

Electronic supplementary information (ESI)

Visible-light/temperature dual-responsive hydrogel constructed by α -
cyclodextrin and an azobenzene linked surfactant

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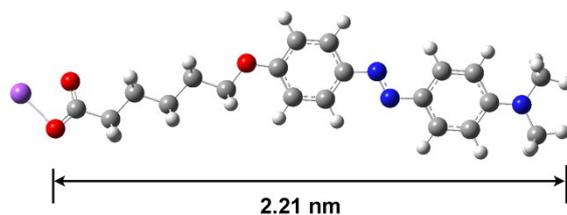


Fig. S1 B3LYP/6-31G (d, p)-optimized structure of a *trans*-DAH.

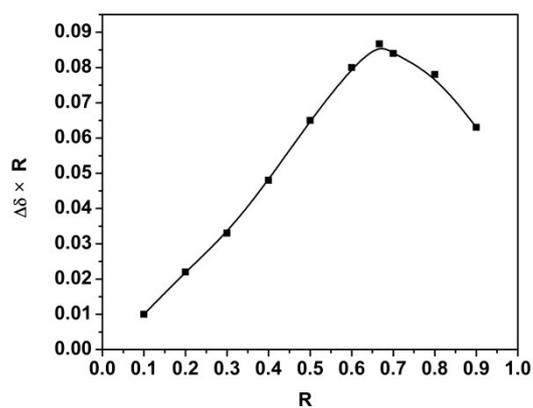


Fig. S2 ^1H NMR Job's plot corresponding to the chemical shift of H-5 of $\alpha\text{-CD}$ in D_2O . $[\text{DAH}] + [\alpha\text{-CD}] = 15 \text{ mM}$.

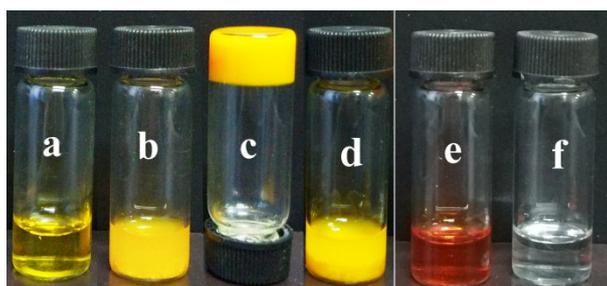


Fig. S3 Representative sample appearance images of DAH@2 $\alpha\text{-CD}$ at different concentrations: (a) 5 mM, (b) 15 mM, (c) 30 mM, (d) 60 mM, and the individual component of DAH (30 mM, e) and $\alpha\text{-CD}$ (60 mM, f).

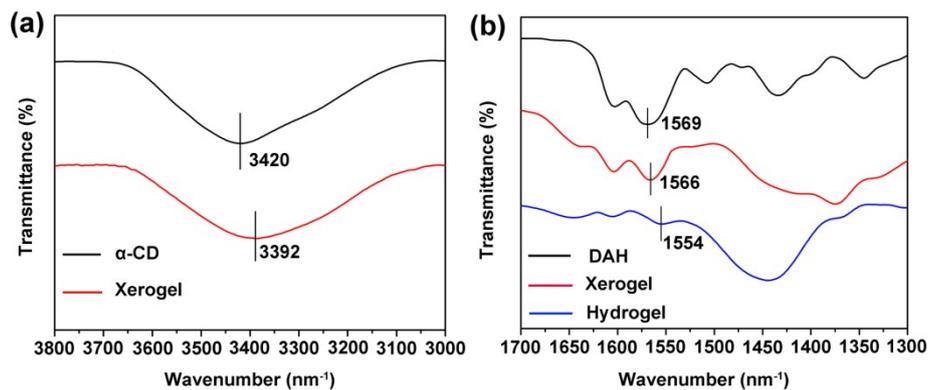


Fig. S4. FTIR spectra of pure α -CD, DAH, xerogel and hydrogel at 30 mM in D₂O.



Fig. S5 Visual appearance of gel-sol transition caused by addition of urea.

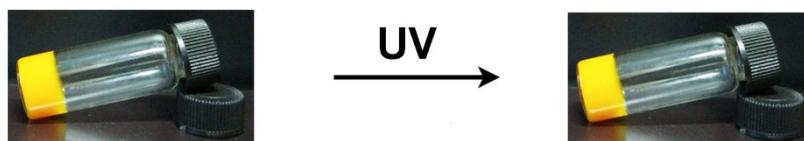


Fig. S6 Visual appearance of hydrogel upon UV light (350 or 365 nm) irradiation.

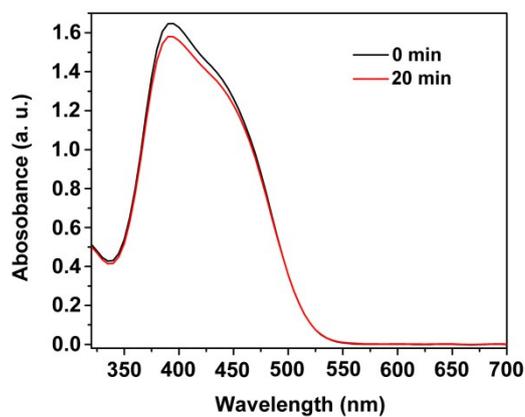


Fig. S7 UV-Vis spectra of 8 mM DAH in aqueous solution before and after UV-365 nm light irradiation.

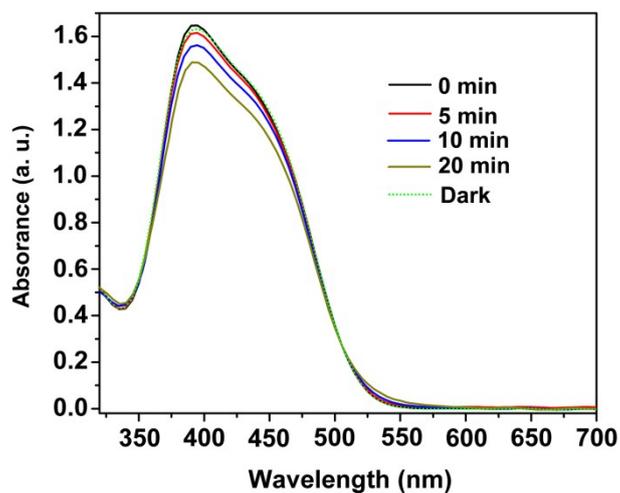


Fig. S8 UV-Vis spectra of 8.0 mM DAH at different times of visible light (420 nm) irradiation and stirring in the dark.

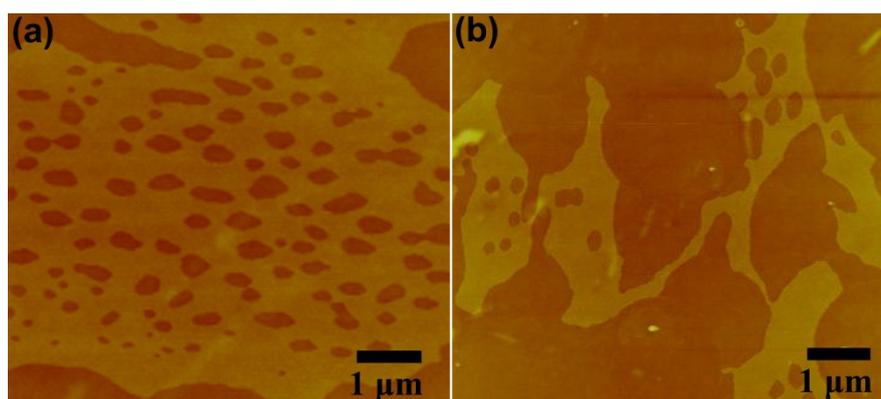


Fig. S9 Tapping-mode AFM images of a 30 mM sample at 35 (a) and 50 °C (b).

AFM measurements were conducted by a Nanoscope IIIA from Digital Instruments in tapping mode under ambient conditions. For the sample preparation, 20 μL of aggregate solution was placed on a freshly cleaved mica surface and the excess was removed by absorption onto filter paper. The resultant substrates were dried under vacuum.

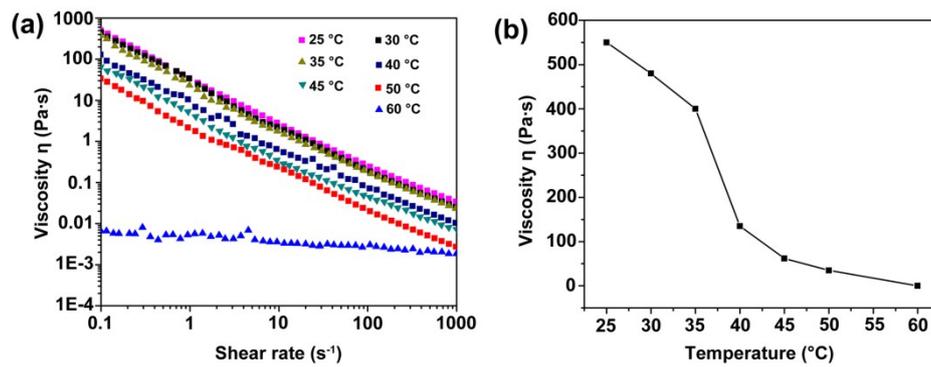


Fig. S10. Viscosity-shear rate curves (a) and viscosity changes (b) for a 30 mM sample at different temperatures.