-17. Energy diagram for the transformation of **4a,b** into **6a,b**. Blue: preferred path. Red: Rate-limiting step for the non-preferred path. Green: Rate-limiting step for the preferred step.



Figure ESI-18. Optimized (M06-2X/cc-pVTZ/PCM=CHCl₃) geometries of transition structures corresponding to enolization of **4a,b** and **6a,b**. Distances are given in angstroms

Since the different energy barriers found for the enolization processes of compounds **4** were in the range of 1-2 kcal/mol, within the error of DFT methods (estimated in ca. 3 kcal/mol),³⁰ we also studied acidity of **4a,b** and introduced $\mathbf{R} = CF_3$ (**4c**) to validate the model. In order to achieve reliable results it is necessary to consider an isodesmic process and to estimate solvation of the species. Thus, we considered the thermodynamic cycle indicated in Scheme ESI-14 corresponding to the deprotonation of compounds **4a-c** by formate anion. The results are collected in Table ESI-14.

The values collected in Table S7 clearly confirm the tendency previously observed (**4a** more easily enolizable than **4b**) and point out that *the acidity of the enolizable proton strongly depends on the R* substituent at β position, predicting that electron-withdrawing substituents like CF₃ should epimerize more easily.



Scheme ESI-14. Thermodynamic cycle for studying deprotonation of compounds 4a,b. For enolate EL the more stable *E*/Z-isomer has been chosen. Calculation level: M06-2X/cc-pVTZ (gas phase and PCM=chloroform).

Table SI.14. Calculated (M06-2X/cc-pVTZ/PCM=CHCl₃) free energies (kcal/mol) and pK_a values for the deprotonation of compounds **4a-c**.

	$\Delta G(g)$	$\Delta\Delta G$ (solv)	$\Delta G(r)$	pK_a
4 a	-2.9	15.3	12.4	9.1
4b	-1.1	14.8	13.7	10.1
4 c	-11.3	17.4	6.1	4.5

³⁰ S. N. Pieniazek, F. R. Clemente, K. N. Houk, Angew. Chem. Int. Ed., 2008, 47, 7746-7749.

4.4. Cartesian Coordinates

4.4.1. Achiral System

1

01

0	-2.0311173810 -0.0925260925 0.7783094638
Η	-2.6951863946 -0.7784669319 0.6510643258
С	0.5530463167 -1.2251123381 0.0339010089
С	-0.5260368841 1.3432602192 -0.2830626900
Η	-1.0580599762 2.2842677839 -0.3419137688
С	0.7855108030 1.2938676655 -0.0765869683
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С	3.9546935772 -2.3410580254 -0.9954339807
Н	4.2949738560 -2.1029447325 -1.9993571044
Н	2.0221385580 -1.4718301526 -1.5198427073
С	4.4495880085 -1.3529480530 0.0632072379
Н	4.3300359049 -0.3280129668 -0.2911333518
Н	5.4879053811 -1.5087357769 0.3429209161
С	3.4932023899 -1.6044027324 1.2213871278
Н	3.3619448406 -0.7569367628 1.8895671458
Н	3.7831085796 -2.4918383986 1.7870691947
Н	4.2890268165 -3.3490737615 -0.7495211190
0	1.6869534644 1.0973560493 1.8717210329
Н	2.0247512588 1.9943672839 1.9229078256
Ν	2.2209930787 -1.8585666699 0.5269818158
С	1.0571816689 -1.5746558069 1.0383052954
С	-0.1631616609 -1.5988125749 0.3557074833
С	-1.2454943272 -1.0479383612 0.9919346257
Н	-1.1713237522 -0.8737031427 2.0561117705
Н	-0.1986064449 -1.8782087576 -0.6869947876
0	-0.8021615365 1.1589529459 1.0747240492
Н	0.7327051470 1.1867552155 1.5680430191
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С	-4.9368314037 -0.5086319788 0.7999912951
Η	-3.4483673200 -0.5002988175 2.3441010514
С	-4.1265416784 -0.9333224570 -1.4258871557

Η	-2.0192142900 -1.2345372997 -1.6212958768
С	-5.1786743252 -0.6612260234 -0.5594347661
Η	-5.7514728179 -0.2870767793 1.4756375010
Η	-4.3119474078 -1.0354841457 -2.4862947641
Η	-6.1837440813 -0.5588334962 -0.9458096610
Η	1.0732040231 -1.2282310561 2.0650851067
Η	1.9200946573 -3.1859809179 -1.0860504998
С	1.3842473055 2.5809912356 -1.1852159562
С	0.8042644326 1.1897388072 -1.3336399640
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0	2.5253678072 2.8156094330 -1.5351772023
0	-0.5961528371 1.1423696673 -1.2620429400
Η	1.1077897309 0.7941657319 -2.3031774480
Η	1.2723375338 0.5876063274 -0.5441827006
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Η	0.9214942706 4.5825418144 -0.4417224247
С	-1.1316439373 1.7713373410 -0.0726062488
Η	-1.2552023078 3.9021868982 0.5430316678
Н	-2.2142240938 1.7527146797 -0.2772268181

TS-s5b-S

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Н	4.7508300306 -2.0558372203 -2.4977785918
Н	2.6171996709 -1.1500276602 -1.8081945791
С	4.9988716455 -1.8568736259 -0.3205538449
Н	5.0845217700 -0.7750161120 -0.4235089960
Н	5.9831305758 -2.2649405535 -0.1083691011
С	3.9892839642 -2.1946426934 0.7683525512
Н	3.9981369385 -1.5205237267 1.6212038053
Н	4.1155547477 -3.2207337780 1.1176586322
Н	4.4933543125 -3.5396818098 -1.5777231715
0	1.4354155368 0.1289138821 2.8721712229
Н	1.4732387603 0.6865145413 3.6515772010
Ν	2.7011412803 -2.0775143080 0.0648526025
С	1.5694848656 -1.8435348456 0.6645474874
С	0.3397796675 -1.6348097204 0.0259818853
С	-0.7226540216 -1.2596114339 0.8033038837
Н	-0.5917630352 -1.2858935836 1.8769911764
Н	0.2771926391 -1.6503092283 -1.0519158520
Н	0.7038311278 0.5155683458 2.2966908894
С	-2.0884152865 -1.0519034779 0.3340281371
С	-3.0569120175 -0.6568787303 1.2582505022
С	-2.4585632173 -1.2248680647 -1.0002691512
С	-4.3649495348 -0.4356373277 0.8612591782

Н	-2.7630085740 -0.4978248841 2.2877056013
С	-3.7666206840 -1.0065122493 -1.3986060487
Η	-1.7218253919 -1.5202648154 -1.7338500488
С	-4.7222150505 -0.6101537405 -0.4700487180
Н	-5.1046844513 -0.1234836518 1.5855567793
Н	-4.0426589238 -1.1406945596 -2.4356386381
Η	-5.7422385776 -0.4355598971 -0.7846237663
Η	1.6145985112 -1.7561181849 1.7449989210
Η	2.2045947116 -2.8724268719 -1.8255683510
0	-0.3946864436 0.9832509496 1.2491082626
С	-1.3444543129 3.5046022762 -0.9672104657
С	-1.6219655958 2.0227347197 -1.1172094927
С	-0.1450987107 3.8570632778 -0.1935816259
0	-2.1118470743 4.3383416757 -1.4070844577
0	-0.4486120175 1.2498935248 -1.0654610695
Η	-2.1071340581 1.8461983690 -2.0759405821
Η	-2.3201818851 1.7371770526 -0.3184981688
С	0.5975528809 2.8923715378 0.3497579823
Η	0.0478486863 4.9109217897 -0.0366556782
С	0.2701530065 1.4250403689 0.1723322918
Η	1.4298093451 3.1241603351 1.0055179403
Η	1.2064170409 0.8689468397 -0.0290722643

TS-s6a

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С	-3.4848600000 -2.4913750000 0.0291370000
Н	-3.9628660000 -2.7291840000 0.9751940000
Н	-1.9355950000 -1.5086940000 1.1865810000
С	-4.1432880000 -1.3009560000 -0.6700060000
Н	-4.3733260000 -0.5266850000 0.0592690000
Н	-5.0524960000 -1.5681060000 -1.2024080000
С	-3.0576850000 -0.8047730000 -1.6137210000
Н	-3.1242590000 0.2636860000 -1.8158060000
Н	-3.0532760000 -1.3545740000 -2.5565430000
Н	-3.5085760000 -3.3719450000 -0.6136540000
Ν	-1.8155960000 -1.0883460000 -0.8743390000
С	-0.6413060000 -0.6589730000 -1.2575890000
С	0.5311550000 -0.7270600000 -0.5085200000
С	1.7504110000 -0.0968330000 -1.1369900000
Н	1.8817810000 -0.4266400000 -2.1702350000
Н	0.6471680000 -1.5259940000 0.2100130000
0	1.4778800000 1.2971780000 -1.2450710000
С	3.0318660000 -0.3914870000 -0.3800240000
С	3.9303720000 0.6238620000 -0.0741350000
С	3.3459840000 -1.7007450000 -0.0218830000

С	5.1126250000 0.3395340000 0.5978430000
Η	3.7067980000 1.6386960000 -0.3732500000
С	4.5292600000 -1.9866060000 0.6422750000
Η	2.6667170000 -2.5068210000 -0.2707790000
С	5.4144420000 -0.9646110000 0.9606270000
Η	5.7995480000 1.1410100000 0.8340840000
Η	4.7600320000 -3.0083720000 0.9122530000
Η	6.3347230000 -1.1855680000 1.4837670000
Η	-0.6412610000 -0.0628700000 -2.1652250000
Η	-1.3154810000 -2.8377050000 0.1797710000
С	$-1.9157660000 \ 1.6932950000 \ 0.5408300000$
С	-1.3176110000 2.4590900000 -0.6263070000
С	-1.0015280000 0.9534930000 1.3064080000
0	-3.1514300000 1.7703710000 0.7182760000
0	0.0264890000 2.8529910000 -0.4064660000
Η	-1.8902720000 3.3717950000 -0.7722770000
Η	-1.4053030000 1.8637370000 -1.5441710000
С	0.2943450000 0.7615690000 0.8161780000
Η	-1.3510750000 0.4179280000 2.1770280000
С	0.9137870000 1.8408500000 -0.0631830000
Η	1.0130040000 0.3118310000 1.4923470000
Η	1.7068670000 2.3335390000 0.5042810000
Η	-4.0055940000 0.0753920000 3.2864960000
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Η	-3.6464120000 0.5131910000 1.8518300000

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С	2.9318490111 -2.5016798654 -1.8742029710
Н	3.5948520246 -1.9974021065 -2.5715990577
Н	2.2465924460 -0.5446333251 -1.2041573169
С	3.6449474767 -3.0046512317 -0.6180868406
Н	4.3121286814 -2.2335504111 -0.2317495767
Н	4.2197370568 -3.9118900165 -0.7833545267
С	2.4914768705 -3.2230257299 0.3515976198
Н	2.7755001215 -3.1569727557 1.3991390562
Н	2.0049258631 -4.1845216148 0.1739674792
Н	2.4580762501 -3.3363769762 -2.3915134982
0	2.6125803764 4.6758054052 -1.9919881468
Н	2.8191638037 3.9686015807 -1.3465606268
Ν	1.5656014008 -2.1314011811 0.0091353314
С	0.5371029044 -1.7878651257 0.7505251792
С	-0.2421699234 -0.6585096078 0.5428632759
С	-1.5676258457 -0.4768191846 1.2276310899
Н	-1.6561217689 -1.1916159001 2.0512383446

Η	-0.1461170397 -0.1475799882 -0.4064090047
0	-1.6171330580 0.8447066913 1.7943031029
Η	2.5440101585 5.4722581159 -1.4615048867
С	-2.7302911064 -0.6717113957 0.2827492988
С	-3.1734148496 0.3745567163 -0.5191320918
С	-3.3289818823 -1.9212064060 0.1708435783
С	-4.2029725262 0.1670278523 -1.4262289768
Η	-2.7082170419 1.3465180256 -0.4206052682
С	-4.3578473054 -2.1290699559 -0.7372426070
Н	-2.9887463453 -2.7356045011 0.8001631047
С	-4.7969908923 -1.0835452229 -1.5385622131
Н	-4.5443754090 0.9853288160 -2.0460956754
Η	-4.8205030487 -3.1038463816 -0.8147899465
Н	-5.6017917001 -1.2410905162 -2.2437617082
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Н	0.9573767513 -1.5283115994 -1.9158794251
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С	0.7115067664 2.9732126017 0.1578270185
С	1.9735249269 1.0599864315 1.1662041320
0	3.0399809610 2.6739391816 -0.1747090958
0	-0.4632242439 2.2210180234 0.4089545181
Н	0.7510014777 3.8524459414 0.8154826415
Н	0.6550856656 3.3276302868 -0.8689405299
С	0.7571712739 0.6164420674 1.6974352631
Η	2.8993136911 0.5419804139 1.3770266375
С	-0.4331168859 1.5451593092 1.6435838399
Н	0.7813412148 0.0185729989 2.6027021455
Н	-0.3784559593 2.2713632052 2.4641735289

4.4.2. Chiral system

TS-s5-r1

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Н	0.4821130000 -0.3138390000 0.8164690000
С	2.8181560000 -1.2755270000 0.5529140000
Н	2.1550660000 -1.5468580000 -1.4802830000
Н	2.4573520000 -1.0881540000 1.5555120000
С	4.1572580000 -1.8530730000 0.4566930000
С	4.8958850000 -2.0336060000 1.6278400000
С	4.7275070000 -2.2202020000 -0.7632810000
С	6.1671590000 -2.5784030000 1.5857960000
Н	4.4627400000 -1.7314620000 2.5728030000

С	6.0040620000 -2.7587130000 -0.8070990000
Н	4.1782960000 -2.0864100000 -1.6850690000
С	6.7249650000 -2.9414720000 0.3657020000
Н	6.7262270000 -2.7157720000 2.5011610000
Н	6.4372070000 -3.0382530000 -1.7577840000
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C	3.511/220000 1.2751360000 0.53/9230000
0	2 282020000 1 0064250000 0 060506000
C C	2.2835500000 1.9004550000 -0.9055900000 4.6660010000 2.2220440000 0.6061500000
U U	4.0009910000 2.2530440000 -0.0901500000
H	3.6230100000 0.4730870000 -1.2891560000
C	2.0560290000 3.1/16610000 -0.3953610000
C	4.5220440000 3.5555580000 -0.7732960000
Н	5.6501430000 1.7756820000 -0.6976890000
С	3.1864190000 4.1482910000 -0.6376010000
Н	1.1431890000 3.5790030000 -0.8277710000
Н	1.9124600000 3.0975860000 0.6894520000
Н	5.3583480000 4.2369100000 -0.8647410000
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и П	1.0161500000 2.3614870000 2.0080550000
0	1.0500710000 1.0180060000 0.8220240000
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C	-5.0088010000 -1.1251980000 -1.5112950000
U U	-3.01323/0000 0.0/92930000 0.3509150000
H	-1.8284780000 -0.1601180000 -3.4992050000
H	-0.7327480000 1.1133780000 -4.0265480000
Si	-1.9844010000 -1.0171980000 2.4996070000
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С	-3.5305540000 2.0628100000 0.4543290000
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С	-2.0992240000 0.7029480000 3.2211790000
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С	-0.3499180000 -1.7920300000 2.9719390000
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Н	-4.5915110000 0.1793330000 -1.9596810000
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Н	0.4492660000 -1.0515110000 2.9068990000
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Н	-2.2001290000 -4.0487420000 -2.8251710000
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Н	-4.4693720000 3.8666400000 1.1244210000
Н	-6.6442050000 0.2476080000 1.8197470000
Н	-4.0402940000 -3.3396450000 -4.3205050000
Н	-6.4658790000 2.7113180000 2.0202790000

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Н	-2.0113980000 -0.5062710000	1.7169640000
Н	-1.8856660000 1.8100550000	-0.3065070000
Н	-1.9165030000 -2.2172380000	0.1218640000
С	-3.6959130000 0.8438620000	1.6012890000
С	-4.4139570000 0.1148000000	2.5501830000
С	-4.2400090000 2.0315620000	1.1091050000
С	-5.6486120000 0.5566390000	2.9967310000
Н	-4.0017180000 -0.8165830000	2.9162010000
С	-5.4727490000 2.4754450000	1.5555380000
Н	-3.7080250000 2.6058650000	0.3641090000
С	-6.1806120000 1.7391900000	2.4986590000
Н	-6.1982510000 -0.0210590000	3.7269320000
Н	-5.8869590000 3.3948920000	1.1647980000
Н	-7.1454890000 2.0864500000	2.8425410000
Н	-0.1095660000 -0.5973520000	0.5023510000
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С	-3.9023940000 -2.1251910000 -2.1496230000
Н	-5.6000630000 -2.9742800000 -3.1118260000
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Н	-2.3753510000 -0.6968840000 -1.6624960000
Ν	0.4829310000 0.6776770000 -0.9579140000
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Н	3.0909030000 0.3847700000 -2.9061510000
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C	4.0188710000 -0.7546500000 -0.4612770000
Н	2.1327020000 2.5508100000 -2.4897160000
Н	1.1314890000 2.0316620000 -3.8419620000
Si	2.3386380000 -1.3277510000 2.1867210000
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С	2.4830030000 2.5678620000 0.4744680000
С	4.0454340000 -1.7687090000 -1.4111590000
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C	2.6293830000 -3.0144320000 1.4348350000
С	3.6666310000 -0.9224250000 3.4419700000
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C	4.5666110000 3.4465840000 -1.1277320000
Н	4.8957610000 1.3839360000 -1.5791770000
C	2.7598880000 3.9245200000 0.3754480000
Н	1.6801290000 2.2238560000 1.1120290000
C	5.1371420000 -2.6262880000 -1.4999830000
Н	3 2191490000 -1 9205600000 -2 0908970000
C	6 2016500000 -1 4626560000 0 3058960000
Н	5 1138600000 0 1929130000 1 1164450000
Н	2.4408780000 -3 7707990000 2 2003540000
Н	3.6492140000 -3.1437020000 1.0716620000
н	1 9353790000 -3 1954480000 0 6113640000
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Η	3.6431520000 0.1378600000 3.6990970000
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Η	0.6946310000 -1.8345170000 3.9556150000
Η	-0.0812240000 -1.7735620000 2.3699370000
Η	0.3432840000 -0.2686270000 3.2158850000
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Η	5.3877460000 3.7802570000 -1.7476050000
Η	2.1647850000 4.6343330000 0.9342380000
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Η	-0.9816711382 -2.2332510379 -1.8588064875
С	-2.7368115312 -0.4779649859 -1.3245943750
Η	-1.1232763472 0.3373469016 -0.1529130857
Η	-2.9508888849 -1.2164415842 -2.0855934475
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С	-4.9619942352 0.4785879083 -1.7953629786
С	-3.3310022737 1.8822509426 -0.7254593413
С	-5.8795216354 1.5143620139 -1.7623327502
Η	-5.2368222704 -0.4870171101 -2.1987814355
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Н	-2.3406569099 2.0382080594 -0.3195653965
С	-5.5252349421 2.7383456834 -1.2055484432
Η	-6.8719190720 1.3699540173 -2.1670388606
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0	-3.4084157358 -2.5804466794 1.9724031527
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C	2.3005677590 3.6515527943 -1.4196873	658
C	-0.2570911800 2.4719217187 -2.4395771	938
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Н	1.8512727737 4.5399536814 -1.8686482	759
Н	2.9836678461 3.2198724933 -2.1538572	406
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Н	-1.2426450000 -0.7904680000 -2.4897460000
С	-2.6089230000 1.0030410000 -1.3571830000
Н	-0.8364720000 1.1676710000 -0.1327350000
Н	-2.9683190000 0.5910970000 -2.2913370000
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Η	-5.5958130000 5.0064670000 0.5207540000
Η	-3.4324970000 -1.5071470000 -1.7632310000
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6a

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Η	-4.9673253336 -0.1915874429 -1.5692661962
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Η	1.5034117914 -0.2822112339 -0.8045384509
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0	1.9107945280 -2.4606923299 0.4504152658
Η	2.0806268762 -3.3113814046 -1.3562377451
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С	-0.3540419388 -0.0023852489 0.5370336307
С	0.5202697518 1.9387321767 -0.5990309176
С	-1.2855350224 1.2528653858 0.6329531918
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Η	2.9022252028 3.8050272126 -1.4097897623
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4.4.4. Racemization of 4

C-r1

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Н	1.1745792717 2.8404455448 0.4327343290
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Н	3.1931881153 1.4863490847 0.3343192108
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Н	2.8444097453 0.4951600445 1.8799227581
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Η	-3.0210643700 1.4363928220 -0.3035792761
0	-0.3059714386 -1.0211695513 -1.0620598036
С	-2.0898048188 -0.3585136863 0.4651572009
0	-2.9790453833 -0.6554717986 1.2279182888
Н	-1.6787460850 -2.3069816942 -0.1877334496
Η	-0.4723822231 -1.6470315817 0.9084504185
С	0.3983338623 1.4079555622 0.1473957652
Η	0.5282362074 0.4840055863 0.7277657135
Η	2.6635213076 -1.0392691083 0.1424047755
Η	3.7030177741 0.6016479765 1.4814110586
Η	2.6442267925 -2.3890914808 -0.6241864719
IN-r2

11	
С	0.9455964090 -0.6280377647 0.0839376173
С	-1.0907604323 1.0362347623 -0.6010840520
Н	-1.9239302229 1.6612239227 -0.8807504004
С	0.1246151847 1.5778847235 -0.2107512929
Н	0.2561435099 2.6523838213 -0.1809271438
0	-0.2196077808 -1.0990426389 -0.2909564732
С	1.1759332173 0.7323666047 0.1428268846
0	2.3884314098 1.1200563171 0.5342805975
Н	1.6728056477 -1.3863144619 0.3305419472
С	-1.2285255664 -0.3241108566 -0.6296826050
Η	-2.1104426354 -0.8762968840 -0.9110487971
Н	2.4758044094 2.0806461046 0.5514808378

IN-r3

11	
С	0.6396582065 -1.2409244243 0.0412307060
С	-0.4568145953 1.3538043144 -0.4726893554
Н	-0.9761077022 2.2726305999 -0.6985601430
С	0.8127564194 1.3120925903 -0.0444489010
Н	1.4119576043 2.2005538866 0.1066046008
0	-0.7166417076 -1.0192786204 -0.4264226034
С	1.4598215241 0.0101041489 0.2458213995
0	2.5922954820 -0.0695395735 0.6300519708
Н	0.5397627306 -1.7910908822 0.9755462951
Н	1.1032495974 -1.8858780393 -0.7035556369
С	-1.1787109985 0.1285490885 -0.6460353893
Н	-2.2076465708 0.1233339111 -0.9910339333

TS-r1

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0	1.1281433455 1.6059537756 -0.3359656842
Н	2.2311170907 1.0190011620 -0.0056741230
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0	2.1271400102 -1.4617285086 -0.1346505137
С	3.1149259296 -0.7918323749 0.1818055542
0	3.1761574195 0.4783623338 0.2840614987
Н	4.0530649361 -1.3089473930 0.4090055268
С	-1.1412140449 -1.2662675643 -0.0039718347
С	-1.2021299206 1.6744227989 0.0664675323
Н	-1.1845245421 2.7567186134 0.0229114726
С	-2.3463369634 1.0029043190 -0.0120335185

Η	-3.2936629718 1.5141728699 -0.1196688923
0	-0.1262454975 -0.4834728785 -0.6172509535
С	-2.4288997451 -0.4701033647 0.0751815322
0	-3.4802718107 -1.0425014111 0.2350186976
Н	-1.3207414932 -2.1672191019 -0.5875897823
Н	-0.8497654716 -1.5553165299 1.0116318015
С	0.1458826518 1.0495124836 0.2344753694
Н	0.3148669471 0.5696389839 1.2098869358

TS-r1t

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0	-1.4057725770 4.5243149525 -0.1017601385
Η	-0.4705504681 3.7930550517 -0.0388895763
Н	-2.3679014909 1.9900822330 -0.6290729757
С	0.2073079819 1.2399934733 -0.1518254665
С	-4.1551657778 1.8453264713 0.1800860209
С	-3.5832625716 4.8292006162 0.7232455430
Н	-3.4515001947 5.9137803549 0.7128973642
С	-4.7994369761 4.2862714767 0.8512238410
Н	-5.6710301306 4.9303809943 0.9841794209
0	-3.2454311397 2.4932554926 -0.6765792249
С	-5.1231786114 2.8273094172 0.8239874333
0	-6.1604785989 2.4332187775 1.3076836363
Н	-4.7633079668 1.0964413055 -0.3539672466
Н	-3.6362869289 1.3162552002 1.0045494223
С	-2.3384778984 4.0535639139 0.5734686128
Н	-2.1372374182 3.1859944784 1.2231339451
Ν	-1.0849937807 1.0371470902 -0.2504333431
S	0.8663285006 2.8279197452 0.1428606667
С	-1.6220220838 -0.2630230730 -0.2779660693
С	-1.4024163142 -1.1870381941 0.7563364541
С	-2.5187635072 -0.6063910774 -1.2996189265
С	-2.0413374209 -2.4235635372 0.7411893169
Н	-0.7412690304 -0.9246285248 1.5852128318
С	-3.1754428531 -1.8330552715 -1.2744749997
Н	-2.7106563913 0.1084192243 -2.1024153096
С	-2.9389356305 -2.7640875609 -0.2667798539
Н	-3.4468225648 -3.7266222920 -0.2623339469
Н	0.6545463729 -0.7231084218 -0.3733767234
С	2.4795930275 0.1531093938 -0.0522011937
С	3.3598859208 1.0576569008 -0.6491305236
С	2.9850151197 -0.8853163305 0.7371409493
С	4.7273077049 0.9387555991 -0.4121253510
Н	2.9783995699 1.8448724360 -1.2965363397
С	4.3565974425 -1.0037531173 0.9313083671
Н	2.3001904130 -1.5986465368 1.1992882743

С	5.2430164889 -0.0857207488 0.3738880422	
Η	6.3151995816 -0.1731238310 0.5435530071	
Ν	1.0854951899 0.1913283228 -0.2749593236	
С	4.8982877070 -2.1574491935 1.7281188638	
С	5.6484129880 1.9569863958 -1.0253450975	
С	-4.1910920913 -2.1128167212 -2.3464807592	
С	-1.7323808747 -3.3869991688 1.8531821930	
F	4.0094070431 -2.6109463858 2.6219565342	
F	6.0104617644 -1.8204939578 2.3953324669	
F	5.2184146811 -3.1936826479 0.9367479646	
F	5.3236972593 2.2100565289 -2.3021938596	
F	6.9257555409 1.5550913632 -1.0060485914	
F	5.5923174167 3.1289013604 -0.3743182954	
F	-3.7077684300 -1.8535323545 -3.5701108293	
F	-4.6044923161 -3.3850650782 -2.3355326054	
F	-5.2797009077 -1.3386729475 -2.1925424530	
F	-2.4534591842 -4.5094386397 1.7688398557	
F	-1.9674043760 -2.8435062536 3.0568235274	
F	-0.4352932084 -3.7410167044 1.8426324683	

TS-r1w

01

0	1.3889971921 1.4262239282 -0.4024907107
Н	2.5872599742 0.8807740216 0.0727553045
Η	1.0439940822 -1.1433529213 -0.3664526781
0	2.1062599309 -1.7271439902 -0.1666179787
0	3.4132286895 0.2791761768 0.3765236281
С	-1.0438319326 -1.1812100845 0.1668587673
С	-0.9271523187 1.6625164754 0.0156609633
Η	-0.8058740727 2.7383612704 -0.0235765780
С	-2.1281460797 1.1005500216 -0.0963967901
Н	-3.0281276401 1.6833647432 -0.2417105758
0	0.0514408905 -0.5170646005 -0.4477003654
С	-2.3064169377 -0.3539405034 0.0357402775
0	-3.3923880494 -0.8830313851 0.0960865354
Η	-1.1923768751 -2.1447787152 -0.3152793646
Н	-0.8478310053 -1.3445858229 1.2333830606
С	0.3694251195 0.9122854326 0.1963519192
Н	0.5142522744 0.6093499867 1.2526497753
Н	2.7744228268 -0.9502329875 0.1113409886
Н	3.6169188416 0.4674430473 1.2955982854
Н	2.4379660893 -2.1488690933 -0.9658514638

TS-r2

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С	0.0625474231 -0.9067438122 0.0181288918
Н	1.9203350714 -1.0540537643 -0.5443793188
Н	1.0957288343 0.8346320902 0.3958376763
0	2.2603563779 1.1991767864 0.8327264027
С	3.1772144524 0.7153856423 0.1106276715
0	3.0666483818 -0.2453292900 -0.6756055343
Н	4.1588542312 1.1991096256 0.1899945547
С	0.0122958734 0.4582291145 -0.3045049180
Н	0.3976076651 0.7363911748 -1.2845545451
С	-1.0181214886 -1.5077699173 0.7791397019
Н	-0.8597141625 -2.4809762337 1.2211115254
С	-2.1793547767 -0.8591802899 0.8197018546
Н	-3.0538370911 -1.2782660765 1.2987721595
С	-2.3301589004 0.4701822635 0.1286640461
Н	-3.0090740945 1.1188504396 0.6824610361
0	-1.1213487652 1.1805440609 0.0933374124
0	-2.8291172258 0.1890039892 -1.1486979031
Η	-3.1241569517 1.0121794400 -1.5525197901







Figure ESI-20. NMR spectra of compound 4j.



Figure ESI-21. NMR spectra of compound 4k.



Figure ESI-22. NMR spectra of compound 4l.



Figure ESI-23. NMR spectra of compound 4m.







Figure ESI-25. NMR spectra of compound 40.



Figure ESI-26. NMR spectra of compound 4p.















Figure ESI-30. NMR spectra of compound 6d.











Figure ESI-33. NMR spectra of compound 6g.







Figure ESI-35. NMR spectra of compound 7i.







Figure ESI-37. NMR spectra of compound 7k.



















Figure ESI-42. NMR spectra of compound 7p.











Figure ESI-45. NMR spectra of compound 8c.



Figure ESI-46. NMR spectra of compound 8d.



Figure ESI-47. NMR spectra of compound 8e.



Figure ESI-48. NMR spectra of compound 8f.



Figure ESI-49. NMR spectra of compound 8g.



Figure ESI-50. NMR spectra of compound 8h.


Figure ESI-51. NMR spectra of compound 9a.



Figure ESI-52. NMR spectra of compound 10a.



Figure ESI-53. NMR spectra of compound 11a.



Figure ESI-54. NMR spectra of compound 6a-²H.



Figure ESI-55. NMR spectra of compound 8a-²H.



Figure ESI-56. NMR spectra of compound 12k.



Figure ESI-57. HPLC traces of compound 7i.



Figure ESI-58. HPLC traces of compound 7j.



Figure ESI-59. HPLC traces of compound 7k.



Figure ESI-60. HPLC traces of compound 71.



Figure ESI-61. HPLC traces of compound 7m.



Figure ESI-62. HPLC traces of compound 7n.



Figure ESI-63. HPLC traces of compound 70.



Figure ESI-64. HPLC traces of compound 7p.



Figure ESI-65. HPLC traces of compound 8a.



Figure ESI-66. HPLC traces of compound 8b.



Figure ESI-67. HPLC traces of compound 8c.





	Peak Results					
		RT	Area	Height	% Area	
	1	66,802	2713503	17758	9,25	
	2	90,286	173844	1075	0,59	
	3	187,351	1121331	3241	3,82	
	4	203,836	25321440	56376	86,33	

Figure ESI-68. HPLC traces of compound 8d.



Figure ESI-69. HPLC traces of compound 8e.



Figure ESI-70. HPLC traces of compound 8f.





Figure ESI-71. HPLC traces of compound 8g.



Figure ESI-72. HPLC traces of compound 8h.



Figure ESI-73. HPLC trace of compound 8a-²H.