

## Supporting Information

### **Photoelectrochemical H<sub>2</sub> evolution using TiO<sub>2</sub>-coated CaFe<sub>2</sub>O<sub>4</sub> without an external applied bias under visible light irradiation at 470 nm based on device modeling**

Shintaro Ida,<sup>a, c, \*</sup> Kara Kearney,<sup>b, c</sup> Takamitsu Futagami,<sup>a</sup> Hidehisa Hagiwara,<sup>a, c</sup> Takaaki Sakai,<sup>a</sup> Motonori Watanabe,<sup>c</sup> Angus Rockett<sup>c, d</sup> and Tatsumi Ishihara<sup>a, c</sup>

<sup>a</sup>Department of Applied Chemistry, Faculty of Engineering, Kyushu University, 744 Motooka, Nishi-ku, Fukuoka 819-0395, Japan

<sup>b</sup>Department of Materials Science and Engineering, University of Illinois, 1304 West Green Street, Urbana, IL, 61801, USA

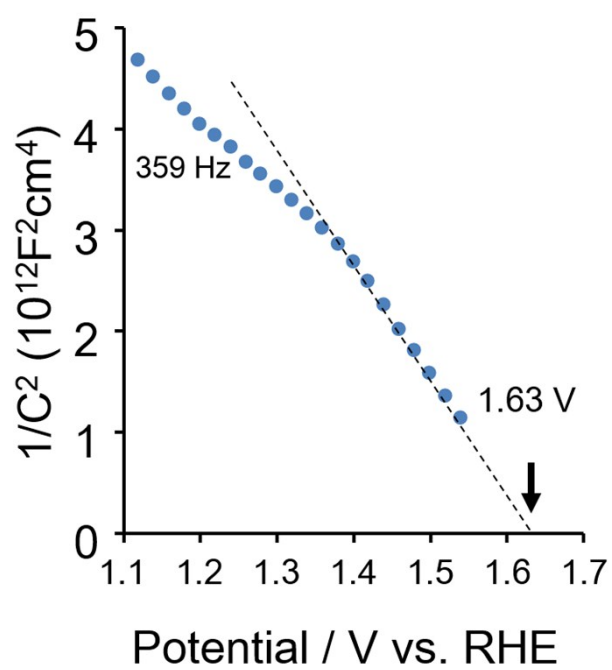
<sup>c</sup>International Institute for Carbon Neutral Energy Research (I2CNER), Kyushu University, 744 Motooka, Nishi-ku, Fukuoka 819-0395, Japan

<sup>d</sup>Department of Metallurgy and Materials Engineering, Colorado School of Mines, 201 Hill Hall, 1500 Illinois Street, Golden, CO, 80401, USA

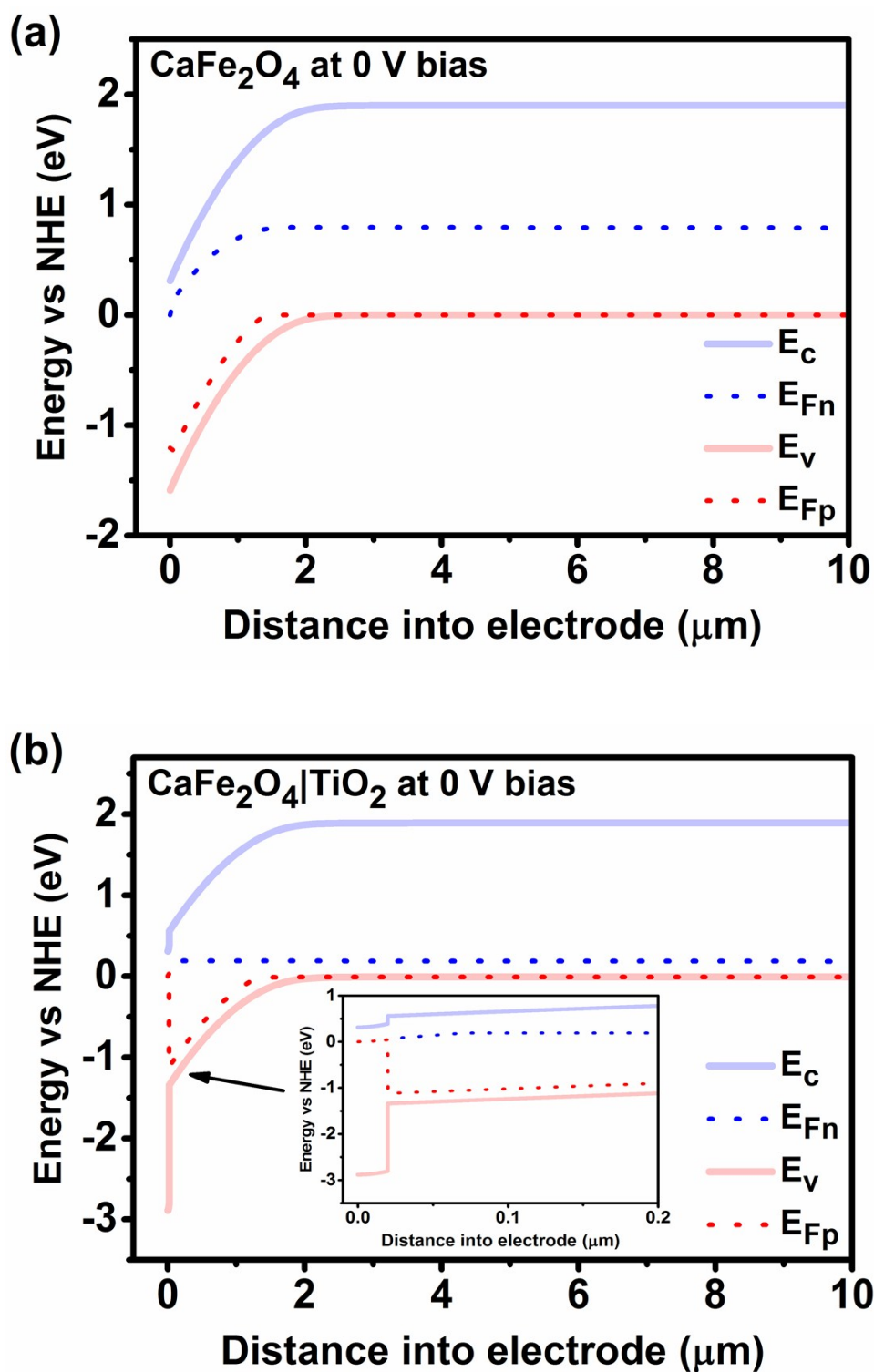
\*Author to whom correspondence should be addressed;

E-Mail: s-ida@cstf.kyushu-u.ac.jp (S.I.);

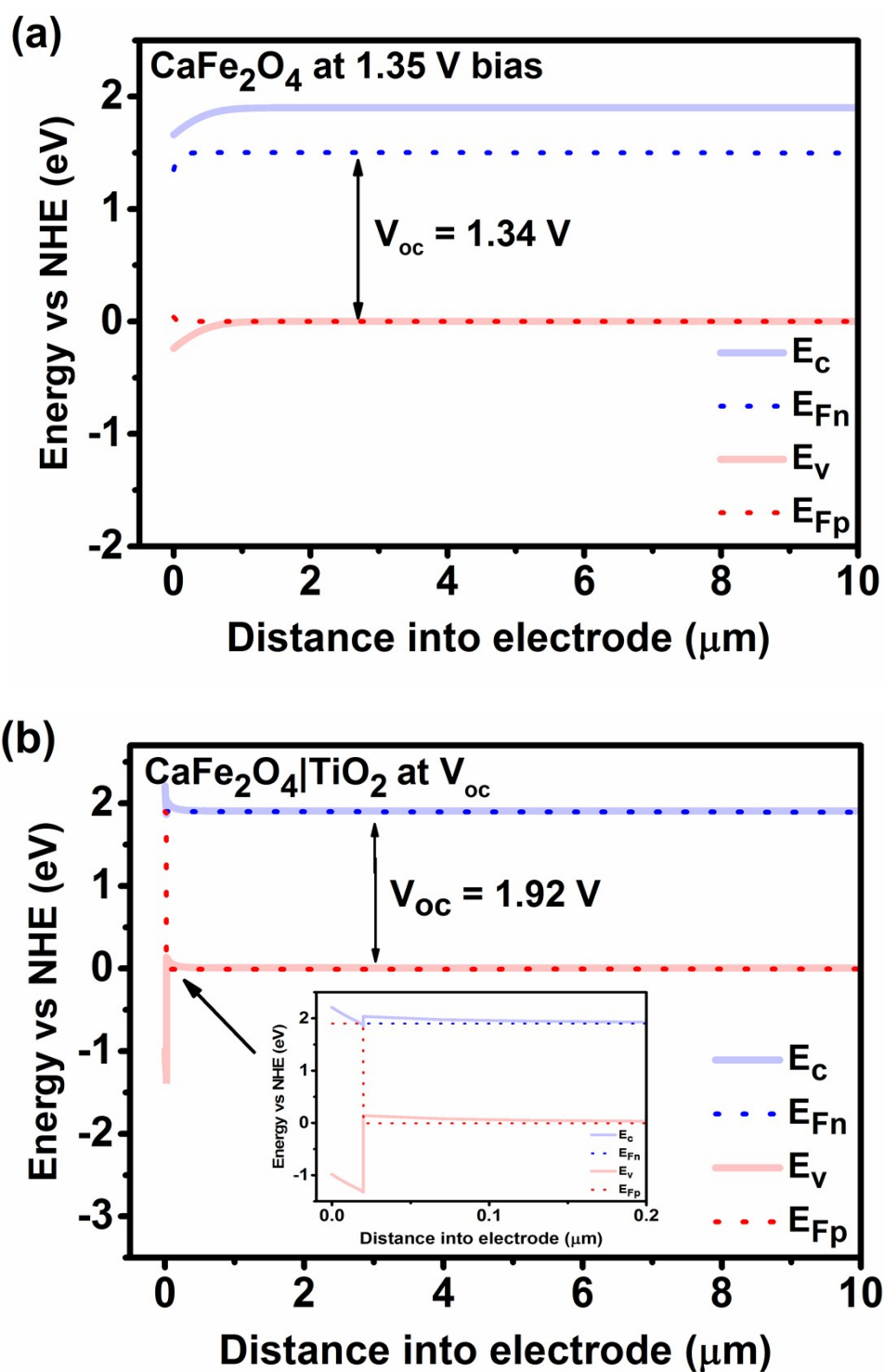
Tel.: +81-92-802-2869; Fax: +81-92-802-2871.



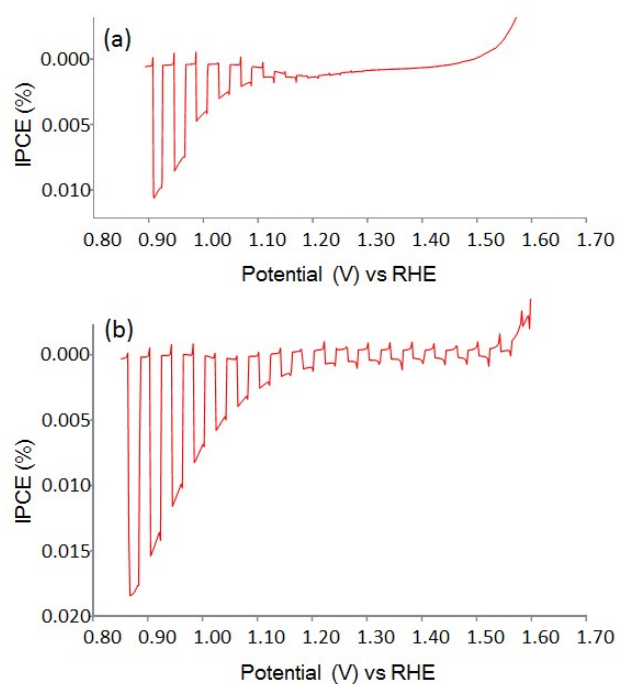
**Figure S1** Mott-Schottky plot of a bare CFO electrode in 0.1 M NaOH solution.



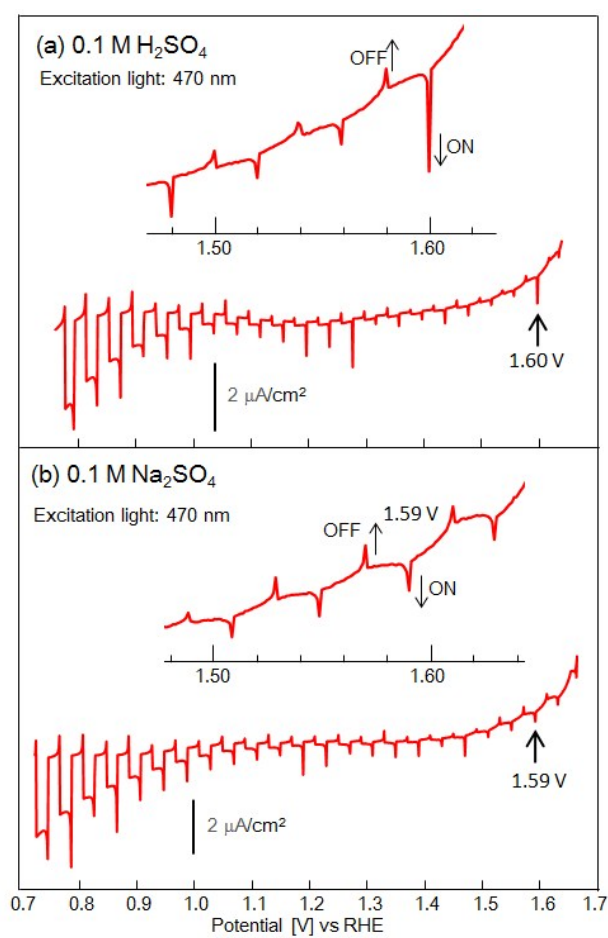
**Figure S2.** Band structures of CaFe<sub>2</sub>O<sub>4</sub> and TiO<sub>2</sub>-coated CaFe<sub>2</sub>O<sub>4</sub> electrodes at 0 V vs NHE under 470 nm excitation.



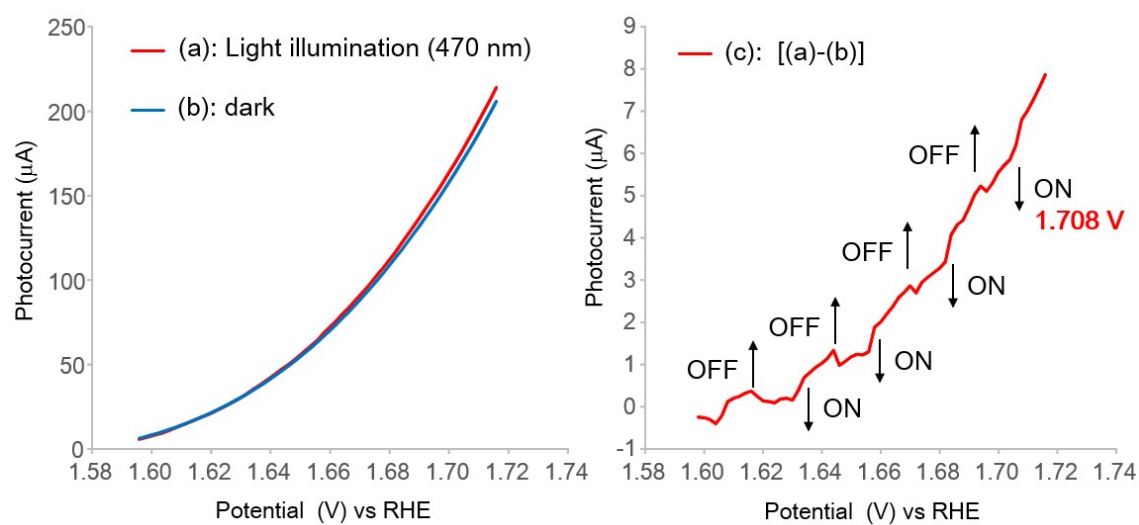
**Figure S3.** Band structures of CaFe<sub>2</sub>O<sub>4</sub> and TiO<sub>2</sub>-coated CaFe<sub>2</sub>O<sub>4</sub> electrodes at the onset potential under 470 nm excitation.



**Figure S4.** IPCE-potential curves of (a) CFO and (b) TiO<sub>2</sub>-coated CFO electrodes in 0.1 M NaOH solution under chopped 470 nm light illumination.



**Figure S5.** Current-potential curve of  $\text{TiO}_2$ -coated CFO electrode under chopped light illumination of 470 nm in (a)  $0.1 \text{ M H}_2\text{SO}_4$  and (b)  $0.1 \text{ M Na}_2\text{SO}_4$ .



**Figure S6** Current-potential curves of  $\text{TiO}_2$ -coated CFO electrodes in 0.1 M NaOH solution under chopped light illumination (470 nm); (a) light illumination, (b) dark condition and (c) difference spectrum [(a)-(b)].

**Table S1.** Electronic parameters used as inputs for CaFe<sub>2</sub>O<sub>4</sub> and CaFe<sub>2</sub>O<sub>4</sub>/TiO<sub>2</sub> photoelectrodes.

Electronic Parameter	Units	CaFe <sub>2</sub> O <sub>4</sub> (50 $\mu$ m)	TiO <sub>2</sub> (20nm)
Permittivity		287	50
Band Gap ( $E_g$ )	eV	1.9	3.2
Electron Affinity	eV	4.12	4.3
Conduction Band Density of States ( $N_c$ )	cm <sup>-3</sup>	10 <sup>16</sup>	10 <sup>19</sup>
Valence Band Density of States ( $N_v$ )	cm <sup>-3</sup>	10 <sup>16</sup>	10 <sup>19</sup>
Electron Mobility ( $\mu_n$ )	cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup>	0.01	0.001
Hole Mobility ( $\mu_p$ )	cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup>	0.05	0.0001
Donor Dopant Density ( $N_d$ )	cm <sup>-3</sup>	0	10 <sup>18</sup>
Acceptor Dopant Density ( $N_a$ )	cm <sup>-3</sup>	10 <sup>16</sup>	0

**Table S1-2.** Defect parameters used as inputs for TiO<sub>2</sub> layer.

Defect Parameter	Units	Value
Type		Donor-like <sup>1</sup>
Density	cm <sup>-3</sup>	10 <sup>18</sup>
Energy Level	eV	1.6
Capture Cross Section for Electrons	cm <sup>2</sup>	5e-12
Capture Cross Section for Holes	cm <sup>2</sup>	5e-12

<sup>1</sup>Discrete energetic distribution