

Chemical bonding at the metal-organic framework / metal oxide interface: simulated epitaxial growth of MOF-5 on rutile TiO₂

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1 Supplementary Information

1.1 Lattice mismatch

The lattice matching data for possible interfaces between metal-organic frameworks and binary inorganic materials within an 8 % mismatch is given in Tables S1, S2, S3 and S4. All binary materials are the experimental structures as taken from the ICSD¹. All MOF structures were cleaned of disorder from experimental .CIF files and converged with the PBEsol+D3 functional using VASP. The planewave cut-off was set to 500 eV and forces converged to 0.001 eV/Å.

Table S1 Lattice matching data for between the surface unit cells of the given binary materials (BM) and MOFs. Given are the Miller indices (hkl), cell expansions, and lattice mismatch percentages between the interfacing surfaces.

BM	MOF	BM index	MOF index	Expansion BM(x,y), MOF(x,y)	Mismatch BM,MOF (%)
SnS ₂	UiO-66 (Zr)	(010)	(110)	(4,5) (1,2)	0.1, 1.1
ZnSe	UiO-66 (Zr)	(001)	(110)	(5,5) (2,2)	2.2, 2.2
Al ₂ O ₃	UiO-67 (Zr)	(010)	(011)	(4,3) (1,2)	2.6, 2.5
		(110)	(011)	(4,3) (1,2)	2.6, 2.5
TiO ₂ (A)	UiO-67 (Zr)	(001)	(001)	(5,3) (1,2)	2.7, 0.7
		(101)	(011)	(5,5) (1,4)	2.6, 0.7
SiO ₂	UiO-67 (Zr)	(001)	(111)	(4,4) (1,1)	3.6, 3.6
ZnS	UiO-67 (Zr)	(111)	(111)	(5,5) (2,2)	1.6, 1.6
SnS ₂	UiO-66 (Ti)	(010)	(110)	(4,5) (1,2)	3.1, 1.8
ZnSe	UiO-66 (Ti)	(001)	(110)	(5,5) (2,2)	0.7, 0.7
Al ₂ O ₃	MIL-125(Ti)	(010)	(001)	(4,3) (1,2)	2.7, 0.5
		(100)	(001)	(4,3) (1,2)	2.7, 0.5
		(010)	(011)	(4,2) (1,1)	0.6, 0.5
		(100)	(010)	(4,2) (1,1)	0.6, 0.5
TiO ₂ (R)	MIL-125(Ti)	(011)	(110)	(4,5) (1,1)	2.2, 2.2
TiO ₂ (A)	MIL-125(Ti)	(001)	(001)	(5,3) (1,2)	2.4, 0.5
		(101)	(001)	(5,5) (1,4)	2.8, 0.5
		(100)	(011)	(5,4) (1,1)	1.4, 0.5
		(101)	(011)	(5,5) (1,3)	0.5, 0.5
SiO ₂	MIL-125(Ti)	(010)	(011)	(4,5) (1,1)	3.1, 3.3
SnS ₂	MIL-125(Ti)	(110)	(010)	(3,3) (1,1)	0.3, 1.8
Al ₂ O ₃	MOF-5(Zn)	(010)	(011)	(4,3) (1,2)	6.2, 4.0
		(110)	(011)	(4,3) (1,2)	6.2, 4.0
		(010)	(110)	(4,3) (1,2)	6.2, 4.0
		(100)	(100)	(4,3) (1,2)	6.2, 4.0
TiO ₂ (R)	MOF-5(Zn)	(001)	(011)	(4,4) (1,1)	3.9, 3.9
		(001)	(110)	(4,4) (1,1)	3.9, 3.9
		(110)	(011)	(6,4) (1,1)	3.0, 4.9
		(110)	(110)	(6,4) (1,1)	3.0, 4.9
TiO ₂ (A)	MOF-5(Zn)	(001)	(011)	(5,3) (1,2)	1.0, 3.0
		(001)	(110)	(5,3) (1,2)	1.0, 3.0
SnS ₂	MOF-5(Zn)	(010)	(011)	(5,3) (1,1)	3.5, 0.7
		(010)	(110)	(5,3) (1,1)	3.5, 0.7
		(110)	(011)	(3,3) (1,1)	3.5, 3.2
		(110)	(110)	(3,3) (1,1)	3.5, 3.2
ZnSe	MOF-5(Zn)	(001)	(110)	(3,3) (1,1)	5.9, 5.9
		(001)	(110)	(3,3) (1,1)	5.9, 5.9

Table S2 Lattice matching data for between the surface unit cells of the given binary materials (BM) and MOFs. Given are the Miller indices (hkl), cell expansions, and lattice mismatch percentages between the interfacing surfaces.

BM	MOF	BM index	MOF index	Expansion BM(x,y), MOF(x,y)	Mismatch BM,MOF (%)
TiO ₂ (R)	MOF-646(Zn)	(001)	(001)	(4,4) (1,1)	0.0, 2.2
		(001)	(100)	(4,4) (1,1)	2.2, 2.2
SnS ₂	MOF-646(Zn)	(110)	(101)	(3,4) (1,1)	3.4, 2.6
		(001)	(001)	(3,3) (1,1)	2.0, 0.2
ZnSe	MOF-646(Zn)	(001)	(100)	(3,3) (1,1)	0.2, 0.2
		(001)	(101)	(3,3) (1,1)	0.2, 0.2
ZrO ₂	DMOF-1(Zn)(orth)	(001)	(001)	(3,3) (1,1)	1.5, 0.2
		(110)	(001)	(3,3) (1,1)	1.8, 2.2
		(101)	(001)	(3,2) (1,1)	2.2, 2.2
		(101)	(010)	(3,5) (1,2)	1.4, 1.5
		(101)	(100)	(3,5) (1,2)	1.4, 2.2
Al ₂ O ₃	DMOF-1(Zn)(orth)	(010)	(110)	(4,5) (1,3)	1.1, 2.7
		(110)	(110)	(4,5) (1,3)	1.1, 2.7
TiO ₂ (A)	DMOF-1(Zn)(orth)	(001)	(001)	(4,5) (1,4)	0.8, 2.2
		(101)	(001)	(4,1) (1,1)	0.5, 2.2
		(100)	(010)	(4,3) (1,1)	1.6, 2.9
		(101)	(010)	(4,5) (1,4)	0.4, 2.9
		(001)	(011)	(4,2) (1,1)	1.1, 2.9
		(100)	(100)	(4,3) (1,1)	1.6, 2.2
		(101)	(100)	(4,5) (1,4)	0.4, 2.2
		(001)	(101)	(4,2) (1,1)	1.3, 2.2
		(010)	(011)	(2,4) (1,1)	1.2, 0.8
		(010)	(101)	(2,4) (1,1)	1.2, 0.8
ZnO	DMOF-1(Zn)(rhom)	(010)	(001)	(3,5) (1,2)	1.9, 0.3
		(010)	(101)	(4,4) (1,1)	3.8, 0.9
ZrO ₂	DMOF-1(Zn)(rhom)	(101)	(101)	(5,5) (2,2)	1.3, 2.6
Al ₂ O ₃	DMOF-1(Zn)(rhom)	(010)	(001)	(2,1) (1,1)	1.1, 1.4
		(100)	(001)	(2,1) (1,1)	1.1, 1.4
		(010)	(010)	(2,4) (1,3)	1.8, 1.4
		(100)	(010)	(2,4) (1,3)	1.8, 1.4
		(010)	(011)	(2,5) (1,3)	0.7, 1.4
		(100)	(011)	(2,5) (1,3)	0.7, 1.4
		(110)	(100)	(3,4) (2,3)	1.8, 3.4
		(110)	(101)	(3,3) (2,2)	1.8, 3.4
		(110)	(110)	(2,4) (1,3)	1.8, 2.8
		(110)	(110)	(2,4) (1,3)	1.8, 2.8
TiO ₂ (R)	DMOF-1(Zn)(rhom)	(011)	(110)	(3,4) (1,1)	0.2, 3.1
TiO ₂ (A)	DMOF-1(Zn)(rhom)	(001)	(001)	(5,1) (2,1)	1.4, 2.8
		(100)	(001)	(5,2) (2,1)	3.4, 2.4
		(101)	(001)	(5,4) (2,5)	2.5, 2.4
		(010)	(110)	(4,4) (1,1)	1.3, 0.0
SiO ₂	DMOF-1(Zn)(rhom)	(010)	(110)	(4,4) (1,1)	1.3, 0.0
SnS ₂	DMOF-1(Zn)(rhom)	(110)	(001)	(5,2) (3,1)	1.6, 16
GaN	DMOF-1(Zn)(rhom)	(010)	(001)	(3,5) (1,2)	2.4, 0.0
		(010)	(011)	(3,4) (1,1)	3.7, 0.0
ZnSe	DMOF-1(Zn)(rhom)	(001)	(010)	(5,3) (3,1)	2.1, 0.8

Table S3 Lattice matching data for between the surface unit cells of the given binary materials (BM) and MOFs. Given are the Miller indices (hkl), cell expansions, and lattice mismatch percentages between the interfacing surfaces.

BM	MOF	BM index	MOF index	Expansion BM(x,y), MOF(x,y)	Mismatch BM,MOF (%)
ZnO	MOF-649(Zn)	(010)	(001)	(5,4) (1,1)	1.9, 0.0
		(010)	(010)	(3,3) (1,1)	3.0, 0.2
		(010)	(100)	(3,4) (1,1)	1.9, 0.2
		(010)	(110)	(3,5) (1,1)	2.3, 0.2
ZrO ₂	MOF-649(Zn)	(001)	(001)	(3,4) (1,1)	1.0, 2.3
		(101)	(011)	(3,3) (1,1)	1.8, 2.3
Al ₂ O ₃	MOF-649(Zn)	(110)	(001)	(2,5) (1,3)	1.7, 2.2
		(010)	(010)	(2,5) (1,4)	0.6, 1.6
		(100)	(010)	(2,5) (1,4)	0.6, 1.6
		(010)	(100)	(2,5) (1,3)	1.7, 1.6
		(100)	(100)	(2,5) (1,3)	1.7, 1.6
		(010)	(101)	(4,5) (1,3)	1.7, 1.2
		(100)	(101)	(4,5) (1,3)	1.7, 1.2
		(010)	(110)	(2,2) (1,1)	2.8, 1.6
		(100)	(110)	(2,2) (1,1)	2.8, 1.6
		(001)	(010)	(5,4) (2,3)	1.8, 2.5
TiO ₂ (A)	MOF-649(Zn)	(100)	(010)	(5,5) (2,2)	2.5, 2.6
		(101)	(010)	(5,1) (2,1)	3.4, 2.5
		(001)	(100)	(5,5) (2,3)	3.5, 2.5
		(101)	(100)	(5,4) (2,3)	2.3, 2.5
		(001)	(101)	(5,5) (1,3)	3.5, 0.3
		(101)	(101)	(5,4) (1,3)	2.3, 0.3
		(001)	(110)	(5,4) (2,1)	0.8, 2.5
		(101)	(110)	(5,5) (2,3)	2.7, 2.5
		(010)	(010)	(2,3) (1,1)	1.1, 0.4
		(010)	(100)	(2,4) (1,1)	1.1, 1.5
SiO ₂	MOF-649(Zn)	(010)	(101)	(4,4) (1,1)	3.9, 1.5
		(010)	(110)	(2,5) (1,1)	1.1, 1.1
		(110)	(001)	(3,5) (1,2)	0.4, 0.5
		(110)	(011)	(3,3) (1,2)	2.5, 0.2
		(010)	(001)	(5,4) (1,1)	1.4, 0.2
GaN	MOF-649(Zn)	(010)	(010)	(3,3) (1,1)	2.4, 0.1
		(010)	(100)	(3,4) (1,1)	1.4, 0.1
		(010)	(110)	(3,5) (1,1)	1.8, 0.1
		(001)	(001)	(3,4) (1,1)	2.3, 1.5
ZnS	MOF-649(Zn)	(011)	(011)	(3,3) (1,1)	0.9, 1.5
		(011)	(010)	(5,2) (3,1)	0.8, 0.9
ZnSe	MOF-649(Zn)	(011)	(010)		

Table S4 Lattice matching data for between the surface unit cells of the given binary materials (BM) and MOFs. Given are the Miller indices (hkl), cell expansions, and lattice mismatch percentages between the interfacing surfaces.

BM	MOF	BM index	MOF index	Expansion BM(x,y), MOF(x,y)	Mismatch BM,MOF (%)
ZnO	IRMOF-16(Zn)	(110)	(001)	(4,4) (1,1)	3.7, 3.2
		(110)	(010)	(4,4) (1,1)	3.7, 3.2
		(110)	(100)	(4,4) (1,1)	3.7, 3.2
ZrO ₂	IRMOF-16(Zn)	(100)	(001)	(4,4) (1,1)	2.2, 2.5
		(100)	(010)	(4,4) (1,1)	2.2, 2.5
		(100)	(100)	(4,4) (1,1)	2.2, 2.5
TiO ₂ (R)	IRMOF-16(Zn)	(001)	(001)	(5,5) (1,1)	2.0, 2.0
		(011)	(001)	(5,4) (1,1)	2.8, 2.0
		(001)	(010)	(5,5) (1,1)	2.0, 2.0
		(011)	(010)	(5,4) (1,1)	2.7, 2.0
		(001)	(100)	(5,5) (1,1)	2.0, 2.0
		(011)	(100)	(5,4) (1,1)	2.7, 2.0
SiO ₂	IRMOF-16(Zn)	(110)	(001)	(4,5) (1,2)	0.2, 1.8
		(110)	(010)	(4,5) (1,2)	0.2, 1.8
		(110)	(100)	(4,5) (1,2)	0.2, 1.8
GaN	IRMOF-16(Zn)	(110)	(001)	(4,4) (1,1)	3.5, 2.7
		(110)	(010)	(4,4) (1,1)	3.5, 2.7
		(110)	(100)	(4,4) (1,1)	3.5, 2.7
ZnS	IRMOF-16(Zn)	(001)	(001)	(4,4) (1,1)	1.3, 1.3
		(001)	(010)	(4,4) (1,1)	1.3, 1.3
		(011)	(011)	(4,4) (1,1)	1.3, 1.3
		(001)	(100)	(4,4) (1,1)	1.3, 1.3
		(011)	(101)	(4,4) (1,1)	1.3, 1.3
		(011)	(110)	(4,4) (1,1)	1.3, 1.3
SnS ₂	IRMOF-14(Zn)	(110)	(011)	(4,4) (1,1)	3.4, 3.3
		(110)	(101)	(4,4) (1,1)	3.4, 3.3
ZnSe	IRMOF-14(Zn)	(001)	(110)	(4,4) (1,1)	5.9, 5.9
		(001)	(110)	(4,4) (1,1)	5.9, 5.9
		(011)	(011)	(4,3) (1,1)	5.9, 0.3
		(001)	(101)	(4,3) (1,1)	5.9, 0.3
		(111)	(001)	(4,4) (1,1)	0.3, 0.3
		(111)	(010)	(4,4) (1,1)	0.3, 0.3
		(111)	(100)	(4,4) (1,1)	0.3, 0.3
		(111)	(111)	(4,4) (1,1)	0.3, 0.3
		(001)	(001)	(5,3) (1,2)	0.7, 1.3
TiO ₂ (A)	HKUST-1(Cu)	(001)	(011)	(5,3) (1,2)	0.7, 1.3
		(001)	(110)	(5,3) (1,2)	0.7, 1.3
ZnS	HKUST-1(Cu)	(111)	(001)	(5,5) (2,2)	3.5, 2.5
		(111)	(100)	(5,5) (2,2)	3.5, 3.5

Table S5 Lattice matching data for between the surface unit cells of the given binary materials (BM) and MOFs. Given are the Miller indices (hkl), cell expansions, and lattice mismatch percentages between the interfacing surfaces.

BM	MOF	BM index	MOF index	Expansion BM(x,y), MOF(x,y)	Mismatch BM,MOF (%)
ZnO	COF-1(-)	(110)	(010)	(3,1) (2,4)	1.3, 1.5
		(110)	(100)	(3,1) (2,4)	1.3, 1.5
ZrO ₂	COF-1(-)	(101)	(110)	(2,5) (3,1)	1.1, 0.8
		(010)	(110)	(3,3) (4,1)	0.8, 1.2
Al ₂ O ₃	COF-1(-)	(100)	(110)	(3,3) (4,1)	0.8, 1.2
		(010)	(010)	(4,5) (5,1)	3.0, 0.3
TiO ₂ (R)	COF-1(-)	(001)	(010)	(5,5) (4,1)	3.0, 0.9
		(010)	(010)	(4,5) (5,1)	3.0, 0.3
		(001)	(100)	(5,5) (4,1)	3.0, 0.9
		(010)	(100)	(5,5) (4,1)	3.0, 0.9
SnS ₂	COF-1(-)	(010)	(010)	(1,4) (1,1)	3.9, 2.9
		(010)	(100)	(1,4) (1,1)	3.9, 2.9
GaN	COF-1(-)	(110)	(010)	(2,4) (3,1)	1.6, 0.9
		(110)	(110)	(2,4) (3,1)	1.6, 0.9
ZnS	COF-1(-)	(001)	(010)	(2,4) (3,1)	3.8, 3.0
		(011)	(010)	(2,3) (3,1)	2.1, 3.0
		(001)	(100)	(2,4) (3,1)	3.8, 3.0
		(011)	(100)	(2,3) (3,1)	2.1, 3.0
		(011)	(110)	(2,5) (3,1)	1.7, 3.0
		(001)	(010)	(3,4) (5,1)	1.5, 2.3
		(001)	(100)	(3,4) (5,1)	1.5, 2.3
		(011)	(110)	(3,5) (5,1)	3.5, 2.3

1.2 First principles reference state energies

The total energy of reference states considered for the calculation of binding energies of BDC on the surface of TiO₂ are given in Table S6.

Table S6 First-principles reference energies used for calculating reported binding energies of BDC ligand to TiO₂ (110) surface.

System	Energy (eV)
BDC-H ₂	-3070.515
BDC-H ₁ ⁻	-3056.345
TiO ₂ (110) surface	-442099.193
TiO ₂ (110) surface + H ⁺	-442114.702

1.3 Coulomb energy correction

An electrostatic energy correction was required for calculating the proton transfer energy between BDC ligand and TiO₂ surface. We place an F⁻ anion in the centre of a cubic unit cell and calculate the internal energy as a function of inverse unit cell length, a . The internal energy of a charged unit cell with infinite dimensions can be extrapolated from the intercept of $1/a$ vs internal energy plot. The difference in energy of an F⁻ anion in the unit cell used for the BDC on TiO₂ surface calculations and the total energy of the charged unit cell with infinite dimensions is the correction per +/- 1 charge on a system (-0.408 eV). As two charges need to be accounted for (from surface and ligand charged reference states) in the correction, a factor of two is applied when calculating the final proton transfer energy. The charge correction was calculated as the difference between the infinite box limit (intercept of the plot given in Figure S1) and the energy of the anion in the unit cell used for the first-principles calculations of BDC on TiO₂.

Table S7 Total energy of F⁻ anion in a cubic unit cell of dimensions (a) used to calculate the electrostatic Coulomb energy correction of charged reference states.

a	$1/a (\text{\AA}^{-1})$	Total energy (Ha)	Total energy (eV)
10	0.1	-24.227	-659.253
15	0.066666667	-24.205	-658.660
20	0.05	-24.194	-658.341
25	0.04	-24.186	-658.145
30	0.033333333	-24.182	-658.012
35	0.028571429	-24.178	-657.916
40	0.025	-24.175	-657.844
45	0.022222222	-24.173	-657.788
50	0.02	-24.172	-657.743

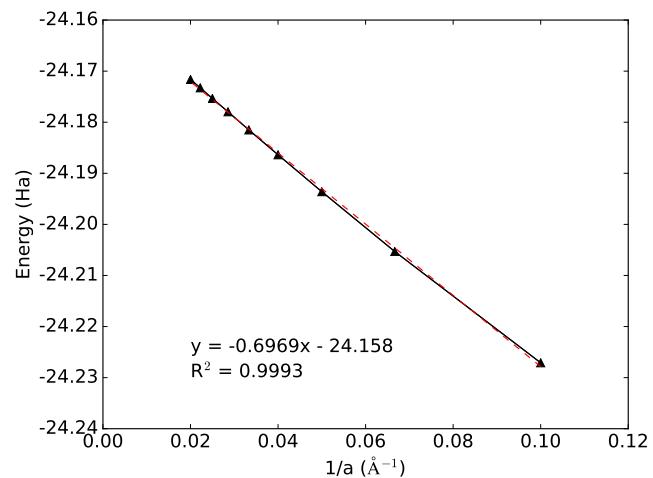


Fig. S1 Electrostatic Coulomb energy correction plot from data given in Table S7.

Table S8 Calculation of electrostatic Coulomb energy correction from the infinite box limit as determined from data reported in Table S7.

Description	Energy
Intercept (infinite box limit) energy (Ha)	-24.158
In box same size as simulations (Ha)	-24.173
Δ E (Coulomb) (Ha)	0.015
Δ E (Coulomb) (eV)	0.408

1.4 Thermodynamic cycle of proton transfer

The thermodynamic cycle considered for calculating the proton transfer energy between a BDC ligand and TiO₂ surface is given in Figure S2. Note the inclusion of the electrostatic Coulomb energy correction.

1.5 Forcefield parameters and structure files

All forcefield parameter files can be obtained from <https://github.com/WMD-group/VMOF>. The charges of the BDC ligands depending on where they are located in relation to the surface and charge terminating ends are given in Figure S3.

1.6 Proton migration across the surface of TiO₂

The energy required to move the protons from ligand a and ligand b across the surface of TiO₂ was calculated (Table S9), with trialled positions depicted in Figure S4. Two different structure models were considered and are depicted in Figure S5. For ligand b, reconstruction of the surface was observed when the proton was moved around neighbouring oxide sites

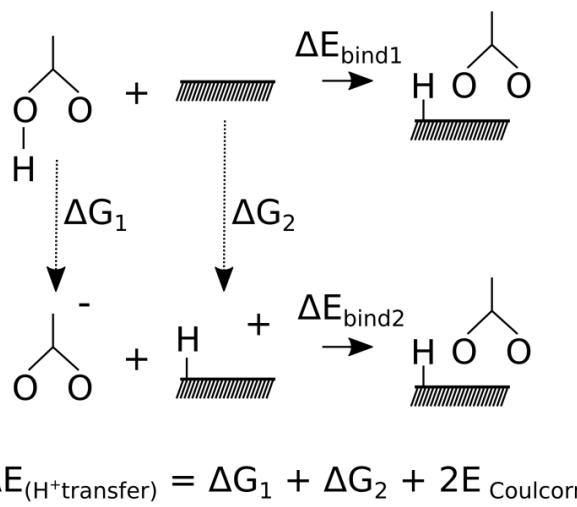


Fig. S2 Thermodynamic cycle considered for calculating proton transfer energies between BDC and the TiO₂ surface.

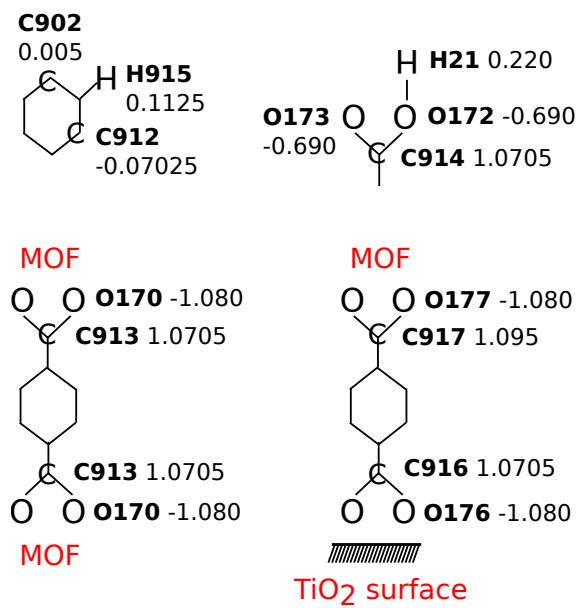


Fig. S3 Charges and atom types correlating to supplied parameter files of the BDC ligand in different regions of the MOF with regards to distance from the TiO₂ surface and charge terminating ends of the ligand.

to the BDC ligand position, which has resulted in the large differences in energies for the reconstructed and un-reconstructed surfaces.

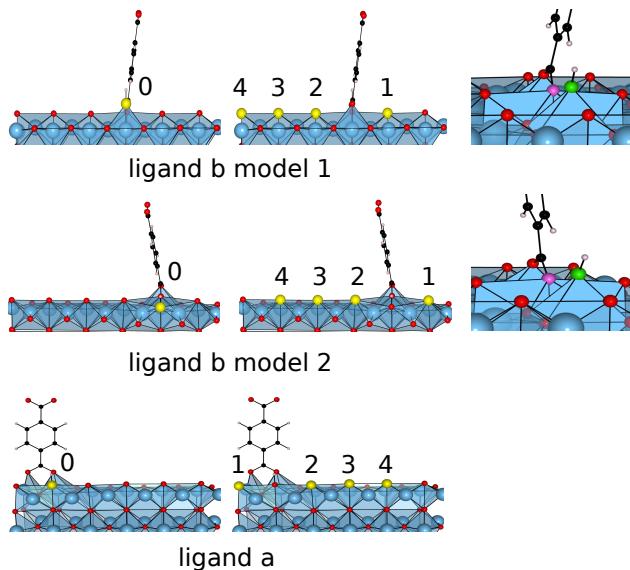


Fig. S4 Locations considered for the movement of a proton across the surface for ligand **b** model 1 and 2 and ligand **a**. Initial converged geometries are given, where the proton is located on the displaced oxide anion (left) with proton movements across the oxide bridge shown (centre). The closer view of the reconstruction at the site of ligand **b** models is given (right), where the protonated oxide is highlighted in green and O (carb) that has become incorporated into the surface is highlighted in pink. Note that the labels of ligand **a** and ligand **b** correspond to Figure 5 in the main manuscript text but the orientation of the cell is rotated.

Table S9 Energy of moving protons associated with ligand **a** and **b** considering structures 1 and 2 (Figure S5) of MOF-5 on TiO₂. Positions 1–4 refer to an increasing distance of the proton from the ligand along down the oxide bridge as depicted in Figure S4.

Ligand a proton movement	Relative energy (eV)			
Proton position	Structure 1		Structure 2	
	Ligand b model 1	Ligand b model 2	Ligand b model 1	Ligand b model 2
	0.000	0.000	0.000	0.000
	0.018	0.001	0.077	0.005
	0.235	0.234	0.236	0.334
Ligand b proton movement	Relative energy (eV)			
	Structure 1		Structure 2	
	Ligand b model 1	Ligand b model 2	Ligand b model 1	Ligand b model 2
	0.000	0.018	0.000	0.000
	2.218	0.202	2.267	0.234
Proton position	2.300	0.000	2.349	0.025
	2.372	2.681	2.410	2.541
	2.346	2.797	2.381	2.471

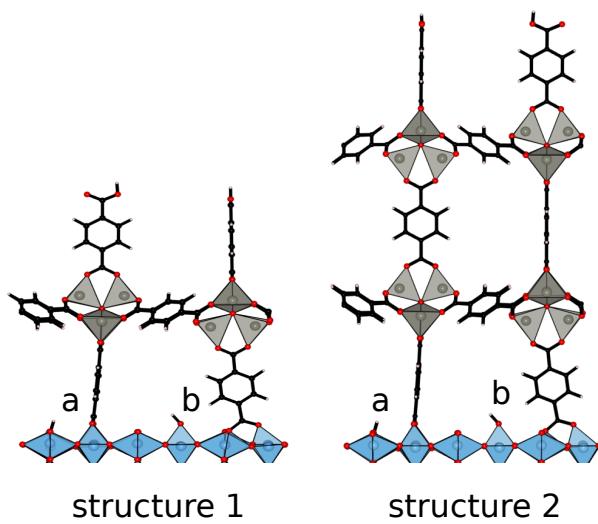


Fig. S5 Structure models considered for different layer thickness of MOF-5 on the (110) surface of TiO₂ with ligand **a** and ligand **b** labelled. Note that the full thickness of the TiO₂ has not been depicted but is identical for both models as described in the main manuscript.

1.7 Energy of adsorption of MOF-5 on TiO₂

To calculate the energy of adsorption of MOF-5 on TiO₂ with the forcefield, the thickness of the two layers were incrementally increased. The extrapolation of a plot of internal energy as a function of layer thickness (Figure S6) allows the bare adsorption energy of the interfacing layers to be calculated.

Table S10 Calculation of the energy of adsorption between the interfacing surfaces of MOF-5 and TiO₂.

Number of layers	Internal energy (eV)
2	-3525.369
3	-5439.346
4	-7357.126
Surface area of interface (m ⁻²)	3.449×10^{-17}
Intercept (zero interface limit)	307.020
Energy of adsorption (Jm ⁻²)	1.426

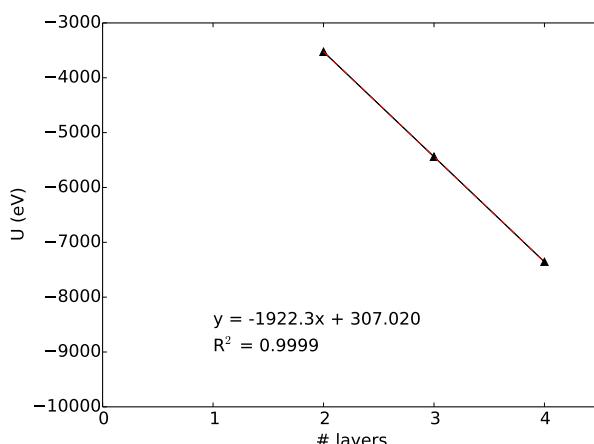


Fig. S6 Thickness of MOF-5 and TiO₂ vs internal energy of the interfacing systems plot used to calculate the energy of adsorption between the interfacing surfaces.

References

- 1 A. Belsky, M. Hellenbrandt, V. L. Karen and P. Luksch,
Acta Cryst. B, 2002, **58**, 364–369.