

## Supporting Information

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### Quick one-pot synthesis of amorphous carbon-coated cobalt–ferrite twin elliptical frustums for enhanced lithium storage capability

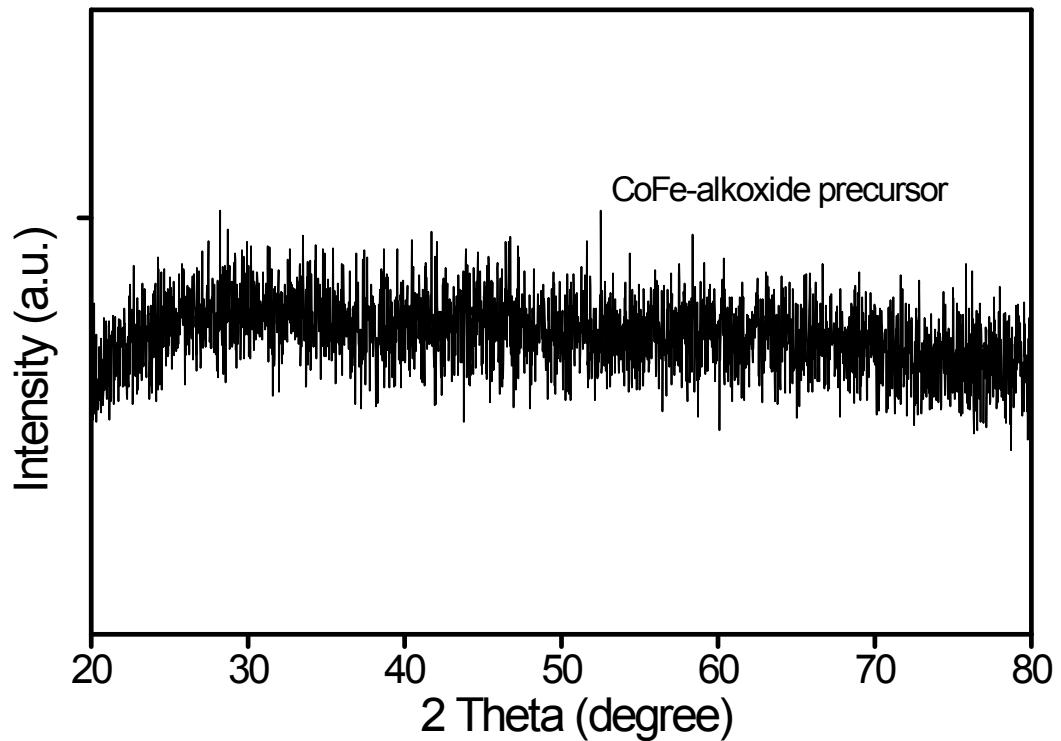
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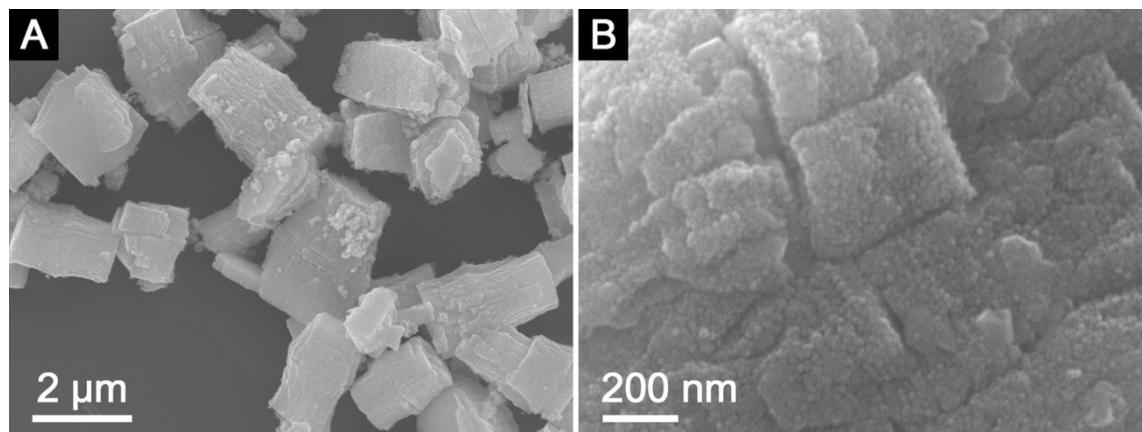
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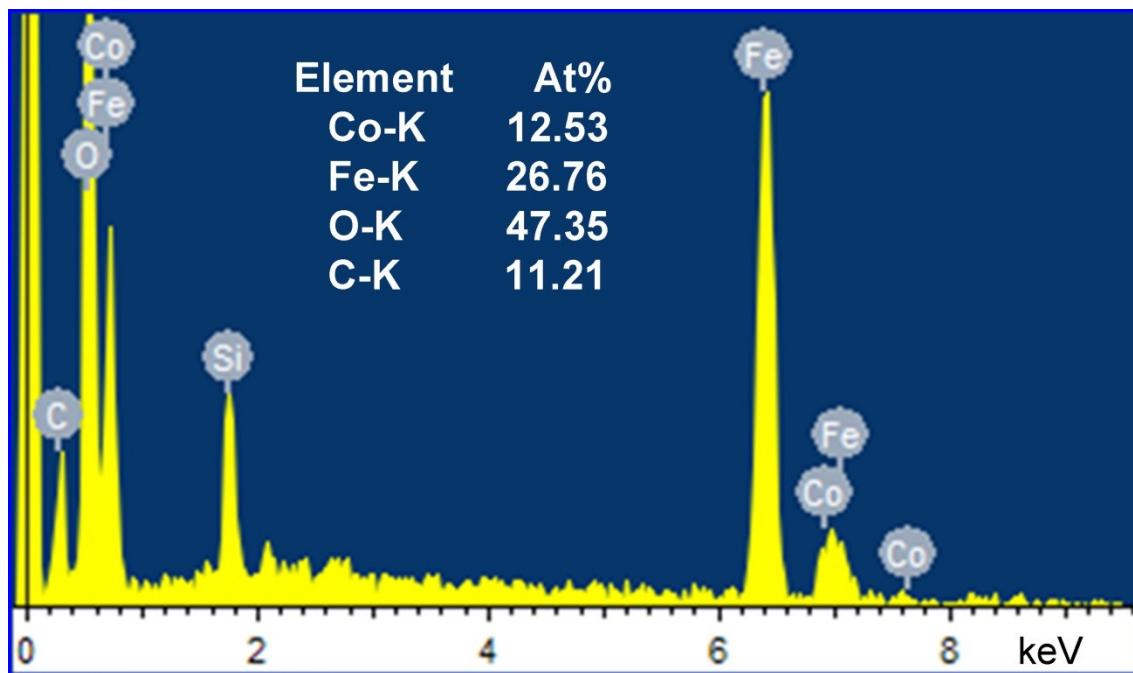
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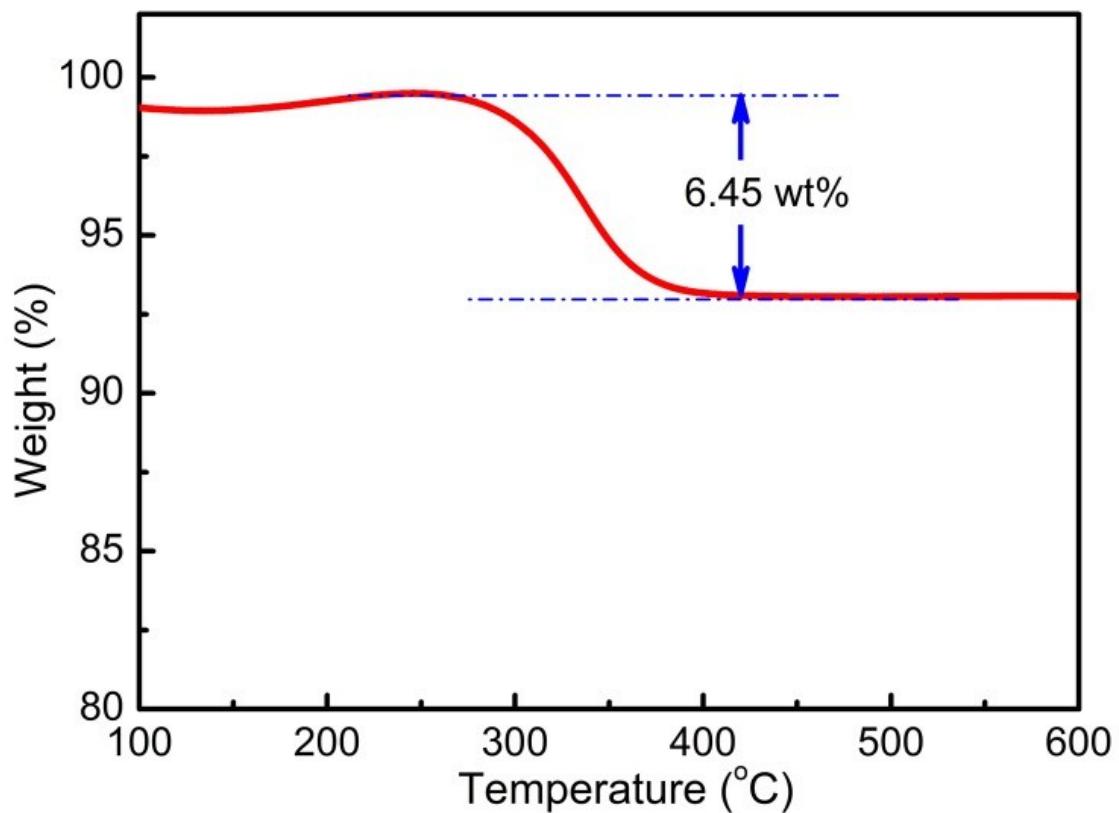
**Figure S1.** XRD pattern of as-prepared CoFe-alkoxide precursor



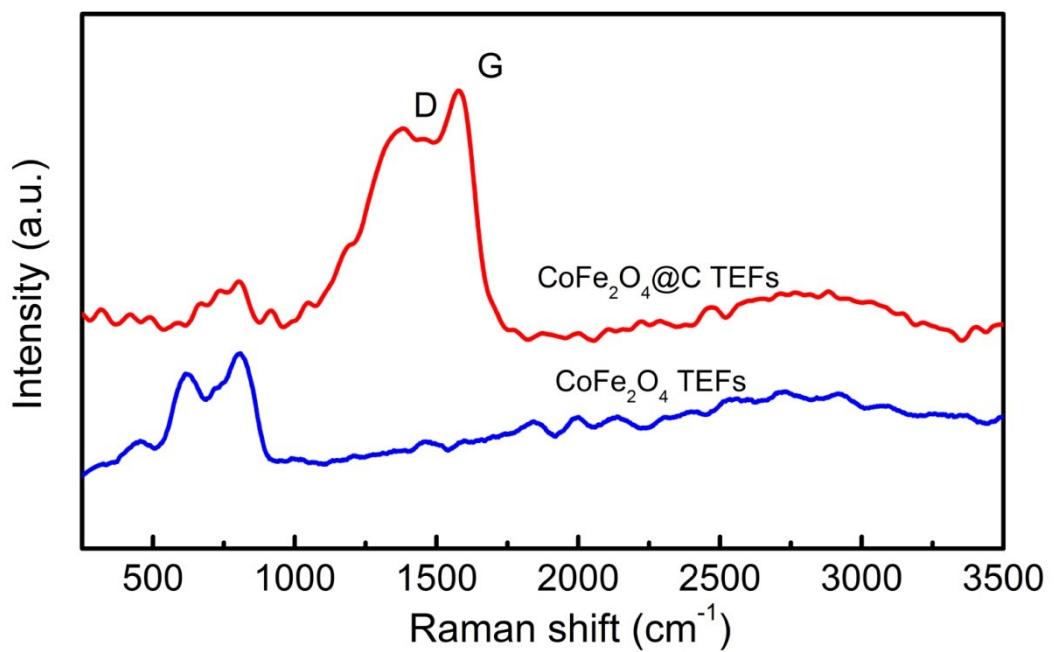
**Figure S2.** FESEM images of  $\text{CoFe}_2\text{O}_4$  TEFs obtained after annealing the CoFe-alkoxide precursor in air at  $500\text{ }^\circ\text{C}$  with a heating rate of  $1\text{ }^\circ\text{C min}^{-1}$ .



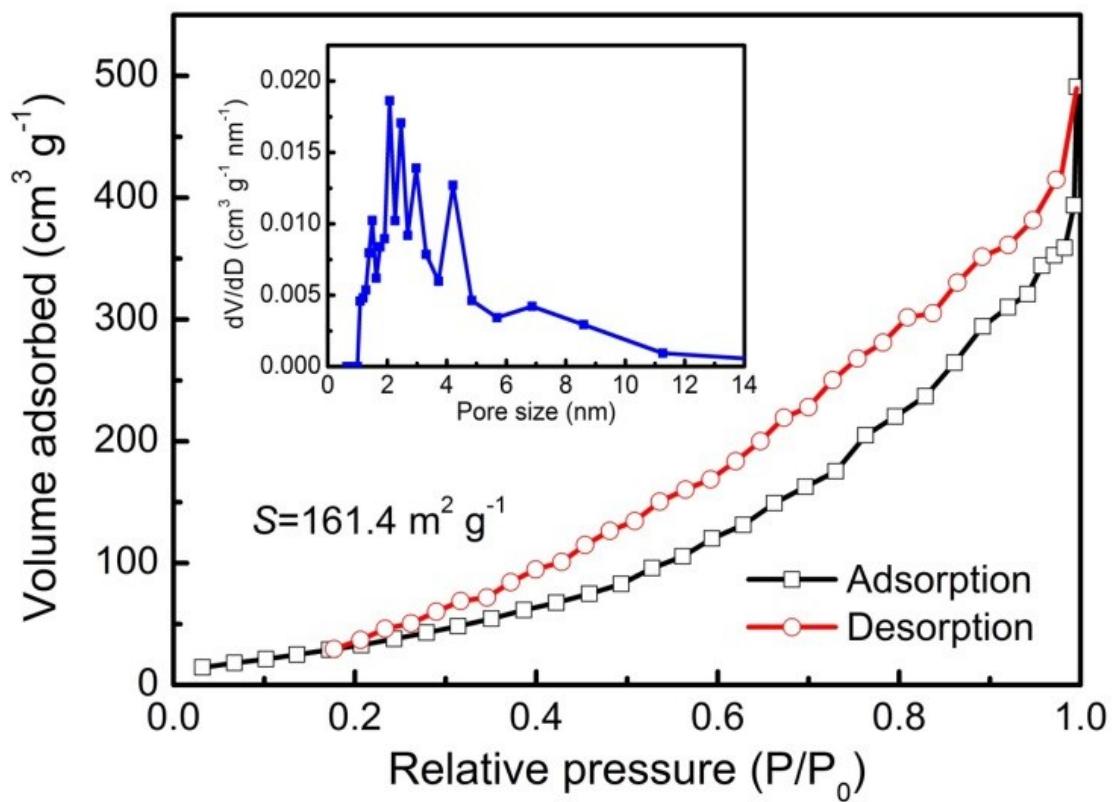
**Figure S3.** EDX pattern of as-prepared  $\text{CoFe}_2\text{O}_4@\text{C}$  TEFs.



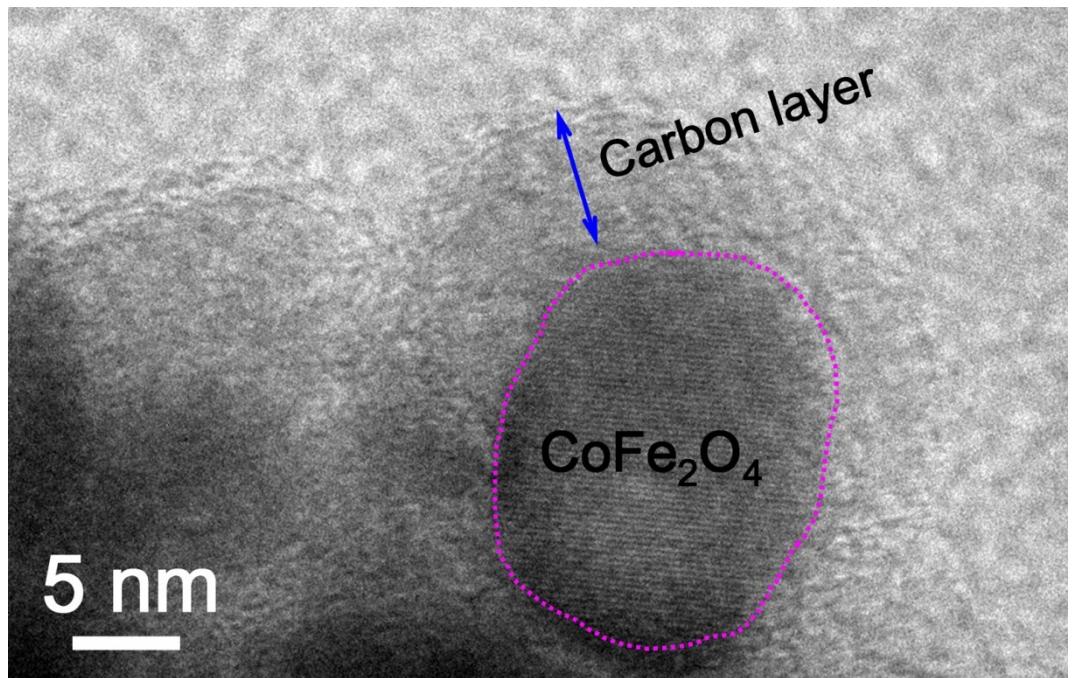
**Figure S4.** TGA curve of the as-prepared  $\text{CoFe}_2\text{O}_4@\text{C}$  TEFs in air between 100 and 600 °C with a heating rate of  $10\text{ }^{\circ}\text{C min}^{-1}$ .



**Figure S5.** Raman spectroscopy of as-prepared  $\text{CoFe}_2\text{O}_4@\text{C}$  TEFs and  $\text{CoFe}_2\text{O}_4$  TEFs.



**Figure S6.**  $N_2$  adsorption-desorption isotherm of the as-prepared  $\text{CoFe}_2\text{O}_4@\text{C}$  TEFs and the pore size distribution curve (inset) obtained by the BJH method.



**Figure S7.** High resolution TEM image of as-prepared CoFe<sub>2</sub>O<sub>4</sub>@C TEFs.

Table S1. Electrochemical comparison with other reported CoFe<sub>2</sub>O<sub>4</sub>-based anodes for LIB.

Electrode materials	Discharge capacity in the second cycle	Capacity retention after long cycles	Ref.
CoFe <sub>2</sub> O <sub>4</sub> @C TEFs	823 mAh g <sup>-1</sup> at 500 mA g <sup>-1</sup>	91.4% after 600 cycles	This work
CoFe <sub>2</sub> O <sub>4</sub> @CNTs	801 mAh g <sup>-1</sup> at 400 mA g <sup>-1</sup>	99% after 50 cycles	<sup>1</sup>
CoFe <sub>2</sub> O <sub>4</sub> @rGO NPs	906 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	100.4% after 50 cycles	<sup>2</sup>
C@CoFe <sub>2</sub> O <sub>4</sub> @rGO NPs	956 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	96.8% after 50 cycles	<sup>3</sup>
CoFe <sub>2</sub> O <sub>4</sub> @rGO NPs	938 mAh g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	69.5% after 50 cycles	<sup>4</sup>
Porous CoFe <sub>2</sub> O <sub>4</sub> Octahedrons	722 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	96.4% after 50 cycles	<sup>5</sup>
CoFe <sub>2</sub> O <sub>4</sub> @rGO NPs	735 mAh g <sup>-1</sup> at 800 mA g <sup>-1</sup>	76.9% after 300 cycles	<sup>6</sup>
CoFe <sub>2</sub> O <sub>4</sub> @CNT NPs	1153 mAh g <sup>-1</sup> at 200 mA g <sup>-1</sup>	90.6% after 100 cycles	<sup>7</sup>
CoFe <sub>2</sub> O <sub>4</sub> @CNT NPs	712 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	82.2% after 350 cycles	<sup>8</sup>
CoFe <sub>2</sub> O <sub>4</sub> Flowers	942 mAh g <sup>-1</sup> at 200 mA g <sup>-1</sup>	77.9% after 50 cycles	<sup>9</sup>
CoFe <sub>2</sub> O <sub>4</sub> NTs	ca. 901 mAh g <sup>-1</sup> at 50 mA g <sup>-1</sup>	67.7% after 150 cycles	<sup>10</sup>
CoFe <sub>2</sub> O <sub>4</sub> @carbon cloth NWs	Ca. 1298 mAh g <sup>-1</sup> at 500 mA g <sup>-1</sup>	92.8% after 200 cycles	<sup>11</sup>
CoFe <sub>2</sub> O <sub>4</sub> NSs	ca. 1301 mAh g <sup>-1</sup> at 50 mA g <sup>-1</sup>	88.2% after 30 cycles	<sup>12</sup>
Yolk–Shell CoFe <sub>2</sub> O <sub>4</sub>	900 mAh g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	83.2% after 200 cycles	<sup>13</sup>
C@CoFe <sub>2</sub> O <sub>4</sub> FITMNs	ca. 1036 mAh g <sup>-1</sup> at 200 mA g <sup>-1</sup>	67.6% after 150 cycles	<sup>14</sup>
CoFe <sub>2</sub> O <sub>4</sub> @rGO NPs	ca. 902 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup>	104.8 after 100 cycles	<sup>15</sup>

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