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## Supporting Information for

# **Interlayer expansion of few-layered Mo-doped SnS<sub>2</sub> nanosheets grown on carbon cloth with excellent lithium storage performance for lithium ion batteries**

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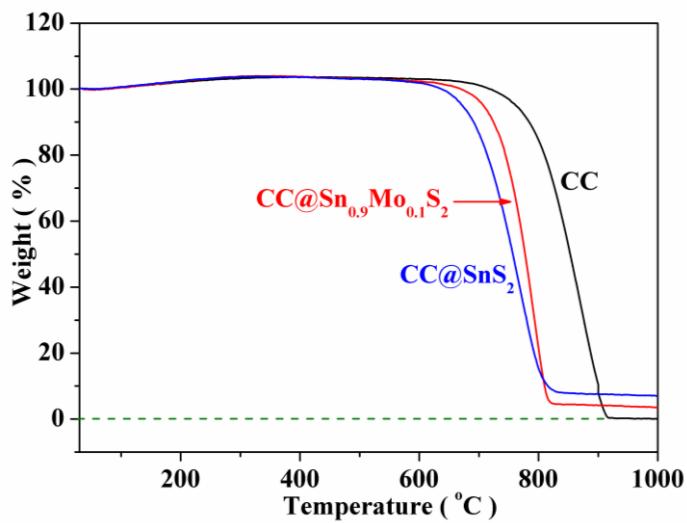


Fig. S1 TGA curves of CC, CC@SnS<sub>2</sub> and CC@Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub>.

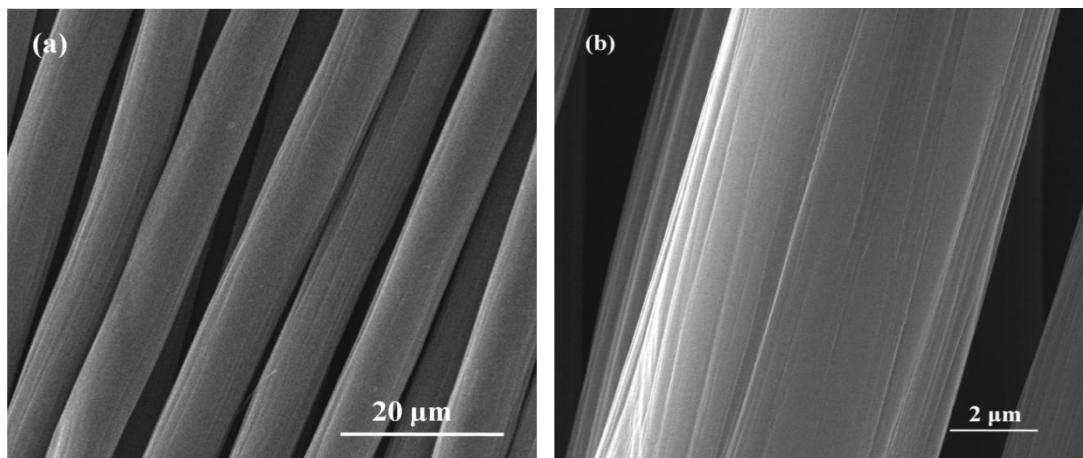


Fig. S2 SEM images of carbon cloth.

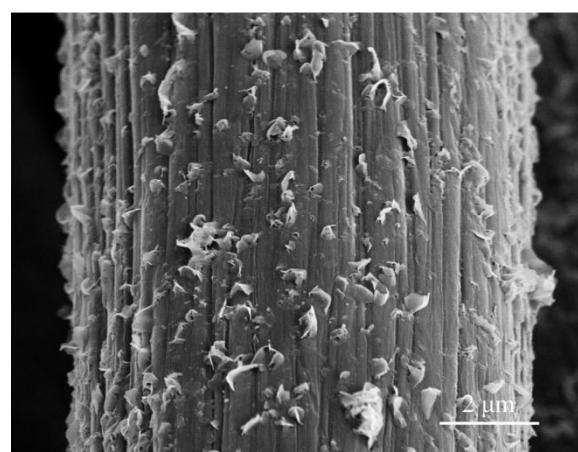


Fig. S3 SEM image of SnS<sub>2</sub> prepared at 200 °C 20 hours.

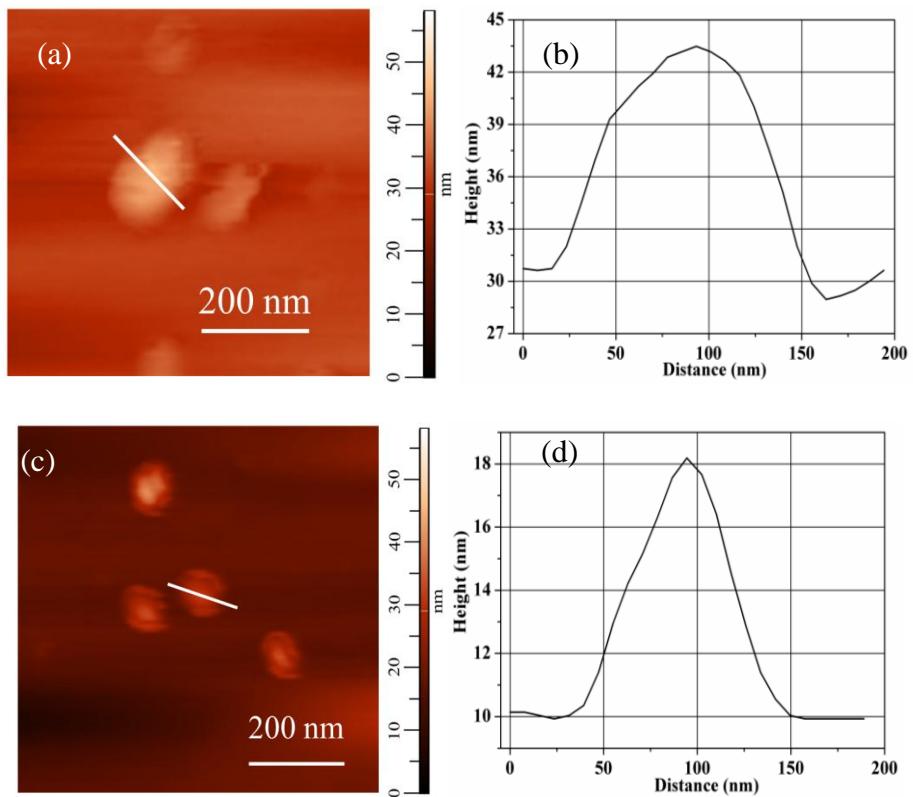


Fig. S4 Typical AFM images (a, c) and thickness profiles (b, d) of SnS<sub>2</sub> (a, b) and Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub> nanosheets after transferred onto mica substrates. The thickness of SnS<sub>2</sub> nanosheets is about 14 nm, and that of Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub> nanosheets is about 8 nm.

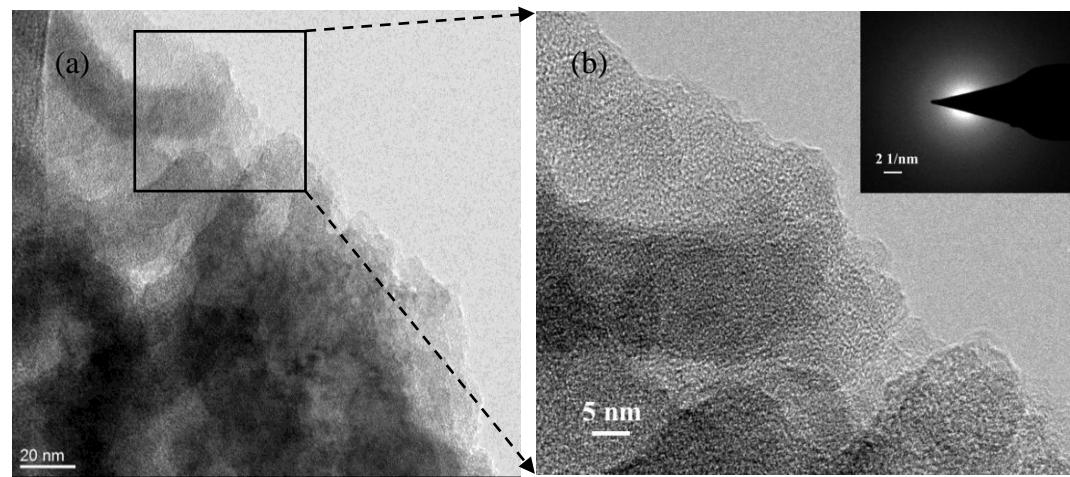


Fig. S5 *Ex-situ* TEM (a) and HRTEM images of CC@Sn<sub>0.9</sub>Mo<sub>0.1</sub>SnS<sub>2</sub> electrode after the first discharge at 1 A g<sup>-1</sup>. Inset of b: showing the corresponding selected area electron diffraction (SAED) pattern.

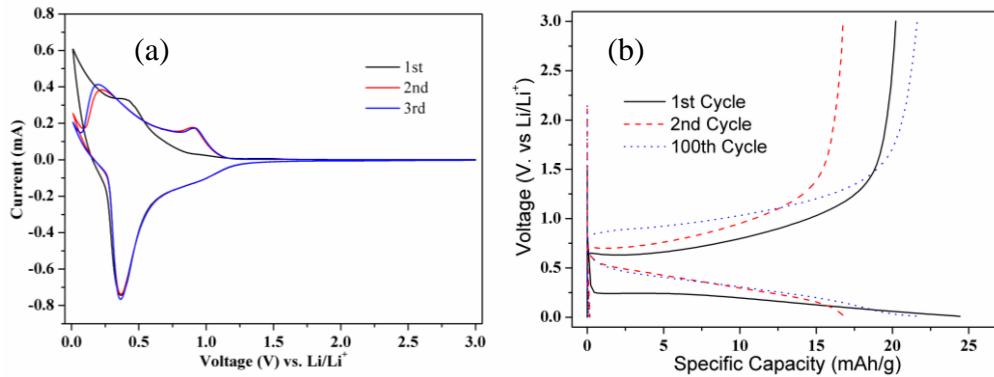


Fig. S6 (a) CV curves of CC measured between 0.01 and 3.0 V at a scan rate 0.1 mV s<sup>-1</sup> at a scan and (b) charging/discharging voltage profiles of CC at a current density of 1 A g<sup>-1</sup>.

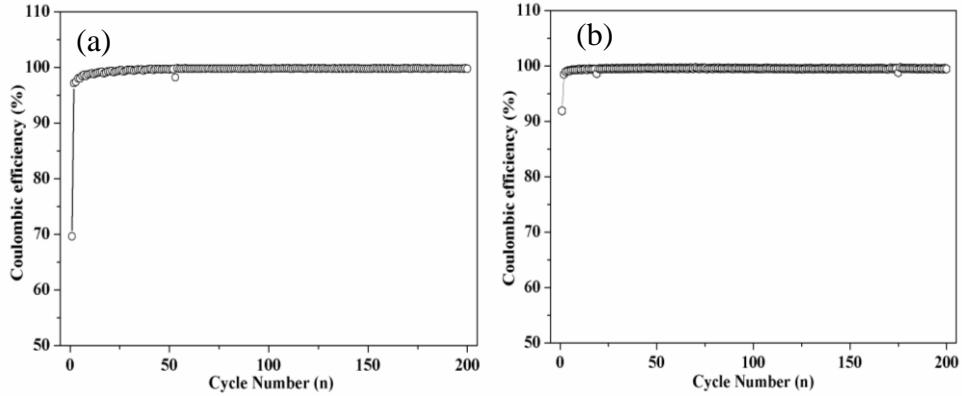


Fig. S7 Coulombic efficiency curves of CC@SnS<sub>2</sub> (a) and CC@Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub> (b) at a current density of 1 A g<sup>-1</sup>.

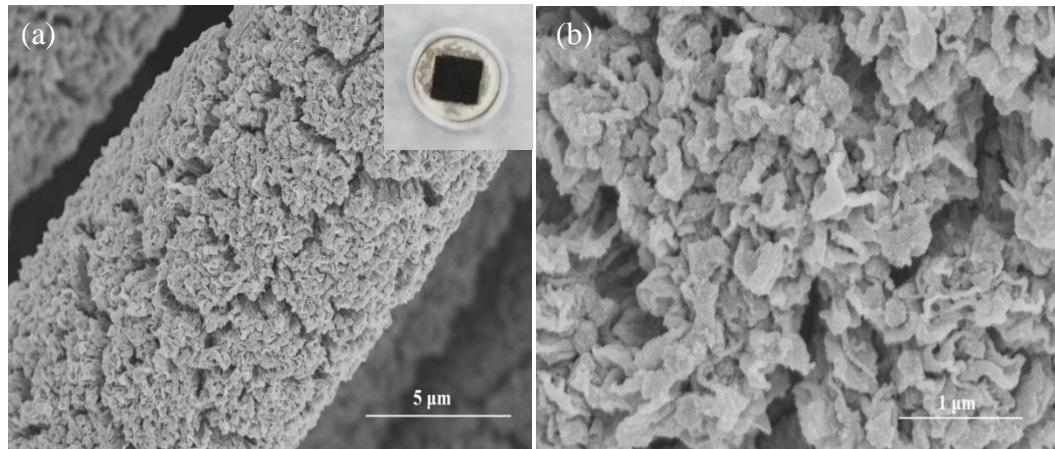


Fig. S8 (a) low and (b) high magnification SEM images of CC@Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub> electrode after 200 cycles at current density of 1 A g<sup>-1</sup>. Inset of (a): Photogragh of CC@Sn<sub>0.9</sub>Mo<sub>0.1</sub>S<sub>2</sub> electrode after 200 cycles.

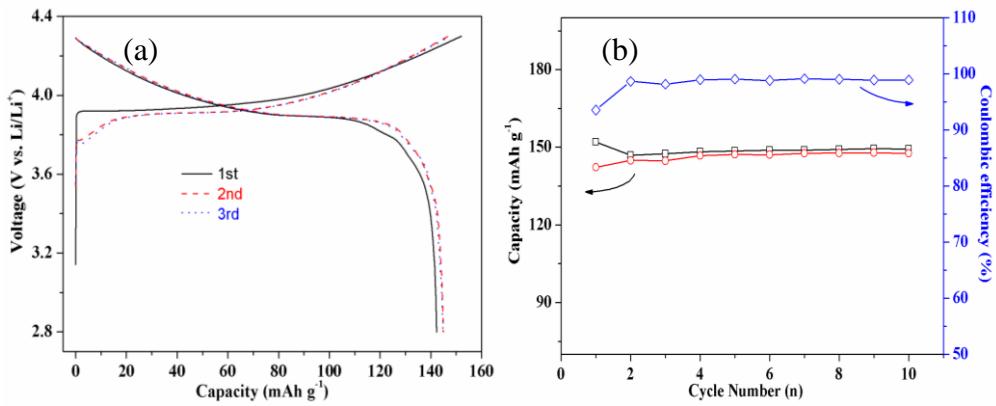


Fig. S9 Electrochemical performance of  $\text{LiCoO}_2$  electrode: The 1<sup>st</sup>, 2<sup>nd</sup> and 3<sup>rd</sup> charge-discharge voltage curves (a) and Cycle performance (b) at a current density of 0.1 C.

Table S1 Electrochemical performance of various  $\text{SnS}_2$ -based anode materials for lithium-ion batteries.

Electrode materials	Current density ( $\text{A g}^{-1}$ )	Initial discharge/charge capacity (mAh g <sup>-1</sup> ), Initial CE	Discharge capacity (mAh g <sup>-1</sup> )	Rate capacipy (mAh g <sup>-1</sup> )	Reference
CC@ $\text{Sn}_{0.9}\text{Mo}_{0.1}\text{S}_2$ NS	1	2033.6/1869.8 91.9 %	1950.8 200 cycles	914.5 $5 \text{ A g}^{-1}$	This work
CC@ $\text{SnS}_2$ NS	1	1272.4/885.9 69.6 %	584.9 200 cycles	171.1 $5 \text{ A g}^{-1}$	This work
Ce doped $\text{SnS}_2$ flowerlike	0.09	998.2/569.7, 57 %	450.7 50 cycles	321.9 $0.9 \text{ A g}^{-1}$	1
$\text{SnS}_2$ nanobelts	0.065	1232.2/645.4 52.4 %	560 50 cycles	273 $1.3 \text{ A g}^{-1}$	2
$\text{SnS}_2$ nanocrystals@RGO	0.065	1596/1257 78.7 %	1034 200 cycles	415 $3.25 \text{ A g}^{-1}$	3
$\text{SnS}_2$ @graphene nanocable	0.2	1334/764 58.3 %	720 350 cycles	580 $1 \text{ A g}^{-1}$	4
$\text{SnS}_2$ NS@MWCNTs	0.1	1497.3/557 37.2	506 50 cycles	513.8 $0.5 \text{ A g}^{-1}$	5
$\text{SnS}_2$ @PANI nanoplates	0.1	1395.8/968.7 69.4 %	730.8 80cycles	356.1 $5 \text{ A g}^{-1}$	6
3D-hierarchical $\text{SnS}_2$ structures	0.65	2283.6/795 34.8 %	570.3 100 cycles	486.2 $3.25 \text{ A g}^{-1}$	7
CNT@ $\text{SnS}_2$ tubular nanosheaths	0.1	1522/713 46.8 %	502 100 cycles	260 $0.4 \text{ A g}^{-1}$	8
CC-VN@ $\text{SnS}_2$	0.65	1098/763	791	349	9

nanocomposites		77 %	100 cycles	$13 \text{ A g}^{-1}$	
2D graphene-SnS <sub>2</sub> nanoplate	0.05	1900 <sup>(a)</sup> /687 36.2 %	650 30 cycles	230 $6.4 \text{ A g}^{-1}$	10
MWCNTs)/SnS <sub>2</sub> NS	0.645	2978/778 26.1 %	432 100 cycles	420 $6.45 \text{ A g}^{-1}$	11
SnS <sub>2</sub> nanocrystal– NGS	0.2	1178.5/1062.1 90.1 %	1407 120 cycles	520 $5 \text{ A g}^{-1}$	12
SnS <sub>2</sub> /VACNTs	0.1	1646/601 36.5 %	551 100 cycles	223 $2 \text{ A g}^{-1}$	13
3D porous SnS <sub>2</sub> /carbon clothes	0.194	1380/977 70.8 %	878 60 cycles	223 $3.23 \text{ A g}^{-1}$	14
3D SnS <sub>2</sub> @GF	1	1361.9/948.4 69.6 %	818.4 500 cycles	160.9 $5 \text{ A g}^{-1}$	15
3D hierarchical SnS <sub>2</sub> nanoflowers	0.1	1455.3/519 35.6 %	519 50 cycles	297 $0.8 \text{ A g}^{-1}$	16
stackes SnS <sub>2</sub> /graphene	0.2	1299/1160 89.3 %	1063 100 cycles	712 $5 \text{ A g}^{-1}$	17
3D SnS <sub>2</sub> nanoplates /graphene	0.1	1677/1195 71.3 %	1060 200 cycles	670 $2 \text{ A g}^{-1}$	18
2D SnS <sub>2</sub> nanoplates	0.2	1416/1028 72.6 %	935 30 cycles	523 $3 \text{ A g}^{-1}$	19
CC@TiO <sub>x</sub> N <sub>y</sub> @SnS <sub>2</sub>	0.645	1082/800 74 %	612 100 cycles	419 $3.25 \text{ A g}^{-1}$	20
SnS <sub>2</sub> NS-graphene	0.066	1505/1077.6 71.6 %	896 40 cycles	934 $0.66 \text{ A g}^{-1}$	21
SnS <sub>2</sub> nanoplates	0.1	1438/590 41 %	521 50 cycles	340 $3 \text{ A g}^{-1}$	22
C-SnS <sub>2</sub>	0.05	1724.4/707 41 %	668 50 cycles	650 $0.5 \text{ A g}^{-1}$	23
3D SnS <sub>2</sub> /graphene aerogels	0.05	4030/1485 37 %	656 30 cycles	240 $1 \text{ A g}^{-1}$	24
RGO–SnS <sub>2</sub>	0.065	1278.4/811 63.44 %	733 10 cycles	200 $3.25 \text{ A g}^{-1}$	25
few-layer SnS <sub>2</sub> /graphene	0.1	1664/705 42.4 %	920 50 cycles	600 $1 \text{ A g}^{-1}$	26
3D SnS <sub>2</sub> nanoflowers	0.4	2130/1140 53.5 %	455.5 100 cycles	495.5 $1.6 \text{ A g}^{-1}$	27
SnS <sub>2</sub> /Co <sub>3</sub> O <sub>4</sub>	0.1	1344/985 73 %	715 100 cycles	530 $1 \text{ A g}^{-1}$	28
SnS <sub>2</sub> /RGO	0.1	2190.5/1382.4 63.1 %	939.9 30 cycles	650.5 $0.8 \text{ A g}^{-1}$	29
SnS <sub>2</sub> /CNTs	0.1	1430.5/410 28.7 %	373.1 50 cycles	370 $1 \text{ A g}^{-1}$	30

graphene-CNT/SnS <sub>2</sub>	0.1	1774.3/1118.2 63 %	1017.5 100 cycles	634.6 2 A g <sup>-1</sup>	31
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CE respents Coulombic efficiency, NS respents nanosheets. (a): Values are estimated from graph.

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