

Electronic supplementary information

Cubic quantum dot/hexagonal microsphere ZnIn₂S₄ heterophase junction for exceptional visible-light-driven photocatalytic H₂ evolution

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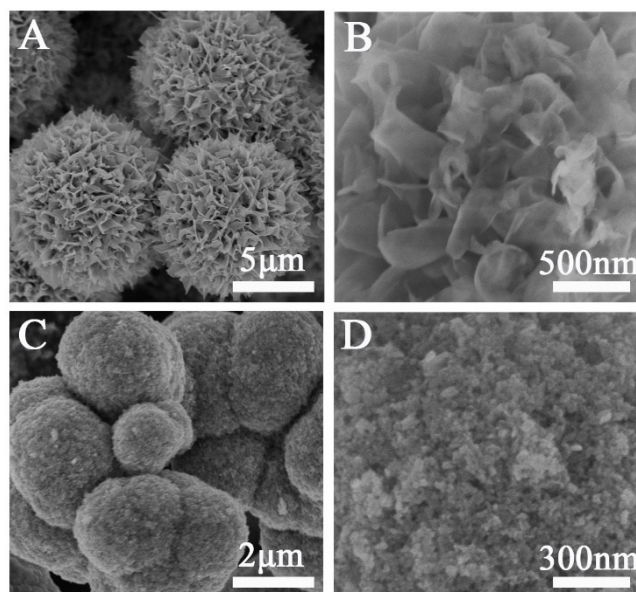


Fig. S1. SEM images of the H-ZnIn₂S₄ (A,B) and C-ZnIn₂S₄ (C,D) with different magnifications.

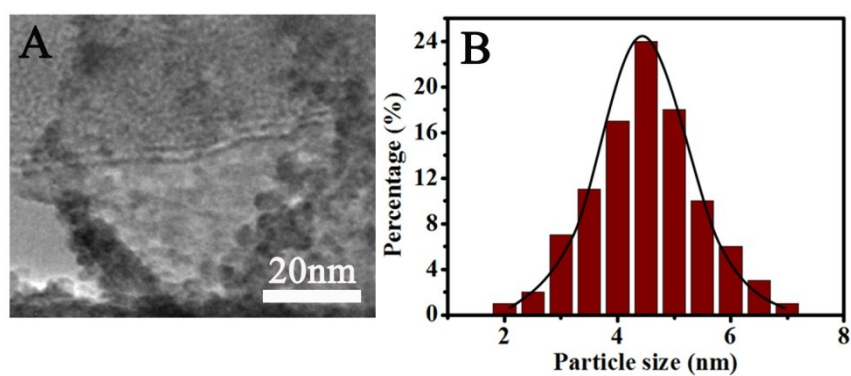


Fig. S2. The TEM image (A) of the 1/6J-ZnIn₂S₄ and corresponding size distribution histogram (B) of the cubic quantum dots in the 1/6J-ZnIn₂S₄.

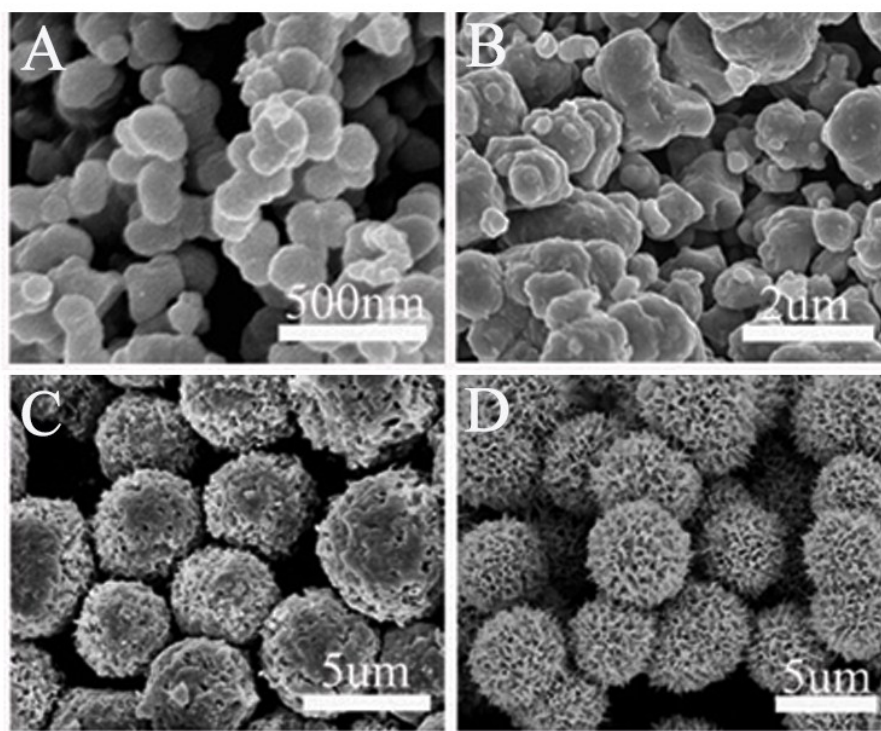


Fig. S3. SEM patterns of the prepared ZnIn_2S_4 samples obtained from different solvothermal reaction time: (A) 0.5 h, (B) 1 h, (C) 6 h, and (D) 12 h.

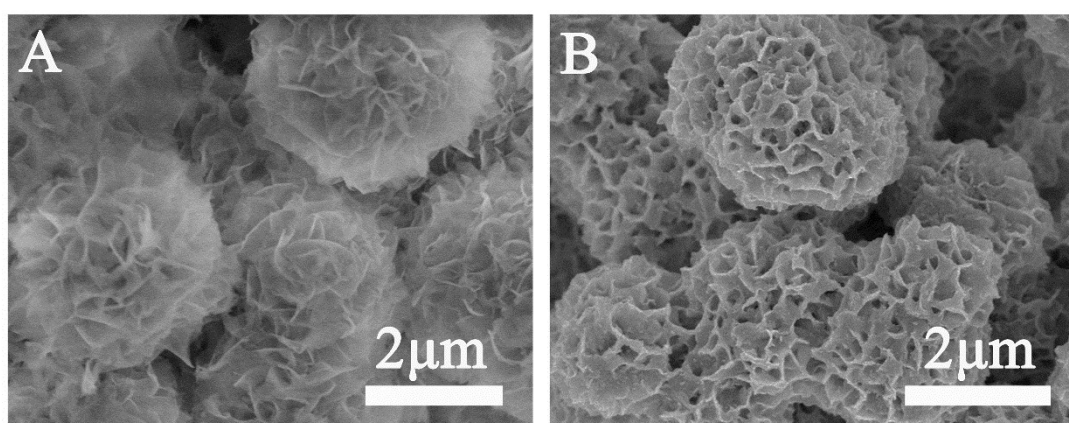


Fig. S4. (A) and (B) are the SEM images of the $1/8\text{J-ZnIn}_2\text{S}_4$ and $1/4\text{J-ZnIn}_2\text{S}_4$, respectively.

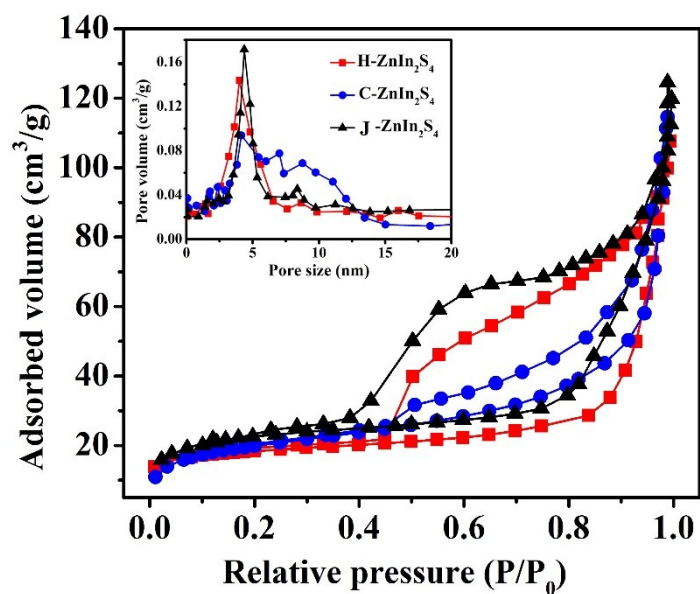


Fig. S5. Nitrogen adsorption–desorption isotherms and the pore size distribution plots (inset) of H-ZnIn₂S₄, C-ZnIn₂S₄ and J-ZnIn₂S₄ (1:6).

Table S1 The results of N₂ adsorption–desorption isotherm test of the different samples.

Samples	S _{BET} (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Average pore size (nm)
H-ZnIn ₂ S ₄	63.8	0.15	3.2
J-ZnIn ₂ S ₄	75.5	0.17	3.0
C-ZnIn ₂ S ₄	32.3	0.10	2.8

Table S2 Summary of the photoluminescence decay time (τ) and their relative intensities of the different samples.

sample	τ_1 (ns)	τ_2 (ns)	I_1 (%)	I_2 (%)	Average lifetime (τ , ns)
J-ZnIn ₂ S ₄	2.19	9.78	27.97	72.03	9.35
H-ZnIn ₂ S ₄	1.68	8.16	32.68	67.32	8.47
C-ZnIn ₂ S ₄	1.55	7.65	33.55	66.45	5.48

The average lifetime was calculated using equation: $\langle\tau\rangle=(I_1\tau_1^2 + I_2\tau_2^2)/(I_1\tau_1 + I_2\tau_2)$

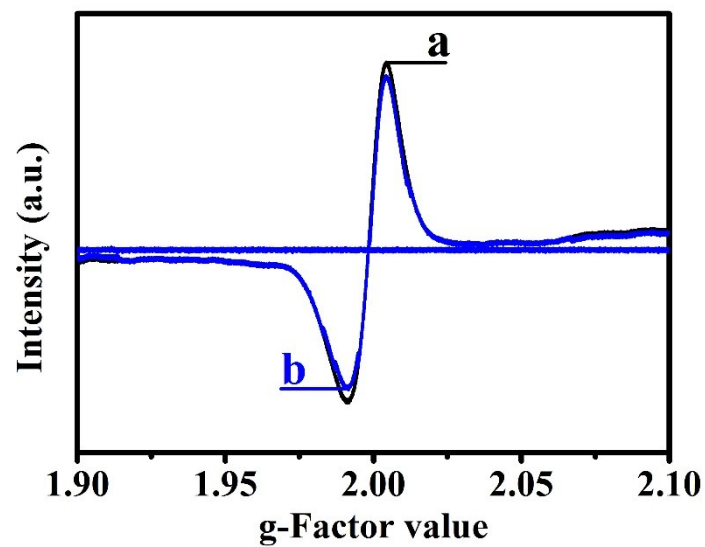


Fig. S6. Room-temperature electron spin resonance (ESR) lines of H-ZnIn₂S₄ (a) and the control sample after further hydrothermal process under pure water of H-ZnIn₂S₄ (b).

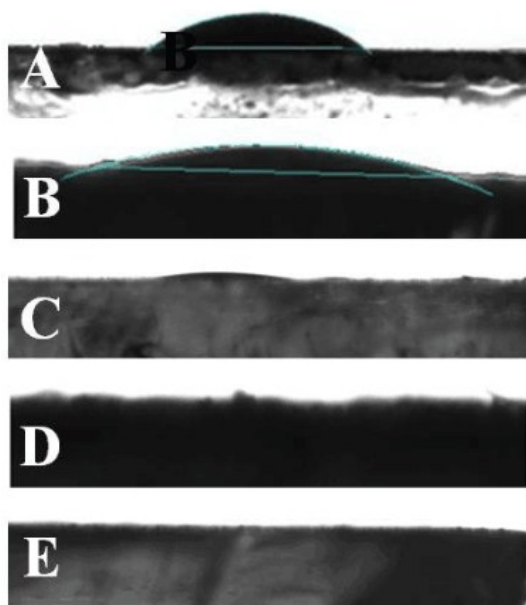


Fig. S7. (A)-(E) are the water contact angle photographs of C-ZnIn₂S₄, H-ZnIn₂S₄, 1/8J-ZnIn₂S₄, 1/6J-ZnIn₂S₄, and 1/4J-ZnIn₂S₄ under dark, respectively.

Table S3 Values of the parameters resulted from fitting the impedance spectra of the different samples using the equivalent circuit Figure 5C.

Samples	R_s ($\Omega \text{ cm}^2$)	CPE1 (F cm^{-2})	R_1 ($\Omega \text{ cm}^2$)	CPE2 (F cm^{-2})	R_{ct} ($\Omega \text{ cm}^2$)
J-ZnIn ₂ S ₄	58.03	7.565×10^{-4}	18.86	5.065×10^{-4}	150.5
H-ZnIn ₂ S ₄	59.35	2.967×10^{-3}	20.55	6.416×10^{-4}	241.2
C-ZnIn ₂ S ₄	59.98	4.856×10^{-3}	22.73	6.815×10^{-4}	320.2

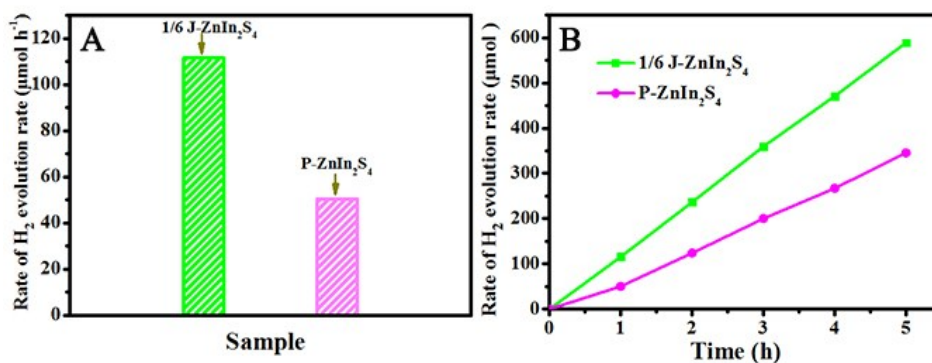


Fig. S8. Comparison of the photocatalytic H₂ evolution rate of 1/6J-ZnIn₂S₄ and the P-ZnIn₂S₄ (physically mixed C-ZnIn₂S₄ and H-ZnIn₂S₄) (A), different reaction times (B).

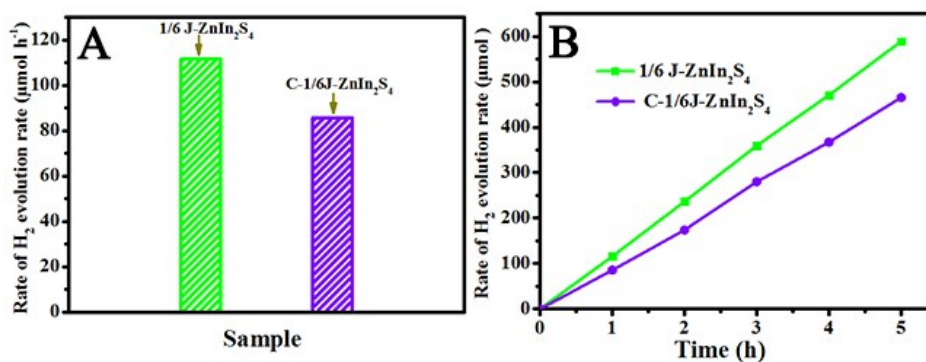


Fig. S9. Comparison of the photocatalytic H₂ evolution rate of 1/6J-ZnIn₂S₄ and the crushed 1/6J-ZnIn₂S₄ (C-1/6J-ZnIn₂S₄) (A), different reaction times (B).

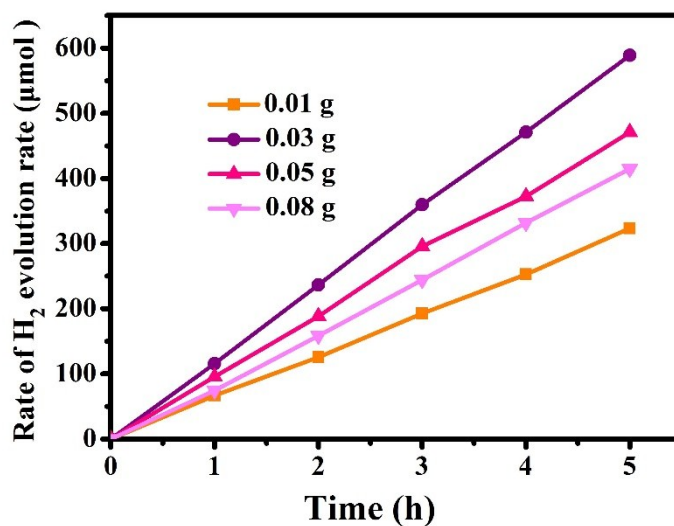


Fig. S10. Photocatalytic H₂ evolution of the 1/6J-ZnIn₂S₄ samples with different amounts.

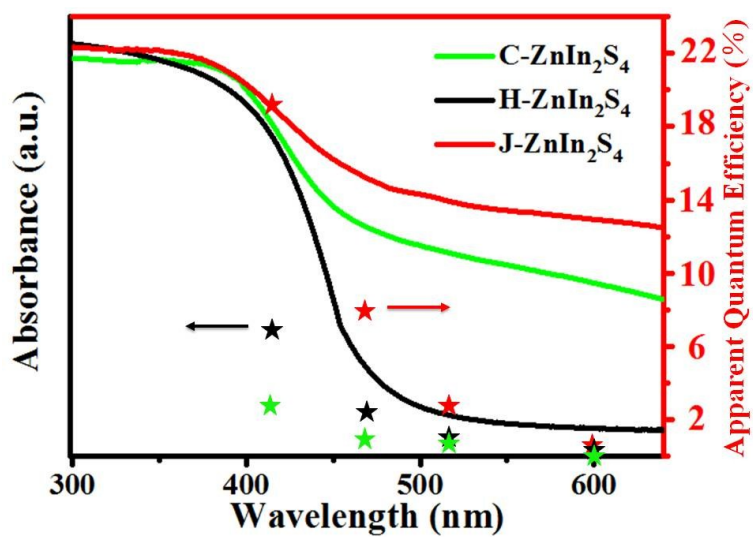


Fig. S11. The H₂ evolution rates of the different samples plotted against wavelength of monochromatic light.

Table S4 The apparent quantum efficiency (AQE) of the different samples under different illumination wavelength.

Sample	Apparent Quantum Efficiency (AQE)			
	420 nm	470 nm	520 nm	600 nm
J-ZnIn ₂ S ₄	18.67%	8.20%	2.12%	0.91%
H-ZnIn ₂ S ₄	6.43%	2.01%	0.69%	0.15%
C-ZnIn ₂ S ₄	3.18%	1.17%	0.12%	0.08%

Table S5 The performance comparison of the catalysts from the different references.

Catalyst	Apparent Quantum Efficiency (AQE) at 420 nm	Photocatalytic H ₂ Evolution Rate (μmol h ⁻¹)	Reference
J-ZnIn ₂ S ₄	18.67%	114.2	This Work
La-ZnIn ₂ S ₄	8.83%	116.68	(1)
ZnIn ₂ S ₄	4.11%	49.78	(2)
RGO-ZnIn ₂ S ₄	--	81.6	(3)
ZnIn ₂ S ₄ -CTAB	11.9%	122.2	(4)
MoS ₂ /ZnIn ₂ S ₄	--	47.71	(5)
Ag/ZnIn ₂ S ₄ /TiO ₂	0.18%	33.1	(6)
ZnIn ₂ S ₄ /g-C ₃ N ₄	0.28%	14.1	(7)

References

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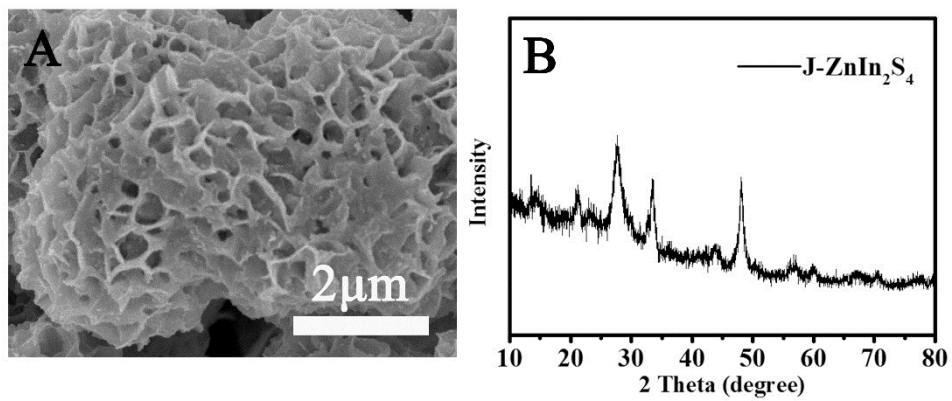


Fig. S12. SEM image (A) and XRD pattern of the J-ZnIn₂S₄ after photocatalytic H₂ evolution reaction.