## **Electronic Supporting Information**

## One-step extend strategy with ionic liquid assisted synthesis of

## Ni<sub>3</sub>S<sub>4</sub>-MoS<sub>2</sub> heterojunction electrodes for supercapacitors

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Figure S1. EDS mapping images of the whole  $Ni_3S_4$ -MoS<sub>2</sub> nanoflower.



Figure S2. EDS of Ni<sub>3</sub>S<sub>4</sub>-MoS<sub>2</sub>



**Figure S3. (**a) SEM (c) TG (d) XRD of Mo precursor and (b) FT-IR of [BMIM]SCN and Mo precursor.

Element	Content (wt%)	
N	6.54	
С	22.32	
н	3.72	

Table S1. Elemental analyzer result of Mo precursor



Figure S4. (a) SEM, (b) XRD of Ni precursor



Figure S5. (a) SEM, (b) TEM and (c)XRD of  $Co_3S_4$ -MoS<sub>2</sub>.



Figure S6. (a,b) SEM, (c) XRD and (d) EDS-mapping of MnS-MoS<sub>2</sub>.



**Fig.S7** The common reaction progress schematic of  $M_xS_y$ -MoS<sub>2</sub>.



Figure S8. (a) SEM, (b) XRD, (c) XPS spectra of S 2p peaks, (d) XPS spectra of Mo 3d and S 2s peaks of  $MoS_2$  for SCs performance comparison.



Figure S9. CV curves of (a)  $MoS_2$  (b)  $Ni_3S_4$ - $MoS_2$  (c)  $Co_3S_4$ - $MoS_2$  at scan rate of 50 mVS<sup>-1</sup> and (d) impedance Nyquist plots of  $MoS_2$ ,  $Ni_3S_4$ - $MoS_2$  and  $Co_3S_4$ - $MoS_2$  at open circuit potential.



Figure S10. (a,b) Plots of anodic and corresponding cathodic peak current densities vs the square root of scan rate of the  $Ni_3S_4$ - $MoS_2$ , respectively.

performances				
	Cycling stability	Rate capability	Reference	
MoS <sub>2</sub> /PANI	6000 cycles	496 F/g at 1A/g	ACS Appl. Mater. Interfaces 7 (2015) 28294-28302	
MoS₂/rGO	50000 cycles	416 F/g at 1A/g	ACS Appl. Mater. Interfaces 8 (2016) 32842-32852	
MoS <sub>2</sub> /Oleylamine	5000 cycles	50.65 mF/cm <sup>2</sup> at 0.37 A/g	Chem. Mater.28 (2016) 657-664	
MoS <sub>2</sub> /Graphene	15min	4.29 mF/cm <sup>2</sup> at 5 mV/s	ACS Appl. Mater. Interfaces 7 (2015) 17388-17398	
$MoS_2/C$ ball	5000 cycles	201 F/g at 1A/g	Int. J. Hydrogen Energy 40 ( 2015) 10150.	
MoS <sub>2</sub> /MWCNT	1000 cycles	452.7 F/g at 1 A/g	Energy 67 (2014) 234	
MoS <sub>2</sub> -C aerogel	1500 cycles	260 F/g at 1A/g	J. Electroanal. Chem. 752(2015)33	
MoS <sub>2</sub>	5000 cycles	368 F/g at 5mV/s	J. Power Sources 285(2015)63.	

Table S2. Selected MoS <sub>2</sub>	, X-MoS <sub>2</sub> hybrid materials and	l its electrochemical
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Figure S11. (a)Comparative CV curves of  $Ni_3S_4$ -MoS<sub>2</sub> and AC electrodes. (b) CV curves of the  $Ni_3S_4$ -MoS<sub>2</sub> //AC asymmetric supercapacitor at different potential windows at a scan rate of 20 mV s<sup>-1</sup> in a two-electrode system.



Figure S12. A LED lamp lighted by two new AA batteries.