

Supporting Information

Repurposing Paper By-Product Lignosulfonate as Sulfur Donor/Acceptor for High Performance Lithium-Sulfur Batteries

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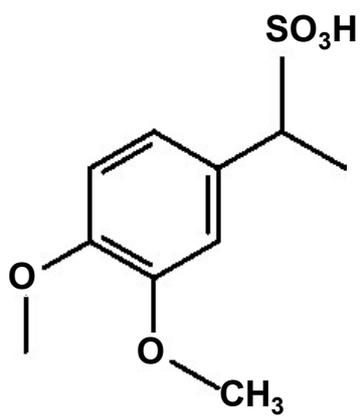


Fig. S1. Partial chemical structure of lignosulfonate.



Fig. S2. Photo of lignosulfonate “brown liquor”.

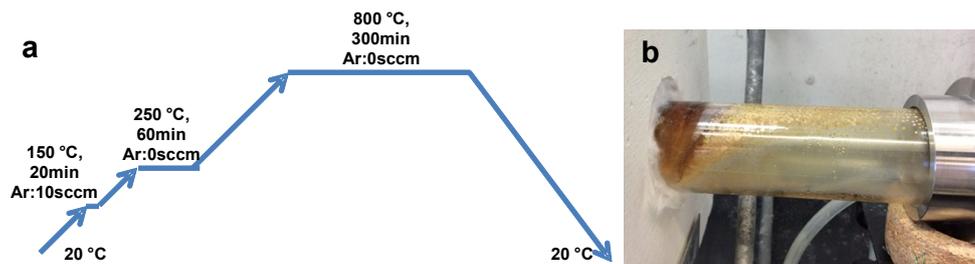


Fig. S3. (a) Pyrolytic reaction flow diagram of lignosulfonate. (b) Photo of reduced S condensed on the reaction quartz tube during the lignosulfonate pyrolysis process.

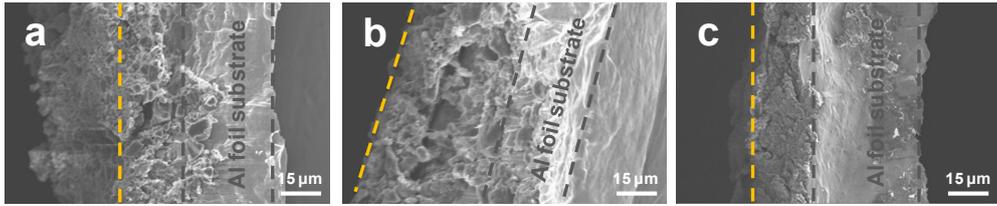


Fig. S4. Cross-sectional SEM images of (a) Act PyLs-Cap S+S electrode (56%S), (b) Act PyLs-Cap S electrode (34%S) and (c) pure S electrode (56%S).

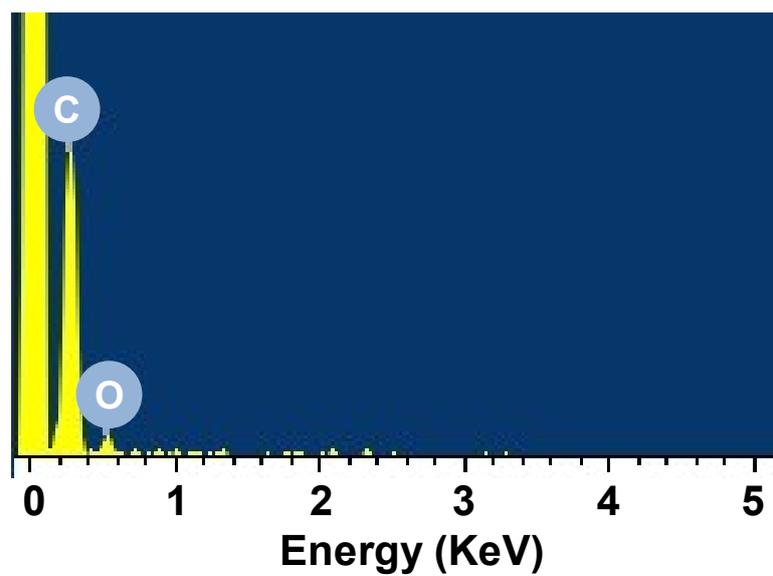


Fig. S5. EDX result of Act PyLs.

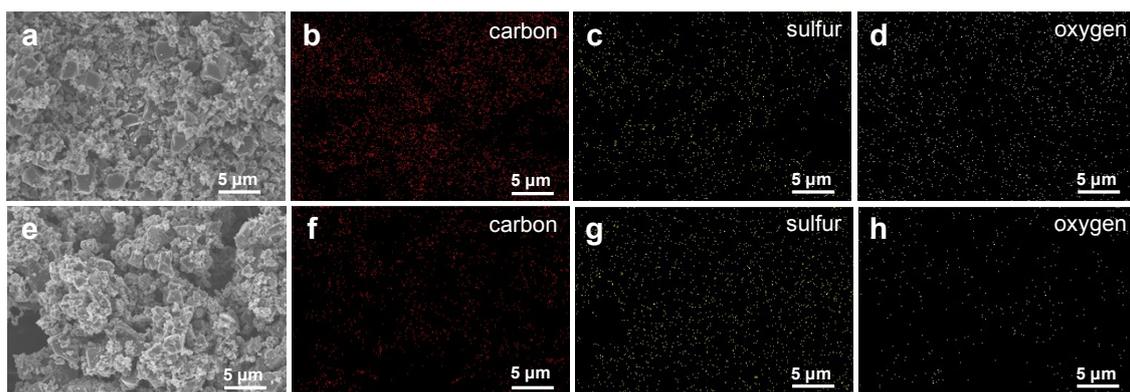


Fig. S6. (a) SEM image of Act PyLs-Cap S composite. (b-d) EDX element mappings of Act PyLs-Cap S composite. (e) SEM image of Act PyLs-Cap S+S composite. (f-h) EDX element mappings of Act PyLs-Cap S+S composite.

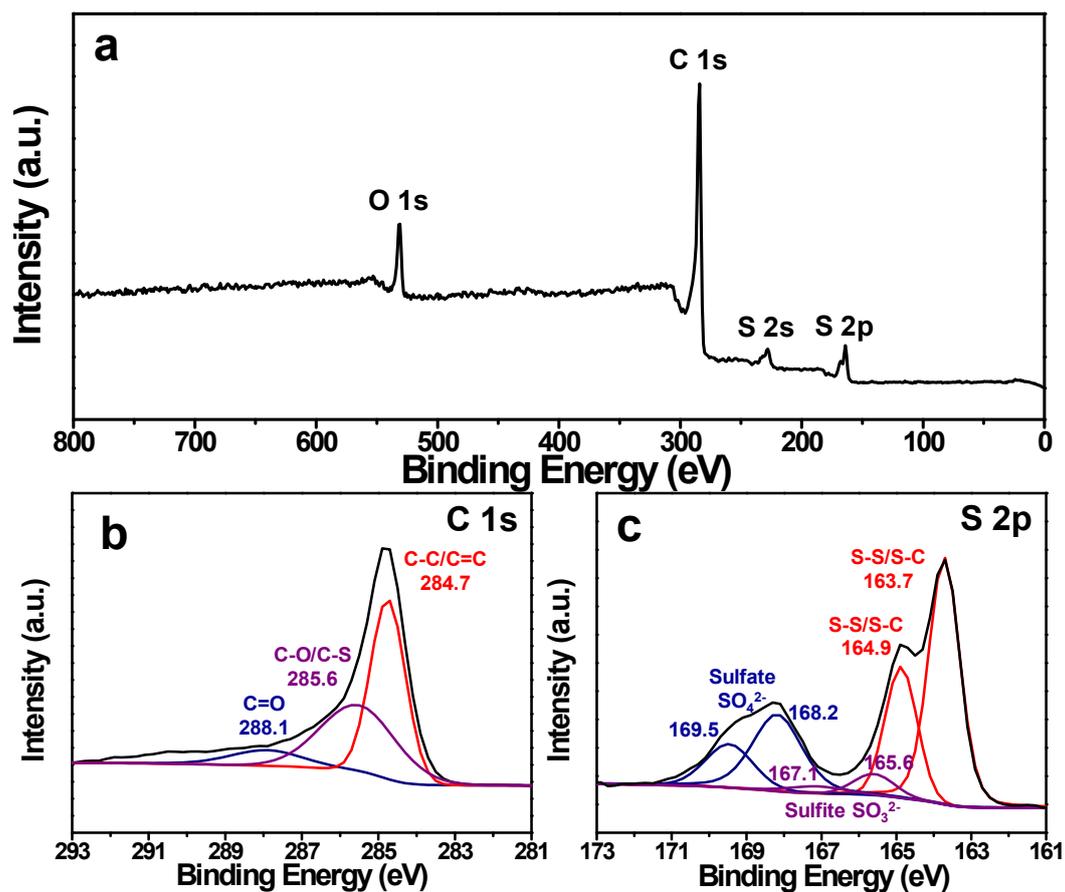


Fig. S7. (a) XPS survey spectra of Act PyLs-Cap S. High-resolution (b) C 1s and (c) S 2p XPS spectra of Act PyLs-Cap S.

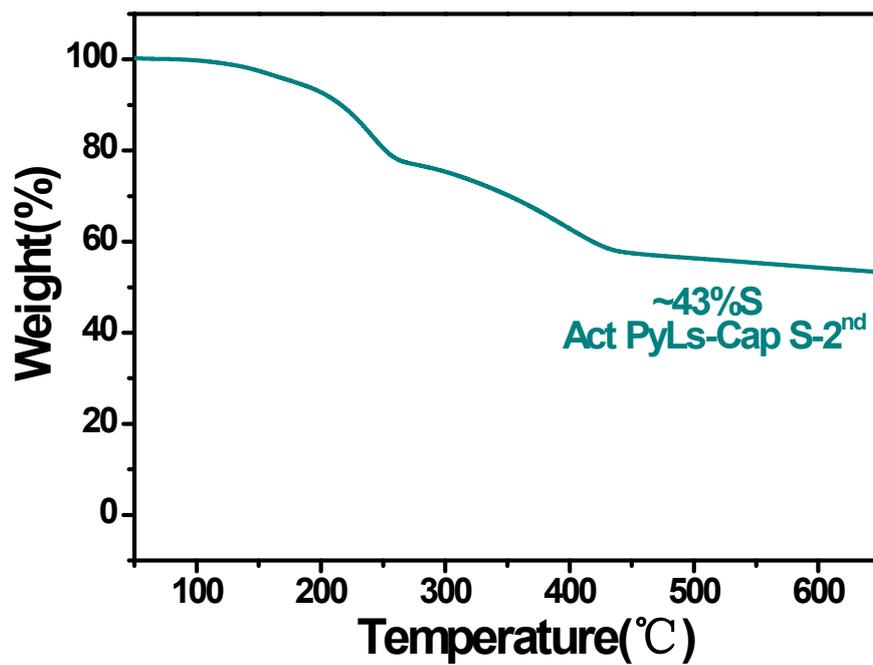


Fig. S8. TGA result of sulfur loading in the sample of Act PyLs-CapS-2nd from second capture process.

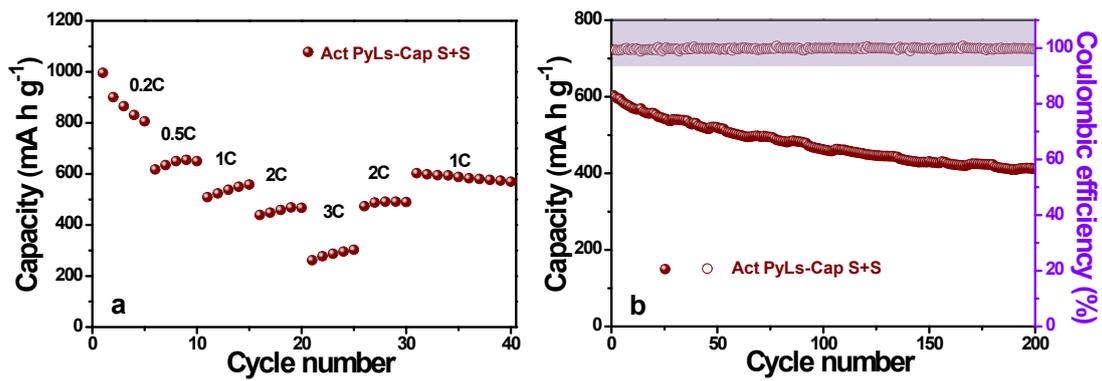


Fig. S9. (a) Rate capability and (b) Cycling performance and Coulombic efficiency of the Li-S batteries with Act PyLs-Cap S+S electrode at high sulfur loading (~3.5 mg/cm²).