

Electronic supplementary information

Ionic Conductivity Promotion of Polymer Electrolyte with Ionic Liquid Grafted Oxides for All-Solid-State Lithium-Sulfur Batteries

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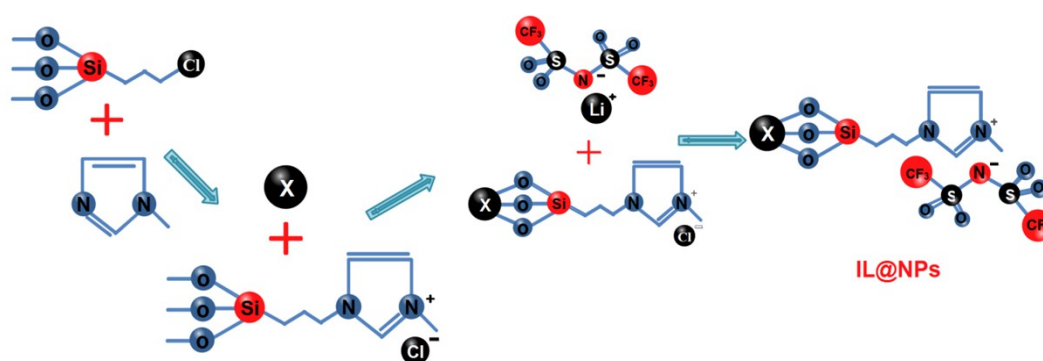


Fig. S1 Schematic illustration of the fabrication process of IL@NPs based on ion exchange method.

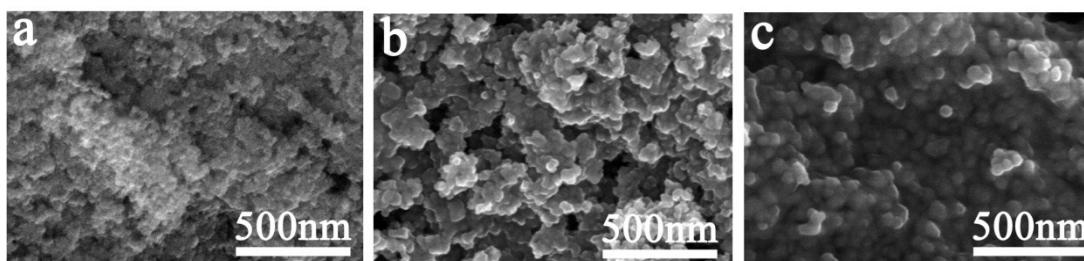


Fig. S2 SEM morphology of (a) IL@SiO₂, (b) IL@TiO₂, (c) IL@ZrO₂.

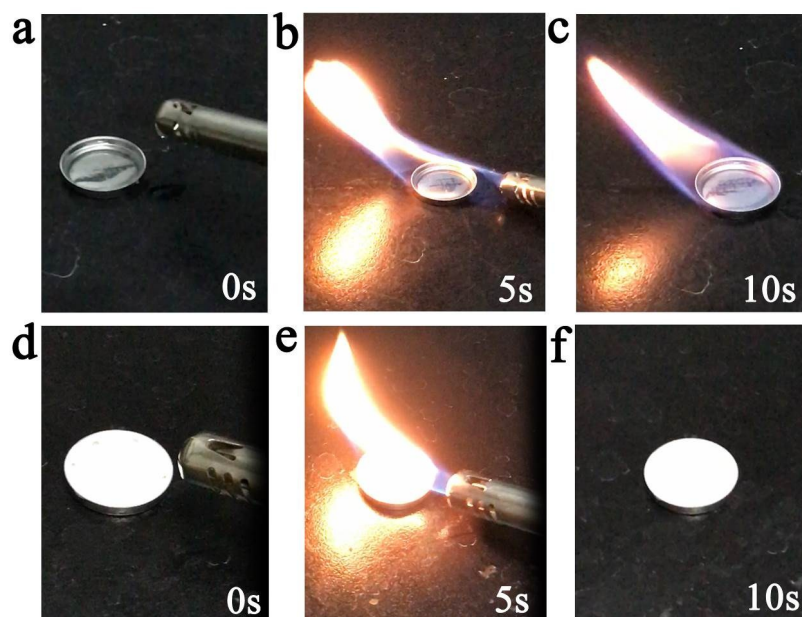


Fig. S3 Flammability tests of (a-c) 1 M LiTFSI and 0.1 M LiNO₃ dissolved in a mixture of 1,3-dioxolane (DOL) and dimethoxymethane (DME) (v/v=1:1) and (d-f) PEO-Li-Zr electrolytes using a burning torch.

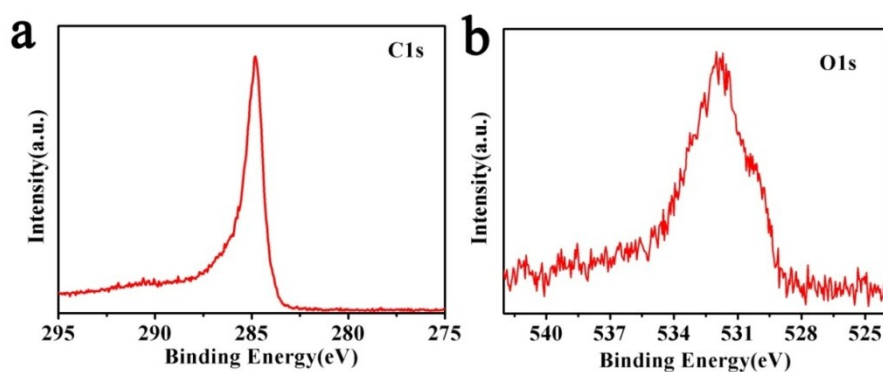


Fig. S4 XPS of (a) C1s and (b) O1s respectively.

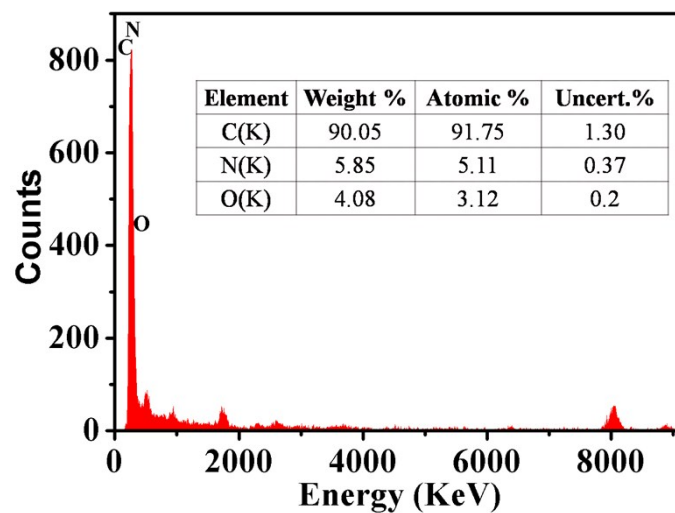


Fig. S5 EDX full elements analysis of N-CNs.

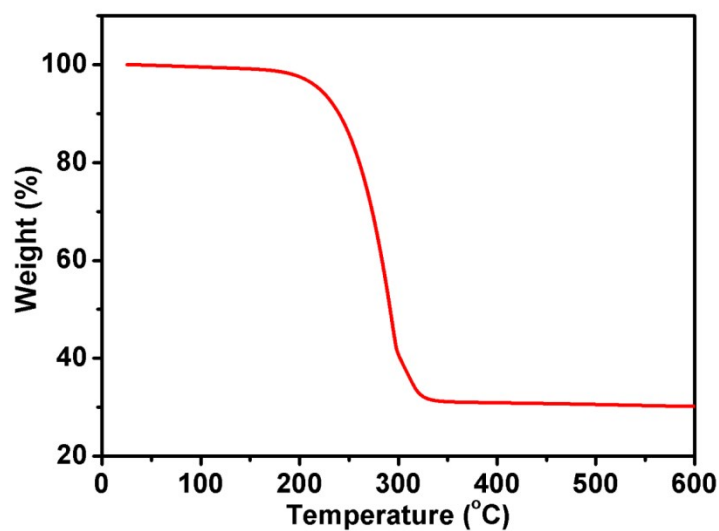


Fig. S6 TG curves of N-CNs/S materials.

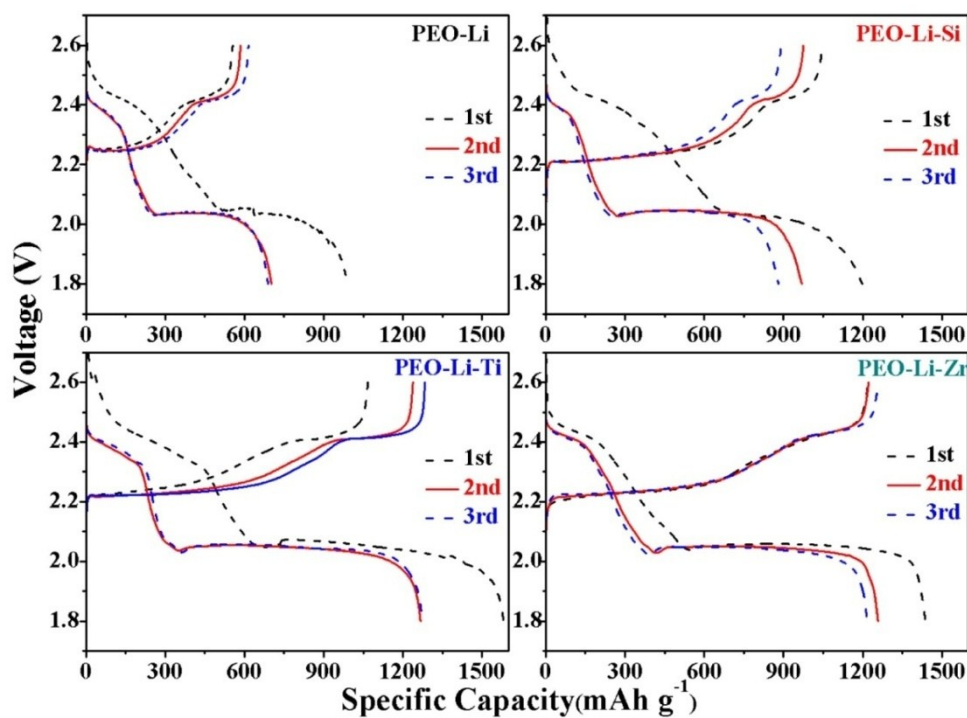


Fig. S7 The discharge/charge curves of the first three cycles for the battery based on PEO-Li, PEO-Li-Si, PEO-Li-Ti, PEO-Li-Zr electrolytes at 50 °C.

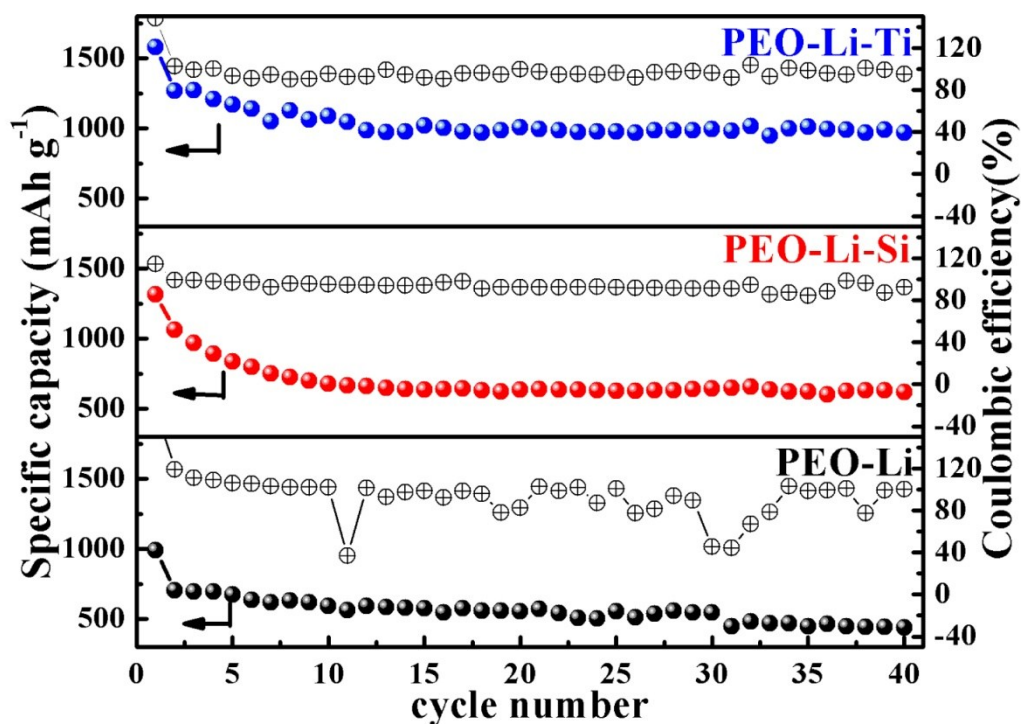


Fig.S8 Discharge capacity and Coulombic efficiencies of PEO-Li, PEO-Li-Si, PEO-Li-Ti battery at 50 °C.

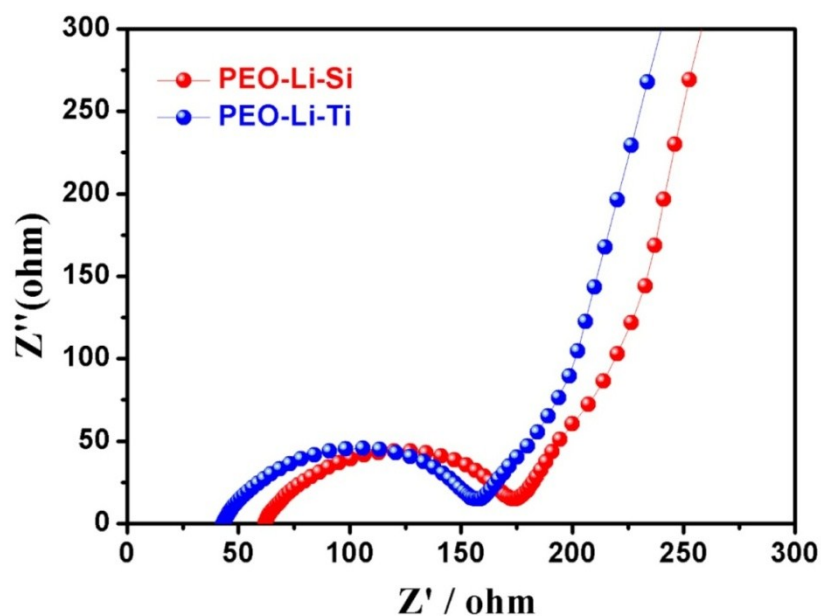


Fig. S9 Electrochemical impedance spectroscopy (EIS) of PEO-Li-Si, PEO-Li-Ti battery at 50 °C.

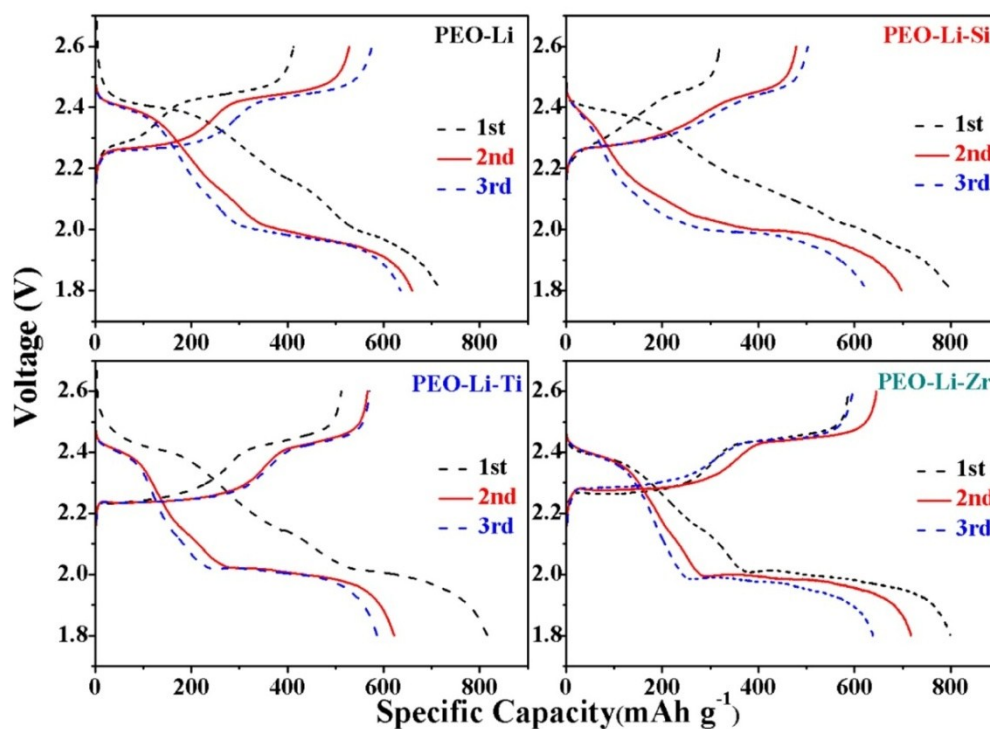


Fig. S10 The discharge/charge curves of the first three cycles for the battery based on PEO-Li, PEO-Li-Si, PEO-Li-Ti, PEO-Li-Zr electrolytes at 37 °C.

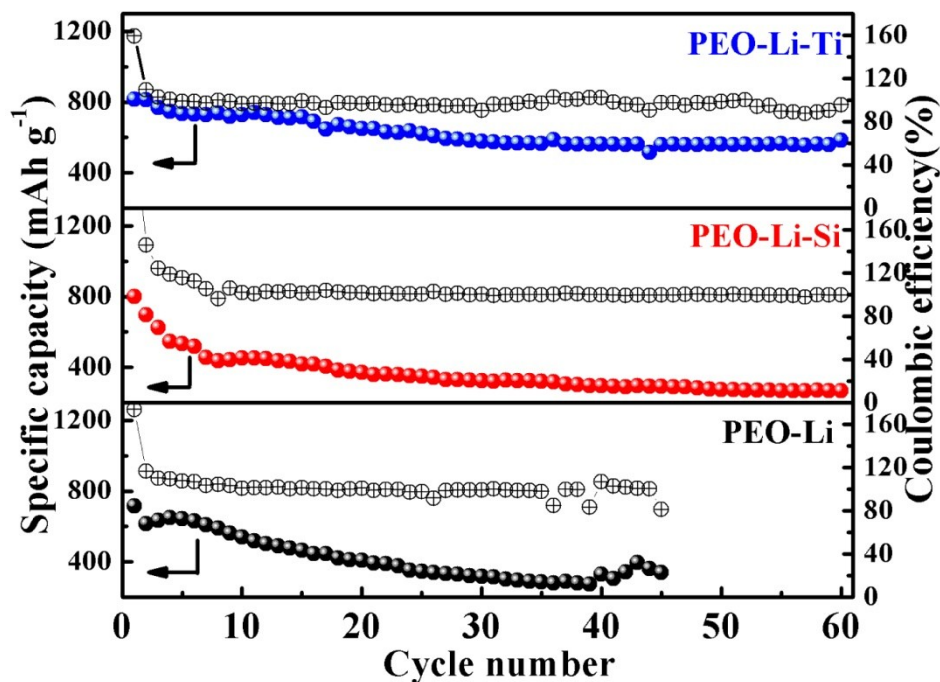


Fig. S11 Discharge capacity and Coulombic efficiencies of PEO-Li, PEO-Li-Si, PEO-Li-Ti battery at 37 °C.

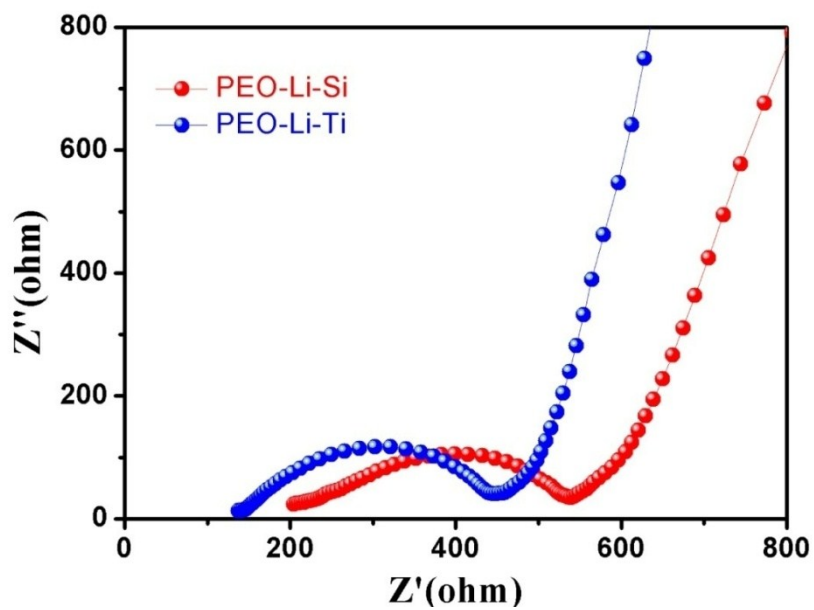


Fig. S12 Electrochemical impedance spectroscopy (EIS) of PEO-Li-Si, PEO-Li-Ti battery at 37 °C.

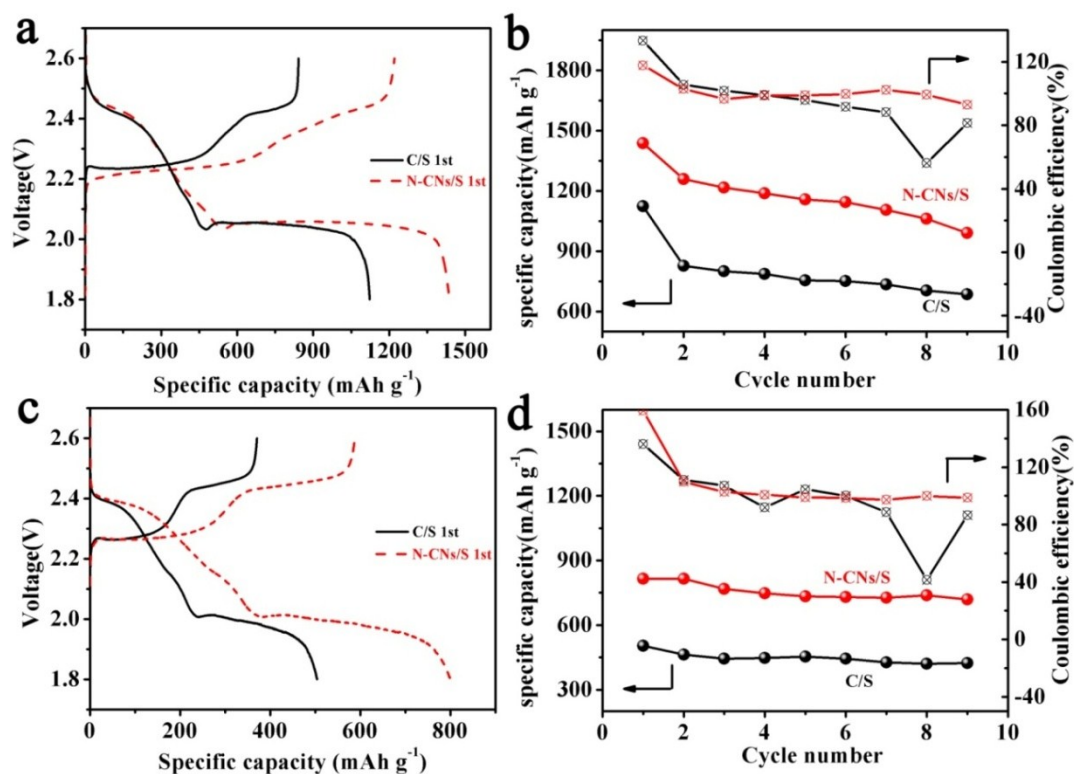


Fig. S13 Typical discharge/charge curves of Carbon black/S cathode and N-CNs/S cathode using PEO-Li-Zr electrolyte at 50 °C. b) Discharge/charge capacity and Coulombic efficiencies of the battery with Carbon black/S cathode or N-CNs/S cathode at 50 °C. c) Typical discharge/charge curves of Carbon black/S cathode and N-CNs/S cathode using PEO-Li-Zr electrolyte at 37 °C. d) Discharge/charge capacity and coulombic efficiencies of the battery with Carbon black/S cathode or N-CNs/S cathode at 37 °C.

Table S1. Comparison of the electrochemical performance of different all-solid-state batteries

Electrolyte	Cathode/ Anode	Working Voltage (V)	Working Temp(°C)	Capacity (mAh g ⁻¹)	Year	Ref
PEO-LiTFSI -Pyr ₁₄ TFSI	LiFePO ₄ /Li	3.0-4.0	40	160 (After 100cycles)	2014	[1]
PEO-LiTFSI -HMOP	LiFePO ₄ /Li	2.9-3.8	65	120 (After 100cycles)	2016	[2]
PEO-LiClO ₄ -SiO ₂	LiFePO ₄ /Li	2.5-4.1	90	105 (After 80cycles)	2016	[3]
PEO-LiClO ₄ -LLZTO	LiFePO ₄ /Li	2.6-3.8	60	105 (After 200cycles)	2016	[4]
PEO-LiCF ₃ SO ₃ -Li ₂ S-ZrO ₂	Li ₂ S-C /Li	1.5-3.2	70	600 (After 50cycles)	2010	[5]
PEO-LiTFSI -MIL-53(Al)	PANI@C/S- 280/Li	1.0-3.0	80	876 (After 60cycles)	2015	[6]
PEO-LiTNSFI	CMK-3/S /Li	1.5-3.0	60	450 (After 200cycles)	2016	[7]
PEO-LiTFSI -IL@ZrO₂	N-CNs/S/Li	1.8-2.6	37	600 (After 80cycles)	Our	
			50	986 (After 40cycles)	work	

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