

Electronic Supplementary Information

Manganese Complex-Catalyzed Oxidation and Oxidative Kinetic Resolution of Secondary Alcohols by Hydrogen Peroxide

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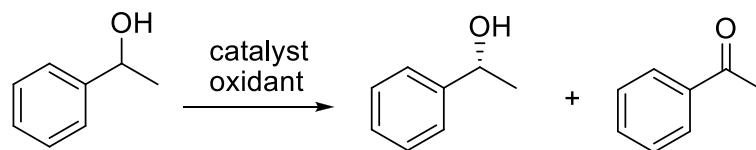
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Table S1 Oxidative kinetic resolution (OKR) of 1-phenylethanol catalyzed by manganese complex with N4 ligand^a

entry	catalyst	T (°C)	oxidant	conversion (%) ^b	ee (%) ^b
1	1	0	H ₂ O ₂	54	50
2	2	0	H ₂ O ₂	57	64
3	3	0	H ₂ O ₂	trace	-
4	2	25	H ₂ O ₂	55	58
5	2	-20	H ₂ O ₂	58	63
6	2	-40	H ₂ O ₂	51	52
8	2	0	70% TBHP	58	65
9	2	0	<i>m</i> -CPBA	50	55
10 ^c	2	0	oxone	20	16
11	2	0	PhI(OAc) ₂	58	70

^a Reaction conditions: An CH₃CN (1.0 or 1.5 mL) solution containing 1-phenylethanol (0.50 mmol), catalyst (0.20 mol%), and H₂SO₄ (1.0 mol%) was added into the Schleck tube at 0 °C. Then, an CH₃CN (0.50 mL) solution containing oxidant (0.60 equiv.) was added dropwise into the Schleck tube by using a syringe pump for 1 h (for H₂O₂ and TBHP) or added at once and stirred for 1 h (for *m*-CPBA, oxone, and PhIOAc). ^b The conversion yields of 1-phenylethanol and the ee values of unreacted 1-phenylethanol were determined by GC with a CP-Chirasil-Dex CB column. ^c 0.30 equiv. of oxone was added.

Table S2 Screening the amount of the acid and reaction time for oxidative kinetic resolution of 1-phenylethanol catalyzed by **2**^a



entry	H ₂ SO ₄ (mol%)	t (h)	conversion (%) ^b	ee (%) ^b
1	1.0	1	69	90
2 ^c	500	1	70	37
3 ^d	1.0	1	3	trace
4	1.0	0.50	61	70
5	1.0	2.0	74	92
6	0.25	1	52	57
7	0.50	1	65	84
8	3.0	1	66	79

^a Reaction conditions: An CH₃CN (1.0 mL) solution containing 1-phenylethanol (0.50 mmol), catalyst (0.20 mol%), and H₂SO₄ (0.25 – 3.0 mol%) was added into the Schleck tube at 0 °C. Then, an CH₃CN (0.50 mL) solution containing 30% H₂O₂ (0.80 equiv.) was added dropwise into the Schleck tube by using a syringe pump for 1 h (except for entries 4 and 5). ^b The conversion yields of 1-phenylethanol and the ee values of unreacted 1-phenylethanol were determined by GC with a CP-Chirasil-Dex CB column. ^c 5.0 equiv. (500 mol%) of AcOH was used instead of H₂SO₄. ^d 1.0 mol% of AcOH was used instead of H₂SO₄.

Table S3 Comparison of the reaction energy barriers for the α -CH activation of both *S*- and *R*-1-phenylethanol (*S*-Sub and *R*-Sub, respectively) by the high-valent $[(\text{P-MCP})\text{Mn}^{5+}(\text{O}^{2-})(\text{SO}_4^{2-})]^+$ species (**I**). Energies were calculated using the B3LYP functional at different levels (i.e., B1 represents a LACVP(Mn), 6-31G*(C, H, O, N, and S) basis set, and B2 represents a LACV3P(Mn), 6-311+G***(C, H, O, N, and S) basis set). Absolute energies are in a.u. unit and relative ones are in kcal/mol

	OPT/B1	ΔE	OPT+ZPE	ΔE	SPE/B2	ΔE	SPE+ZPE	ΔE	ZPE/B1
<i>S</i>-enantiomer									
³ I + <i>S</i> -Sub	-2724.214617	0.0	-2723.404002	0.0	-2724.896918	0.0	-2724.086303	0.0	0.810615
⁵ I + <i>S</i> -Sub	-2724.199664	9.4	-2723.389608	9.0	-2724.881228	9.8	-2724.071172	9.5	0.810056
³ TS _{<i>S</i>}	-2724.194168	12.8	-2723.386492	11.0	-2724.881688	9.6	-2724.074012	7.7	0.807676
⁵ TS _{<i>S</i>}	-2724.196030	11.7	-2723.389233	9.3	-2724.884267	7.9	-2724.077470	5.5	0.806797
<i>R</i>-enantiomer									
³ I + <i>R</i> -Sub	-2724.214613	0.0	-2723.404005	0.0	-2724.896919	0.0	-2724.086311	0.0	0.810608
⁵ I + <i>R</i> -Sub	-2724.199660	9.4	-2723.389611	9.0	-2724.881229	9.8	-2724.071180	9.5	0.810049
³ TS _{<i>R</i>}	-2724.187034	17.3	-2723.379333	15.5	-2724.877133	12.4	-2724.069432	10.6	0.807701
⁵ TS _{<i>R</i>}	-2724.192413	13.9	-2723.386093	11.2	-2724.881785	9.5	-2724.075465	6.8	0.806320

Table S4 Mulliken spin densities and charges of the oxidant-substrate complex (**I** + *S-/R-Sub*) as well as the transition states in the α -CH activation of both *S*- and *R*-1-phenylethanol by the $[(P\text{-MCP})\text{Mn}^{5+}(\text{O}^{2-})(\text{SO}_4^{2-})]^+$ species (**I**) calculated at the UB3LYP/LACVP(Mn), 6-31G*(C, H, O, N, and S) level in solvent

	spin density						charge					
	Mn	P-MCP	O	SO_4	H	PA-H	Mn	P-MCP	O	SO_4	H	PA-H
<i>S</i>-enantiomer												
$^3\mathbf{I}+\text{S-Sub}$	2.84	-0.18	-0.44	-0.21	0.00	0.00	0.82	1.49	-0.34	-0.97	0.13	-0.13
$^5\mathbf{I}+\text{S-Sub}$	2.79	-0.19	0.87	0.52	0.00	0.00	0.82	1.44	-0.39	-0.88	0.13	-0.13
$^3\mathbf{TS}_S$	2.81	-0.16	-0.35	-0.07	0.01	-0.24	0.83	1.48	-0.48	-1.03	0.19	0.01
$^5\mathbf{TS}_S$	2.68	-0.11	0.59	0.07	0.09	0.68	0.82	1.34	-0.61	-1.17	0.35	0.27
<i>R</i>-enantiomer												
$^3\mathbf{I}+\text{R-Sub}$	2.84	-0.18	-0.44	-0.21	0.00	0.00	0.82	1.49	-0.34	-0.97	0.13	-0.13
$^5\mathbf{I}+\text{R-Sub}$	2.79	-0.19	0.87	0.52	0.00	0.00	0.82	1.44	-0.39	-0.88	0.13	-0.13
$^3\mathbf{TS}_R$	2.95	-0.17	-0.49	-0.14	0.01	-0.15	0.88	1.51	-0.50	-0.99	0.26	-0.15
$^5\mathbf{TS}_R$	2.63	-0.12	0.62	0.10	0.09	0.69	0.80	1.36	-0.59	-1.16	0.33	0.26

Additional Data: NMR Spectra of the Products

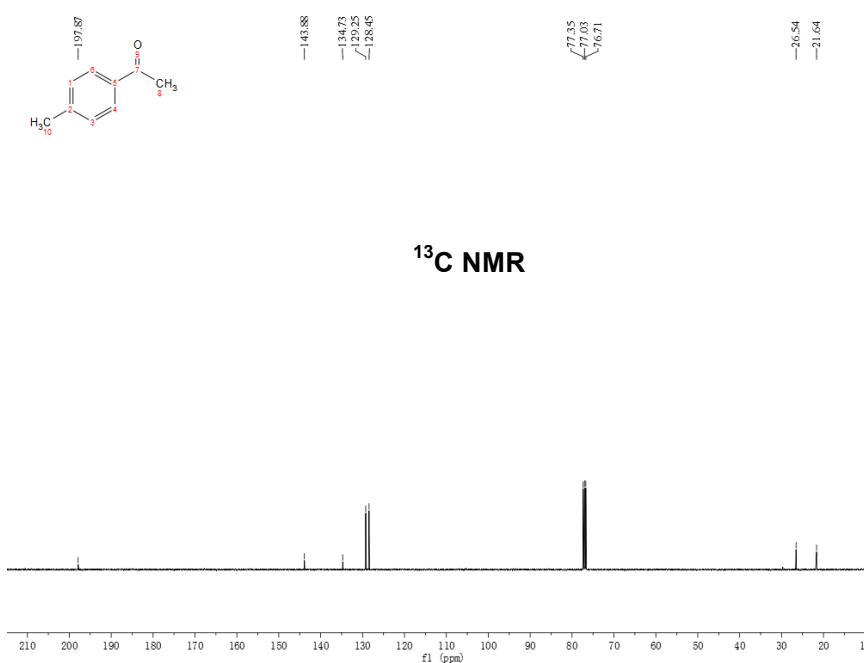
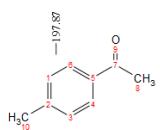
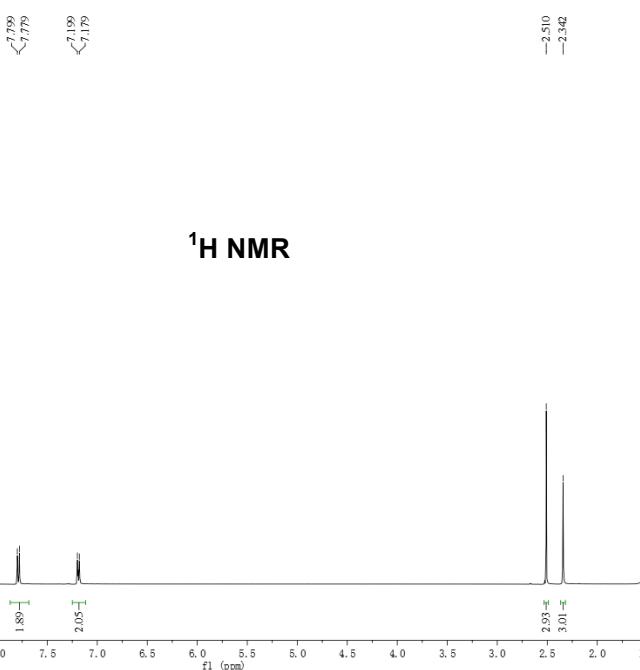
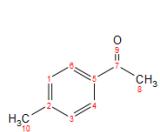


Fig. S1 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*p*-tolyl)ethan-1-one produced in the catalytic oxidation of 1-(*p*-tolyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.79 (d, *J* = 8.0 Hz, 2H), 7.19 (d, *J* = 8.0 Hz, 2H), 2.51 (s, 3H), 2.34 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 197.9, 143.9, 134.7, 129.3, 128.5, 26.5, 21.6.

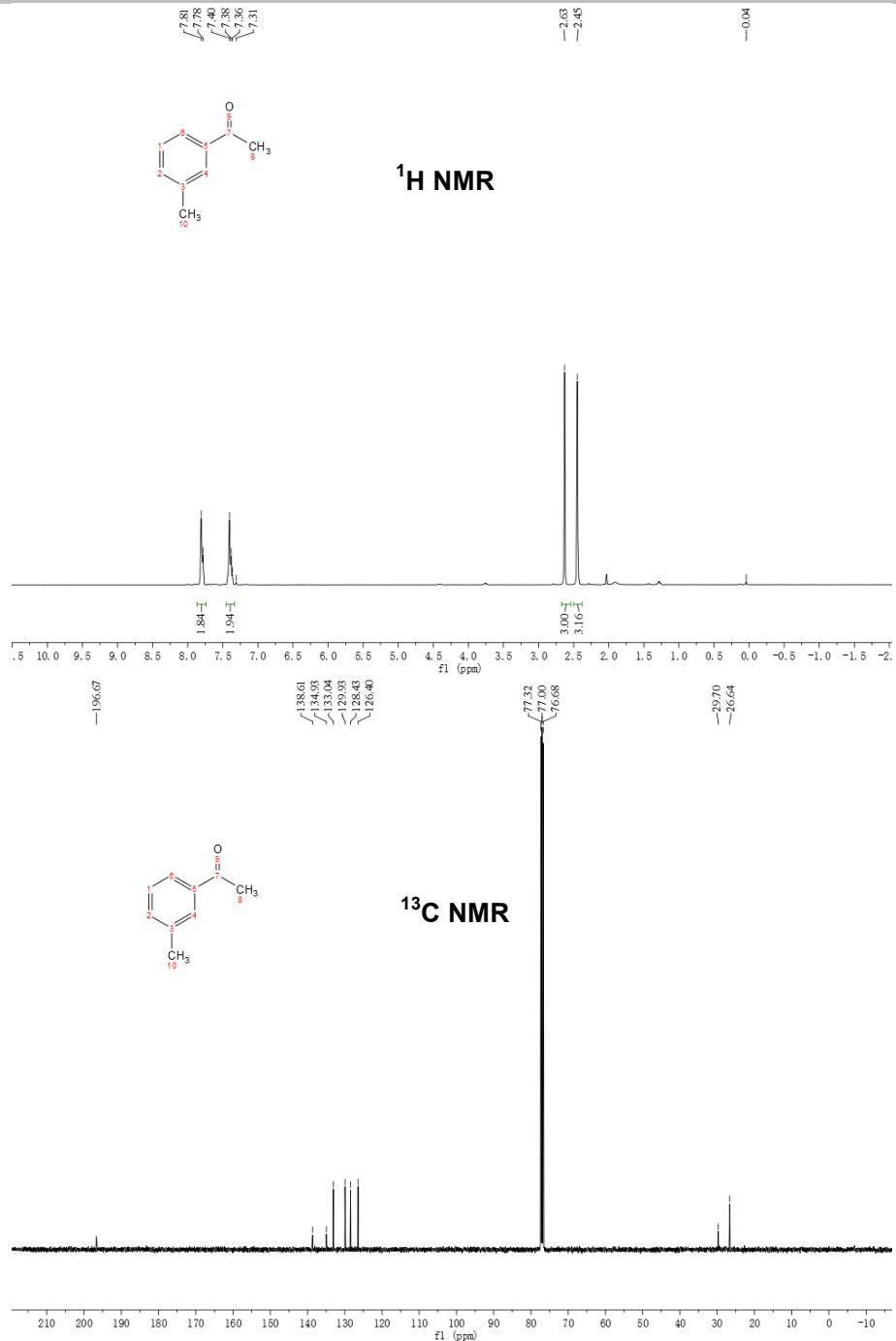


Fig. S2 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*m*-tolyl)ethan-1-one produced in the catalytic oxidation of 1-(*m*-tolyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.74 (d, *J* = 7.6 Hz, 1H), 7.42 (t, *J* = 7.2 Hz, 1H), 7.31 (m, 2H), 2.62 (s, 3H), 2.57 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 196.7, 138.6, 134.9, 133.0, 129.9, 128.4, 126.4, 29.7, 26.6.

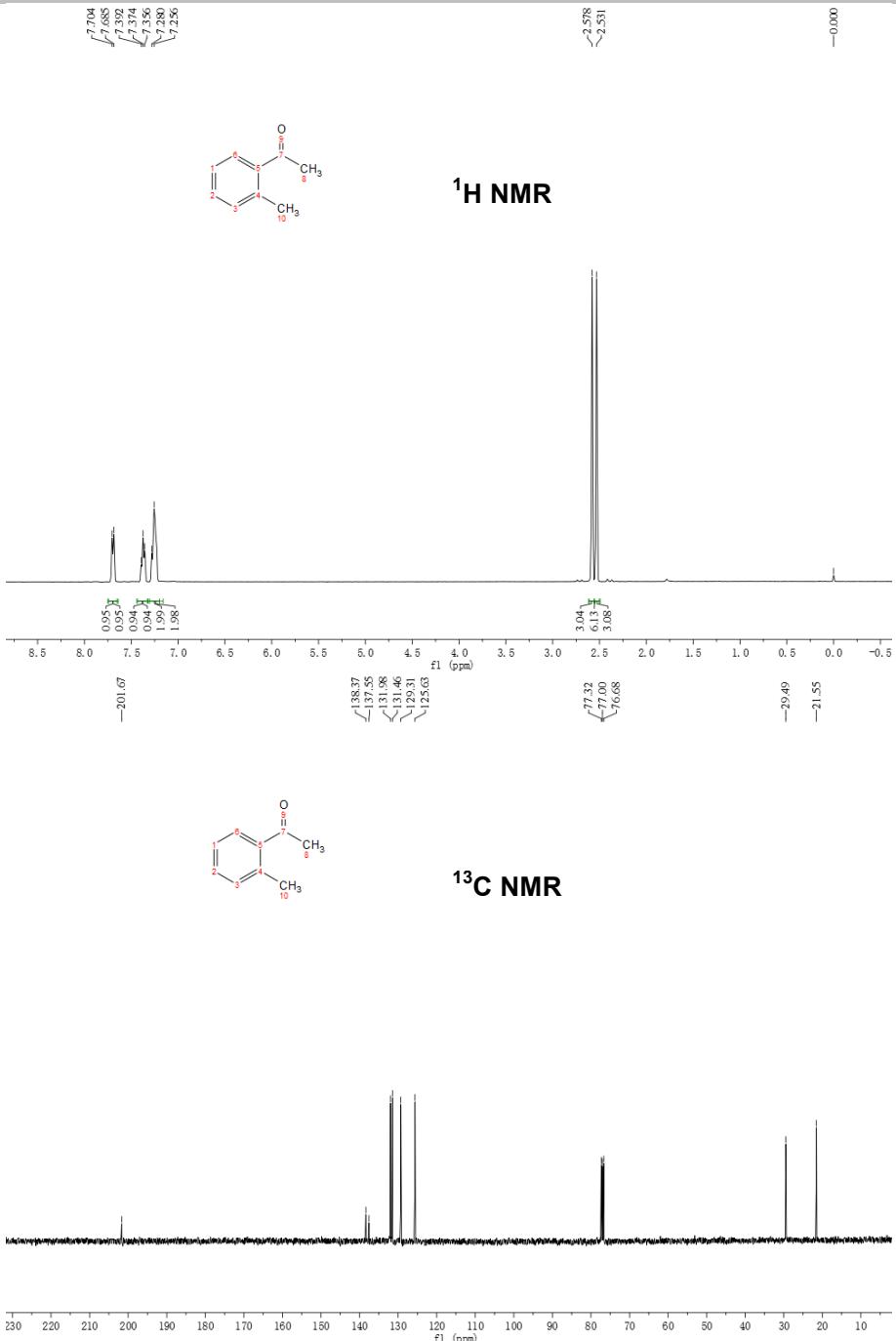


Fig. S3 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*o*-tolyl)ethan-1-one produced in the catalytic oxidation of 1-(*o*-tolyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.69 (d, *J* = 7.6 Hz, 1H), 7.37 (t, *J* = 7.2 Hz, 1H), 7.27 (d, *J* = 9.6 Hz, 2H), 2.58 (s, 3H), 2.53 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 201.7, 138.4, 137.6, 132.0, 131.5, 129.3, 125.6, 29.5, 21.6.

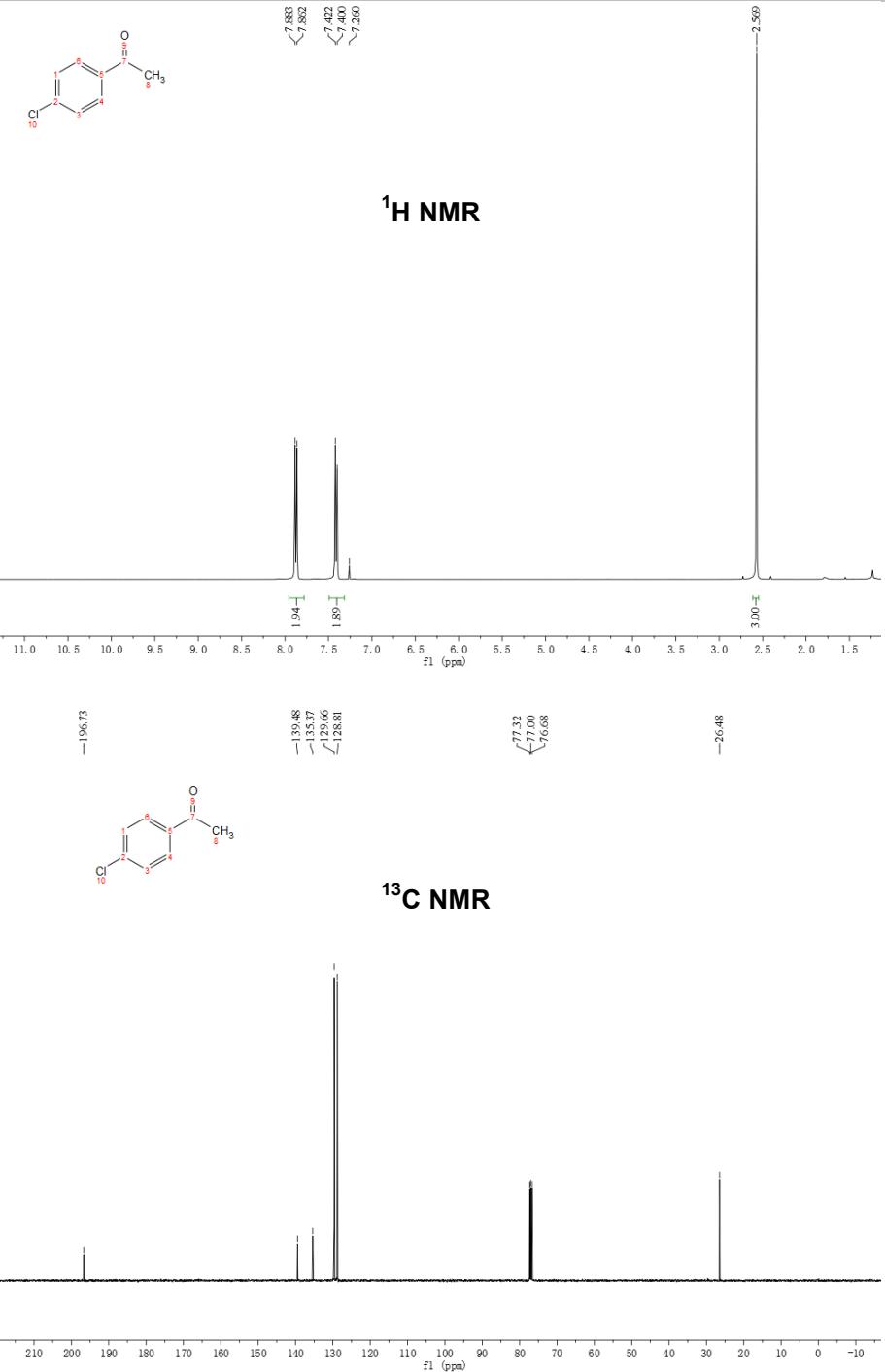


Fig. S4 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*p*-chlorophenyl)ethan-1-one produced in the catalytic oxidation of 1-(*p*-chlorophenyl)ethan-1-ol (0.50 mmol) by **1** (0.03 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.89 (d, *J* = 8.4 Hz, 2H), 7.43 (d, *J* = 8.8 Hz, 2H), 2.59 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 196.8, 139.6, 135.4, 129.7, 128.9, 26.5.

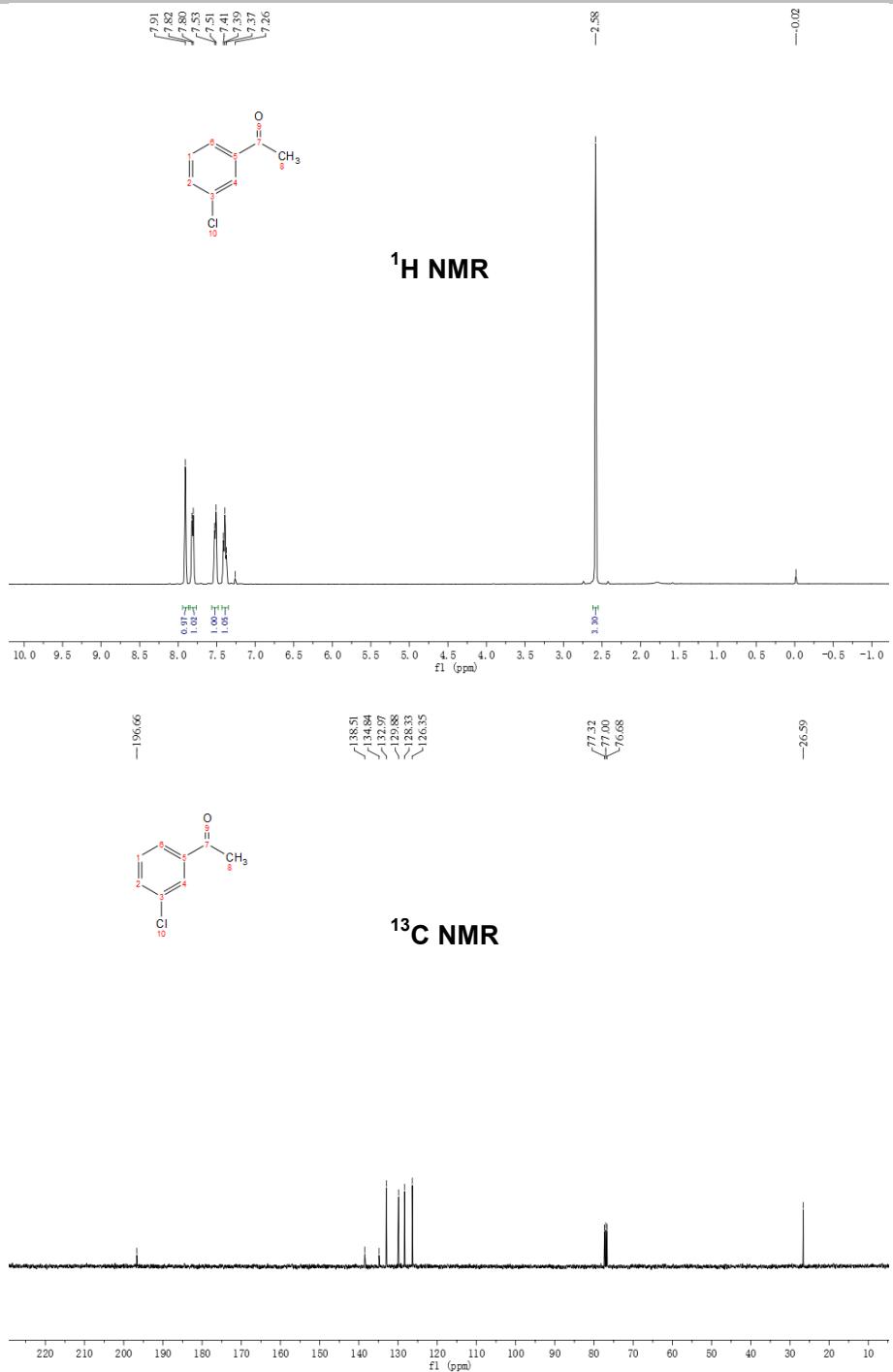


Fig. S5 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*m*-chlorophenyl)ethan-1-one produced in the catalytic oxidation of 1-(*m*-chlorophenyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.91 (s, 1H), 7.81 (d, *J* = 7.6 Hz, 1H), 7.52 (d, *J* = 7.6 Hz, 1H), 7.39 (t, *J* = 7.6 Hz, 1H), 2.58 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 196.7, 138.5, 134.8, 133.0, 129.9, 128.3, 126.4, 26.6.

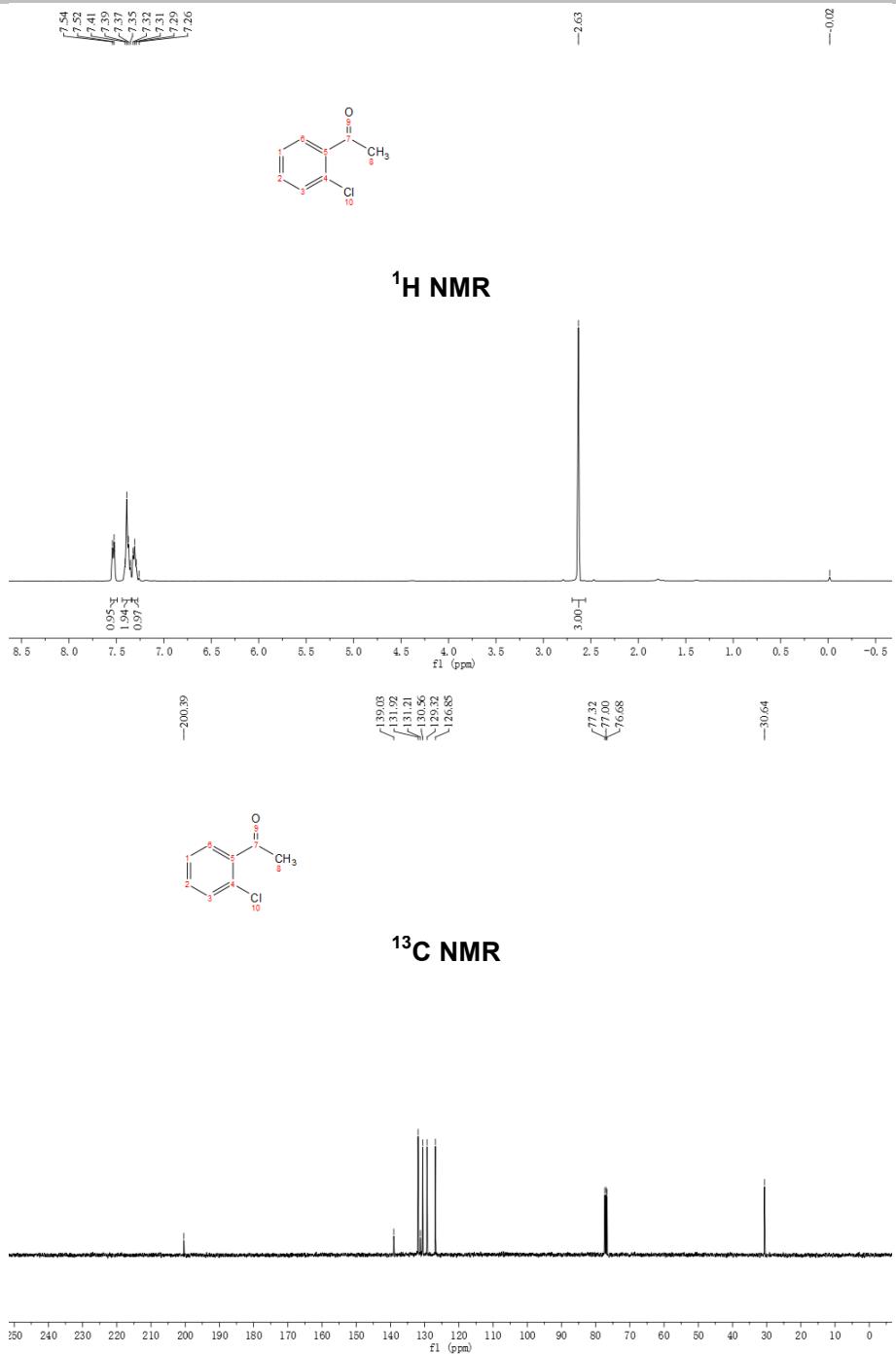


Fig. S6 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*o*-chlorophenyl)ethan-1-one produced in the catalytic oxidation of 1-(*o*-chlorophenyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.53 (d, *J* = 7.2 Hz, 1H), 7.40-7.35 (m, 2H), 7.29 (t, *J* = 6.8 Hz, 1H), 2.63 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 200.4, 139.0, 131.9, 131.2, 130.6, 129.3, 126.9, 30.6.

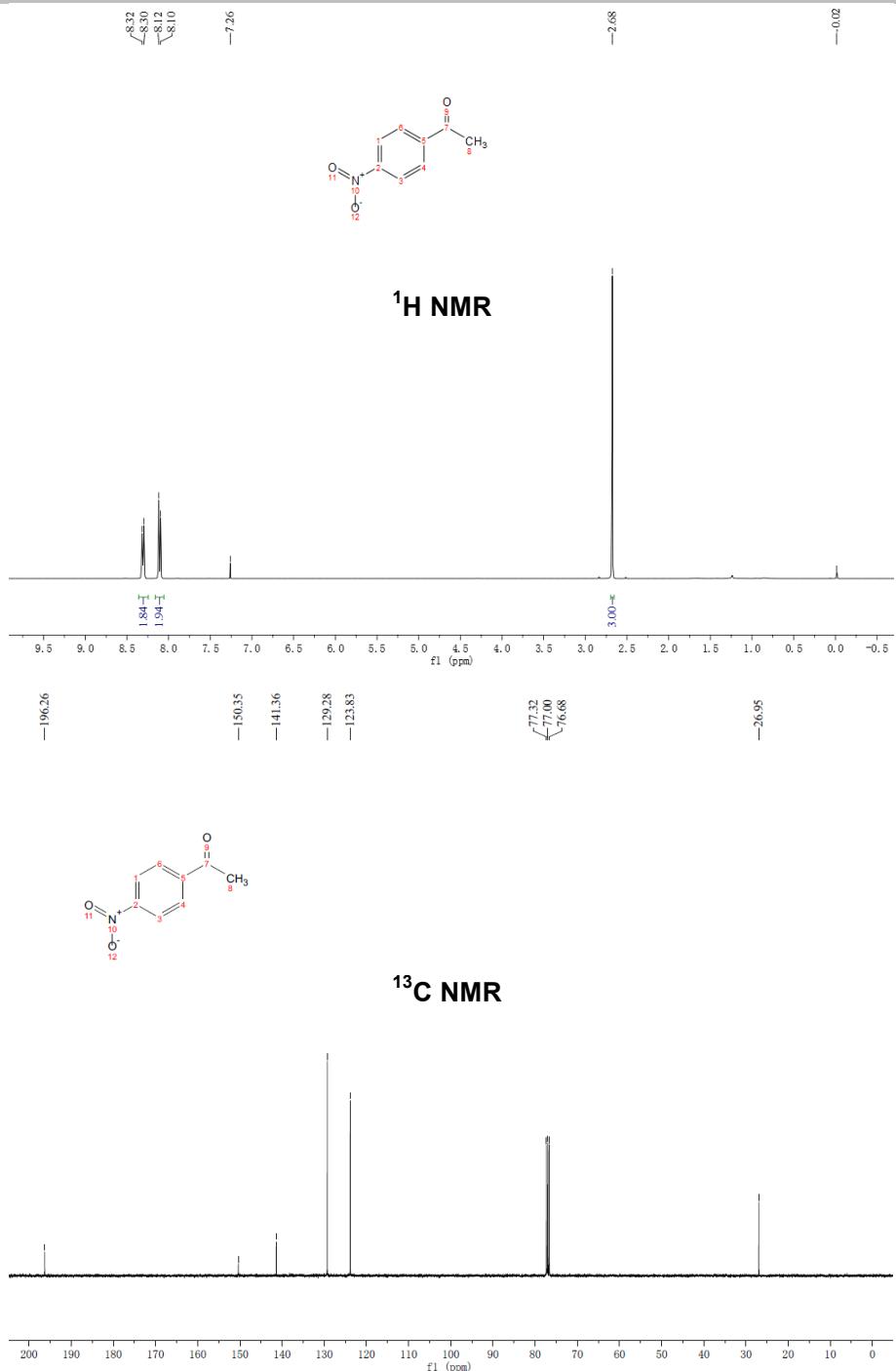
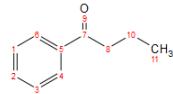


Fig. S7 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*p*-nitrophenyl)ethan-1-one produced in the catalytic oxidation of 1-(*p*-nitrophenyl)ethan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 8.31 (d, *J* = 7.2 Hz, 2H), 8.11 (d, *J* = 8.4 Hz, 2H), 2.68 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 196.3, 150.4, 141.4, 129.3, 123.8, 27.0.

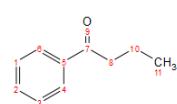
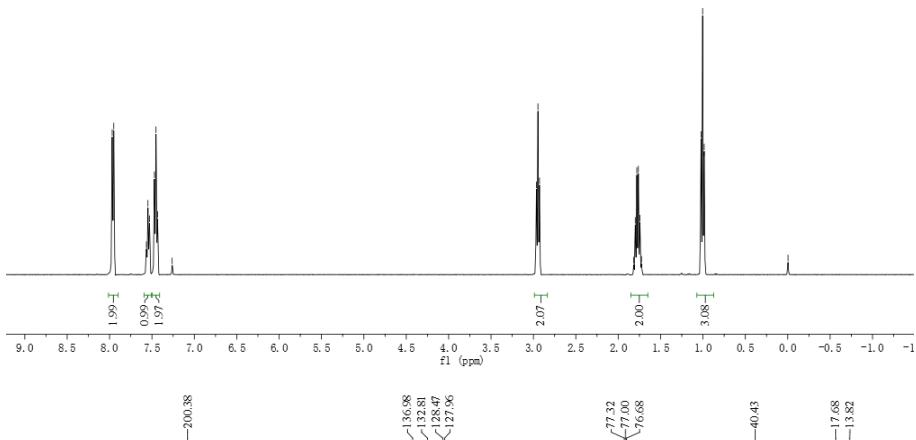
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7.45
7.43
7.26

2.96
2.95
2.93
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1.76
1.74
1.73
1.02
1.00
0.99

-0.00



¹H NMR



¹³C NMR

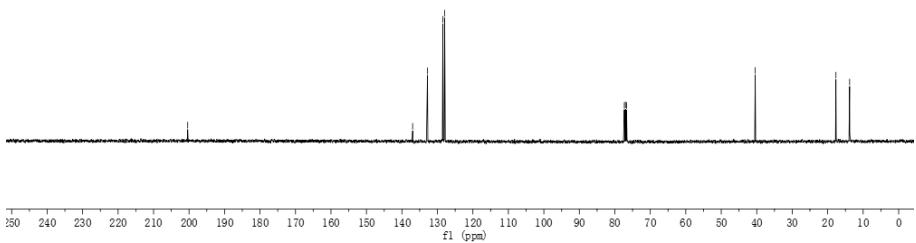


Fig. S8 ¹H (top) and ¹³C (bottom) NMR spectra of 1-phenylbutan-1-one produced in the catalytic oxidation of 1-phenylbutan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.96 (d, *J* = 7.6 Hz, 2H), 7.55 (t, *J* = 7.2 Hz, 1H), 7.45 (t, *J* = 7.2 Hz, 2H), 2.95 (t, *J* = 7.2 Hz, 2H), 1.77 (m, 2H), 1.00 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 200.5, 137.1, 132.9, 128.6, 128.0, 40.5, 17.8, 13.9.

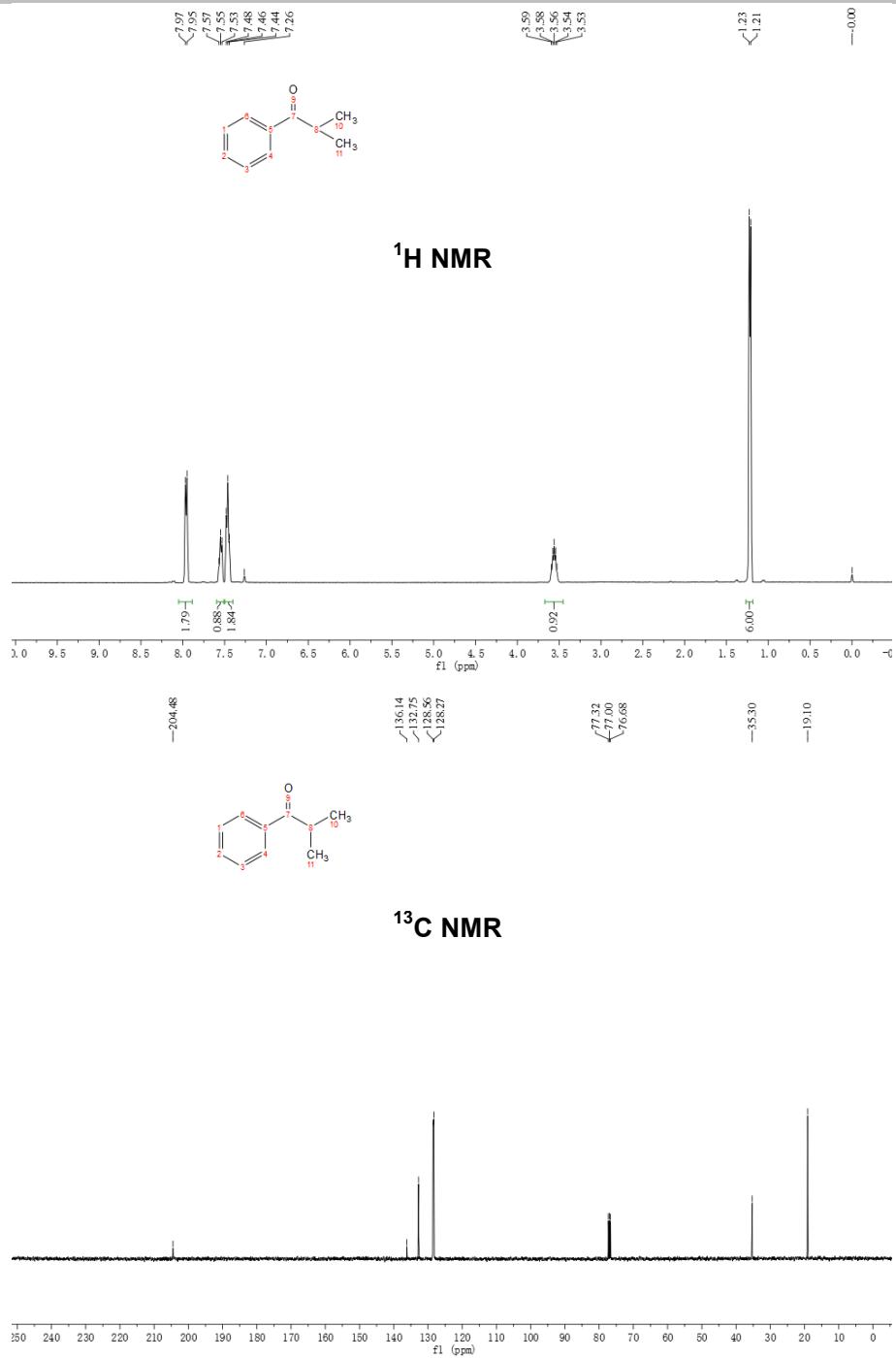


Fig. S9 ¹H (top) and ¹³C (bottom) NMR spectra of 2-methyl-1-phenylpropan-1-one produced in the catalytic oxidation of 2-methyl-1-phenylpropan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.96 (d, *J* = 7.6 Hz, 2H), 7.55 (t, *J* = 7.2 Hz, 1H), 7.46 (t, *J* = 7.2 Hz, 2H), 3.56 (m, 1H), 1.22 (d, *J* = 6.8 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 204.5, 136.2, 132.8, 128.6, 128.3, 35.4, 19.2.

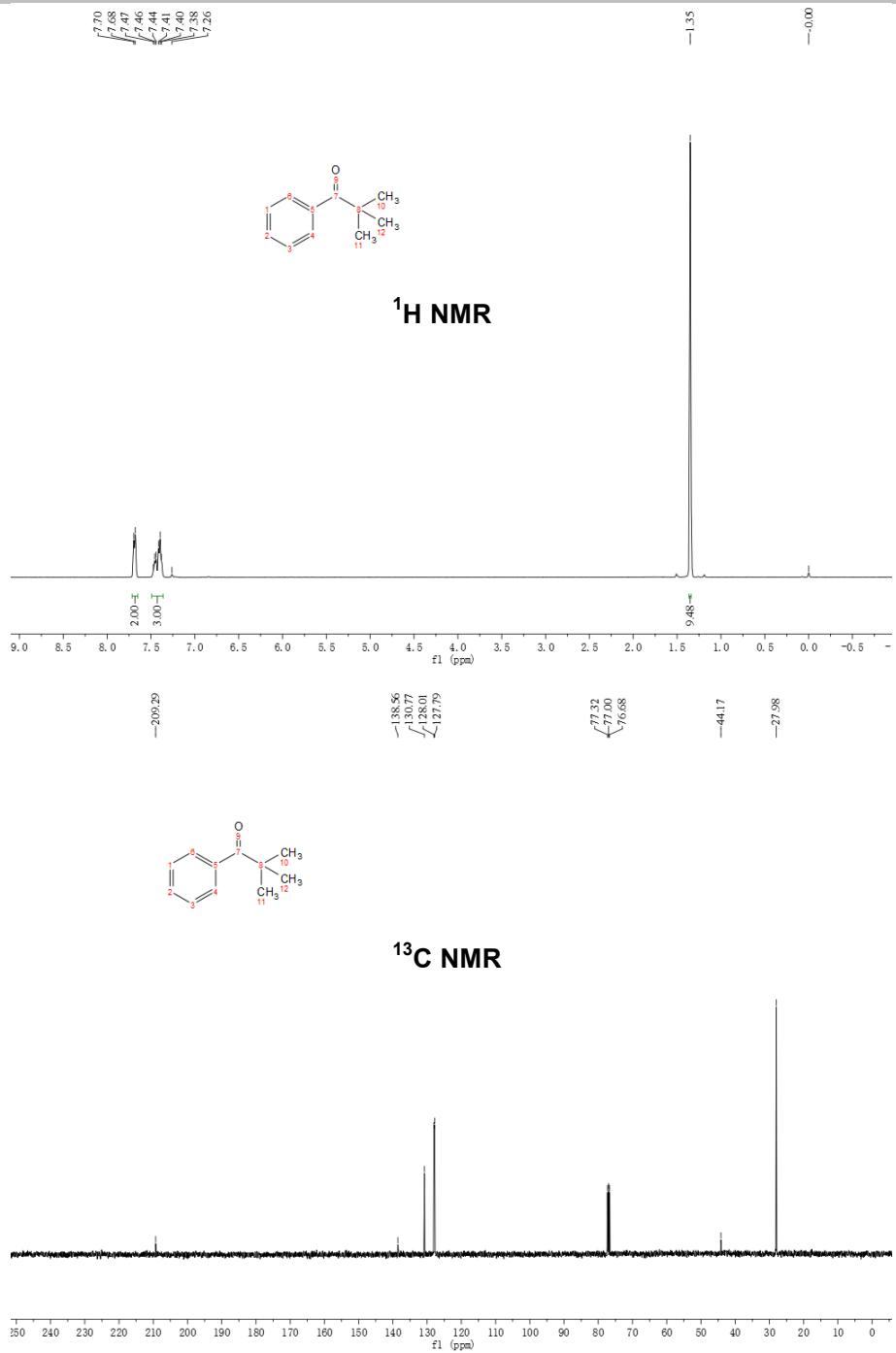


Fig. S10 ¹H (top) and ¹³C (bottom) NMR spectra of 2,2-dimethyl-1-phenylpropan-1-one produced in the catalytic oxidation of 2,2-dimethyl-1-phenylpropan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at room temperature for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.69 (d, *J* = 7.2 Hz, 2H), 7.43 (m, 3H), 1.35 (s, 9H). ¹³C NMR (100 MHz, CDCl₃) δ 209.3, 138.6, 130.8, 128.0, 127.8, 44.2, 28.0.

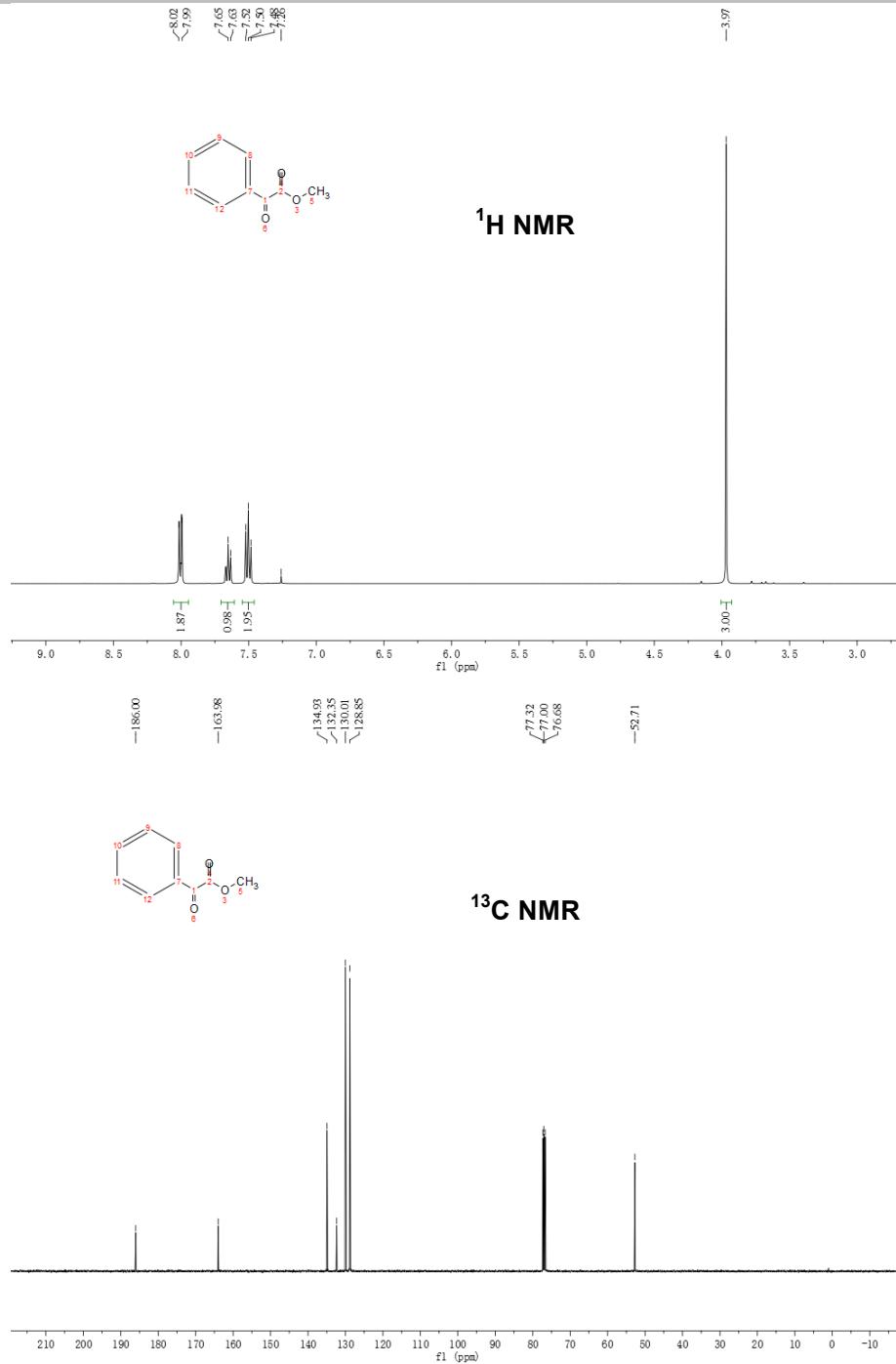


Fig. S11 ¹H (top) and ¹³C (bottom) NMR spectra of methyl 2-oxo-2-phenylacetate produced in the catalytic oxidation of methyl 2-hydroxy-2-phenylacetate (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.00 (d, *J* = 9.6 Hz, 2H), 7.65 (t, *J* = 7.2 Hz, 1H), 7.50 (t, *J* = 8.0 Hz, 2H), 3.97 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 186.1, 164.1, 135.0, 132.4, 130.1, 128.9, 52.8.

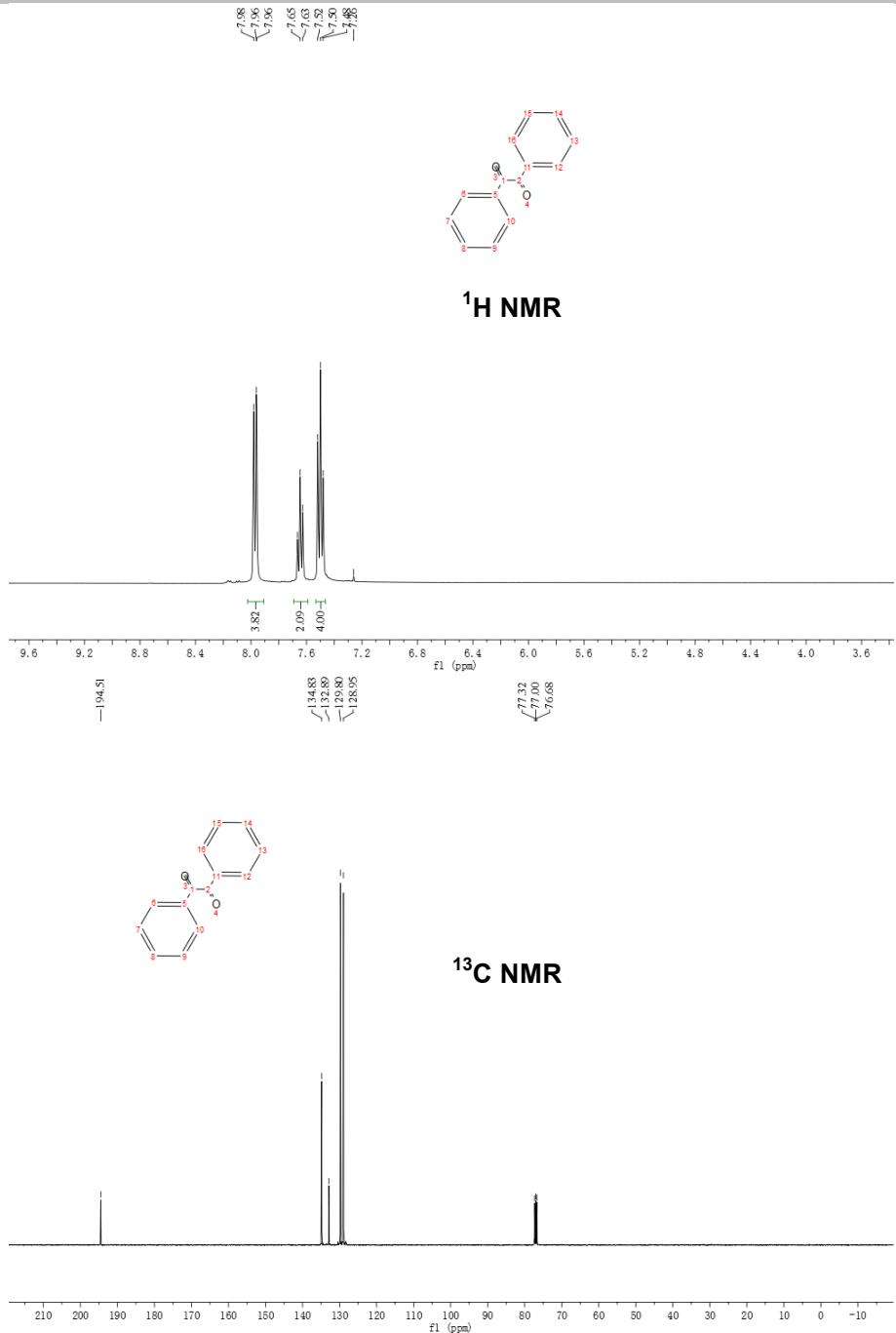


Fig. S12 ¹H (top) and ¹³C (bottom) NMR spectra of benzil produced in the catalytic oxidation of 2-hydroxy-1,2-diphenylethan-1-one (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 8.02–7.91 (m, 4H), 7.65 (t, *J* = 7.6 Hz, 2H), 7.50 (t, *J* = 8.0 Hz, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 194.6, 134.9, 133.0, 129.9, 129.1.

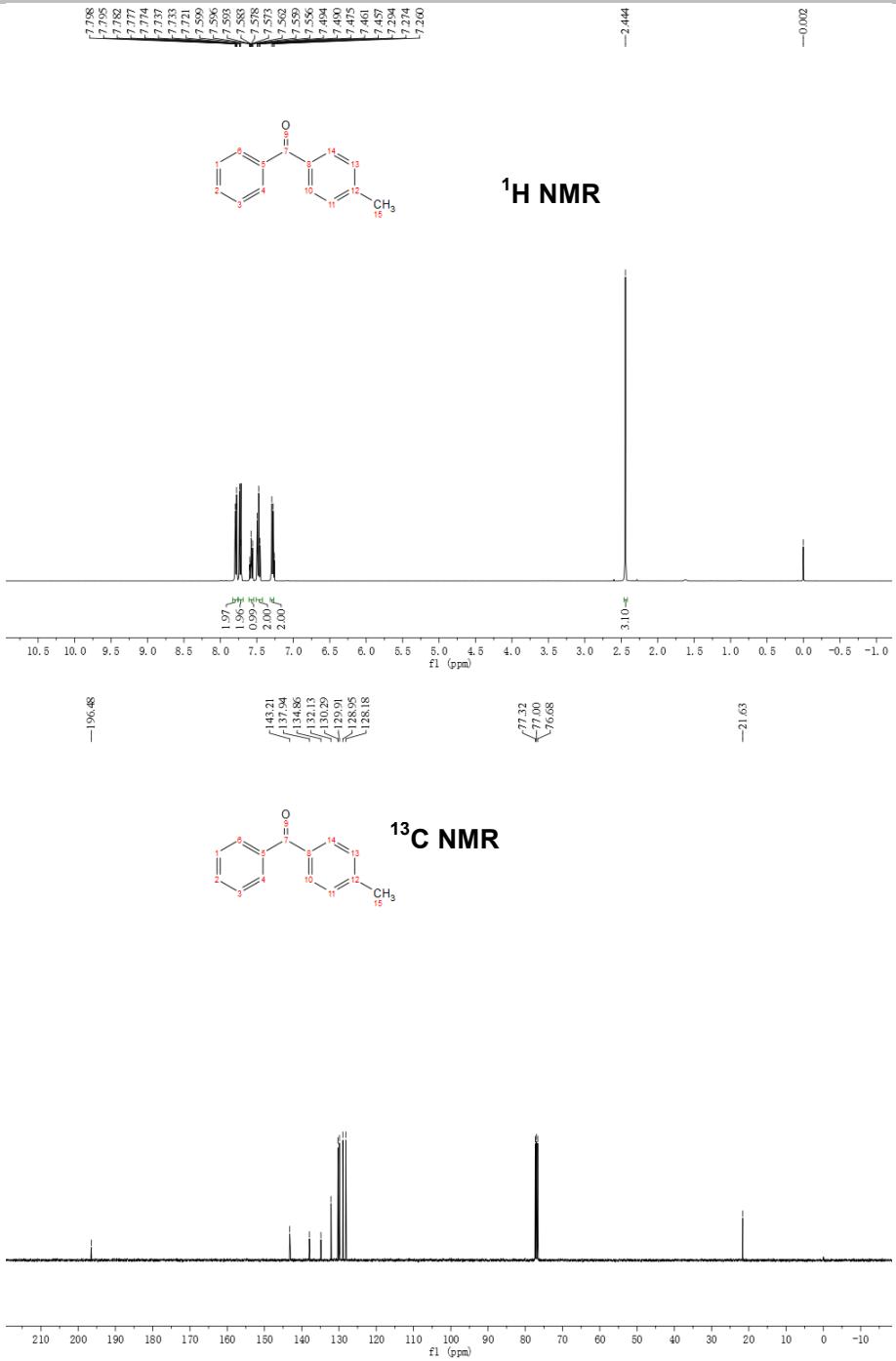


Fig. S13 ¹H (top) and ¹³C (bottom) NMR spectra of phenyl(*p*-tolyl)methanone produced in the catalytic oxidation of phenyl(*p*-tolyl)methanol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.80–7.77 (m, 2H), 7.74–7.72 (m, 2H), 7.59–7.55 (m, 1H), 7.49–7.46 (m, 2H), 7.28 (d, *J* = 8.0 Hz, 2H), 2.44 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 196.5, 143.2, 137.9, 134.9, 132.1, 130.3, 129.9, 129.0, 128.2, 21.6.

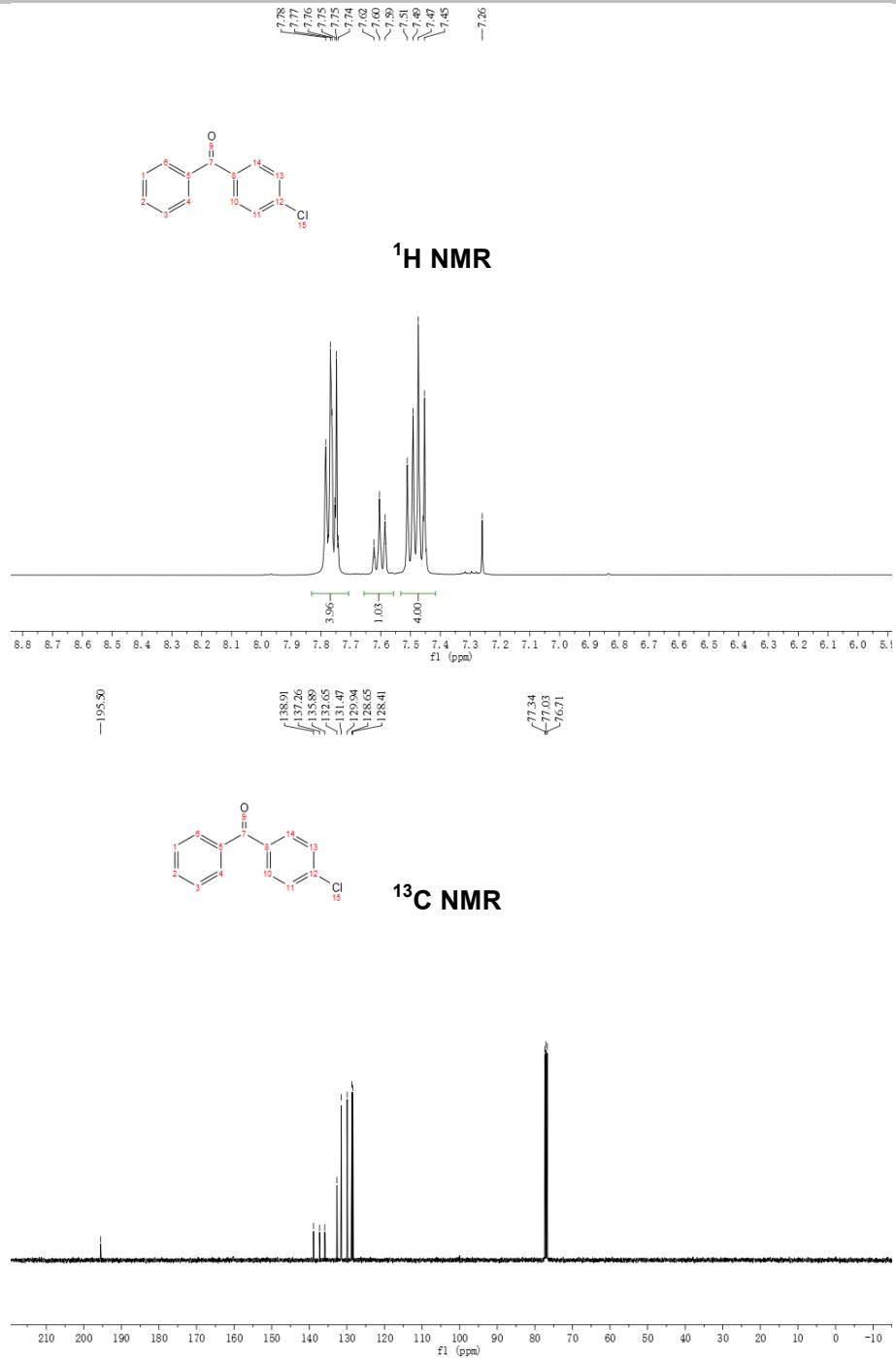
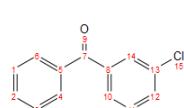
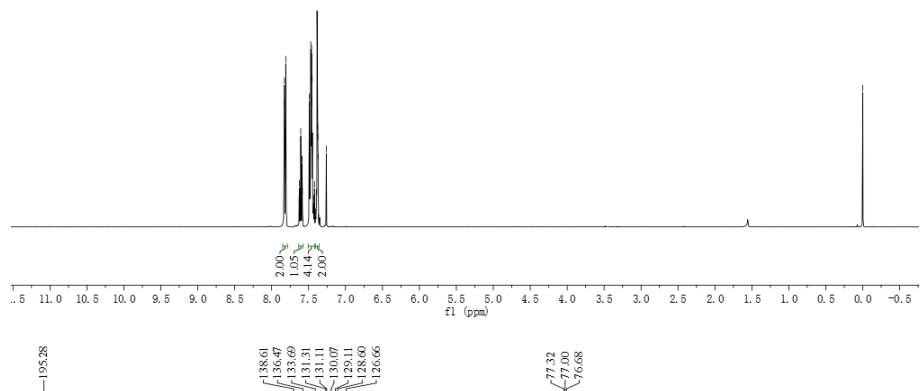


Fig. S14 ¹H (top) and ¹³C (bottom) NMR spectra of (p-chlorophenyl)(phenyl)methanone produced in the catalytic oxidation of (p-chlorophenyl)(phenyl)methanol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.79–7.75 (m, 4H), 7.63–7.57 (m, 1H), 7.52–7.45 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 195.5, 138.9, 137.2, 135.9, 132.6, 131.5, 129.9, 128.6, 128.4.



¹H NMR



¹³C NMR

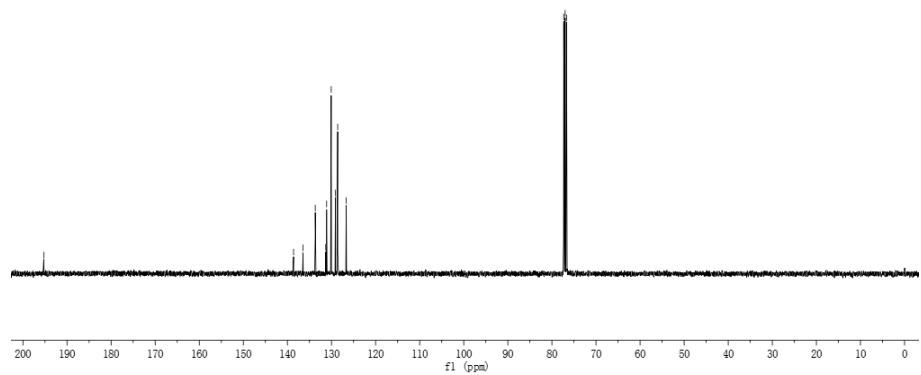


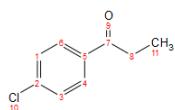
Fig. S15 ¹H (top) and ¹³C (bottom) NMR spectra of (*m*-chlorophenyl)(phenyl)methanone produced in the catalytic oxidation of (*m*-chlorophenyl)(phenyl)methanol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.83–7.81 (m, 2H), 7.62–7.58 (m, 1H), 7.49–7.42 (m, 4H), 7.41–7.36 (m, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 195.3, 138.6, 136.5, 133.7, 131.3, 131.1, 130.1, 129.1, 128.6, 126.7.

<7.910
<7.889
<7.436
<7.415
~7.260

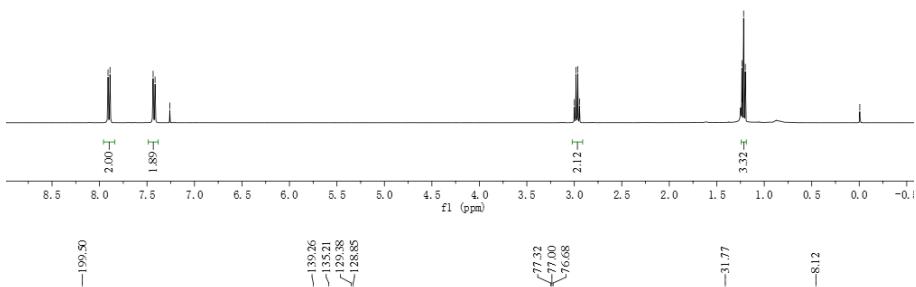
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2.882
2.864
2.846

1.235
1.217
1.199

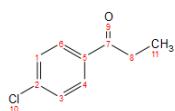
-0.006



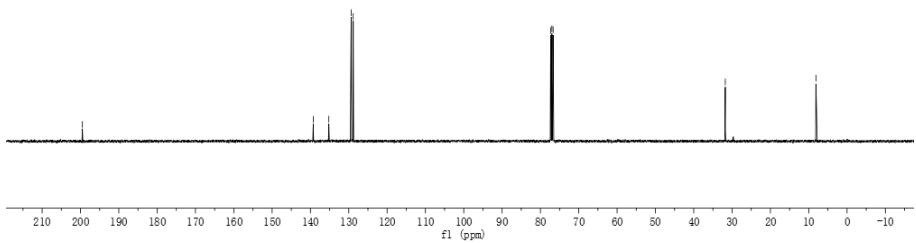
¹H NMR



-199.50



¹³C NMR



-199.50

-139.26
-135.21
-129.38
-128.85

77.32
77.00
76.68

21.12
1.77

-8.12

Fig. S16 ¹H (top) and ¹³C (bottom) NMR spectra of 1-(*p*-chlorophenyl)propan-1-one produced in the catalytic oxidation of 1-(*p*-chlorophenyl)propan-1-ol (0.50 mmol) by **1** (0.30 mol%) and H₂O₂ (1.2 equiv. based on the substrate concentration) in the presence of H₂SO₄ (0.30 mol%) in CH₃CN at 25 °C for 1 h. NMR spectra were recorded in CDCl₃ at 25 °C: ¹H NMR (400 MHz, CDCl₃) δ 7.90 (d, *J* = 8.4 Hz, 2H), 7.43 (d, *J* = 8.4 Hz, 2H), 2.97 (q, *J* = 7.2 Hz, 2H), 1.22 (t, *J* = 7.2 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 199.5, 139.3, 135.2, 129.4, 128.9, 31.8, 8.2.

Additional Data: GC Chromatograms in the OKR of Secondary Alcohols

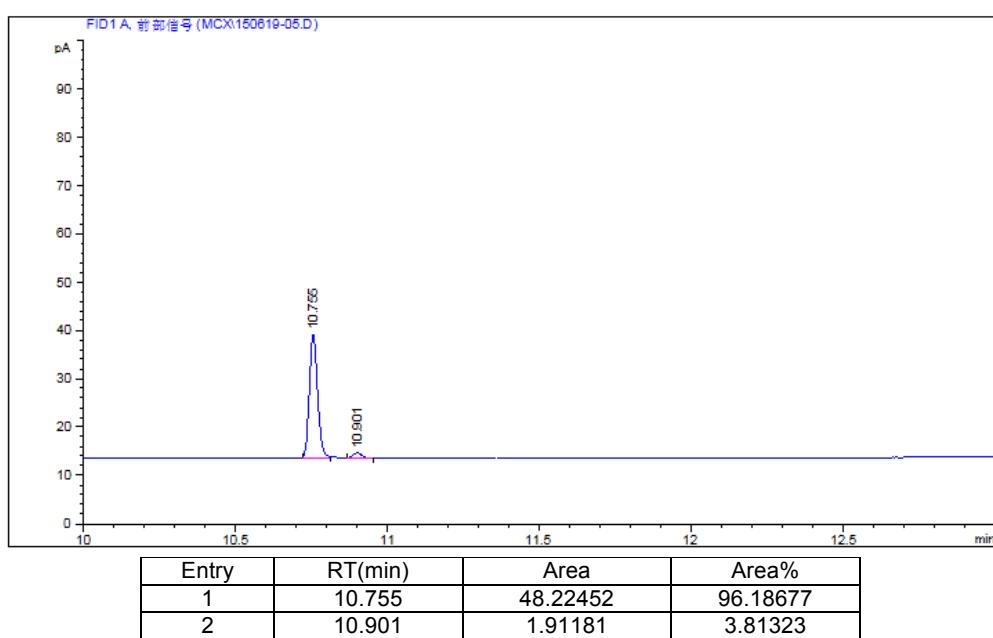
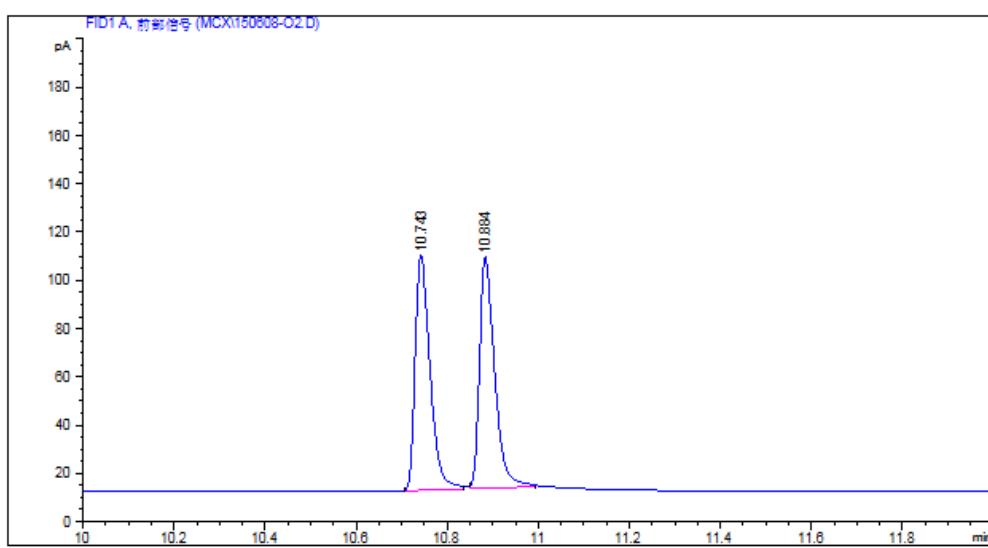


Fig. S17 GC chromatograms of racemic 1-phenylethanol (top) as an authentic compound and the unreacted 1-phenylethanol (bottom) remained in the oxidative kinetic resolution (OKR) of 1-phenylethanol (0.50 mmol) by 30% H₂O₂ (0.80 equiv based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 10.76, t_R(minor) = 10.90.

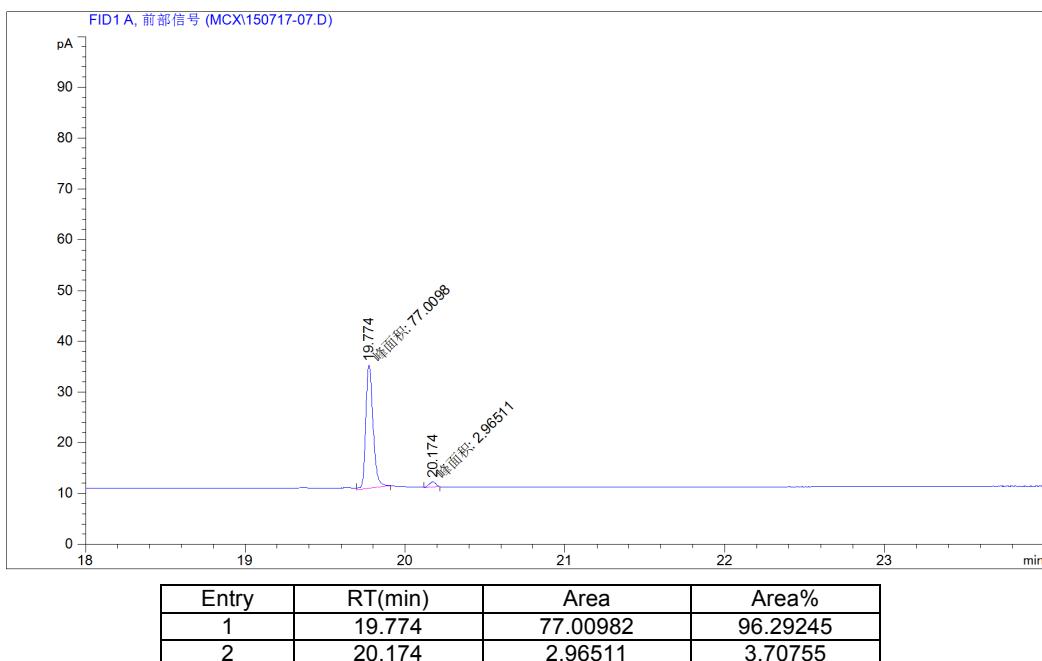
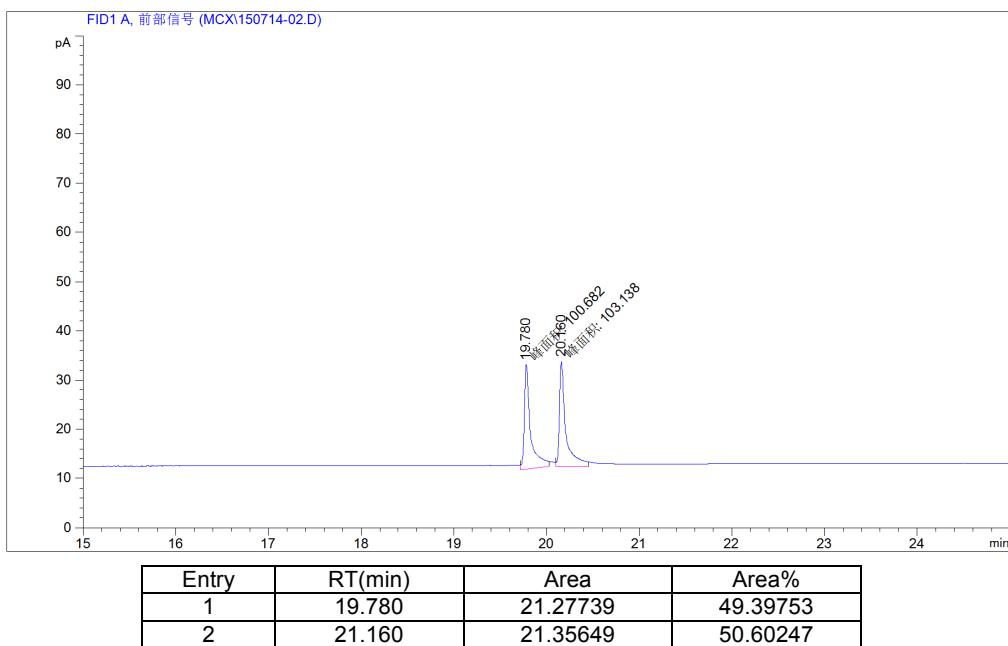


Fig. S18 GC chromatograms of racemic 1-(*p*-chlorophenyl)ethan-1-ol (top) as an authentic compound and the unreacted 1-(*p*-chlorophenyl)ethan-1-ol (bottom) remained in the OKR of 1-(*p*-chlorophenyl)ethan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 19.77, t_R(minor) = 20.17.

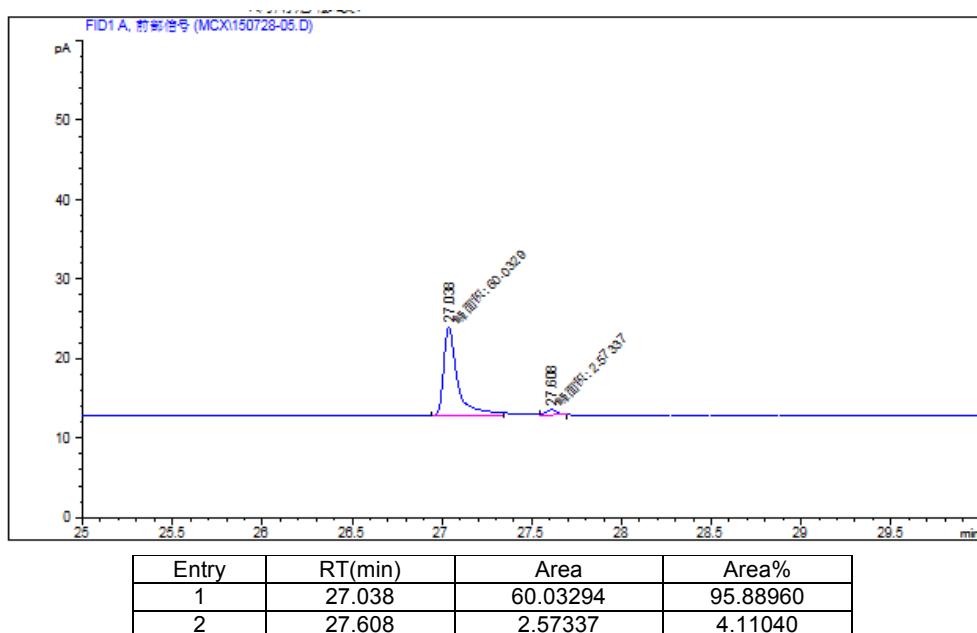
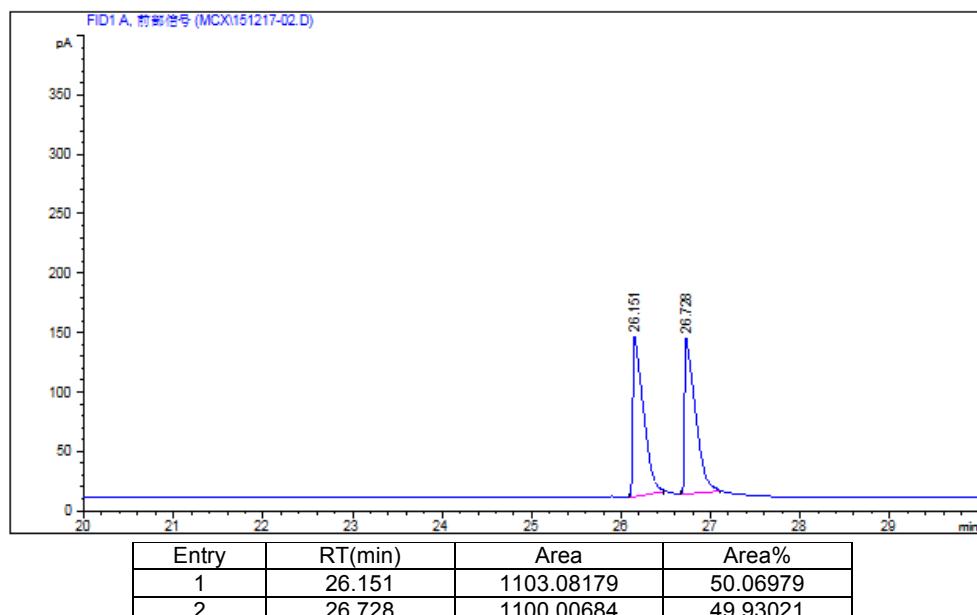


Fig. S19 GC chromatograms of racemic 1-(*m*-chlorophenyl)ethan-1-ol (top) as an authentic compound and the unreacted 1-(*m*-chlorophenyl)ethan-1-ol (bottom) remained in the OKR of 1-(*m*-chlorophenyl)ethan-1-ol (0.50 mmol) by 30% H₂O₂ (0.90 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 27.04, t_R(minor) = 27.61.

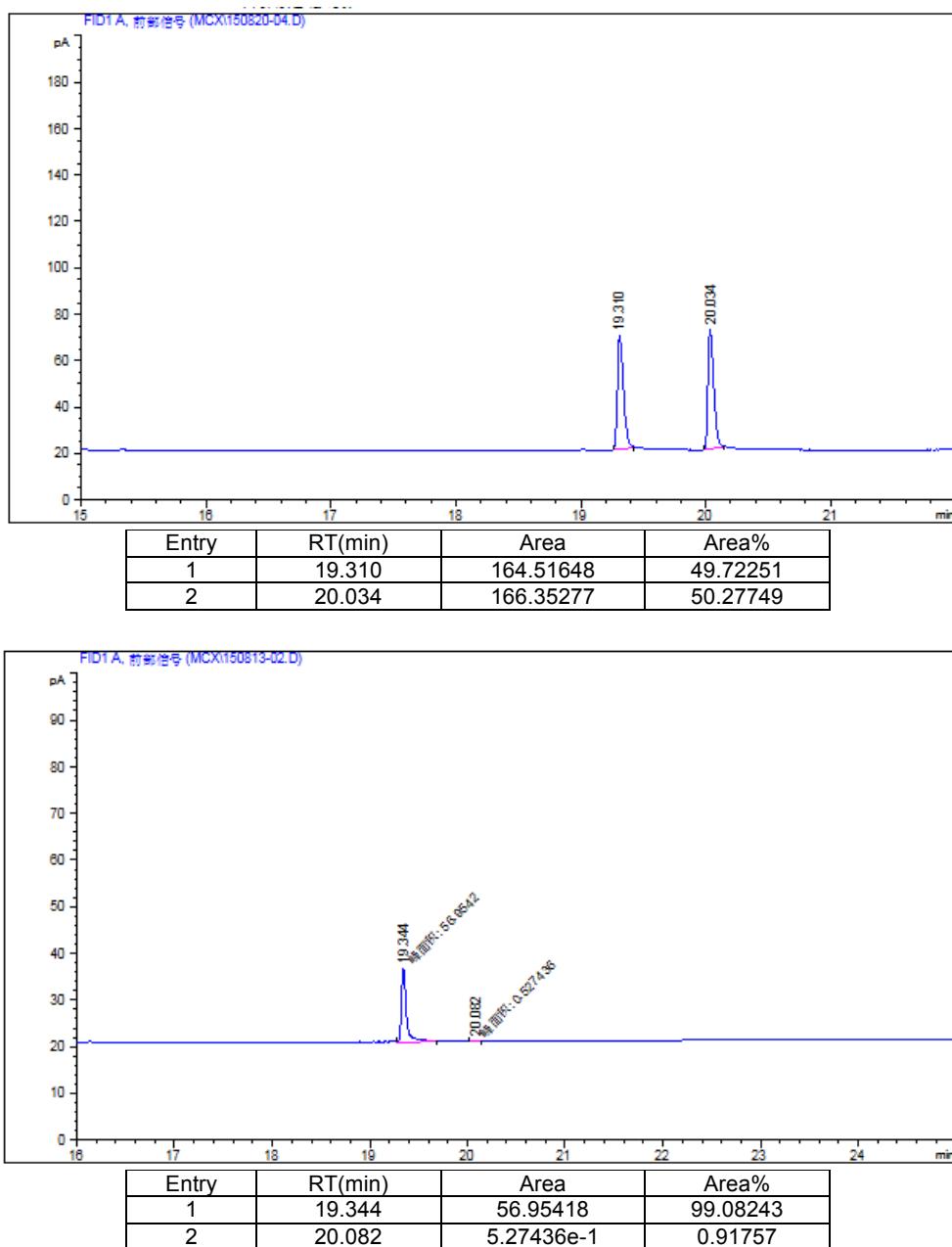


Fig. S20 GC chromatograms of racemic 1-(*o*-chlorophenyl)ethan-1-ol (top) as an authentic compound and the unreacted 1-(*o*-chlorophenyl)ethan-1-ol (bottom) remained in the OKR of 1-(*o*-chlorophenyl)ethan-1-ol (0.50 mmol) by 30% H₂O₂ (0.90 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 19.34, t_R(minor) = 20.08.

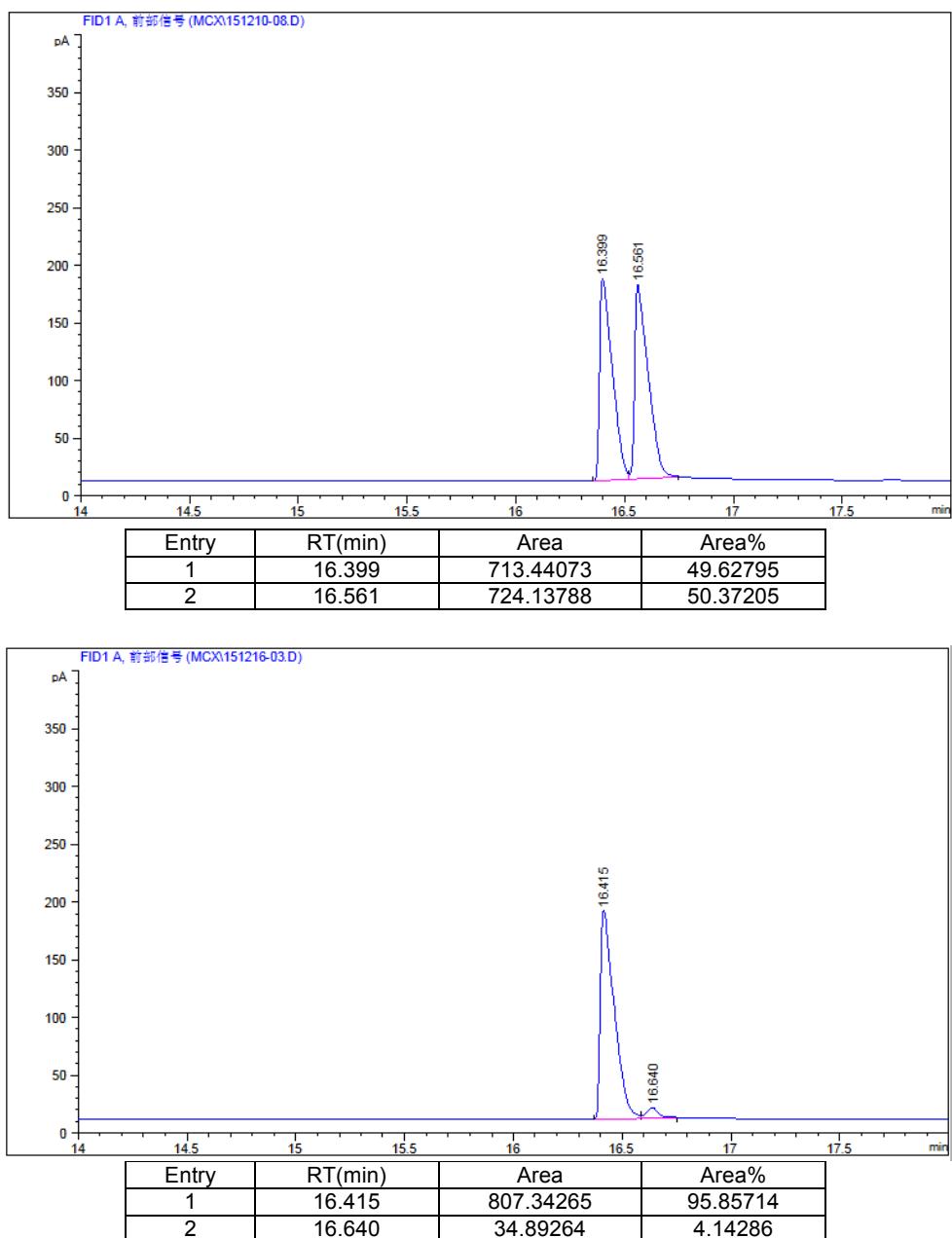


Fig. S21 GC chromatograms of racemic 1-phenylpropan-1-ol (top) as an authentic compound and the unreacted 1-phenylpropan-1-ol (bottom) remained in the OKR of 1-phenylpropan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 16.42, t_R(minor) = 16.64.

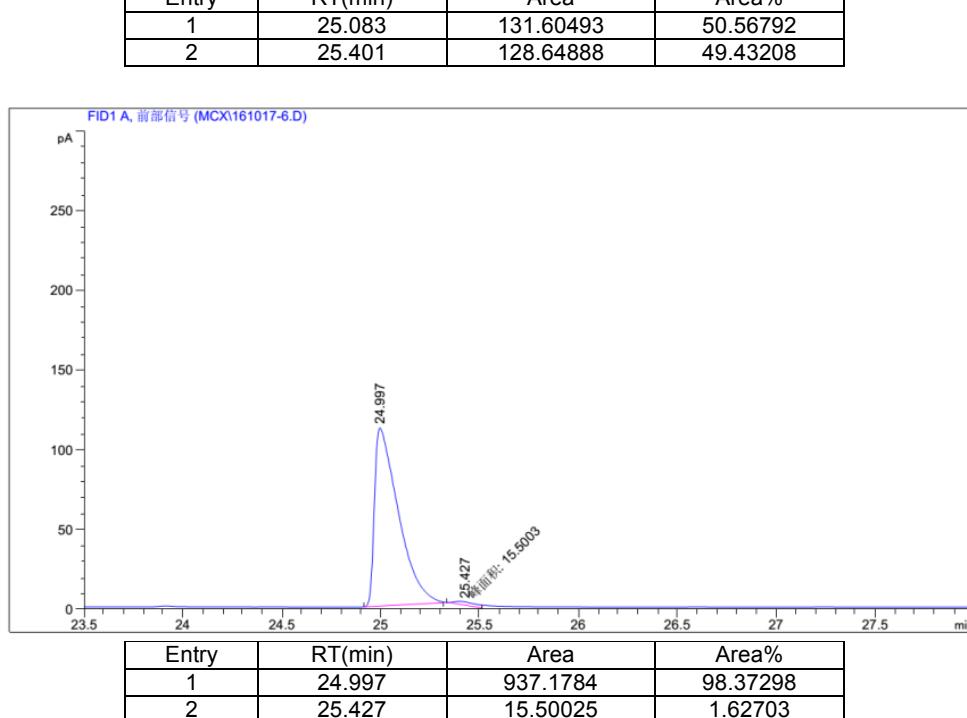
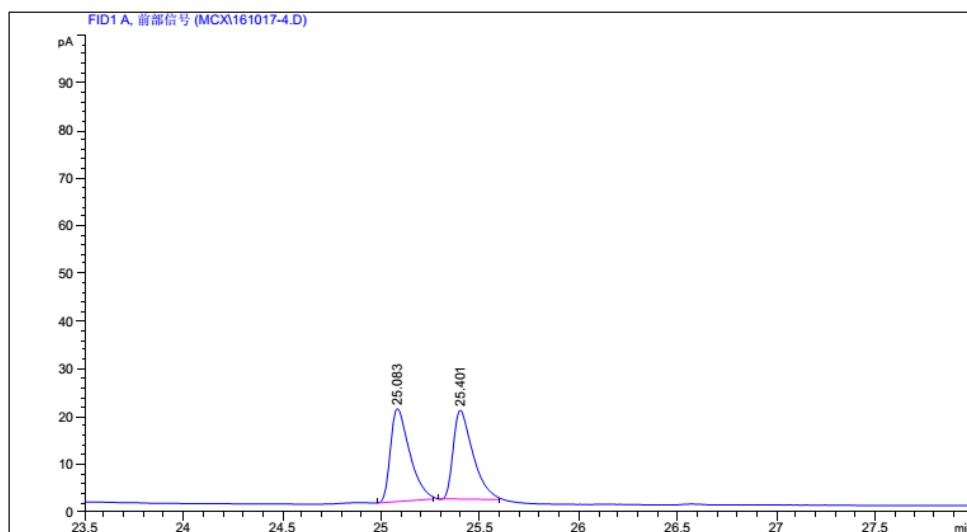


Fig. S22 GC chromatograms of racemic 1-(*p*-bromophenyl)propan-1-ol (top) as an authentic compound and the unreacted 1-(*p*-bromophenyl)propan-1-ol (bottom) remained in the OKR of 1-(*p*-bromophenyl)propan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 25.00, t_R(minor) = 25.43.

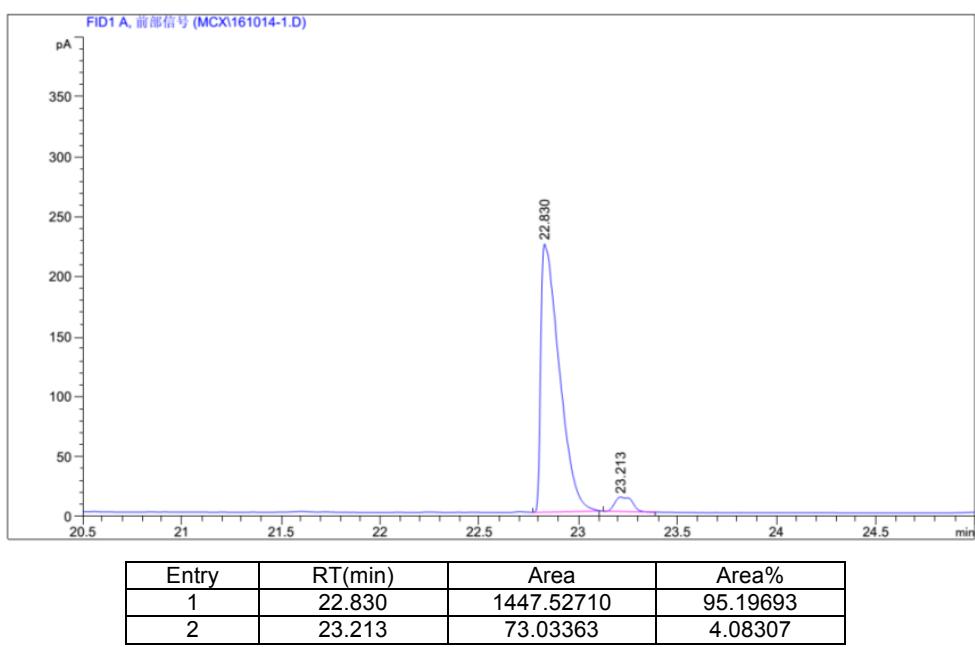
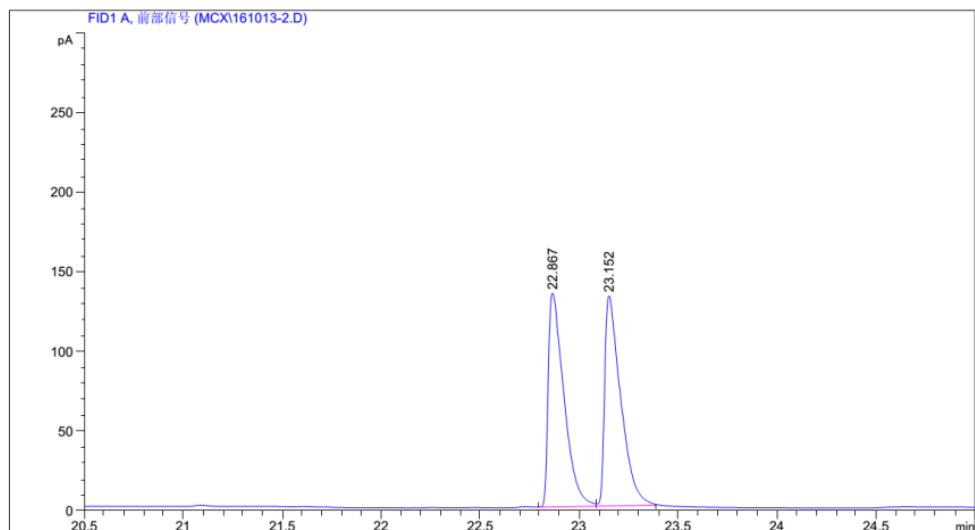


Fig. S23 GC chromatograms of racemic 1-(*p*-chlorophenyl)propan-1-ol (top) as an authentic compound and the unreacted 1-(*p*-chlorophenyl)propan-1-ol (bottom) remained in the OKR of 1-(*p*-chlorophenyl)propan-1-ol (0.50 mmol) by 30% H₂O₂ (0.70 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 5 °C/min to 170 °C, hold for 20 min. t_R(major) = 22.83, t_R(minor) = 23.21.

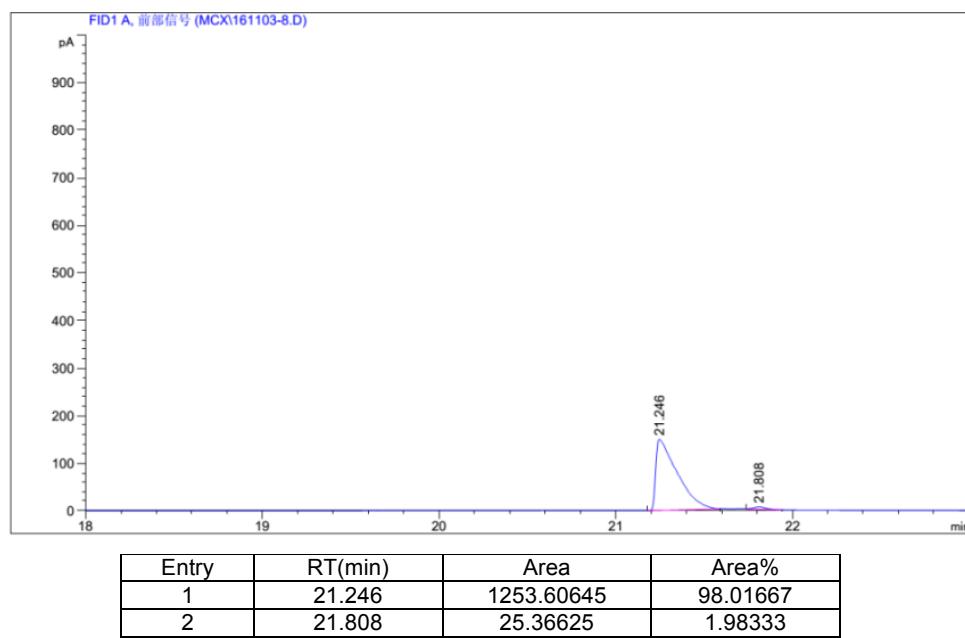
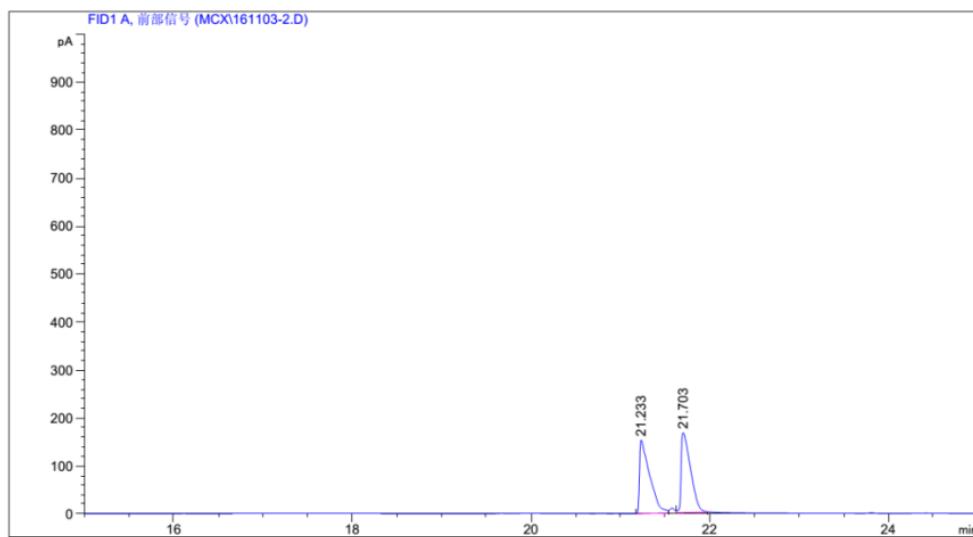


Fig. S24 GC chromatograms of racemic 1-(*p*-fluorophenyl)propan-1-ol (top) as an authentic compound and the unreacted 1-(*p*-fluorophenyl)propan-1-ol (bottom) remained in the OKR of 1-(*p*-fluorophenyl)propan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 4 °C/min to 170 °C, hold for 20 min. t_R(major) = 21.25, t_R(minor) = 21.81.

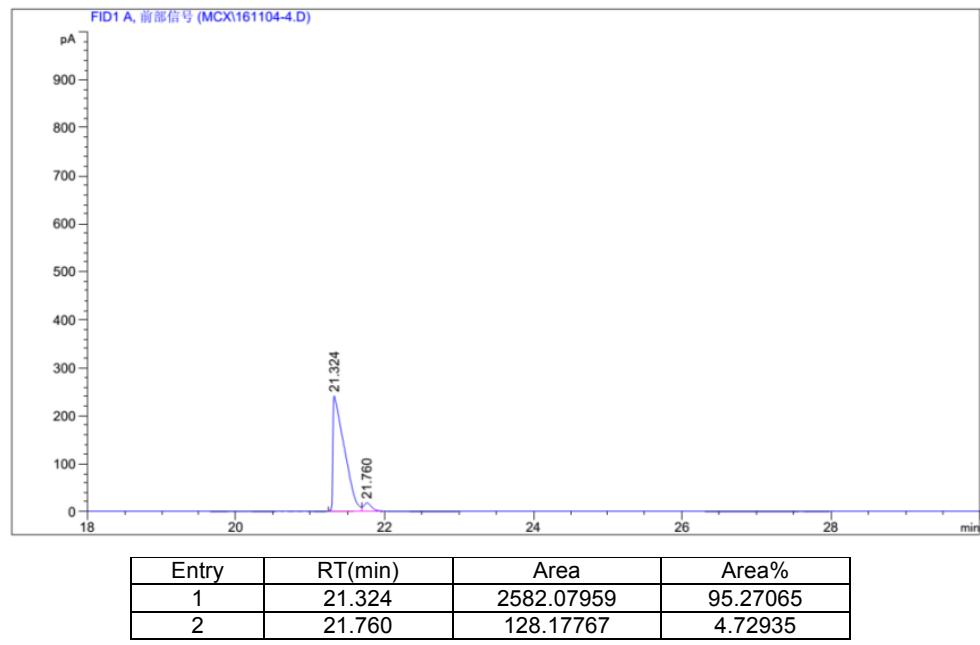
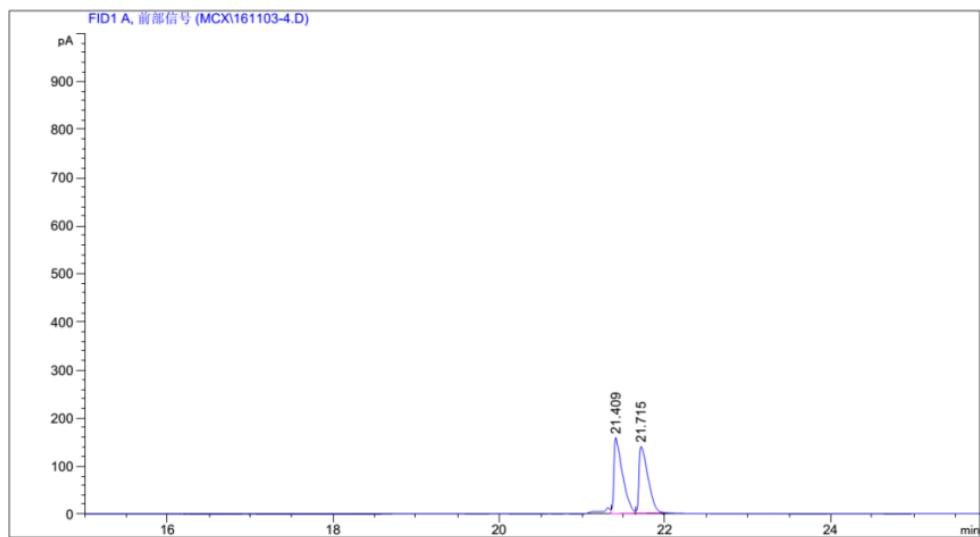


Fig. S25 GC chromatograms of racemic 1-(*m*-fluorophenyl)propan-1-ol (top) as an authentic compound and the unreacted 1-(*m*-fluorophenyl)propan-1-ol (bottom) remained in the OKR of 1-(*m*-fluorophenyl)propan-1-ol (0.50 mmol) by 30% H₂O₂ (0.90 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 80 °C, hold for 5 min, ramp: 4 °C/min to 170 °C, hold for 20 min. t_R(major) = 21.32, t_R(minor) = 21.76.

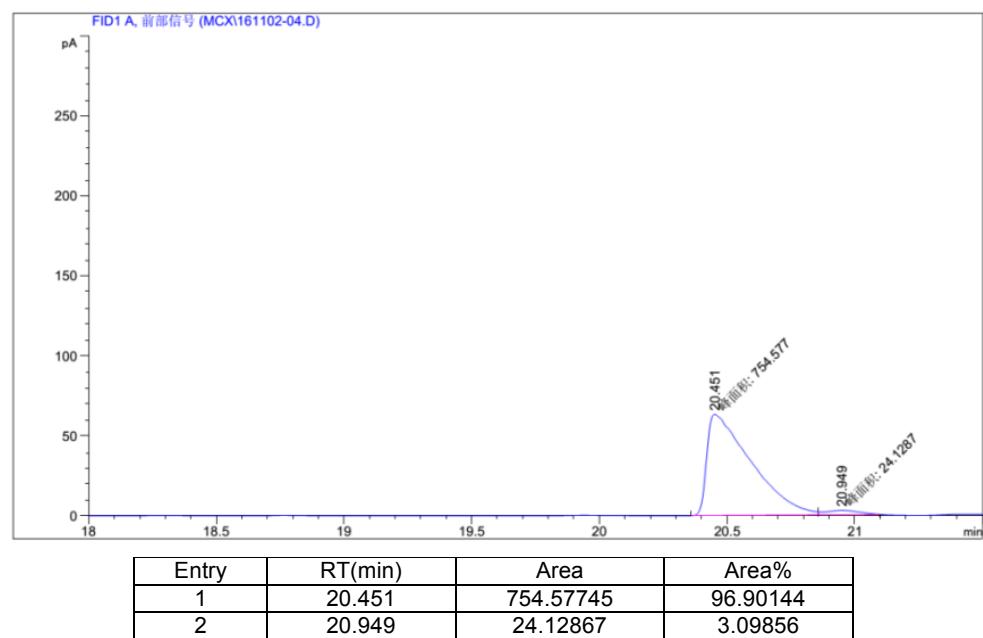
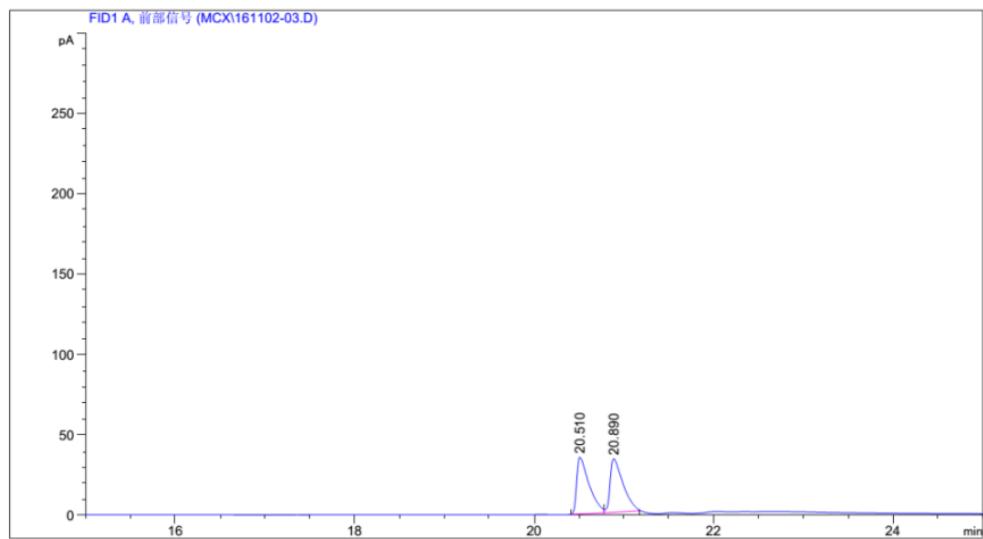


Fig. S26 GC chromatograms of racemic 1-(*o*-fluorophenyl)propan-1-ol (top) as an authentic compound and the unreacted 1-(*o*-fluorophenyl)propan-1-ol (bottom) remained in the OKR of 1-(*o*-fluorophenyl)propan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 100 °C, hold for 5 min, ramp: 2 °C/min to 170 °C, hold for 20 min. t_R(major) = 20.45, t_R(minor) = 20.95.

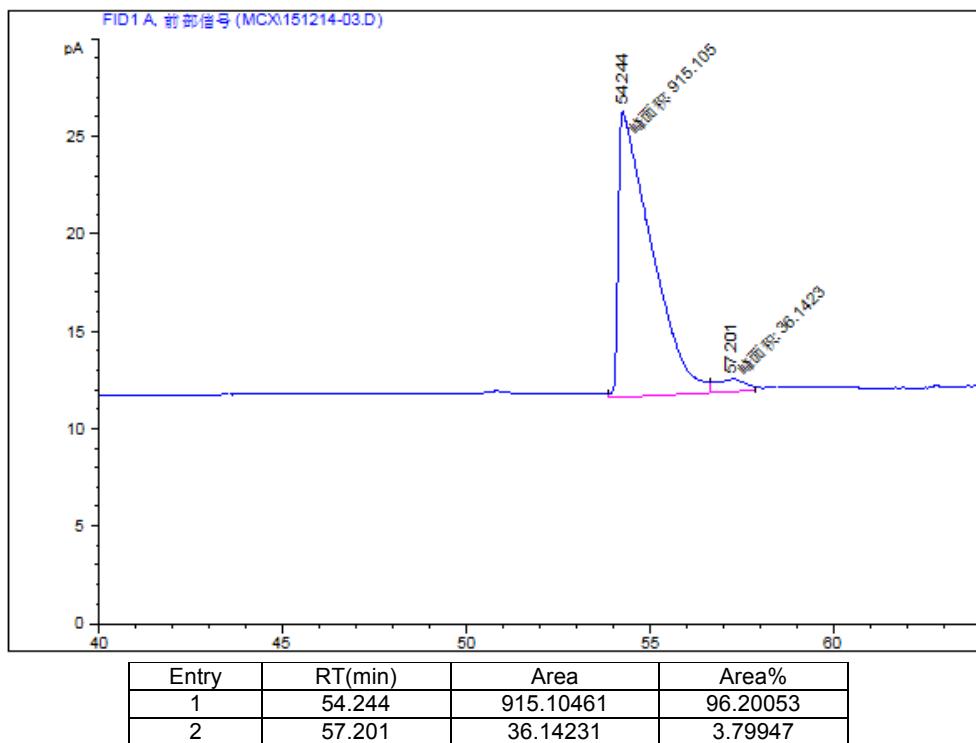
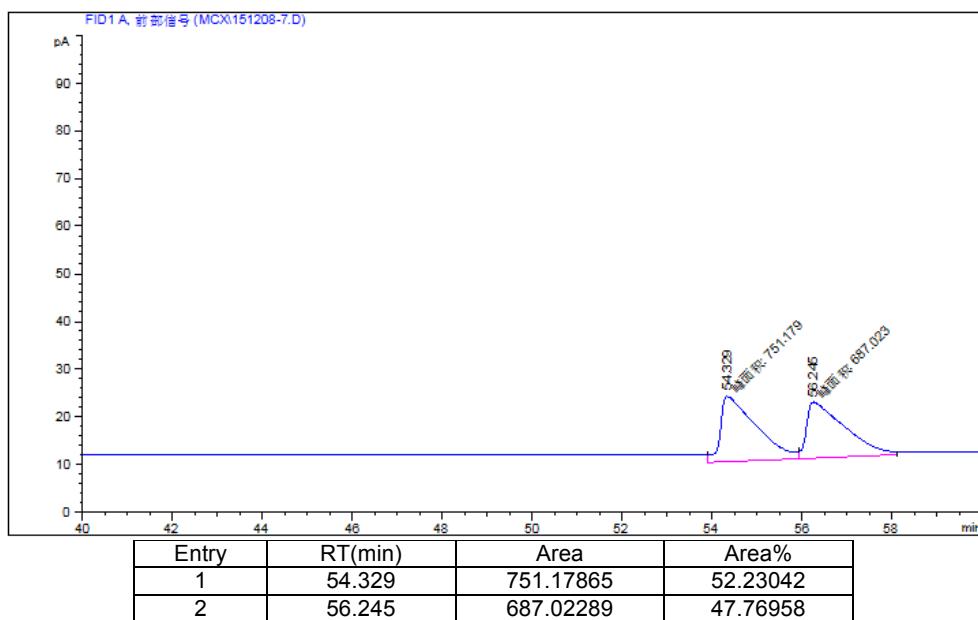


Fig. S27 GC chromatograms of racemic 2-methyl-1-phenylpropan-1-ol (top) as an authentic compound and the unreacted 2-methyl-1-phenylpropan-1-ol (bottom) remained in the OKR of 2-methyl-1-phenylpropan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and 2 (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 140 °C, hold for 25 min, ramp: 10 °C/min to 170 °C, hold for 20 min. t_R(major) = 54.24, t_R(minor) = 57.20.

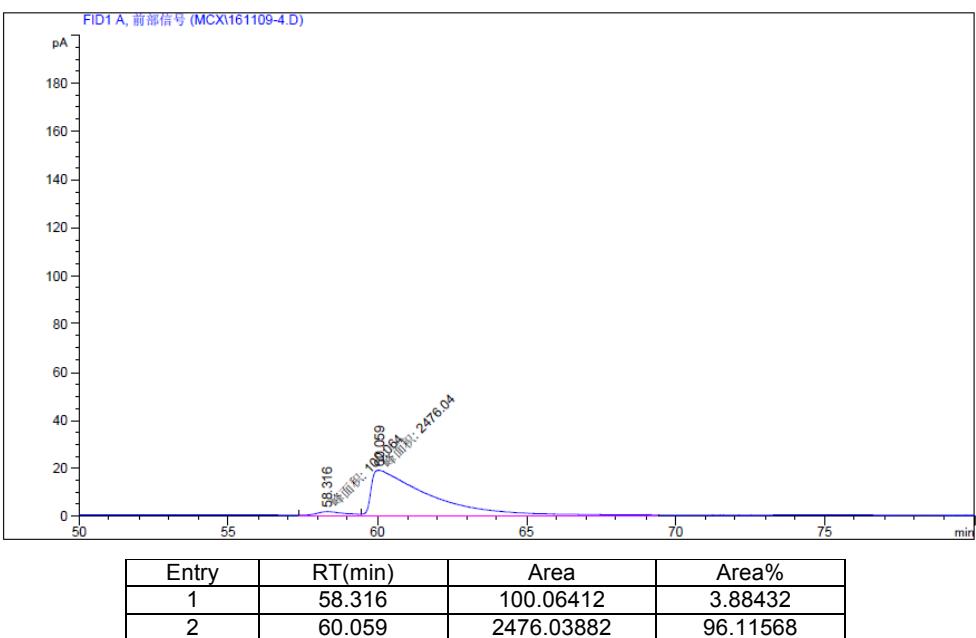
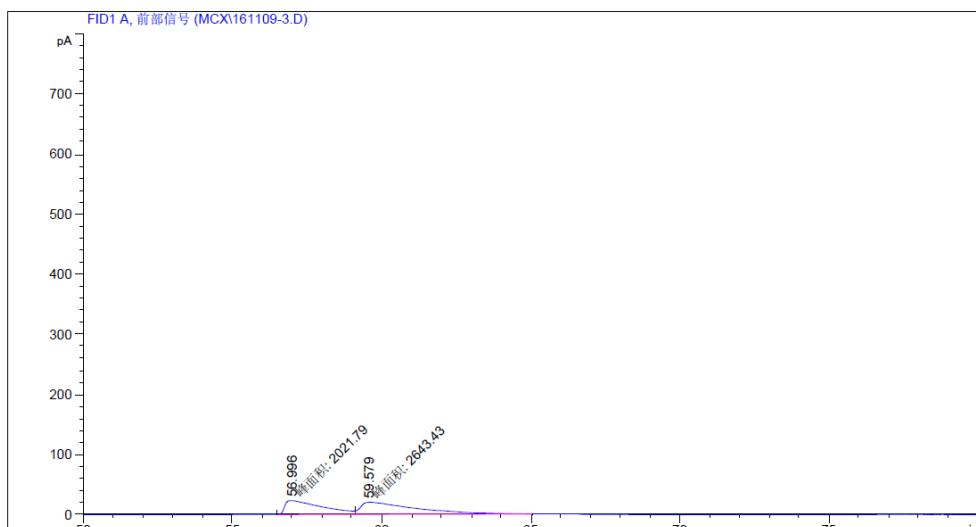
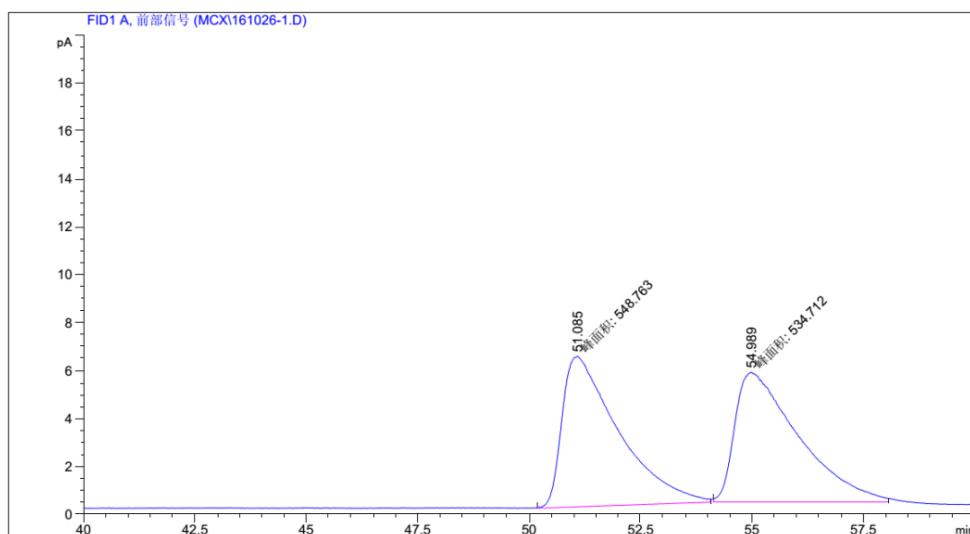
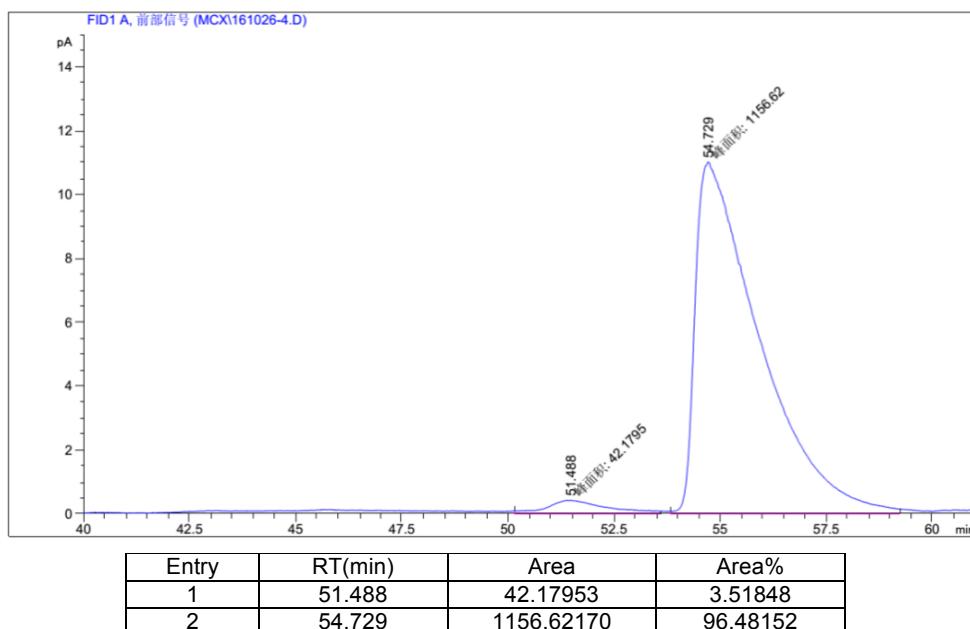


Fig. S28 GC chromatograms of racemic cyclohexyl(phenyl)methanol (top) as an authentic compound and the unreacted cyclohexyl(phenyl)methanol (bottom) remained in the OKR of cyclohexyl(phenyl)methanol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 110 °C, hold for 3 min, ramp: 10 °C/min to 135 °C, hold for 80 min, then ramp: 10 °C/min to 170 °C, hold for 20 min. t_R(major) = 58.32, t_R(minor) = 60.06.



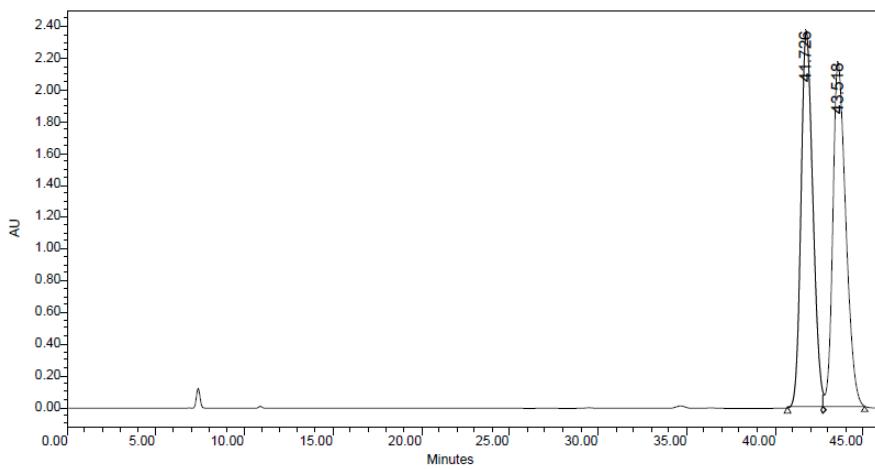
Entry	RT(min)	Area	Area%
1	51.085	548.76337	50.64846
2	54.989	534.71161	49.35154



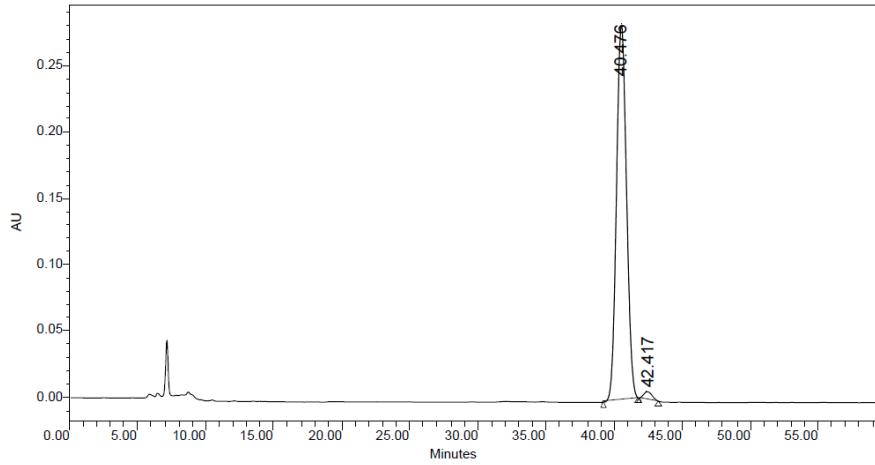
Entry	RT(min)	Area	Area%
1	51.488	42.17953	3.51848
2	54.729	1156.62170	96.48152

Fig. S29 GC chromatograms of racemic 1-phenylheptan-1-ol (top) as an authentic compound and the unreacted 1-phenylheptan-1-ol (bottom) remained in the OKR of 1-phenylheptan-1-ol (0.50 mmol) by 30% H₂O₂ (0.80 equiv. based on the substrate concentration), and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. GC-7890, CP-Chirasil-Dex CB chiral column; Column temperature: 130 °C, hold for 60 min, ramp: 10 °C/min to 170 °C, hold for 20 min. t_R(major) = 51.49, t_R(minor) = 54.73.

Additional Data: HPLC Chromatograms in the OKR of Secondary Alcohols



	RT (min)	Area (AU * sec)	% Area	Height (AU)	% Height
1	41.726	110476089	49.93	2373301	52.19
2	43.518	110770129	50.07	2173780	47.81



	RT (min)	Area (AU * sec)	% Area	Height (AU)	% Height
1	40.476	14488446	98.37	282963	98.04
2	42.417	240423	1.63	5663	1.96

Fig. S30 HPLC chromatograms of racemic 1-(4-nitrophenyl)ethan-1-ol (top) as an authentic compound and the unreacted 1-(4-nitrophenyl)ethan-1-ol (bottom) remained in the oxidative kinetic resolusion (OKR) of 1-(4-nitrophenyl)ethan-1-ol (0.50 mmol) by 30% H₂O₂ (0.90 equiv. based on the substrate concentration) and **2** (0.20 mol%) in the presence of H₂SO₄ (1.0 mol%) in CH₃CN at 0 °C. HPLC-separation conditions: Chiralcel IA, 25 °C, 254 nm, 90/10 hexane/i-PrOH, 1.0 mL/min; t_{major} = 40.48 min, t_{minor} = 42.42 min.

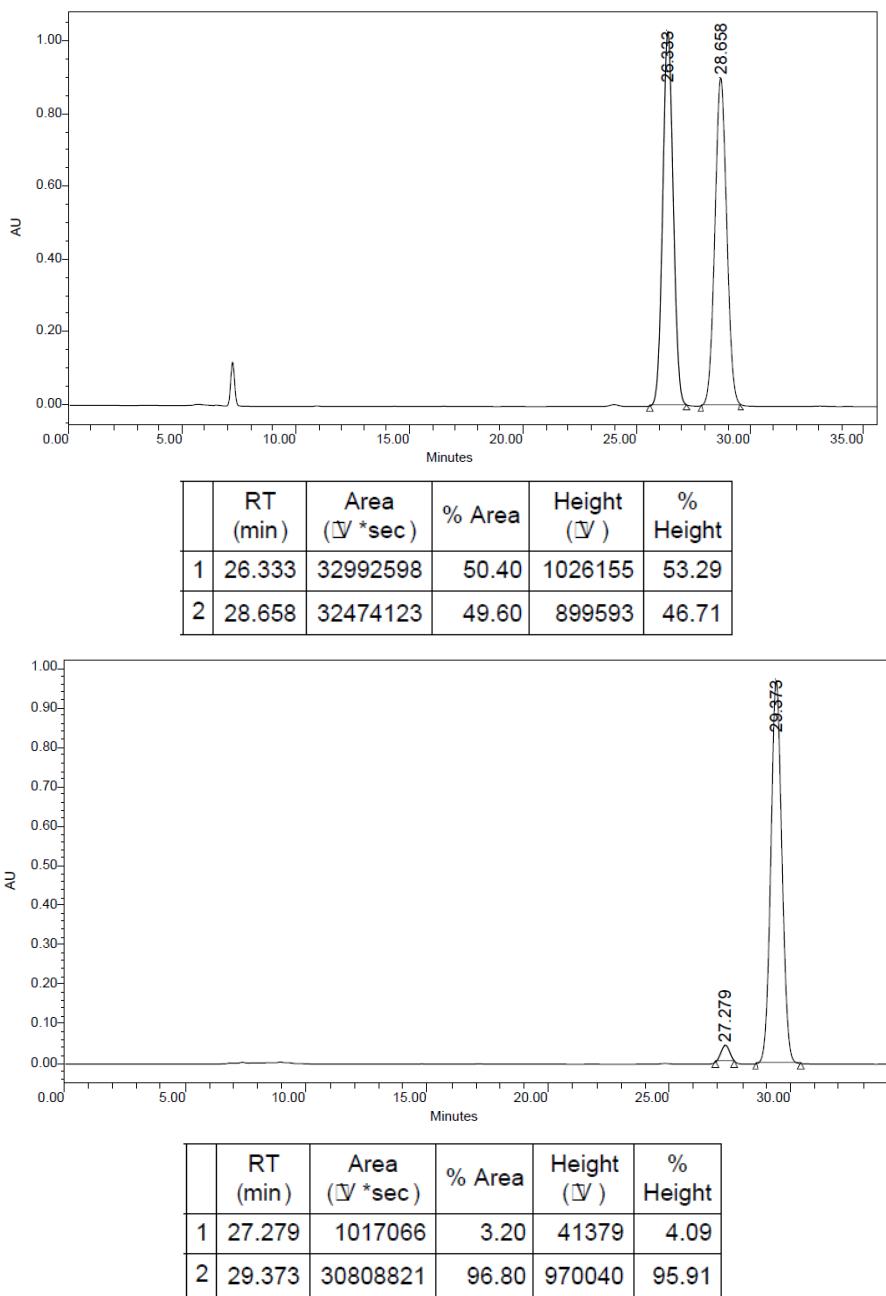


Fig. S31 HPLC chromatograms of racemic 1-([1,1'-biphenyl]-4-yl)ethan-1-ol (top) as an authentic compound and the unreacted 1-([1,1'-biphenyl]-4-yl)ethan-1-ol (bottom) remained in the OKR of 1-([1,1'-biphenyl]-4-yl)ethan-1-ol (0.50 mmol) by 30% H_2O_2 (0.90 equiv. based on the substrate concentration) and **2** (0.20 mol%) in the presence of H_2SO_4 (1.0 mol%) in CH_3CN at 0 °C. HPLC-separation conditions: Chiralcel IA, 25 °C, 254 nm, 90/10 hexane/i-PrOH, 1.0 mL/min; t_{major} = 27.28 min, t_{minor} = 29.37 min.

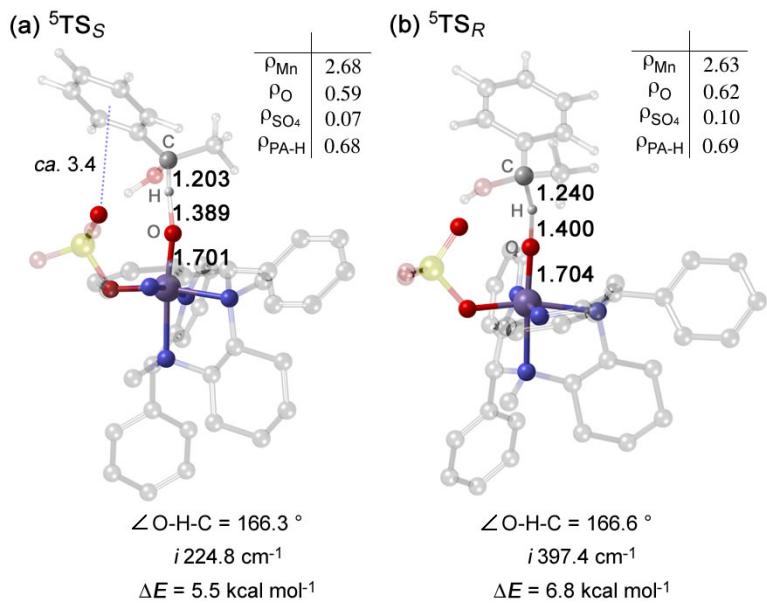


Fig. S32 UB3LYP/LACVP* key geometric information for transition states of $\text{C}_\alpha\text{-H}$ bond activation for both *R*- and *S*-1-phenylethanol mediated by a high-valent Mn-oxo complex ($[\text{LMnO}(\text{SO}_4)]^+$). Length, angle, imaginary frequency are in Å, degree ($^\circ$), and cm^{-1} units. Hydrogen atoms in the ligand are omitted for clarity.

Distance			spin density		
Mn-O ₁	Mn-O ₃	O ₁ -O ₂	Mn	O ₁	O ₂
1.722	1.885	2.253	2.79	0.87	0.42

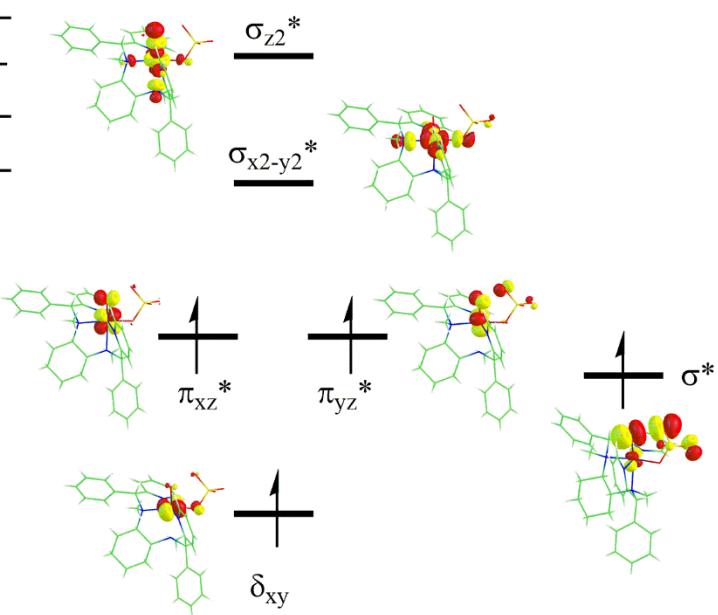
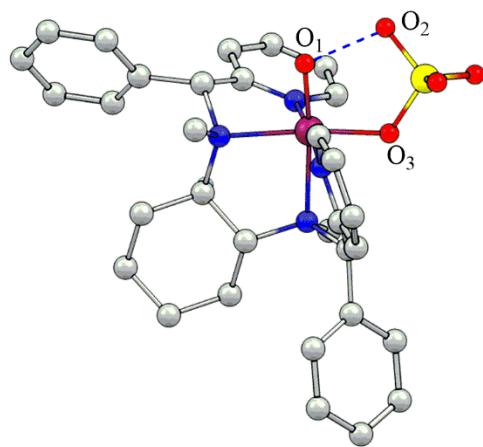


Fig. S33 Geometric and electronic configuration of ⁵I.

1	-1.112743000	0.922764000	-2.025143000
1	0.530815000	0.487628000	-2.538890000
1	-0.017468000	2.175600000	-2.640012000
6	-0.457261000	2.573497000	-0.009812000
1	-1.475939000	2.257086000	-0.256578000
6	-0.357236000	2.441003000	1.497341000
6	-0.627702000	3.477135000	2.388062000
1	-0.841204000	4.468690000	2.011324000
6	-0.629021000	3.220398000	3.758758000
1	-0.838084000	4.021424000	4.460041000
6	-0.368133000	1.929096000	4.212745000
1	-0.363675000	1.684721000	5.268535000
6	-0.123989000	0.932651000	3.277918000
1	0.052670000	-0.094681000	3.566181000
6	-0.286069000	3.985791000	-0.549528000
6	-1.182575000	4.440717000	-1.527991000
1	-1.981662000	3.789459000	-1.872043000
6	-1.069768000	5.727591000	-2.058199000
1	-1.772911000	6.060366000	-2.815755000
6	-0.064176000	6.583124000	-1.606001000
1	0.023341000	7.586110000	-2.013218000
6	0.822824000	6.148899000	-0.616699000
1	1.600565000	6.812363000	-0.250585000
6	0.712367000	4.862269000	-0.090323000
1	1.407011000	4.550329000	0.684064000
6	2.364404000	-1.958858000	0.195766000
1	1.991338000	-2.467766000	1.090962000
6	1.389411000	-2.346974000	-0.900818000
6	1.575539000	-3.433096000	-1.753357000
1	2.500382000	-3.993103000	-1.715116000
6	0.561655000	-3.795035000	-2.639741000
1	0.700955000	-4.638836000	-3.307575000
6	-0.629236000	-3.072220000	-2.645707000
1	-1.450994000	-3.324587000	-3.305665000
6	-0.765027000	-2.012816000	-1.760449000
1	-1.682136000	-1.443961000	-1.689052000
6	3.798858000	-2.426454000	-0.013750000
6	4.472131000	-2.310569000	-1.242063000
1	3.974064000	-1.880460000	-2.105993000
6	5.787355000	-2.755620000	-1.375255000
1	6.291287000	-2.658427000	-2.332280000
6	6.451077000	-3.327415000	-0.285980000
1	7.474586000	-3.674587000	-0.393121000
6	5.788651000	-3.460821000	0.935000000
1	6.290390000	-3.915194000	1.784120000
6	4.470560000	-3.018469000	1.066074000
1	3.957395000	-3.142390000	2.016072000
8	-1.337696000	-2.712588000	3.573540000
16	-1.527943000	-2.287793000	2.176127000
8	-0.098234000	-1.568688000	1.788746000
8	-1.742784000	-3.355207000	1.188672000
8	-2.505939000	-1.138548000	2.063751000
8	-1.687161000	-0.093792000	0.172811000
1	-3.131303000	0.080248000	0.096709000
1	-4.272598000	-0.510287000	1.819779000
6	-4.306211000	0.208174000	0.029427000
8	-4.831247000	0.136099000	1.350921000
6	-4.619849000	1.604570000	-0.511034000
1	-4.171110000	2.365751000	0.136725000
1	-4.235707000	1.742941000	-1.526476000
1	-5.704156000	1.770311000	-0.533030000
6	-4.821192000	-0.909603000	-0.867280000
6	-4.433169000	-0.982257000	-2.217716000
6	-5.663555000	-1.911874000	-0.365666000
6	-4.873526000	-2.022829000	-3.037867000
1	-3.789951000	-0.213535000	-2.639522000
6	-6.107755000	-2.952977000	-1.184738000
1	-5.976831000	-1.860080000	0.671710000
6	-5.713784000	-3.015209000	-2.523594000
1	-4.564526000	-2.055458000	-4.079508000
1	-6.765311000	-3.715927000	-0.775551000
1	-6.058723000	-3.825176000	-3.160499000