Electronic Supplementary Information (ESI)

Controllable Synthesis of Mn₃O₄ Nanodots@Nitrogen-Doped Graphene

and Its Application for High Energy Density Supercapacitors

Li Liu,^{a,c} Lijun Su,^{a,c} Junwei Lang,^a Bin Hu,^b Shan Xu,*,^band Xingbin Yan*,^a

- ^a Laboratory of Clean Energy Chemistry and Materials, State Key Laboratory of Solid Lubrication, Lanzhou Institute of Chemical Physics, Chinese of Academy of Sciences, Lanzhou, 730000, P. R. China.
- ^b State Key Laboratory for Oxo Synthesis and Selective Oxidation, Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences, Lanzhou 730000, P. R. China.
- ^c University of Chinese Academy of Sciences, Beijing 100080, P. R. China
- * The corresponding authors: E-mail: <u>xbyan@licp.cas.cn</u>;Tel.: +86-931-4968055 & E-mail: <u>xushan@licp.cas.cn</u>

Supporting Figures

Table S1 The reaction conditions of synthesizing Mn₃O₄ NDs@NG composites with

	Mn ₃ O ₄					
Sample	NDs@NG-	NDs@NG-	NDs@NG-	NDs@NG-	NDs@NG-	NDs@NG
	0.25	0.4	0.5	0.6	0.75	-1
$m[Mn_2(CO)_5] (mg)$	10	16	20	24	30	40
The volume of						
Mn ₂ (CO) ₅ /DMF solution	10					
(mL)						
m[GO] (mg)	40					
The volume of GO and DMF						
(mL)	20					
$m[Mn_2(CO)_{10}]:m[GO]$	0.25:1	0.4:1	0.5:1	0.6:1	0.75:1	1:1
The volume of octylamine						
(mL)	0.25					

different Mn₃O₄ mass loading.



Figure S1 TEM images of the (a) $Mn_3O_4 NDs@NG-0.25$, (b) $Mn_3O_4 NDs@NG-0.4$, (c) Mn_3O_4

NDs@NG-0.5, (d) Mn₃O₄ NDs@NG-0.6, (e) Mn₃O₄ NDs@NG-0.75, (f) Mn₃O₄ NDs@NG-1.



Figure S2 The TEM images of Mn₃O₄ NDs@NG-0.5 composite prepared by adding different volume of





Figure S3 The TEM images of (a) Mn₃O₄ NDs@NG-Cl and (b) Mn₃O₄ NDs@NG-CO; (c) FESEM and

(d) TEM of Mn₃O₄ @NG composite.



Figure S4 Electrochemical properties of NG, Mn₃O₄@NG and different proportions of Mn₃O₄ NDs@NG

composites in EMIMBF₄ electrolyte: (a) GCD curves at 2 A g⁻¹; (b) Specific capacitances at different



current densities.

Figure S5 Electrochemical properties of Mn₃O₄ NDs@NG-based SC: (a) CV curves in different voltage

range at 20 mV s⁻¹; (b) GCD curves in different voltage range at 1A g⁻¹; Electrochemical properties of Mn₃O₄@NG-based SC: (c) CV curves at different scan rates; (d) GCD curves at different current densities;



Figure S6 (a) The CV curves of Mn₃O₄ NDs@NG and APDC performed in a three-electrode cell in EMIMBF₄ electrolyte at 20 mV s⁻¹; (b) Specific capacitances of APDC at different current densities.
Electrochemical properties of Mn₃O₄ NDs@NG//APDC ASC: (c) GCD curves in different voltage range at

1A g⁻¹; (d) Specific capacitances in different voltage range at different current densities.



Figure S7 Electrochemical properties of APDC//APDC SC: (a) CV curves in different voltage range at 10 mV s⁻¹; (b) GCD curves in different voltage range at 1 A g⁻¹; (c) Specific capacitances in different voltage range at different current densities; (d) The GCD curves at 0.5 A g⁻¹ of APDC//APDC and Mn₃O₄

NDs@NG//APDC ASC;