Graft of Nano-Ag Particles on -SO₃⁻-coated Nanopolymers for Promoting Methane Hydrate Formation

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Fig. S1 Schematic diagram of the methane hydrate formation apparatus



Fig. S2 Pyrene fluorescence spectra of deionized water, -SO₃-@PSMM emulsion of 0.5 mmol/L and -SO₃-@PSNS emulsion of 0.5 mmol/L



Fig. S3 XRD spectrum of the precipitate of Ag&-SO₃-@PSMM-25%



Fig. S4 Evolution of methane storage capacity during methane hydrate formation with Ag&-SO₃-@PSNS-25% of 0.5 mmol/L and deionized water at unstirred condition (the initial pressure was 6 MPa, the temperature was 275.15 K)



Fig. S5 Evolutions of methane storage capacity during methane hydrate formation with Ag&-SO₃-@PSNS-25% of 0.5 mmol/L and SDS of 1 mmol/L (the initial pressure was 6 MPa, the temperature was 275.15 K, the stirring was 300 rpm)



Fig. S6 TEM photos of Ag& SO₃-@PSNS-25% after hydrate formation-dissociation

Table S1 Induction time, growth period, growth rate and methane storage capacity of methane hydrate formation with Ag&-SO₃-@PSMM of different Ag amounts (the initial pressure was 6 MPa, the temperature was 275.15 K, the stirring was 300 rpm)

Sample	Run 1	Run 2	Run 3	av ^a	sd ^b			
	Induction time (min) ^c							
Ag&-SO ₃ -@PSMM-25%	59.9	43.9	78.7	60.8	14.2			
Ag&-SO ₃ -@PSMM-50%	44.9	54.0	59.3	52.7	5.9			
Ag&-SO ₃ -@PSMM-75%	43.3	25.1	28.2	32.2	7.9			
	Growth rate (mmol gas·mL water ⁻¹ ·min ⁻¹)							
Ag&-SO ₃ -@PSMM-25%	0.069	0.082	0.075	0.075	0.006			
Ag&-SO ₃ -@PSMM-50%	0.061	0.094	0.062	0.072	0.015			
Ag&-SO ₃ -@PSMM-75%	0.066	0.043	0.036	0.048	0.013			
	Growth period (min) ^d							
Ag&-SO ₃ -@PSMM-25%	118.1	98.1	110.3	108.8	8.2			
Ag&-SO ₃ -@PSMM-50%	133.2	88.2	125.7	115.7	19.7			
Ag&-SO ₃ -@PSMM-75%	127.8	180.5	223.0	177.1	38.9			
	Storage capacity (v/v) e							
Ag&-SO ₃ -@PSMM-25%	144.6	144.1	146.8	145.2	1.2			
Ag&-SO ₃ -@PSMM-50%	144.7	148.0	139.0	143.9	3.7			
Ag&-SO ₃ -@PSMM-75%	149.9	140.0	142.4	144.1	4.2			

^a av - average value; ^b sd - standard deviation; ^c induction time was calculated as the time from charging methane into the reactor to obvious pressure decrease was observed; ^d growth period was viewed as the time from the end of induction period to new pressure balance was reached; ^e methane storage capacity was calculated as the storage capacity at 200 min during hydrate growth period.

Table S2 Induction time, growth period, growth rate and methane storage capacity of methane hydrate formation with Ag&-SO₃-@PSNS of different Ag amounts (the initial pressure was 6 MPa, the temperature was 275.15 K, the stirring was 300 rpm)

Sample	Run 1	Run 2	Run 3	av	sd			
	Induction time (min)							
Ag&-SO ₃ ⁻ @PSNS-25%	12.6	45.3	0.4	19.4	19.0			
Ag&-SO ₃ -@PSNS-50%	28.5	38.8	48.2	38.5	8.0			
Ag&-SO ₃ -@PSNS-75%	21.2	17.7	14.4	17.8	2.8			
	Growth rate (mmol gas·mL water ⁻¹ ·min ⁻¹)							
Ag&-SO ₃ -@PSNS-25%	0.20	0.21	0.23	0.21	0.02			
Ag&-SO ₃ -@PSNS-50%	0.20	0.21	0.20	0.20	0.01			
Ag&-SO ₃ -@PSNS-75%	0.20	0.19	0.23	0.21	0.02			
	Growth period (min)							
Ag&-SO ₃ -@PSNS-25%	42.8	39.9	36	39.6	2.8			
Ag&-SO ₃ -@PSNS-50%	43.3	41.2	41.8	42.1	0.9			
Ag&-SO ₃ -@PSNS-75%	41.4	43.1	38.8	41.1	1.8			
	Storage capacity (v/v)							
Ag&-SO ₃ -@PSNS-25%	149.4	147.9	150.8	149.3	1.2			
Ag&-SO ₃ -@PSNS-50%	153.3	153.5	147.1	151.3	3.0			
Ag&-SO ₃ -@PSNS-75%	146.5	146.7	156.9	150.0	4.9			