## Highly stable non-noble metal $Ni_2P$ co-catalyst for increased $H_2$ generation by g-C<sub>3</sub>N<sub>4</sub> under visible light irradiation

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Fig. S1 FT-IR spectra of pure  $g-C_3N_4$  and  $g-C_3N_4/Ni_2P$  composites with various amount of Ni<sub>2</sub>P (i.e. 2wt%, 4wt%, and 8wt%). FT-IR for Ni<sub>2</sub>P is also provided for reference.



Fig. S2 XPS spectra of Ni 2p binding energy in (a) pure  $Ni_2P$  and (b)  $g-C_3N_4/Ni_2P(8wt\%)$  sample.



Fig. S3 XPS spectra of P 2p binding energy in (a) pure  $Ni_2P$  and (b) g-C<sub>3</sub>N<sub>4</sub>/Ni<sub>2</sub>P(8wt%) samples.



**Fig. S4** XPS spectra of C 1s binding energy in pure  $g-C_3N_4$ , a physical mixture of  $g-C_3N_4$  and  $Ni_2P$  (2 wt%),  $g-C_3N_4/Ni_2P$  (2 wt%), and  $g-C_3N_4/Ni_2P$  (8 wt%) samples.



**Fig. S5** XPS spectra of N 1s binding energy in pure  $g-C_3N_4$ , a physical mixture of  $g-C_3N_4$  and Ni<sub>2</sub>P (2 wt%),  $g-C_3N_4/Ni_2P$  (2 wt%), and  $g-C_3N_4/Ni_2P$  (8 wt%) samples.



**Fig. S6** UV-vis absorption spectra of pure  $g-C_3N_4$ ,  $g-C_3N_4/Ni_2P$  (2 wt%), and a physical mixture of  $g-C_3N_4$  and  $Ni_2P$  (2 wt%).



Fig. S7 SEM images of (a) pure  $g-C_3N_4$ , (b)  $g-C_3N_4/Ni_2P(2wt\%)$ , (c)  $g-C_3N_4/Ni_2P(4wt\%)$ , and (d)  $g-C_3N_4/Ni_2P(8wt\%)$  sample. The scale bar is 1  $\mu$ m.



Fig. S8 Elemental mapping for C, N, Ni, and P in  $g-C_3N_4/Ni_2P(2wt\%)$  sample.



Fig. S9 TEM image of a  $g-C_3N_4/Ni_2P(8wt\%)$  sample. The growth and aggregation of  $Ni_2P$  particle is highlighted by the dashed circles.



Fig. S10 XRD patterns of a g-C $_3N_4/Ni_2P(2wt\%)$  sample before and after 24 h long-term H<sub>2</sub> evolution testing.



Fig. S11 TEM image of a  $g-C_3N_4/Ni_2P(2wt\%)$  sample after 24 h long-term  $H_2$  evolution testing. Ni<sub>2</sub>P particles are shown by dashed circles.



Fig. S12 TEM image of a g-C\_3N\_4/Pt (0.5 wt%) sample after 24 h long term  $\rm H_2$  evolution testing.



**Fig. S13** Stable photoluminescence spectra of pure  $g-C_3N_4$ ,  $g-C_3N_4/Ni_2P(2wt\%)$ ,  $g-C_3N_4/Ni_2P(8wt\%)$ ,  $g-C_3N_4/Pt(0.5wt\%)$ , and a physical mixture of  $g-C_3N_4$  and  $Ni_2P(2wt\%)$  measured in the solid state.



**Fig. 14** Time-resolved photoluminescence decay spectra of pure  $g-C_3N_4$ ,  $g-C_3N_4/Ni_2P(2wt\%)$ , and  $g-C_3N_4/Ni_2P(8wt\%)$  measured in the solid state. The fitted lifetime from the fluorescence decays are illustrated in the table (inset).