

Supporting information for:

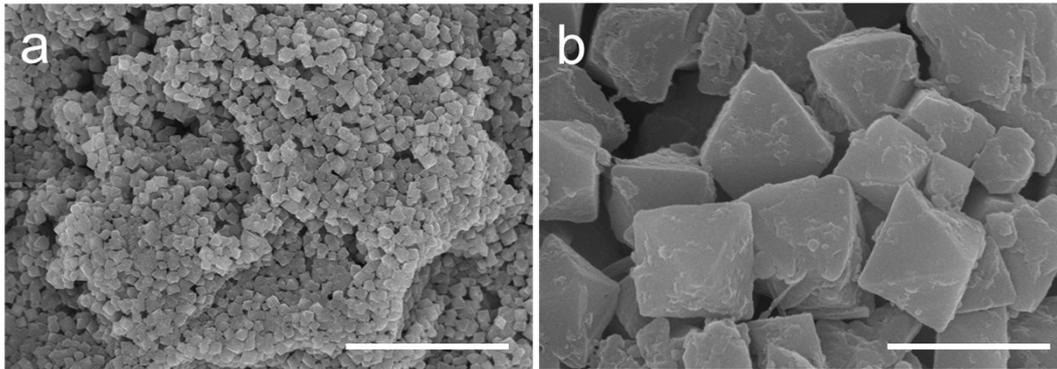
**Interfacial assembly and hydrolysis for synthesizing TiO<sub>2</sub>/metal-  
organic framework composite**

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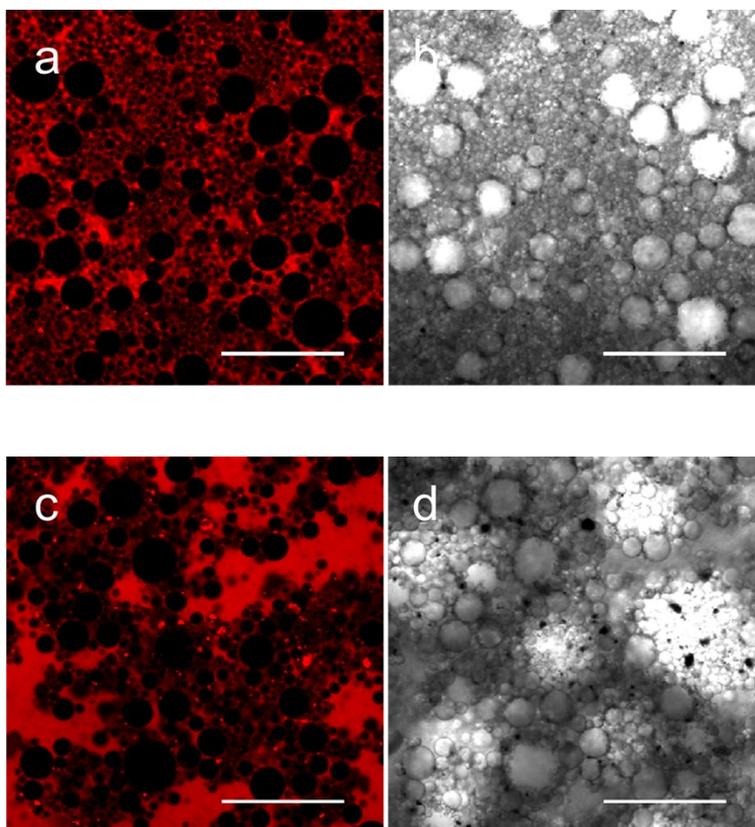
Beijing National Laboratory for Molecular Sciences, CAS Key Laboratory of Colloid,  
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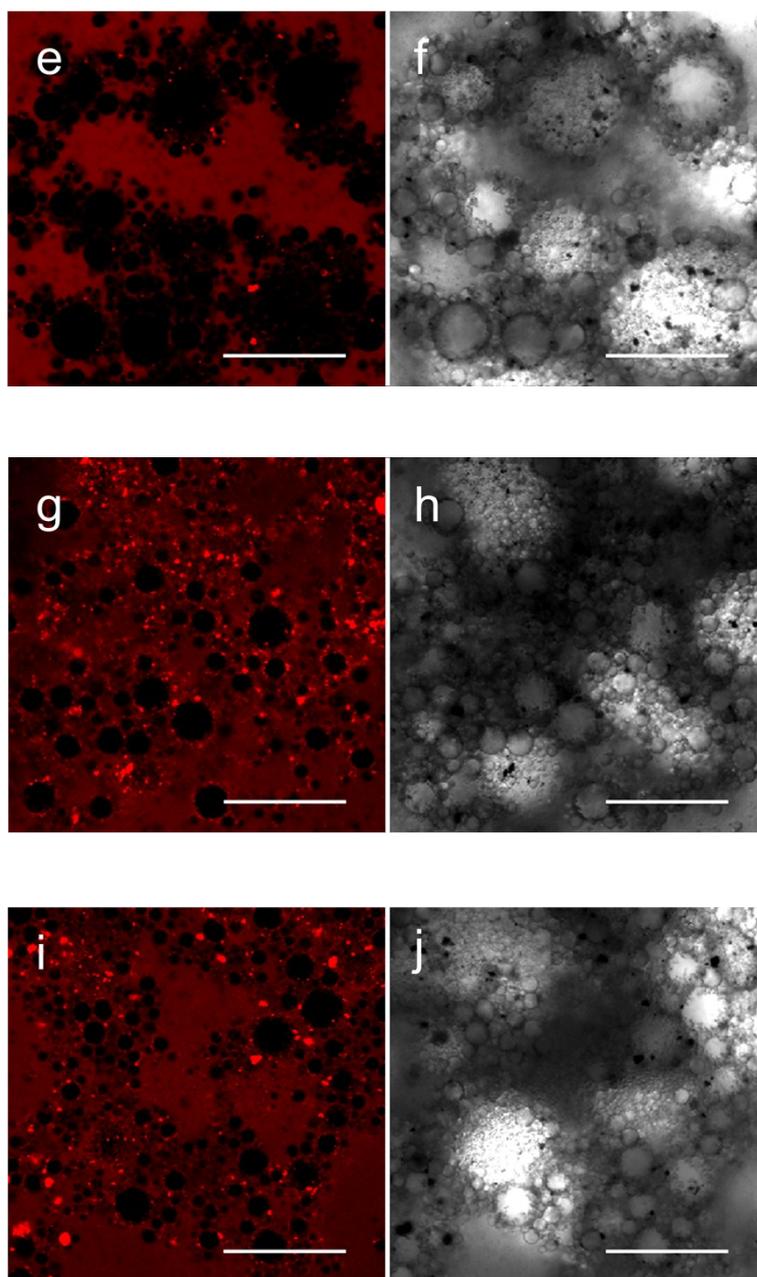
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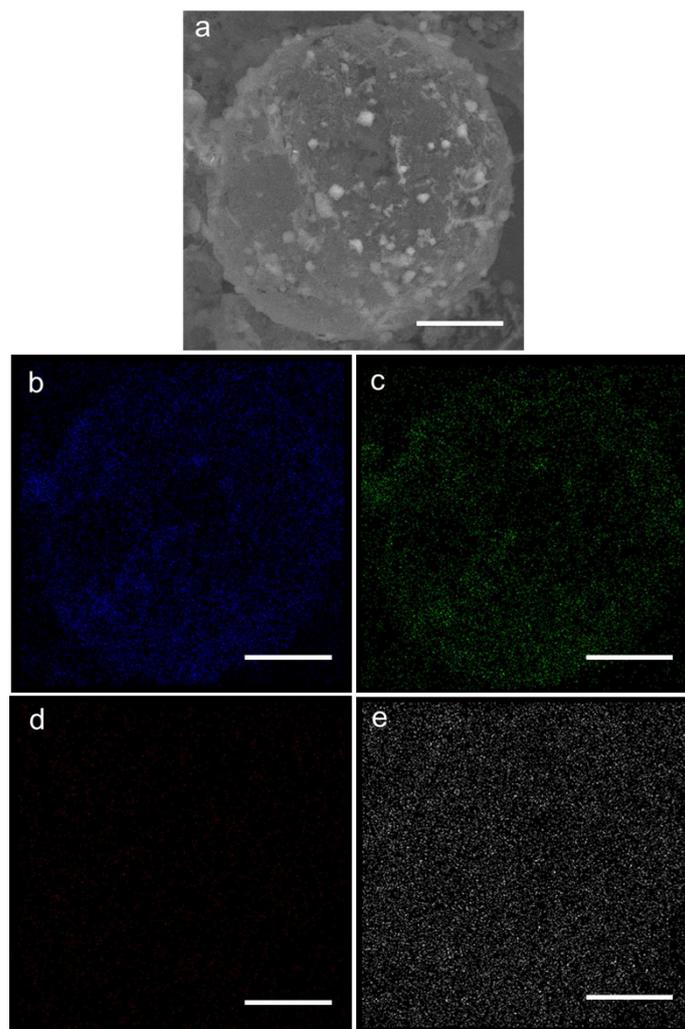


**Figure S1.** SEM images of the pristine UiO-67. Scale bars, 20  $\mu\text{m}$  in a, 2  $\mu\text{m}$  in b.

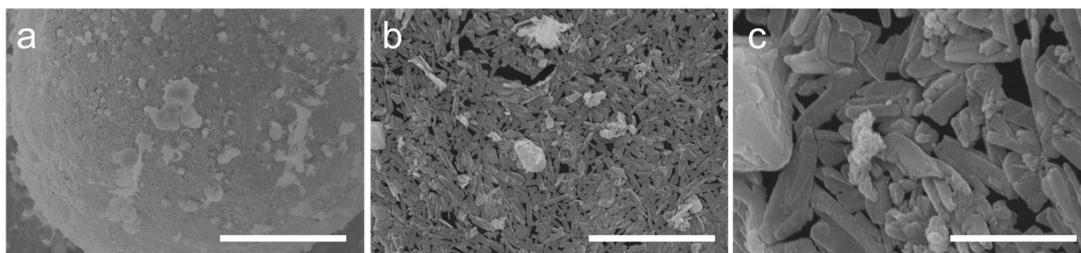




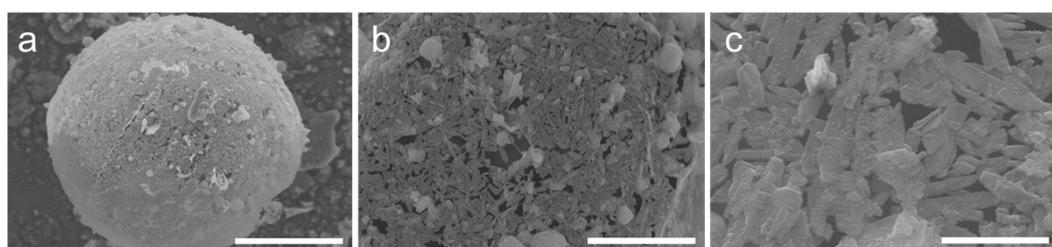
**Figure S2.** CLSM images of the emulsions with UiO-67 concentration of  $10 \text{ mg mL}^{-1}$  and TBT concentration of 0 (a, b),  $2.5 \text{ } \mu\text{L mL}^{-1}$  (c, d),  $3.8 \text{ } \mu\text{L mL}^{-1}$  (e, f),  $6.3 \text{ } \mu\text{L mL}^{-1}$  (g, h),  $12.5 \text{ } \mu\text{L mL}^{-1}$  (i, j), respectively. Scale bars:  $100 \text{ } \mu\text{m}$ .



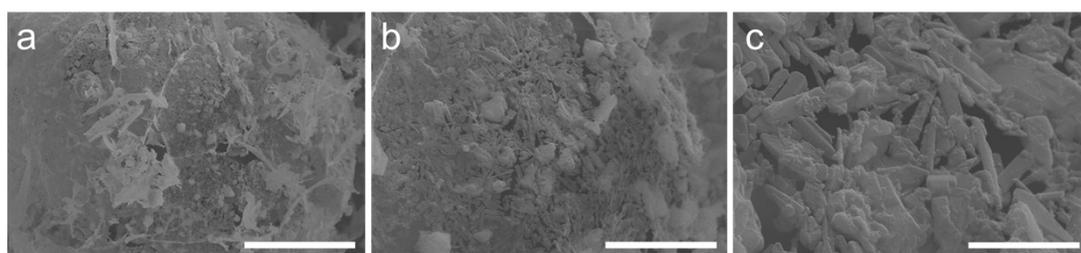
**Figure S3.** Elemental analysis of  $\text{TiO}_2/\text{UiO-67}$  composite obtained from the emulsion with UiO-67 concentration of  $10 \text{ mg mL}^{-1}$  and TBT concentration of  $2.5 \text{ }\mu\text{L mL}^{-1}$ . SEM image (a) and corresponding elemental mappings of C (b), O (c), Ti (d) and Zr (e). There is a uniform distribution of the Ti and Zr atoms throughout the composite structure. The atomic percent of Ti and Zr atoms is 0.98% and 2.23% respectively. The relative content of Ti and Zr atoms is 0.439. Scale bars:  $10 \text{ }\mu\text{m}$ .



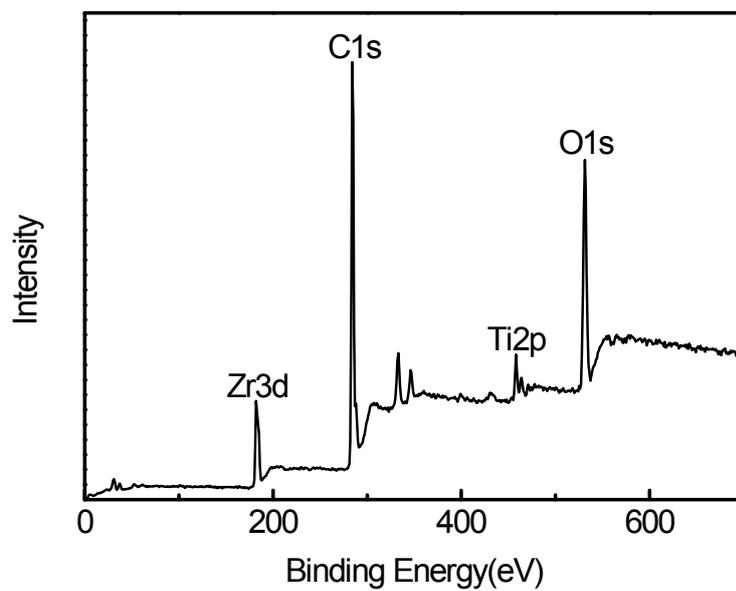
**Figure S4.** SEM images of TiO<sub>2</sub>/UiO-67 composite obtained from the emulsion with UiO-67 concentration of 10 mg mL<sup>-1</sup> and TBT concentration of 2.5 μL mL<sup>-1</sup>. Scale bars, 20 μm in a, 5 μm in b, 1 μm in c.



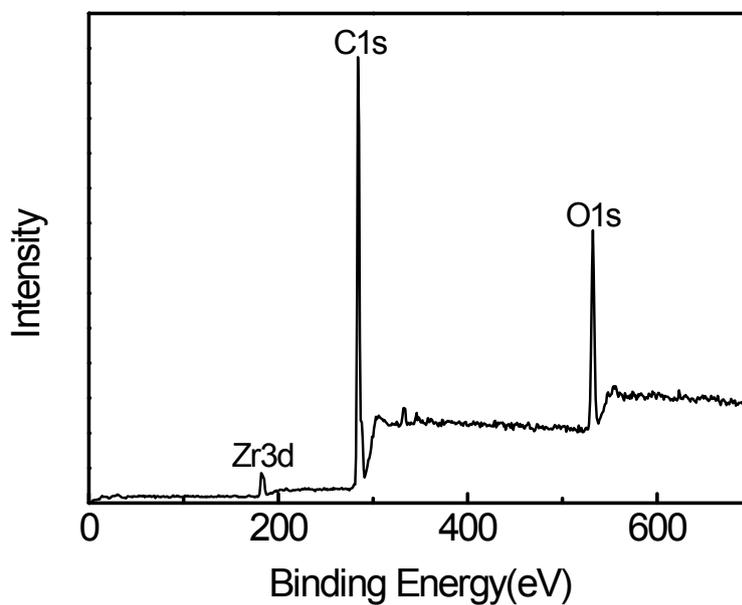
**Figure S5.** SEM images of TiO<sub>2</sub>/UiO-67 composite obtained from the emulsion with UiO-67 concentration of 10 mg mL<sup>-1</sup> and TBT concentration of 3.8 μL mL<sup>-1</sup>. Scale bars, 20 μm in a, 5 μm in b, 1 μm in c.



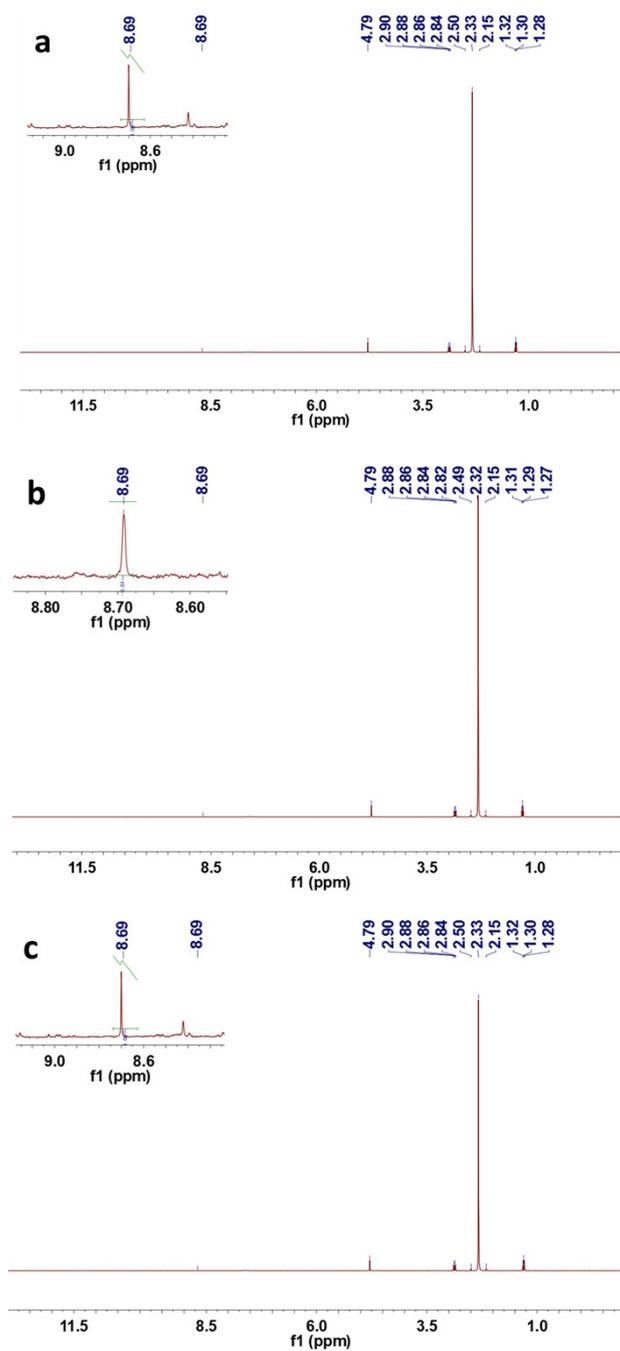
**Figure S6.** SEM images of TiO<sub>2</sub>/UiO-67 composite obtained from the emulsion with UiO-67 concentration of 10 mg mL<sup>-1</sup> and TBT concentration of 6.3 μL mL<sup>-1</sup>. Scale bars, 20 μm in a, 5 μm in b, 1 μm in c.



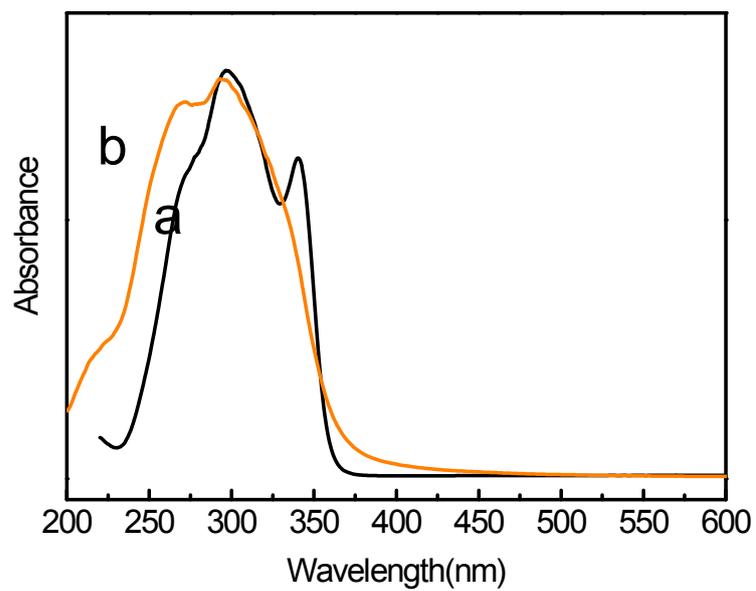
**Figure S7.** XPS spectrum of the  $\text{TiO}_2/\text{UiO-67}$  composite obtained from the emulsion with UiO-67 concentration of  $10 \text{ mg mL}^{-1}$  and TBT concentration of  $12.5 \text{ } \mu\text{L mL}^{-1}$ .



**Figure S8.** XPS spectrum of the pristine UiO-67.



**Figure S9.** The  $^1\text{H}$  NMR spectra for the following samples using  $\text{D}_2\text{O}$  as nuclear magnetism reagent in a NMR tube:  $\text{HCOOH}$  and  $\text{CH}_3\text{CN}$  (a); product (b); product and  $\text{HCOOH}$  (c). A singlet was observed at 8.69 ppm, which was assigned to the inactive hydrogen of  $\text{HCOOH}$ .



**Figure S10.** Diffuse-reflectance spectra of the pristine UiO-67 (a) and the TiO<sub>2</sub>/UiO-67 composite obtained from the emulsion with UiO-67 concentration of 10 mg mL<sup>-1</sup> and TBT concentration of 12.5 μL mL<sup>-1</sup> (b).